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LEAD

By LEWIS A. SMITH

STATISTICAL SUMMARY

Lower prices for lead in 1927 resulted in a decrease in production in the tri-State area but on the whole did not affect the domestic situation as regards production. Instead of being stimulated by the reduction in price, the production of primary metal actually decreased more. The salient statistics of the lead industry in the tri-State area for 1926 and 1927 are compared in the following table:

Salient figures of the lead industry in the United States

Production of refined primary lead:	
From domestic ores.....	sh.
From domestic and foreign ores.....	sh.
Average selling price per pound of refined lead.....	
Value of refined primary lead produced from domestic ores.....	
Imports of lead in ore, base bullion, etc.:.....	
General.....	sh.
For consumption.....	
Exports of lead:	
From domestic ores.....	
From foreign ores.....	
Stocks of lead in bonded warehouse:	
Jan. 1.....	
Dec. 31.....	
Total supply of refined primary lead.....	
Lead withdrawn.....	
Apparent consumption of refined primary lead ¹	
Antimonial lead produced.....	
Recovery of secondary lead.....	

¹ Exclusive of foreign lead exported in manufactures and upon which duty is paid.

The production of refined primary lead from domestic sources decreased 1.8 per cent from 1926, but the decrease in lead from domestic and foreign sources was 1.7 per cent. The selling price of lead decreased 1.7 cents per pound from 1926 and accounted largely for the decrease in value. \$25,000,000 in the value of refined lead produced. Most of the smelter production from domestic sources was in the States, ranking as follows in importance: Missouri (22.7 per cent), Utah (22.7 per cent), Idaho (22.7 per cent),

¹ The statistics on domestic production were prepared by H. M. Smith, and the statistics on foreign production and exports were compiled from records of the Bureau of Foreign and Domestic Commerce and the Bureau of Mines.

(9.6 per cent), Colorado (5.4 per cent), and Montana (3.4 per cent). These six States account for nearly 94 per cent of the total of 684,165 tons from domestic sources; only Idaho and Utah showed substantial increases. The total imports of lead in 1927 were nearly 14,000 tons in excess of those in 1926, and the exports were 53,000 tons more than in 1926. Practically all of the lead recovered from ore produced in the United States is used at home; only about 2,500 tons of lead from domestic ores was exported in 1927.

GENERAL FEATURES OF LEAD INDUSTRY IN 1927

The production and the apparent use of lead both declined in the United States in 1927 but probably did not in the rest of the world. Available figures for world output and consumption indicate a slight increase over those for 1926, a record year. No new factors appeared in the situation. The competition of substitutes for lead and the return of secondary metal continued to trouble producers, but even along these lines that would tend substantially to alter the trend that has been under way for several years there were no outstanding developments. There is reason to believe that the apparent drop in domestic demand represents in part at least nothing more than the adoption of hand-to-mouth buying policies on the part of certain consumers who, in view of market conditions and improved transportation services, considered it no longer necessary to maintain as large stocks as formerly. Flotation, notably selective or differential flotation, made further advances, yielding richer concentrates and higher extractions at lower costs.

CHANGES IN USES

For several years the American Bureau of Metal Statistics has published a table that shows the approximate amounts of lead consumed in the various uses in which that metal is required. By translating these figures into percentages an idea may be gained as to the relative changes in distribution according to the various outlets for lead.

Percentage of total lead consumed in the United States, 1923-1927, by uses

Use	1923	1924	1925	1926	1927
Storage batteries	18.62	20.93	21.02	21.09	20.81
Cable coverings	17.06	17.00	18.21	20.63	19.15
White lead	16.93	18.47	15.29	13.22	14.98
Building	9.70	10.23	10.32	10.40	10.46
Litharge and red lead	5.99	4.19	4.90	4.00	4.52
Solder	3.91	3.69	4.09	4.11	4.16
Ammunition	5.21	3.32	3.68	3.55	4.04
Bearing metal	3.65	3.94	3.97	4.00	3.89
Foil	4.34	4.31	3.80	3.89	3.87
Calking	3.28	3.28	3.60	3.55	3.67
Castings	1.95	1.85	2.10	2.00	2.02
Type metal	1.66	1.60	1.75	1.78	1.90
Automobiles	1.35	1.32	1.49	1.85	1.43
Terneplate	.60	.60	.63	.45	.46
Railway cars	.78	.55	.62	.42	.32
Locomotives	.27	.14	.09	.11	.08
Shipbuilding	.01	.02	.01	.01	.02
Miscellaneous	4.76	4.68	4.73	4.94	4.82

Four of the outlets alone, it will be noted, account for two-thirds of the total consumption. As prepared by the Bureau of Metal Statistics the estimates of consumption in 1927 add to only 840,950 tons, or almost 100,000 tons less than the total lead available for consumption in the United States by production and trade figures now available. The relative proportions of lead used for the various purposes are, but using the larger figures for indicated consumption, required by these principal consuming industries, estimated at approximately 195,000 short tons for storage batteries, 141,000 tons for cable coverings, 141,000 tons in building.

Four years is scarcely long enough for definite trends to be established, but the percentages in the foregoing table are relative to other uses, the consumption of lead in cable coverings and on the decrease in the manufacture of building there is a consistent though slight upward trend. In storage batteries, on the other hand, the upward trend seems to have reached its climax. The demand for lead for the last-named purpose is affected by the use of "socket power" for radio receiving sets, which is available, thus eliminating a large number of uses. A more important factor, however, is the recovery of lead from spent batteries (estimated at about 95 per cent) of the lead returned to the industry within two to three years of initial utilization in this way. Hence, despite the increasing use of automobiles and radio receivers in use, the lead required in storage battery manufacture is not increasing. However, the extent to which old battery metal is replaced by virgin metal is not revealed by available figures. The figures given in the table are based upon estimates of total lead, including lead of both primary and secondary sources.

As compared with 1926 there was a drop in the percentage of lead consumed in cable coverings in 1927. Nevertheless, the outlook for this field, the second largest use of the metal, is favorable. Although undersea cable coverings are being raised, enlarged, and replaced. The lead required is reclaimed; but, since new and larger lines are added as the use of electricity becomes more extensive, consumption in this field should increase, provided no substitutes to replace lead are not found.

In the manufacture of white lead, formerly an important outlet for the metal, there was an apparent drop in consumption in 1927, offsetting in part the drop in the recent years. In Europe the use of white lead has been unfavorably affected by legislation that has been particularly for interior painting, because of the workmen applying it. In the United States the use of white pigment has failed to share in the fast-growing demand for products is simply the result of active corrosion substances, notably zinc-white, lithopone, barytes, and cellulose lacquers. Of these substitutes

which now find such extensive use in the automobile industry, and titanium oxide are newcomers in the field. The latter after becoming moderately well established abroad has made steady advances in this country. The importance of this new competition may be gauged by the fact that the National Lead Co., a leading manufacturer of lead pigments, has purchased the properties and business of one of the largest titanium pigment companies in Europe and has obtained control of one of the largest producers of such pigments in this country. It is evident that the prospects for the wider use of white lead are not bright.

As regards the use of primary lead in the building industry the competition from reclaimed lead and substitute materials is likewise apparent. Because of its resistance to corrosion and its workability lead is used in roofing sheets, cornices, electric cables and wiring, drainage and sewerage pipes, plumbing fixtures, etc. Brass, copper, and zinc are all finding increased application in building, and the brass manufacturers especially are advertising extensively and effectively. However, the American people are beginning to demand more permanence in their buildings. For this reason, even though lead may be replaced by other materials for some purposes, it is likely to hold its own and possibly may find further uses in the construction industries.

In the numerous minor fields of use the only recent development that calls for special comment seems to be the replacement of solid bearings by roller bearings. Even some of the railway companies have equipped their rolling stock with these new bearings which, though much more expensive in initial cost, are deemed advantageous because they reduce friction and consequently contribute to greater operating economy.

Lead is relatively a cheap metal, and for this reason even a decrease in its price does not tend largely to encourage its use to replace other metals. Increases in the price of other metals, however, do tend to encourage the substitution of lead therefor. In the manufacture of foil, for example, either (1) pure tin, (2) pure lead, (3) lead covered with tin on both sides, or (4) aluminum may be used. When the price of tin advances to 50 cents or over per pound, foil manufacturers are disposed to replace it as far as possible by lead. Since ordinary solder contains lead and tin in equal proportion, other solders, such as cadmium solders, which contain 85 to 95 per cent of lead, or "amaloy," a new invention containing 98 per cent or even more of lead and only 1 per cent or less of tin (a small amount of phosphorus being added to produce the desired properties), tend to be used when tin becomes relatively more costly.

NEW USES

In addition to amaloy, which is being used for soldering work in the automobile industry and in the manufacture and installation of electrical equipment, a few other new uses have come into prominence lately, and some of these may develop into important items of consumption. In building construction interest is evoked by the use of lead mattresses to act as shock absorbers between the foundations and the steel framework of skyscrapers. In one building in New York about 55 tons of lead appears to have been used for this purpose.

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In the pigment field a new lead preparation has been introduced. This preparation, which consists of divided particles of lead suboxide (Pb_2O) suspended in oil, can be sprayed or applied with a brush upon a surface. When being applied it undergoes a slow transformation, and the metallic lead held firmly as a protective coat by

SELECTIVE FLOTATION

UNITED STATES

No important new lead districts have been discovered in recent years, but the world's resources have been largely increased by the result of more efficient metallurgical processes. As lead and zinc are typically associated in nature. Ores of lead and zinc are mined together in many parts of the world, but the advent of the Wilfley table in the nineties the separation of lead and zinc was incomplete and wasteful. In the United States lead and zinc have been separated with fair success, but in the western United States the lead and zinc are frequently also copper and iron sulphides are so intimately associated that relatively fine grinding is necessary to liberate the minerals. The development of methods of treating the ores of the Rocky Mountain States is one of the most important in the history of ore dressing. The shaking table, an economic operation of an enormous number of tables, succeeded in effecting a thoroughly satisfactory separation of the blends in particular, being intermediate in specific gravity between the lead and the gangue minerals, was not recovered. The zinc that accompanied the lead concentrate in smelting, was not paid for, and if present in large amount was penalized.

Even before the World War flotation was used to improve the recovery of sulphide minerals, particularly as a necessary process for the treatment of slimes, but it was not until 1918 that methods were devised whereby the separation of lead and zinc could be effected especially by flotation. In North America, the Sullivan mine in Columbia was the first to introduce selective flotation of lead and zinc on a large scale, and it was not until 1924 that this new method became apparent in the production of lead in the United States. The fact that a reasonably clean separation of lead and zinc could be effected by this relatively simple process has made possible the profitable operation of many mines that otherwise would have been forced to close down as a result of the price level that prevailed during the greater part of 1927. Flotation has been brought into the category of commercial processes that was not previously workable.

In the Mississippi Valley and Eastern States, where the process is less complex, the effect of this new method has not been so marked. In consequence, operators in these States have not been able to compete with western competitors. The reasons for this are somewhat complex. By using selective flotation the mills in the West are making cleaner products and recovering more zinc

penalized as formerly the zinc thus saved is now being paid for and contributes to the operating income of the mining companies. Moreover, much of this new western zinc, being refined electrolytically, is produced so cheaply that under existing conditions it introduces serious competition as against the product of retort smelting in Central and Eastern States. This situation relates primarily to zinc but nevertheless reacts upon the lead situation. To be sure, the Western States have increased lead production but did so essentially because they have been able vastly to increase their zinc output. In the Mississippi Valley, on the other hand, the considerable proportion of the lead production coming from small mines that are dominantly zinc producers has been curtailed. The fact that the Mississippi Valley ores do not contribute by-product credits in the form of gold and silver values is a further handicap, as it makes the eastern miners depend even more than the western operators upon the prices of lead and zinc. As noted elsewhere, the output of lead in the Mississippi Valley and Eastern States, as measured by smelter records, decreased from 313,891 tons in 1926 to 288,347 tons in 1927, or 8.1 per cent.

The American Bureau of Metal Statistics has canvassed the principal western producers to determine the amount of lead and zinc ores beneficiated by flotation. The following table shows the rapid increase in lead output by selective flotation and other flotation methods. The production of 220,000 tons of lead in flotation concentrates in the Western States in 1927, accounted for by the incomplete canvass above cited, indicates that at least 56 per cent of the lead from western ores was recovered by flotation.

Lead and zinc concentrates directly obtained in Western States by use of flotation methods, 1921-1927

[Based on figures compiled by the American Bureau of Metal Statistics and published in its Year Book for 1927, p. 70]

Year	Lead concentrates							Lead content of zinc concentrates	
	Total produced			Lead content					
	Short tons	Increase		Percent	Short tons	Increase		Short tons	Percent
		Short tons	Per cent			Short tons	Per cent		
1921	107,237			41.87	44,904			4,467	11.60
1922	117,405	10,168	9.48	41.86	49,131	4,227	9.41	6,105	7.43
1923	166,777	49,372	42.05	42.07	70,166	21,035	42.81	11,181	6.80
1924	235,049	68,272	40.94	41.17	96,774	26,608	37.92	14,368	6.30
1925	320,513	85,464	30.36	43.46	130,264	42,490	43.91	11,064	4.36
1926	392,220	71,707	22.37	46.67	183,047	43,783	31.44	16,722	4.13
1927	451,617	59,397	15.14	48.77	220,236	37,189	20.32	12,293	2.46

The first significant feature is the increase of western concentrates being produced by flotation (mainly selective). The greatest absolute increase was made in 1925, when an increase of 85,464 tons of concentrates was obtained. Following that lesser increases of 71,707 tons in 1926 and 59,397 tons in 1927 were made. The increase of

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1922 over 1921 was 9.48 per cent; of 1923 over 1922, 40.94; of 1925 over 1924, 36.36; of 1926 over 1925, 22.37; and of 1927 over 1926, 15.14 per cent. During 1921 to 1927, the amount of lead in the concentrates increased from 41.87 per cent in 1921 to 41.17 per cent in 1922 to 48.77 per cent in 1927, a net increase of 6.90 per cent. Such an increase indicates that much progress has been attained since 1921, and some further increase may be expected in the future as more improved methods are found. The percentage of lead in the zinc concentrates has increased from 11.6 per cent in 1921 to 2.46 per cent in 1927.

CANADA

The Canadian mills are likewise treating a large amount of lead ores by selective flotation. The production figures are shown in the table that follows.

Flotation concentrates produced in Canada,

[Figures compiled by the American Bureau of Metal Statistics]

Year	Lead concentrates			Zinc concentrates	
	Short tons	Lead content		Short tons	Zinc content
		Short tons	Per cent		Short tons
1924	125,870	80,822	64.21	127,901	55,014
1925	161,103	101,908	63.26	163,252	75,464
1926	182,885	118,243	64.65	206,641	93,899
1927	204,764	137,789	67.29	206,867	99,009

Selective flotation in Canada has also materially increased lead and zinc output as well as the zinc or lead concentrates. The lead content of the zinc concentrates has been reduced from 5.41 per cent in 1924 to 2.94 per cent in 1927. This influence hitherto has been similar to that in the United States and Canada. The production of lead in 1927 was more than three times that in 1924, and the percentage of lead in the zinc concentrates is lower than in Canada, and the percentage of lead in the zinc concentrates is lower than in Canada. The complexity of the Mexican situation, the large number of mines producing, has no doubt retarded the recovery. The increased production of 1927 was largely due to installations using selective flotation.

MEXICO

Many of the Mexican mines have shown a marked increase in depth. Thus, a good price for lead is necessary to insure their continued operation and will probably have an increasingly beneficial influence upon the situation. This influence hitherto has been similar to that in the United States and Canada. The production of lead in 1927 was more than three times that in 1924, and the percentage of lead in the zinc concentrates is lower than in Canada, and the percentage of lead in the zinc concentrates is lower than in Canada. The complexity of the Mexican situation, the large number of mines producing, has no doubt retarded the recovery. The increased production of 1927 was largely due to installations using selective flotation.

The following paragraphs are quoted from the American Bureau of Metal Statistics statement of April 23, 1928:

The returns from 17 operations in this country [Mexico], beginning with 1925, give the following totals:

Flotation concentrates produced in Mexico, 1925-1927

Year	Lead concentrates			Zinc concentrates				
	Short tons	Lead content		Short tons	Zinc content		Lead content	
		Short tons	Per cent		Short tons	Per cent	Short tons	Per cent
1925.....	108,622	34,808	31.58	85,206	41,693	48.87	2,305	2.70
1926.....	142,160	61,124	43.00	190,893	96,680	50.65	2,687	1.41
1927.....	211,107	96,316	45.62	288,160	138,260	51.55	5,327	1.99

By comparison with the Mexican official statistics it appears that our totals as given above are only a little short of the total zinc production of Mexico.

Most of the zinc concentrate of Mexico is beneficiated pyrometallurgically, principally in Europe, but a small part of its ore is distilled domestically and a part is distilled in bond in the United States. Of the exportation of ore to Europe, a small part goes to electrolytic extraction there.

DOMESTIC PRODUCTION

RÉSUMÉ

Refined primary lead produced in the United States in 1927 from domestic ores amounted to 668,320 short tons, valued (at the average selling price of refined lead, 6.3 cents a pound) at \$4,208,000. These figures represent decreases of 1.8 per cent in quantity and 22.7 per cent in value as compared with 1926. Refined lead produced in 1927 from foreign ores and base bullion amounted to 128,210 short tons, valued at \$16,155,000. The total output of refined lead in 1927 from both domestic and foreign ores and bullion was 796,530 tons, valued at \$100,363,000.

Refined lead produced and consumed in the United States, 1926 and 1927, in short tons

	1926	1927
PRODUCTION		
Domestic desilverized lead.....	366,413	378,889
Domestic soft lead.....	258,565	233,944
Domestic desilverized soft lead.....	55,707	55,487
Foreign desilverized lead.....	680,685	668,320
Antimonial lead.....	118,256	128,210
Secondary lead.....	798,941	796,530
CONSUMPTION		
Apparent consumption of primary lead, refinery stocks disregarded.....	717,653	663,412

¹ "Primary lead," which is produced directly from ore, is here distinguished from "secondary lead," which is obtained by refining skimmings, drosses, and old metals. The statistics of secondary lead are given on p. 358. In this report the word "lead" used without qualification means primary lead.

² For method of calculating consumption see p. 371.

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Refined lead produced in the United States from

Year	Short tons	Value	Y
1918.....	539,905	\$76,667,000	1923.....
1919.....	424,433	44,990,000	1924.....
1920.....	476,849	76,296,000	1925.....
1921.....	398,222	35,840,000	1926.....
1922.....	468,746	51,562,000	1927.....

QUANTITY AND VALUE OF RE

The total refined lead produced in the U 1927, from domestic ores and from foreign its value are shown in the following table:

Refined lead produced in the United S

Year	Short tons				
	Desilverized lead ¹	Desilverized soft lead	Soft lead ²	Total production ¹	From domestic ores and base bullion
1720-1912.....	7,176,497	326,991	2,929,180	10,432,668	8,545,38
1913.....	301,160	29,433	131,867	462,460	411,87
1914.....	340,397	43,506	158,219	542,122	512,70
1915.....	344,503	44,001	161,461	550,055	507,02
1916.....	335,376	38,244	167,515	571,134	552,22
1917.....	305,998	56,268	188,503	510,769	548,40
1918.....	352,314	47,418	210,403	610,135	539,04
1919.....	266,538	67,938	147,744	482,220	424,43
1920.....	273,135	66,668	180,854	520,657	476,84
1921.....	238,329	62,747	157,513	458,589	398,22
1922.....	249,107	74,305	209,260	532,662	468,74
1923.....	366,241	61,332	190,749	618,322	543,84
1924.....	423,429	63,449	203,616	690,493	666,40
1925.....	457,477	48,032	200,860	706,369	664,92
1926.....	484,669	55,707	258,565	798,941	680,68
1927.....	507,099	55,487	233,944	796,530	668,32
1927.....	12,512,358	1,162,426	5,798,002	19,473,786	16,500,00

¹ The lead content of antimonial lead is excluded; see page 357.

² Desilverized soft lead is excluded.

Details for years prior to 1913 are shown. The value for years prior to 1917 is based on New York; beginning with 1917 the value is based on each producing company.

PRODUCTION BY ST

SMELTER OUTPUT

The smelter production of primary lead from domestic and foreign sources for 1927 was 668,320 short tons. While the total output was essentially the same, a substantial decrease in the Mississippi Valley tailments in the production from many of the operations. This decrease was balanced, however, by the output of the Western States. The selective flotation methods applied to the tailments increased the output of lead there.

The St. Joseph Lead Co.'s output and sales in 1927 were as follows:

Lead produced and sold by the St. Joseph Lead Co. in 1927, in short tons

Ore	4,527,698
Lead concentrates	203,381
Lead content	151,048
Lead:	
Smelter output	140,878
Sales	193,258

Comparable figures for the National Lead Co. and the United States Smelting, Refining & Mining Co. are not available from the company reports.

SOFT LEAD

The lead ores of the Mississippi Valley are almost free from silver, and the lead produced from them is known as soft lead. The output of the lead pigment known as basic lead sulphate or sublimed lead is derived from these ores. The galena of southern Illinois, however, is notably argentiferous as compared with the rest of the Mississippi Valley ores, and the silver-bearing lead concentrates treated during the period 1918 to 1927 averaged 5.5 ounces of silver to the ton; lead in this district is now produced only incidentally in the concentration of fluorspar, and the output is small compared with that from other districts in the Mississippi Valley States. Lead ores from southeastern Missouri also contain a small amount of silver; over a series of years the silver-bearing lead concentrates smelted have averaged about 1 ounce of silver to the ton.

Zinc ores from Western States, that are made into zinc oxide, contain some lead which goes into the pigment; therefore it is not strictly accurate to speak of the lead content of pigments as wholly soft lead.

In some years small quantities of foreign lead have also been made into pigments in the United States. During the period covered by the following table, lead pigments were made from foreign ores in 1924 only, when the lead so recovered amounted to 545 tons.

Soft lead produced in the United States from domestic ores, 1922-1927, in short tons

Year	Soft pig lead			Soft lead recovered in pigments	Total soft lead	Total domestic lead ¹	Soft lead percentage of domestic lead
	Undesilverized	Desilverized	Total				
1922	209,250	74,305	283,555	13,892	297,447	488,448	61
1923	190,749	61,332	252,081	13,612	265,693	587,247	47
1924	203,616	63,449	267,064	15,338	282,402	593,486	48
1925	280,560	48,932	309,492	17,547	327,039	684,968	48
1926	258,565	55,707	314,272	13,685	327,957	710,990	46
1927	233,944	55,487	289,431	13,136	302,567	695,500	44

¹Includes domestic refined lead, domestic lead in antimonial lead, and domestic lead in pigments.

ANTIMONIAL LEAD

By-product antimonial lead, produced by lead refineries in 1927 from domestic and foreign argentiferous lead ores, amounted to 24,347 short tons. This amount is 8 per cent greater than the

production in 1926 and exceeds any previous annual considerable antimonial lead from secondary sources was at the regular lead smelters in 1927; the figures covering this production are given in the chapter on secondary metals. Antimonial lead from which antimonial lead is obtained is produced in the United States and in Mexico.

By-product antimonial lead produced in the United States, 1918-1927, in short tons

Year	Total lead refined from base bullion (short tons)	By-product antimonial lead (percentage of total)	Antimonial lead				Total antimonial lead (short tons)
			From domestic ore (short tons)	From foreign ore (short tons)	Total		
					Short tons	Value	
1918	382,314	4.9	10,777	7,793	18,570	\$2,820,350	2
1919	266,538	5.2	9,674	4,200	13,874	1,513,968	1
1920	279,135	4.6	9,604	2,931	12,535	1,963,255	2
1921	238,329	4.2	7,881	2,183	10,064	870,069	1
1922	249,107	3.2	7,068	1,007	8,075	844,992	1
1923	366,241	3.9	11,635	2,555	14,190	1,950,870	2
1924	423,429	4.9	13,563	7,194	20,757	3,276,713	2
1925	467,477	4.3	14,472	5,195	19,667	3,788,547	2
1926	484,689	4.6	18,909	3,615	22,524	3,916,714	2
1927	507,099	4.8	16,040	8,307	24,347	3,277,043	2
Average		4.5					

LEAD PIGMENTS

The amount of lead contained in pigments was 37 per cent of the total lead recovered from domestic ores in 1926. By far the largest part was pigments manufactured from pig lead—white lead, mineral, and litharge. In these pigments the lead recovered in the pigment known as basic lead sulphate, white lead and sublimed blue lead, and a still smaller amount recovered from the zinc ores from which the zinc oxide

Lead in pigments, 1922-1927, by sources, in short tons

Year	Lead in pigments	
	Domestic ore	Foreign ore
1922	13,892	
1923	13,612	
1924	15,338	545
1925	17,547	
1926	13,685	
1927	13,136	

¹Includes also lead recovered in zinc oxide and leaded zinc oxide.

SECONDARY LEAD

The recovery of lead from waste material is a branch of the metal industry that has expanded notably in recent years. Lead is a relatively cheap metal; but for various reasons, one of which is that it does not rust, it is rather easily reclaimed. The main sources of secondary lead are old pipe, cable sheathings, battery plates, lead linings for acid tanks, and such alloys as babbitt, solder, and type metal. The most important of these sources now is old battery plates which, as stated elsewhere in this chapter, are likely to be returned within two or three years after being fabricated.

Secondary lead recovered in the United States, 1922-1927, in short tons

Year	Pig lead	Lead in alloys	Total recovered lead	Value
1922.....	75,480	84,080	159,560	\$17,552,000
1923.....	96,430	98,060	194,490	27,220,000
1924.....	90,400	114,100	204,500	32,720,000
1925.....	112,420	114,400	226,820	39,477,000
1926.....	125,000	132,300	257,300	44,368,000
1927.....	110,000	157,000	267,000	34,776,000

A number of years ago the principal use for lead was in pigments, and lead consumed in this way could not be reclaimed. As time passes the increased tendency toward nondestructive uses is building up a large reserve of lead which will eventually become available for reuse. The quantity of secondary lead recovered in the United States has increased largely not only in actual tonnage but also relatively in respect to the primary production of lead. As the following table indicates, the ratio of secondary lead recovery to refined primary lead production in the United States has been increasing quite rapidly since 1918, amounting in 1927 to 41 per cent.

Ratio of secondary lead recovered to refined primary lead produced in the United States from domestic ores, 1913-1927

Year	Refined primary lead		Secondary lead	
	Short tons	Average price per pound (cents)	Short tons	Ratio to refined primary lead (per cent)
1913.....	411,878	4.4	72,834	18
1914.....	512,794	3.9	61,062	12
1915.....	507,026	4.7	78,900	16
1916.....	552,228	6.0	96,300	17
1917.....	548,460	8.0	93,500	17
1918.....	639,905	7.1	97,100	15
1919.....	424,433	6.3	122,100	29
1920.....	476,840	8.0	124,050	26
1921.....	398,222	4.5	104,780	26
1922.....	468,746	6.5	159,560	34
1923.....	543,841	7.0	194,490	36
1924.....	506,407	8.0	204,500	36
1925.....	654,021	8.7	226,820	35
1926.....	680,686	8.0	257,300	41
1927.....	668,320	6.3	276,000	41

LEAD

LEAD SMELTERS AND REFINERIES

UNITED STATES

Arizona: Douglas—Phelps Dodge Corporation. (Smelter and refinery.)
 California: Selby—Selby plant, American Smelting & Refining Co. (Smelter and refinery.)
 Colorado:
 Durango—Durango plant, American Smelting & Refining Co. (Smelter.)
 Leadville—Arkansas Valley plant, American Smelting & Refining Co. (Smelter.)
 Idaho: Kellogg—Bunker Hill & Sullivan smelter. (Smelter.)
 Illinois:
 Chicago—Great Western Smelting & Refining Co. (Smelter and refinery.)
 Collinsville—St. Louis Smelting & Refining Works. (Smelter and refinery.)
 Federal—Federal plant, American Smelting & Refining Co. (Smelter.)
 Indiana:
 East Chicago—International Lead Refining Co. (Smelter and refinery.)
 East Chicago—U. S. S. Lead Refinery (Inc.). (Smelter and electrolytic process.)
 Kansas: Galena—Galena plant, Eagle-Picher Lead Co. (Smelter.)
 Missouri:
 Granby—American Zinc, Lead & Smelting Co. (Smelter and refinery.)
 Hercules—St. Joseph Lead Co. (Smelter and refinery.)
 Joplin—Eagle-Picher Lead Co. (Smelter.)
 Montana: East Helena—East Helena plant, American Smelting & Refining Co. (Smelter.)
 Nebraska: Omaha—Omaha plant, American Smelting & Refining Co. (Smelter and electrolytic process.)
 New Jersey:
 Newark—Balsbach plant, Balsbach Metals Corporation. (Smelter and refinery.)
 Perth Amboy—Perth Amboy plant, American Smelting & Refining Co. (Smelter and refinery.)
 Oklahoma: St. Louis—Ontario smelter, Eagle-Picher Lead Co. (Smelter.)
 Pennsylvania: Carnegie—Pennsylvania Smelting Co. (Smelter.)
 Texas: El Paso—Kansas City Consolidated Smelting & Refining Co. (Smelter.)
 Utah:
 International—International Smelting Co. (Smelter and refinery.)
 Midvale—United States Smelting, Refining & Mining Co. (Smelter and refinery.)
 Murray—Murray plant, American Smelting & Refining Co. (Smelter.)

CANADA

British Columbia: Trail—Consolidated Mining & Smelting Co. (Ltd.). (Smelter and refinery. Betts electrolytic process.)
 Ontario: Galletta—Kingdon Mining, Smelting & Manufacturing Co. (Smelter.)

MEXICO

Chihuahua: Chihuahua—American Smelting & Refining Co. (Smelter.)
 Coahuila: Torreon—American Metal Co. (Ltd.). (Cia. Minera.) (Smelter.)
 Nuevo Leon:
 Monterrey—American Metal Co. (Ltd.). (Cia. Minera.) (Smelter and refinery.)
 Monterrey—American Smelting & Refining Co. (Smelter.)

¹ Smelters of secondary materials but treating some lead ores.

² Subsidiary of Anaconda Copper Mining Co.

³ Subsidiary of the United States Smelting, Refining & Mining Co.

⁴ Plant idle in 1927.

⁵ Subsidiary of American Smelting & Refining Co.

San Luis Potosi: San Luis Potosi—American Smelting & Refining Co. (Cia. Minera Asarco.) (Smelter.)
Zacatecas: La Fé—American Metal Co. (Ltd.). (Cia. Minera de Peñoles—S. A.) (Smelter.)

WORLD PRODUCTION

The accompanying table shows the world smelter production of lead from 1923 to 1927, so far as statistics are available. Official figures are used wherever possible. For countries for which official data are not yet available the figures are taken from the yearbook of the American Bureau of Metal Statistics and other reliable sources, but in later reports the official statistics will be substituted as received.

There is always a possibility of duplication in figures showing world production, for some countries send "work lead" or base bullion to other countries for refining. Furthermore, secondary and antimonial lead are often included in the figures showing the foreign production, so that to put the United States on an equal footing in the statistics it would be necessary to include in its production the domestic output of secondary lead and antimonial lead. The figures for the United States represent refined lead that was smelted in this country, whatever the source of the ore.

World production of lead, 1923-1927, in metric tons¹

(Compiled by E. M. Jones, of the Bureau of Mines).

Country ²	1923	1924	1925	1926	1927
Australia	132,165	130,494	151,260	155,398	167,616
Austria	4,255	4,985	5,408	6,478	5,068
Belgium	65,680	77,950	87,480	87,878	(³)
Canada	46,601	75,503	104,303	123,644	155,854
Chosen	(⁴)	(⁴)	485	1,958	(⁵)
Czechoslovakia	2,452	2,264	2,212	2,316	2,394
France	17,413	18,881	20,543	19,500	24,900
Germany	58,110	67,427	89,458	96,200	81,800
Greece	6,815	5,387	4,811	4,384	6,182
India (Burma)	4,235	5,106	5,356	5,059	4,800
Italy	46,799	52,590	48,034	55,202	67,027
Japan	17,131	22,062	24,475	23,590	23,774
Mexico	2,700	2,941	3,337	3,010	4,000
Northern Rhodesia	150,028	133,440	163,236	170,723	227,870
Norway	11,197	5,909	3,406	3,897	6,042
Poland	242	642	406	385	(⁵)
Rumania	17,195	20,223	28,631	27,924	28,843
Russia ⁶	402	433	489	655	1,850
South America	307	565	1,021	1,342	1,850
Southwest African Protectorate ⁷	5,081	11,705	11,964	17,663	14,604
Spain	622	813	1,019	1,682	8,898
Sweden	127,514	141,849	153,567	149,514	144,023
Tunisia	119	477	723	822	330
Turkey	14,090	14,291	16,013	18,108	10,662
Union of South Africa	1,400	5,104	4,786	6,014	7,832
United States (refined)	5,999	4,609			
Yugoslavia	515,395	555,913	636,700	661,239	635,651
	9,250	11,487	10,956	9,809	3,135
Known production	1,261,000	1,373,000	1,581,000	1,655,000	1,703,000

¹ By countries where smelted but not necessarily refined.

² In addition to the countries listed, China smelts lead, but no reliable data of output are available.

³ Official estimates of production in 1925 range from 1,500 to 3,500 tons. Indo-China, also, reported 1 ton of lead smelted in 1926, and 4 tons in 1927.

⁴ Estimate included in total.

⁵ Data not available.

⁶ Exclusive of Upper Silesia.

⁷ Data for year ended Sept. 30.

⁸ Data for year ended Mar. 31 of year following that stated.

SUMMARY OF PRODUCTION AND CONDITIONAL LEAD-PRODUCING REGIONS

Despite its slight decrease in production from 1926, the United States remained in first position as a lead producer in 1927. In Spain and in a few of the less important countries production also decreased in 1927, but the world as a whole showed an increase as compared with 1926. Mexico retained second place in 1927 and was followed by Australia, Spain, Canada, Germany, India, and the United States. A decrease in Spain was due mainly to the lowering of a serious situation was partly prevented by Government action. Australia increased production through further metallurgical efficiency, that Commonwealth is suffering from labor difficulties which again threaten a shutdown of properties in the Broken Hill district unless they are worked substantially. The entry of the Anaconda Company was to no little extent responsible for the increase in production. This company is now the main producer in Potosi. Other producers were forced to curtail output somewhat in 1927, a condition brought about by a decided shortage of lead, coupled with an improvement in the value of the metal. France and Germany both report a substantial production of lead, the ores from which it is obtained are imported from Mexico, and Greece. France, however, produces lead in the departments of Gard and Lozère. Relatively small quantities are obtained from mines in Argentina, Chile, Russia, Turkey, and Rhodesia.

MEXICO

Mexico, even with the decreased price of lead, has been able to produce some of the mines, showed an increased production of 227,870 metric tons of lead in 1927 compared with 227,870 metric tons of lead in 1926. Many of the mines are exhausted of the bonanza ores and will in consequence work the complex sulphides that usually underlie the bonanza ore. This means that more effective mining methods must be adopted.

By far the greater proportion of the lead ore in Mexico comes from limestone replacement deposits known as great manta deposits of Santa Eulalia, Mapimé, and Mojada continued their outputs, as did a number of other mantas. The great lead region, comprising Chihuahua, western Coahuila, Durango, and eastern Sonora, has reserves sufficient for a few years ahead, even though the ores are very hard to estimate accurately. The mantas contain oxide ores, generally branch from a common group of chimneys, in which the ores are complex, containing lead, zinc, antimony, copper, and free gold. The selective flotation are rapidly gaining a foothold in Mexico, and methods are decidedly applicable to the ores in this region. Mexico's production can be maintained at a comfortable level and under reasonably low market conditions, probably responsible for the added production.