

March 30, 1972

Mr. James R. Gervan
State Supervisor of Art & Music
Education
Department of Public Instruction
Dover, Delaware 19901

Dear Mr. Gervan:

The sampling results obtained from the preliminary survey of the atmosphere adjacent to kilns during the firing of ceramic ware are summarized in the enclosed table. The atmospheric levels of lead are expressed in micrograms (μg) of lead per cubic meter of air and appear in the extreme right column. Three samples were obtained at John Dickinson High School, three samples at William Penn High School, and one sample at the Faulk Road Elementary School. All of the atmospheric levels are below the accepted occupational exposure level of $200 \mu\text{g Pb/m}^3$ of air, which is the standard under the Occupational Safety and Health Act of 1970. This level is accepted as the occupational exposure level at which an employee can be exposed for 8 hours per day for a lifetime with no adverse health effects.

The sample identified as sample no. 1 represents a very high value and if it is indeed representative of the atmosphere, the kiln should be equipped with a hood and exhausted into the outside atmosphere. However, sample no. 3 was obtained at the same sample location as sample no. 1, but approximately one hour later, and with the same sampler. In view of the results of sample no. 3 and the low levels represented by the other samples, I tend to believe that sample no. 1 was contaminated and is not a representative sample. Contamination sometimes does occur, and is significant because of the relatively low levels of lead in the samples. However, I do not know positively that the sample was contaminated and therefore cannot overlook this result. The other atmospheric samples are relatively low and do not represent any significant exposure. Samples nos. 2 and 5, which were obtained with the small Wilson Casella personal sampler, operating at a much lower flow rate, are not as reliable as the samples obtained with Gelman pump (samples # 1, 3, 4, 6, and 7), which samples at a much higher flow rate.

As you know, this was a cursory study to determine whether the atmospheric levels of lead in the vicinity of kilns were high or low, and was not designed to provide details of kiln emissions. Unfortunately, because of the high value of sample no. 1, there can be no definite conclusion without additional information. We plan to do additional sampling around kilns of a similar type here in the New York area, in an effort to obtain conclusive information. This sampling cannot be fit into my schedule before approximately mid April. I will forward the results to you as they become available.

LIA-76943

Any possibility of airborne contamination could be completely eliminated by providing each kiln with an exhaust hood and adequate exhaust system. In order to determine whether the personnel working continuously in the kiln area have excessive absorption of lead, analysis of blood for lead is recommended. The normal concentration of lead in blood is in the range of 15-20 μg of lead per 100 grams of whole blood. These values are found in the general population and are representative of personnel with no occupational exposure. Blood lead concentrations of 80 $\mu\text{g}/100\text{ gm}$ are considered safe for workmen engaged in the lead trades.

If you desire to have blood samples analyzed for lead, I would be happy to recommend qualified laboratories to do the analyses. Also, perhaps Mrs. Evans' physician would like to consult with physicians who have had much experience in dealing with occupational lead exposures. I would recommend Dr. Lloyd Tepper, Department of Environmental Health, University of Cincinnati, Kettering Laboratory, Cincinnati, Ohio 45219, telephone 513-872-5704, as a highly qualified medical expert in the lead area.

I am enclosing a copy of the reprint, "Industrial Lead Poisoning" by Dr. Robert A. Kahoe, which provides much information on occupational exposures to lead.

Thank you very much for your assistance, and if we can provide additional information, please do not hesitate to contact us.

Sincerely,

D. R. Lynam
Assistant Director,
Environmental Health

Enc.

cc: J. F. Cole
J. F. Smith

lp

LIA-76944

SUMMARY OF RESULTS						
Sample Number	School	Date and Time of Sampling Period	Description of Kiln	Kiln Temperature (Pyrex. Thermist)	Sampler Location	Type of Sample
1	John Dickinson High School	3-7-72, 2:12 to 3:12 P.M.	Hexagonal Dry Stacking (3 stacks) L & L Econo Kiln Model K-230 32 inches wide 32" High	1250 at 3:30 P.M. 1325 at 3:40 P.M. 1500 at 3:09 P.M.	1 1/2' feet directly above top of kiln	Colman Pump and glass fiber filter, operating at 22 L.P.M. 13 1/2 46
2	John Dickinson High School	3-7-72, 2:13 to 4:17 P.M.	Same as Sample #1	Same as Sample #1 & Sample #3	Same as Sample #1	Wilson Cassella Residual Samples glass fiber filter operating at 22 P.M. 29.03
3	John Dickinson High School	3-7-72, 3:16 to 4:17 P.M.	Same as #1	1500 at 3:20 1650 at 3:50 1700 at 4:12	Same as Sample #1	Same as Sample #1 1.86
4	William Penn High School	3-8-72, 9:43 to 10:43 A.M.	Fine Art Electric Kiln, Type FAB8 6' high, front loading	1150 at 9:50 A.M. 1350 at 10:35	7' above floor, 1 1/2' horizontally in front of kiln	Same as Sample #1 2.80
5	William Penn High School	3-8-72, 9:44 to 11:59 A.M.	Same as Sample #4	Same as Samples #4 and #6	horizontally 4' from front of kiln, at 6' height	Same as Sample #2 8.52
6	William Penn High School	3-8-72, 10:45 to 12:14 P.M.	Same as Sample #4	1550 at 11:21 1750 at 12:14	Same as Sample #4	Same as Sample #1 2.25
7	Fallen Road Elementary	3-8-72, 1:54 P.M. to 3:24 P.M.	AMACO High Fire Top High Fire #95 30' x 28' x 13 1/4'	750 at 1:54 850 at 2:35 875 at 3:00 900 at 3:20	1 1/2' directly above top of kiln	Same as Sample #1 1.26

LIA-76945

N 1046.01

SUMMARY OF RESULTS

Sample Number	School	Date and Time of Sampling Period	Description of Kiln	Kiln Temperature (Degrees Fahrenheit)	Sampler Location	Type of Sampler	Atmospheric Levels of Lead (ug/m ³)
1	John Dickinson High School	3-7-72 2:12 to 3:12 P. M.	Hexagonal dry stacking, (3 stacks) L & L Econo Kiln model K-230 32 inches wide 32 inches high	1250 at 2:30 P. M. 1325 at 2:40 P. M. 1500 at 3:09 P. M.	1-1/2 feet directly above top of kiln	Gelman pump and glass fiber filter, operating at 22 LPM	134.48
2	John Dickinson High School	3-7-72 2:13 to 4:17 P. M.	Same as Sample # 1	Same as Sample # 1 & Sample # 3	Same as Sample # 1	Wilson Casella personal sampler, glass fiber filter operating at 2 LPM	29.03
3	John Dickinson High School	3-7-72 3:16 to 4:17 P. M.	Same as Sample # 1	1500 at 3:20 P. M. 1650 at 3:50 P. M. 1700 at 4:12 P. M.	Same as Sample # 1	Same as Sample # 1	1.86
4	William Penn High School	3-8-72 9:45 to 10:43 A. M.	Fine art electric kiln Type FA88, 6 inches high, front loading	1150 at 9:50 A. M. 1350 at 10:35 A. M.	7 feet above floor, 1-1/2' horizontally in front of kiln	Same as Sample # 1	2.80
5	William Penn High School	3-8-72 9:44 to 11:59 A. M.	Same as Sample # 4	Same as Samples # 4 and # 6	4 feet horizontally from front of kiln, at 6 feet height	Same as Sample # 2	8.52
6	William Penn High School	3-8-72 10:45 A. M. to 12:14 P. M.	Same as Sample # 4	1550 at 11:21 A. M. 1750 at 12:14 P. M.	Same as Sample # 4	Same as Sample # 1	2.25
7	Faulk Road Elementary	3-8-72 1:54 to 3:24 P. M.	AMACO High Fire, Type High Fire # 95, top loading 28" x 28" x 24"	750 at 1:54 P. M. 850 at 2:35 P. M. 875 at 3:00 P. M. 900 at 3:20 P. M.	1-1/2 feet directly above top of kiln	Same as Sample # 1	1.26

LIA-76946

Sample #1 - $\frac{22 \text{ ppm}}{60 \text{ min}}$
 $\frac{13.2 \text{ liters}}{13.2 \text{ meters}^3} \approx 1.32 \text{ meters}^3$

$$\frac{177.5 \text{ wgs}}{1.32 \text{ meters}^3} \approx 134.46 \text{ wgs/m}^3$$

Sample #2 - $\frac{124 \text{ minutes}}{2 \text{ ppm or } 4 \text{ ppm}}$
 $\frac{248 \text{ liters}}{248} \approx 1.248 \text{ meters}^3$ $\frac{7.2}{248} \approx 29.03 \text{ wgs/m}^3$

Sample #3 - $\frac{22 \text{ ppm}}{61 \text{ min}}$
 $\frac{13.2}{13.42} \approx 1.342 \text{ meters}^3$ $\frac{2.5 \text{ wgs}}{1.342 \text{ meters}^3} \approx 1.86 \text{ wgs/m}^3$

Sample #4 - $\frac{22 \text{ ppm}}{60}$
 $\frac{13.2 \text{ liters}}{13.2} = 1.32 \text{ m}^3$ $\frac{3.7}{1.32} \approx 2.80 \text{ wgs/m}^3$

Sample #5 - $\frac{135 \text{ min}}{2}$
 $\frac{270 \text{ liters}}{270} = 1.270$ $\frac{2.3}{1.270} = 8.52 \text{ wgs/m}^3$

Sample #6 - $\frac{22 \text{ ppm}}{89 \text{ min}}$
 $\frac{19.8}{19.58} \text{ liters} = 1.958$ $\frac{4.4}{1.958 \text{ liters}} = 2.247 \text{ wgs/m}^3$

LIA-76947

Sample # 7 -

8.41

3.29

1.54

1:30

2L

90 min.

198.0 liters = 1.98 m³

1.545

1.98

= 1.26 $\frac{\text{kg}}{\text{m}^3}$

LIA-76948

Four MRS. Carter loan a machine for a day →
 Program → 101 →
 Bring program in tomorrow →

Russell FRANK - FERRO

not around hobby Kilns -

around production equipment -

depl. store - characteristic sweet taste →
 hot plate using Jewellers enamel

both a sweet taste & Selenium

of how a hobbyist can be tricked into an unusual situation -

Russell FRANK

Pot smelters. 55 gal. barrel as high as
 All are gas fired → non circulating inert furnaces

Greatest exposure during

moisture expanded into a vapor
 have a drying action - have a pressure

went to a box type furnace without circulating fan -
~~what the hell was I thinking~~

LIA-76949

Work to identify that condition exists →

Are getting material out

John MARQUIS → of PennCO -
[did some work regarding to circulation air in Kiln -
E₁ atmosphere]

Where present in

Selenium, Cadmium

→ freshly Recreated Crystalline metal ←

Reducing in a furnace:

oxide, reduce, condense when it pours out the end.

how much lead - etc. was in the glazes
one major customer - S. ^{power} HALL WARD. Reprocessed
Rest of people will smelt a lot of their own materials.

National Ceramics hobby manufacturers -

dead

ORGANIC +
colors applied as ^{ORGANIC} of Resin & Resins burns out &
leaves oxide on surface -

Encourage to go on →
Identify 6 families of glazes →

possible contamination

① 177.5 wgs. Joh.

possible contamination

② 7.2 wgs.

③ 2.5 wgs.

④ 3.7 ✓ 1

⑤ 2.3 ✓ 2 1/2

⑥ 4.4 ✓ 12

⑦ 2.5 wgs. P6

22 LPM

60
1320 LPM

$$\frac{177.5 \text{ wgs.}}{1320 \text{ LPM}} = 0.1 \frac{\text{wgs}}{\text{LPM}}$$

0.1 mg/m³

2090

LIA-76951

Casella - hooked up with an y flow meter

of TRIO → at 2 LPM - (top mark) 4 LPM

at a setting of two on Casella meter →

Adj. Flow meter to 2 LPM & brush ball → 1.95 & 2.1

ball on Casella scale → Probably at 4 LPM rather than

2 → flow meters in series here →

calibrate with wet test meter - the flow meters -

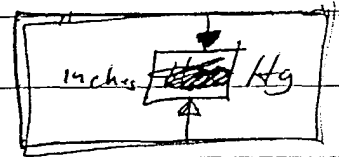
not went over with Bill -

most of curves →

5-25 Lpm scale -

how accurate reading of bell

Data Meter Reading → Wet Meter Result:	
5 Lpm	6.55 Lpm
10	11.80 Lpm
15	17.80 Lpm
18	20.5 Lpm



Wet Test meter before flow meter & pump

fitting - wet test meter pump →

SAMPLING Data

Date: 8-8-72

Weather Conditions:

Time: 1.

~~with~~ Faulk Road Elementary School

Name _____

Alfred I de Post School District

Location	Gauge Reading (H₂O)	Gauge Type	Time	Comments
Sample #7 Gelman Pump	20 LPM High 19 LPM - 2:35 19 LPM - 3:00 19 LPM - 3:20 Sampler located at 1' above top of kiln 3' left of center		begin 1:54 P.M. end 3:24 P.M.	750° Kiln Temp - at 1:54 850° at 2:35 850° at 2:50 875° at 3:00 900° at 3:20 5" Hg - Vacuum Pump begin
Changed position of sample	at 2:11 sample placed 1 1/2' above top of kiln.			5" Hg. 2:35 5" " " " " 3:00
	Kiln - Top Loading 28 x 28 x 24" High.			The thermometer PYrometer - was stuck into viewing port, so there was not a tight seal.
Type Kiln: Amaco High Fire HF 95- Type High Fire 955 Serial No.				
	200 volts 28 Amps.			

LIA-76953

SAMPLING Data

Location: John Dickerson H/S Date: MARCH 7 1972
 Weather Conditions: 5 stars Time: 2:00 P.M.
 Large Hexagonal Dry Stackings 900 watts Name:
 L & L Econo Kiln model K-230 220 volt 4/17 Amps

Location	L & L manuf. Co. Twin Oaks Reading Gauge Reading (Inches H ₂ O)	Time	Comments
Sample #1 <u>20</u>	Gelman Pump (ESB) - begin 20 LPM end: 19 LPM Positioned at 1 1/2 feet directly above TOP of Kiln	begin 2:12 P.M. end 3:12	5" Vacuum on Pump Gauge Kiln Temp → 1250° F - 2:30 1325° F at 2:40 1500° F at 3:00
Sample #2	Wilson Casella positioned at app. 1 1/2 feet above Top of Kiln.	begin 1522.33 minutes - app. 2:13 P.M. ball at Lower Line on Rotameter 1589.13 not 3:22 end: 1644.27 at 4:17 P.M.	
Sample #3	Gelman Pump (ESB) same position as #1	begin - 20 LPM begin 3:16 P.M. end 4:17	5" Vacuum on Pump Gauge Kiln Temp - 1500° F at 3:20 1650 at 3:50 1700 at 4:12
Kiln: 30 inches wide at Top } outside dimensions 32 0.5 High			
AMACO Indianapolis American Art Clay Co. Inc			

LIA-76954

American Art Clay Co.

SAMPLING Data

Indianapolis Ind. 46222

Date: 3-8-1972

Weather Conditions: William Penn High School

Time: 9:30

Newcastle - Det. Gunning Bedford School

Chuck

Teacher

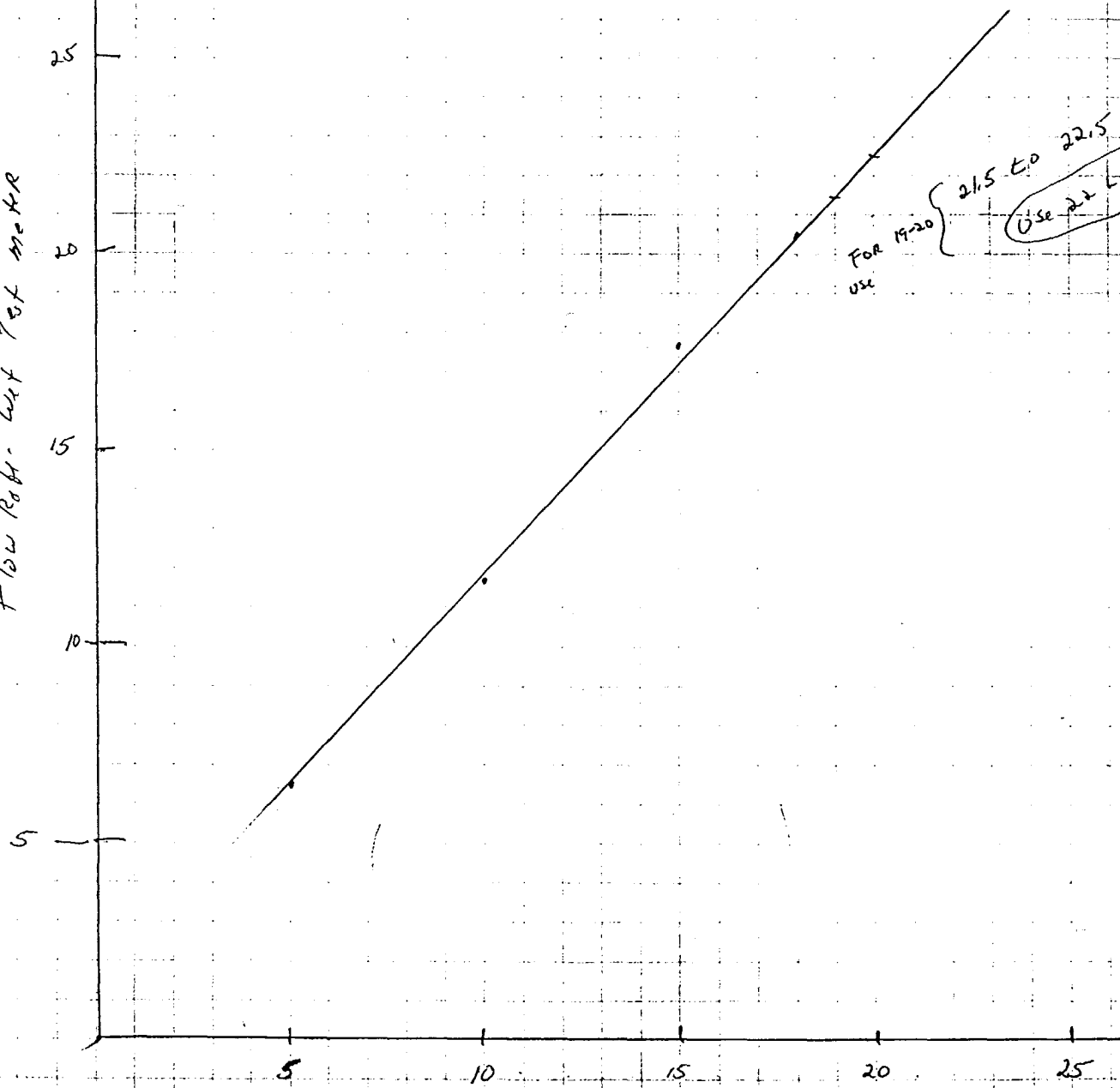
Director

Fine ART Electric Kila, An Amco Product Type FA88

Location	Gauge Reading (Amperes H ₂ O)	Gauge Temp	Time	Comments Serial No. 88492 28.7 Amperes 20.8 Volts
Sample #4 Positioned at 6' above floor level - 1 1/2' in front of Kila - 6' High Kila	begin at 20 LPM 20 LPM - at 10:03 20 LPM - at 10:33	at 10:03 at 10:33	begin 9:43 at 10:43 at 10:43 PM	4 1/2" Hg. on Gelman Pump. Kila Temp - 1150° - 9:50 1350° - 10:35
Sample #5 Wilson Personal Gasella Sampler.			begin 1644.29 min. - 9:44 1663.30 - 10:03 1694.07 - 10:34 1702.93 - 10:48 1779.25 - 12:10	at lower line on Rotameter 126 135 min 136 2:22 11:0 7:46 battery had just about stopped
Sample #6 Gelman	begin at 19.5 LPM. Same position - @ #4. 19.0 LPM at 11:21 19.0 - end at 12:14		begin 10:45	4" Hg. on Gelman Pump 1550° - 11:21 1750 - 12:14

LIA-76955

Flow Rate - Wet Test Meter



For 19-20 use { 21.5 to 22.5 }
Use 22 LPM

Rotameter Reading