

DEFENDANT'S
EXHIBIT

8-31-04

13

98

MONSANTO COMPANY
ORGANIC CHEMICALS DIVISION
ANNISTON, ALABAMA

STANDARD MANUFACTURING PROCESS
FOR
AROCLOR

Departments 826.08
826.17

Supersedes: Standard Manufacturing Process
For
Aroclor
Issued: 10/22/57

Prepared by: M. J. Williams
Prod. Foreman

Supervisor: V. R. Haupt

Production Supt.: R. G. Moody

Issued: September, 1966

COPY NO. 16

CONFIDENTIAL INFORMATION

This process is the property of Monsanto Company and the recipient is responsible for its safekeeping. It contains confidential information of Monsanto Company which must not be reproduced, revealed to unauthorized persons, or sent outside the Company without proper authorization. Either retain in a secure file or return to the Standards Department.

DSW 303353

WATER_PCB-SD0000050904

**STANDARD MANUFACTURING PROCESS
FOR
AROCLOR**

DISTRIBUTION

Technical Information Center - St. Louis Research	2 Copies
Security File	1 Copy
Technical Production Manager	1 Copy
Plant Manager	1 Copy
Production Superintendent	1 Copy
Production Supervisor	1 Copy
Maintenance Superintendent	1 Copy
Technical Services Department Files	3 Copies
Plant Standards Department	Extras

3 - This Copy For -

DSW 303354

STANDARD MANUFACTURING PROCESS
FOR
AROCLOR

TABLE OF CONTENTS

SECTION I	-	SYNOPSIS OF PROCESS
SECTION II	-	FLOW SHEETS
SECTION III	-	PROCESS IN DETAIL
		A. Batch Liquid Chlorinators
		B. Continuous Chlorinators
		C. Biphenyl Mix Tank and Feed Tank
		D. Solid Batch Chlorinators
		E. Blowing Tank Operation
		F. Liquid Stills
		G. Earth Treatment and Filtration
		H. Solid Still
		I. Packaging
SECTION IV	-	COMMENTS ON THE PROCESS
APPENDIX	-	A. Equipment List
		B. Finished Goods, By-Product, & Raw Material Specifications
		C. Physical Constants for Finished Goods, Intermediates, and Major Raw Materials
		D. Utilities
		E. Personnel
		F. Packaging and Shipping
		G. Material Balances
		H. Atmospheric & Stream Discharge & Waste Disposal
		I. Process Control Procedures
		J. Safety and Toxicity Data & Accident Reports
		K. Background Information
		L. Time Cycles
		M. Quality Considerations
		N. Miscellaneous

DSW 303355

STANDARD MANUFACTURING PROCESS
FOR
AROCLOL

AMENDING RECORD SHEET

The disposition of the Amendments to the Standard Manufacturing Process dated October 22, 1957 are as follows:

<u>Amendment No.</u>	<u>Title or Description</u>	<u>Date</u>	<u>Disposition</u>
A	Reclaim Pyranol made with TTCB with high hydrolyzable chlorides	12/21/60	Accepted
B	Circulate chlorinator material during entire chlorination period	10/22/57	Accepted
C-Rev.#1	Use of LVM Attasorb	5/15/59	Rejected
D	Optimum Lime Concentration	12/31/58	Rejected
E	Use of Undried Earth	1959	Rejected
F	Recycle Boildown to Still	5/15/59	Rejected
G	Number of Runs before Boildown	6/16/59	Rejected
H	Earth Treat (0.2%) Aroclor 1242 at 70-80°C. Dry earth at 100-300°C or use as received	9/22/61	Accepted in part
I	Earth Treat (0.1-0.4%) Aroclor 1242 at 50-100°C	11/15/61	Never Tried (Re-submit)
J	N ₂ Blanket Blend Tanks, Filtrate Receivers and Storage Tanks	8/29/63	Rejected due to large losses of N ₂
K-Rev.#1	Remove Catalyst Basket from Chlorinator to make Aroclor 5060	7/29/64	Accepted
L	Optimize Earth Treatment Time	8/18/64	Accepted
M	Eliminate Earth Drying	1/20/65	Rejected
N	Increase Mesh Size of Earth	2/5/65	Rejected
O	Start-Up of No. 9 Solid Aroclor Chlorinator	5/3/65	Accepted

DSW 303356

DSW 303357

STANDARD MANUFACTURING PROCESS
FOR
AROCLOL

I. SYNOPSIS OF PROCESS

Aroclors are the reaction products formed by chlorination of Biphenyl and/or Terphenyls using Ferric Chloride Catalyst. The by-product HCl gas produced is absorbed in water to produce Muriatic Acid. Crude Aroclors are produced in the chlorinators, air blown, distilled, earth treated, filtered, and loaded for shipment.

Aroclor is a trade name used to designate chlorination products of Biphenyl, Terphenyls, or mixtures of the two. The physical properties vary from light free-flowing liquids to crystalline and non-crystalline solids. Depending on the degree of chlorination and the composition of the raw materials, Santowax R, and mixtures of Biphenyl and Santowax R, or Biphenyl and Santowax C, yield products of increasing viscosity up to non-crystalline resins.

A numerical system of nomenclature, consisting of four digits, has been developed for identifying the Aroclors. The thousands digit denotes the raw material used in the chlorination. The figure 1--- indicates 100% technical grade Biphenyl. The hundreds digit denotes whether or not the product is distilled. The -1-- indicates undistilled, and the figure -2-- indicates distilled. The tens and units digit denote the approximate percentage of chlorine contained in the finished product. For example, 1260 is a distilled chlorinated Biphenyl containing 60% chlorine.

There are deviations from the above nomenclature when Solid Aroclors are produced. These are listed below:

1. Aroclors 4065 and 4465

Aroclor 4065 is the crude and 4465 is obtained by distillation. The starting raw material contains 40% Santowax R and 60% technical Biphenyl. This is chlorinated to 65%.

2. Aroclors 5060 and 5460

Aroclor 5060 is the crude from which Aroclor 5460 is distilled. The starting raw material is 100% Santowax R which is chlorinated to 60%.

DSW 303358

3. Aroclor 5042 and 5442

Aroclor 5042 is the crude from which Aroclor 5442 is distilled. The starting raw material is 100% Santowax R which is chlorinated to 42%.

4. Aroclor 2565

This is an undistilled product. The starting raw material consists of 25% Santowax C and 75% technical Biphenyl. This is chlorinated to 65%.

Chlorinator Area

There are four batch operated liquid chlorinators, one batch operated solid chlorinator, and one continuous operated chlorinator system. The continuous system consists of four batch chlorinators cascaded with overflow from one unit to the next. The product from the last stage is the finished crude Aroclor.

Batch operated liquid chlorinators are charged with 3,000 pounds of Biphenyl with 1-1/2 pounds of Ferric Chloride Catalyst. Charges for the solid chlorinator vary dependent upon the Aroclor to be produced. Chlorine gas added through a distributor plate reacts with the raw material to release HCl and heat. The HCl gas goes to the Muriatic Acid Department. Heat is taken away by use of external air cooled heat exchangers. The specific gravity of the crude is checked after chlorine feed has been on for four hours, again when 80% chlorinated, as indicated by alarm, and as needed thereafter.

In the continuous chlorination system, Biphenyl is fed to the first stage and chlorinated. Crude Aroclor overflows to the second stage, etc., until the final product overflows from the last stage. Each chlorinator is held at a different level of chlorination with the last stage being the final crude Aroclor desired. Normally, Aroclor 1142 is produced in this system.

Crude Aroclors from the batch chlorinators and the continuous system are transferred to a blowing tank to be air blown for one (1) hour minimum before distillation.

DSW 303359

Still Area

Liquid Aroclors

Aerated Aroclors from a blowing tank or a feed tank are charged to a still, hydrated lime is added, a vacuum of 720 mm Hg (40 mm Hg absolute) is applied, the still furnace is fired, and the Aroclor is distilled into a receiver. The distillation is semi-continuous, since ten (10) receivers are distilled prior to drawing bottoms of residue from the still. Every two receivers (or approximately 2,000 gallons) are treated with 40 lbs. of earth for two hours and then filtered. The finished product is transferred to storage tanks prior to loading into drums or tank cars.

Solid Aroclors

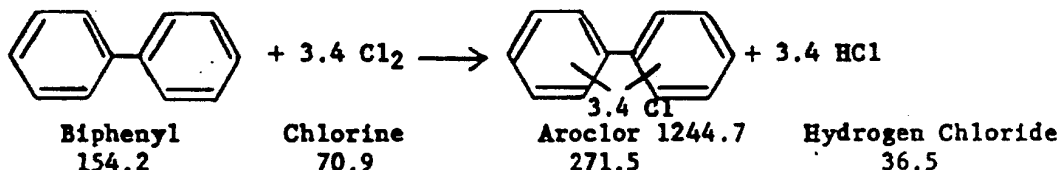
Solid Aroclors are distilled in a similar manner to Liquid Aroclors except for 1168. Aroclor 1168 is not air blown and is charged directly to the still from the chlorinator. A vacuum of 4 mm Hg absolute is applied for distillation. Solid Aroclors have one to four receivers distilled prior to drawing bottoms in the still. This is dependent upon the particular Aroclor being produced.

Solid Aroclors are not earth treated or filtered. They are drummed directly from the receiver or are flaked (Aroclors 1268 and 5460).

Chemical Equations

Liquid Aroclors

Based on the forecasted demands through 1969, the average Liquid Aroclor could be called 1244.7.



<u>Yields per CWT Aroclor</u>	<u>Theory</u>	<u>Standard</u>
Chlorine	89.4	96.1
Biphenyl	56.5	58.2
Hydrogen Chloride	45.7	38.8

DSW 303360

Aroclor Manufacturing Process
Synopsis of Process

4.

Solid Aroclors

Based on the forecasted demands through 1969, the average Solid Aroclor would contain 60.7% chlorine and 11.9% Biphenyl.

0.881 Santowax R	+ 0.119 Biphenyl	+ 10.3 Cl ₂	_____	Solid Aroclor	+ 10.3 HCl
230.3	154.2	70.9		586.0	36.5

Yields per CWT Aroclor

Theory

Standard

Chlorine	121.4	146.3
Biphenyl	5.01	--
Santowax R	36.2	42.3
Hydrogen Chloride	64.1	54.5

DSW 303361

FROM PUBLIC WORKS

DSW 303362

STANDARD MANUFACTURING PROCESS
FOR
AROCLOP

II. FLOW SHEETS

DSW 303363

DSW 303364

HCL HEADER TO MURIATIC ACID ABSORBERS

BIPHENYL

FLOW CONTROL

BIPHENYL MIX TANK

BIPHENYL FEED TANK

FEED PUMP

CHLORINATORS

CHLORINATOR CIRCULATING PUMPS

SEAL LEG

#8

#7

#6

#5

CONTINUOUS AIR BLOW TANK

AIR FROM BLOWERS

HCL TO S.M. DROWNING JET

NO. 1000 BLOWERS

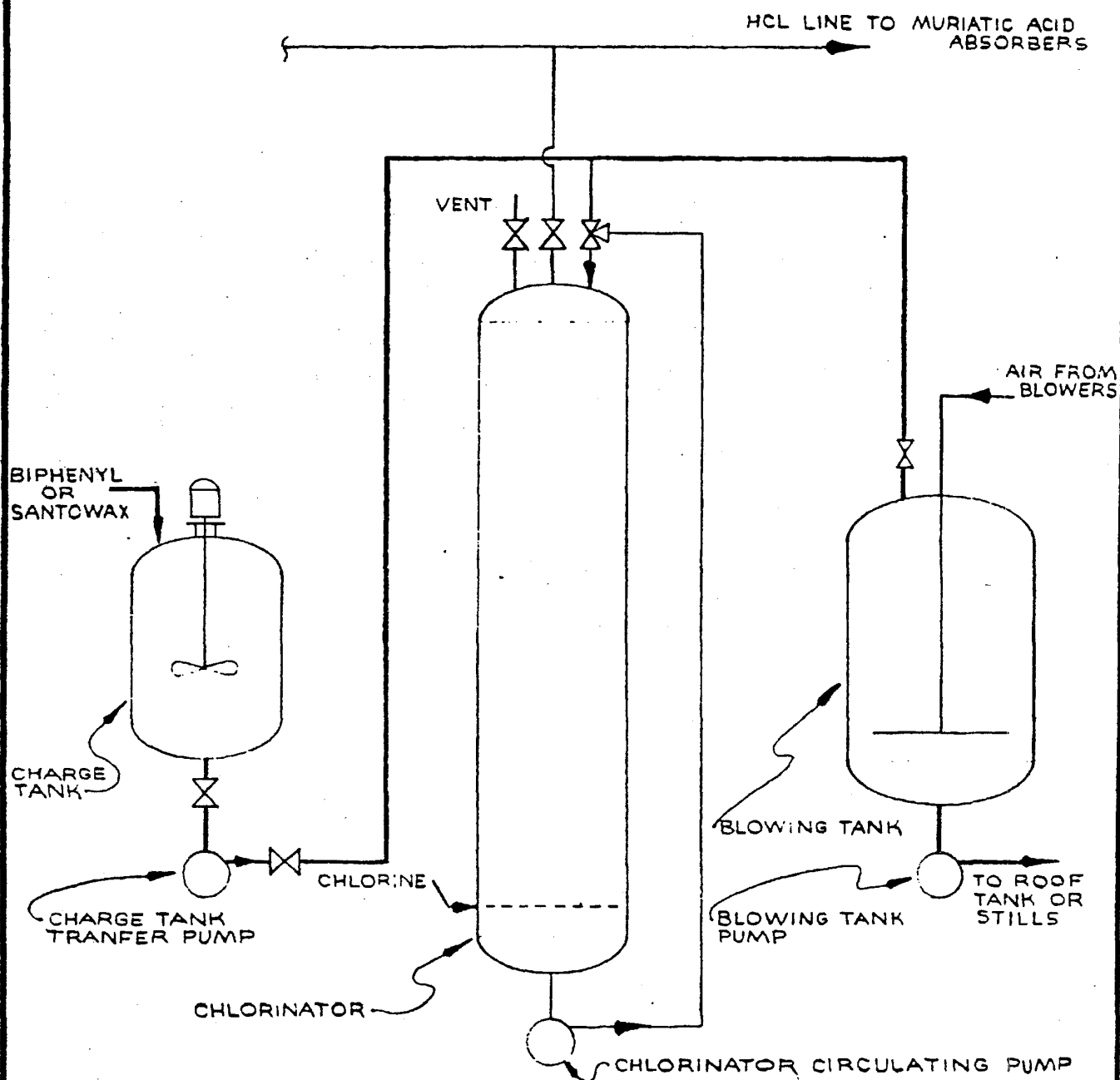
CRUDE ARDCLOR TRANSFER PUMP

TO REFORM STILL

BLOWING TANK PUMP

ARISTON PLANT - MONSANTO CHEMICAL COMPANY - ARISTON, ALA.

TITLE	CONTINUOUS CHLORINATOR - FLOW SHEET			
	DRAWN BY EAE	APPROVED	DATE 8-11-64	BOLE NONE
DESIGN - B-SL				
ARDCLOR				



--- BATCH CHLORINATOR OPERATION ---
 --- FLOW SHEET ---

DSW 303365

ANNISTON PLANT - MONSANTO CHEMICAL COMPANY - ANNISTON, ALA.

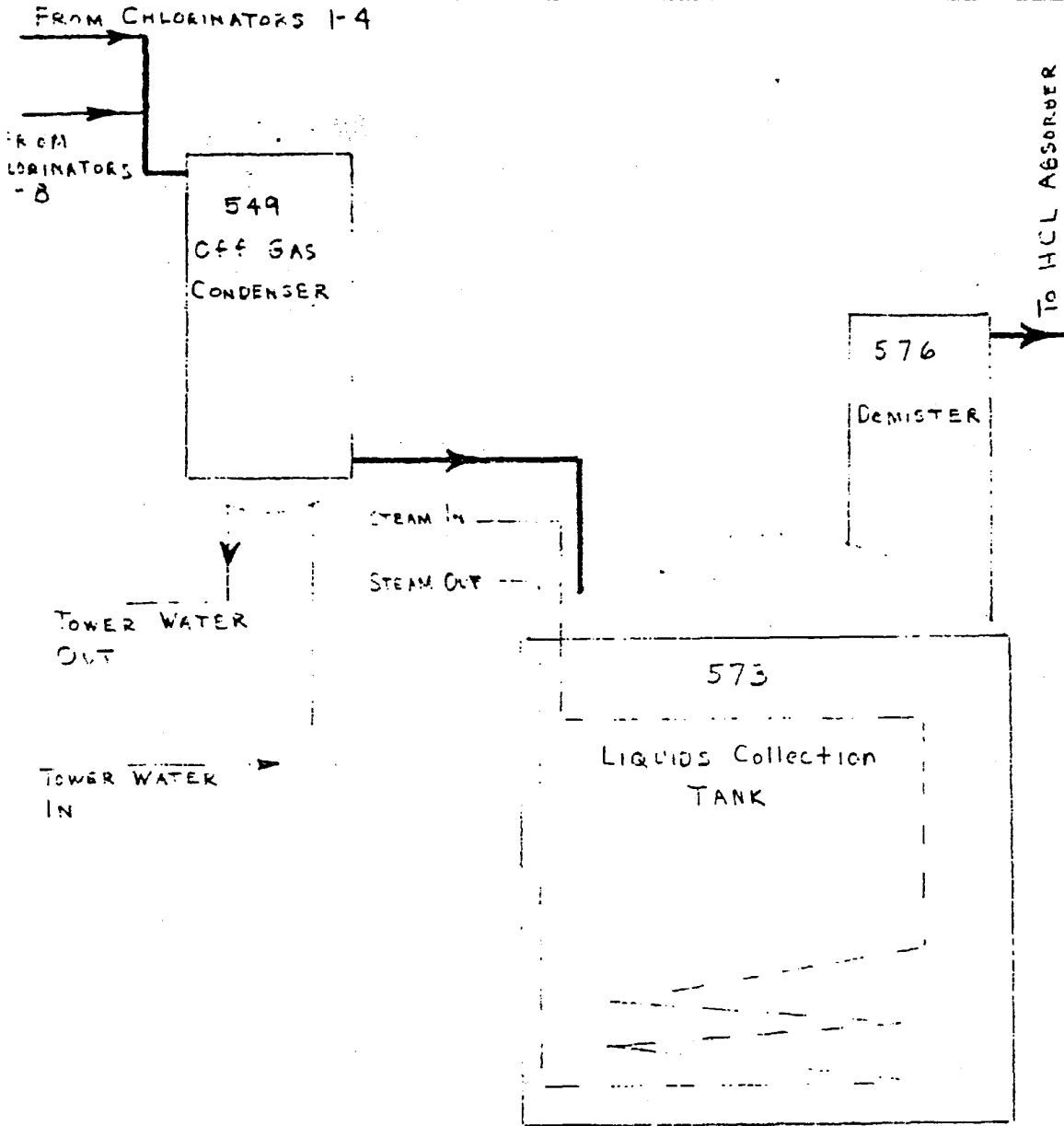
TITLE BATCH CHLORINATOR OPERATION - FLOW SHEET					ALPH. NO.
					JOB NO.
DRAWN BY EAE	APPROVED	DATE 8-7-64	SCALE NONE	DEPT. AROCOR	TS-A-701

SANTO CHEMICAL COMPANY
 NIC CHEMICALS DIVISION
 ENGINEERING DEPARTMENT
 DUIS MISSOURI

LIQUID HCL OFF GAS
 CLEAN UP
 PLANT ANNISTON
 LOCATION ANNISTON, ALA

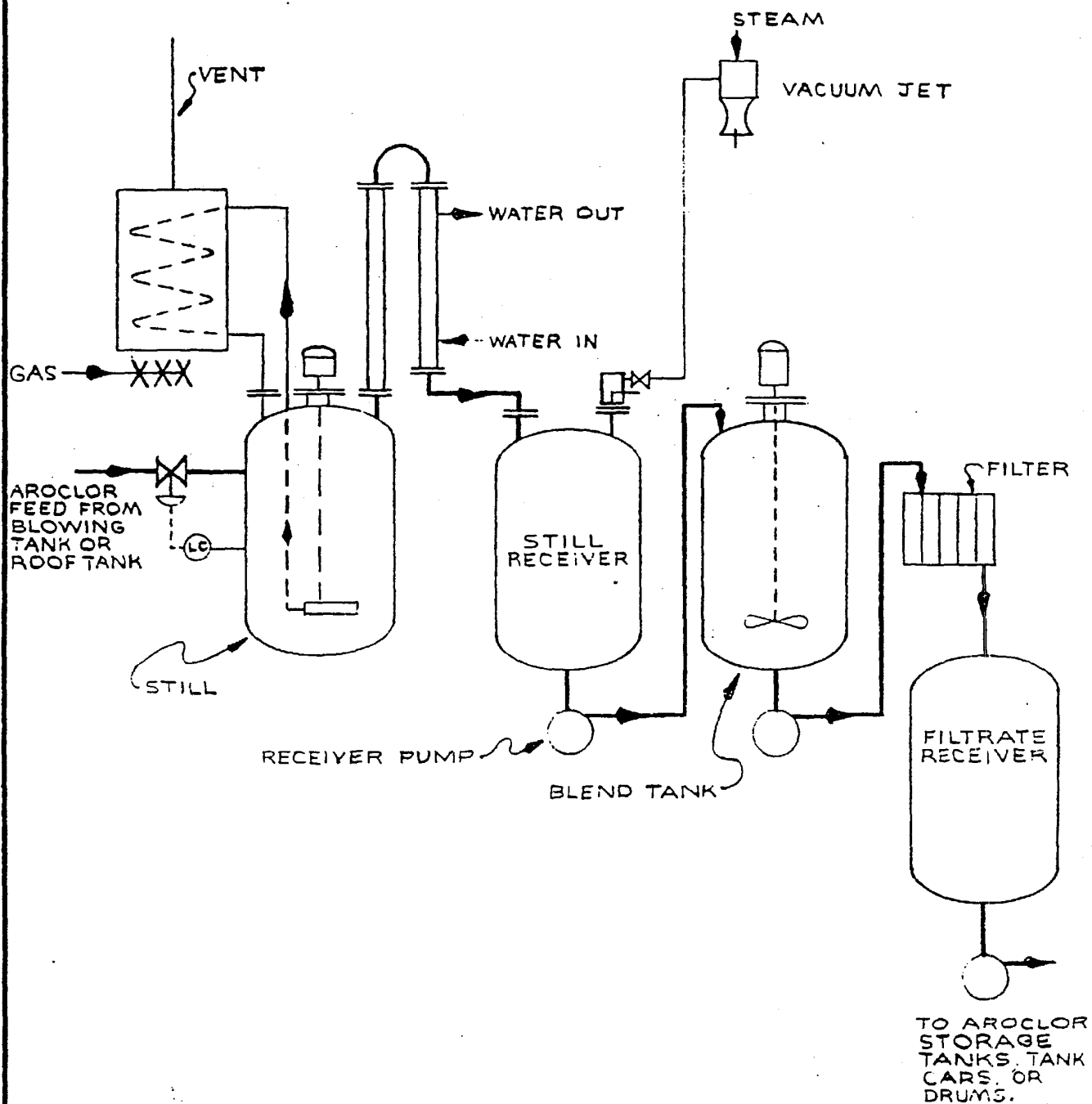
DESIGN MANUAL
 SECTION NO.
 ESTIMATE NO.
 PREPARED BY
 SHEET OF

ITEM NO.
 E. A. NO.
 DATE



DSW 303366

				APPROVED BY					
				DATE					
V. NO.	DATE	REVISION	BY	DATE					

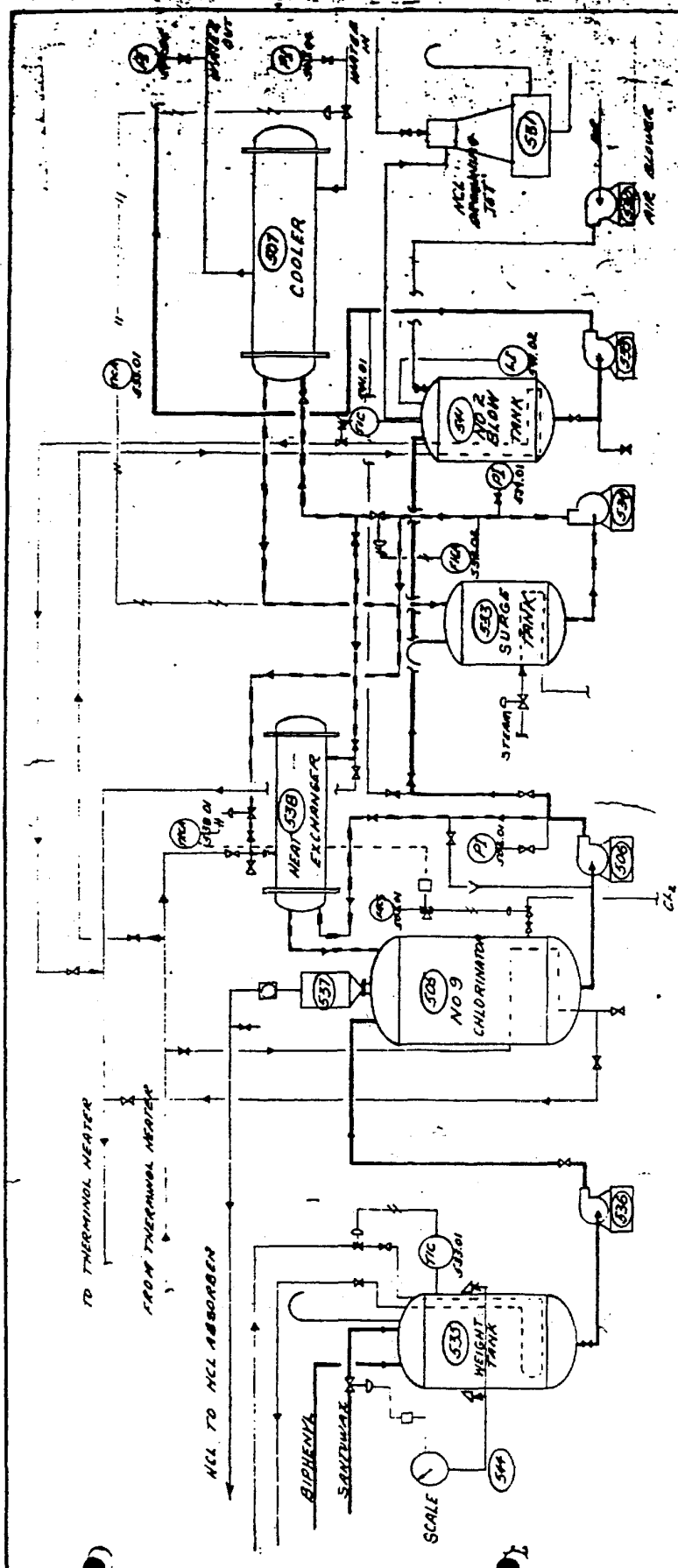


AROCLOR STILL - FLOW SHEET

DSW 303367

ANNISTON PLANT - MONSANTO CHEMICAL COMPANY - ANNISTON, ALA.

TITLE AROCLOR STILL - FLOW SHEET					AUTH. NO.	
					JOB NO.	
DRAWN BY EAE	APPROVED	DATE 8-10-64	SCALE NONE	DEPT. AROCLOR	TS-A-704	



- MAIN ABSORBER ROOM
 - INDUCTANCE COILS
 - CLASS 8000

DSW 303369

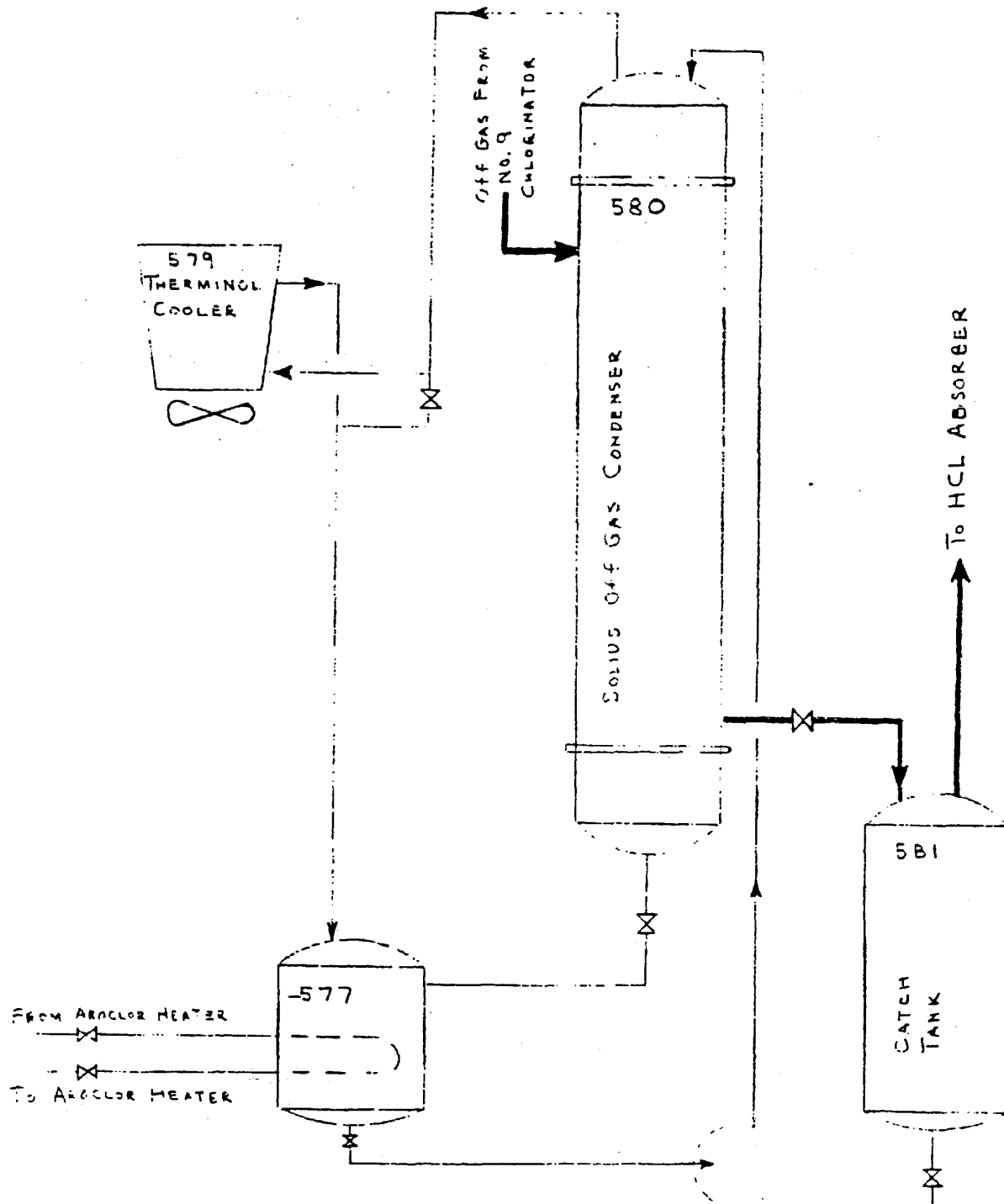
ANNISTON PLANT - MONSANTO CHEMICAL COMPANY - ANNISTON, ALA.			
TITLE	NO 9 AROCLOR CHLORINATOR FLOWSHEET	DATE	TS-B 804-1
DESIGNED BY	APPROVED	DATE	
DRAWN BY		DATE	

ONSANTO CHEMICAL COMPANY
 ORGANIC CHEMICALS DIVISION
 ENGINEERING DEPARTMENT
 ST. LOUIS, MISSOURI

SOLID HCL OFF GAS
 CLEAN UP
 PLANT ANNISTON
 LOCATION ANNISTON, ALA.

DESIGN MANUAL
 SECTION NO.
 ESTIMATE NO.
 PREPARED BY
 SHEET OF

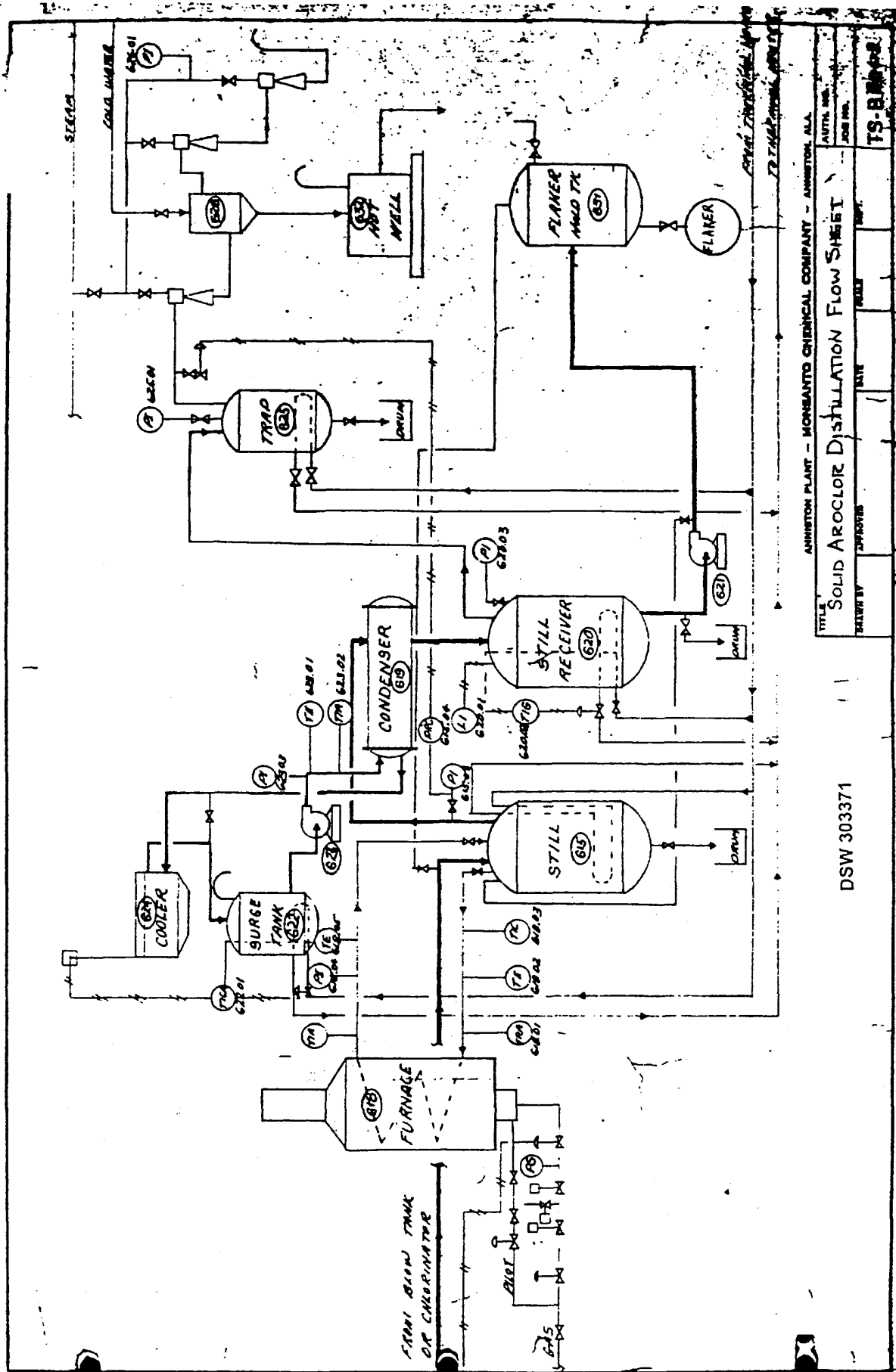
ITEM NO.
 E. A. NO.
 DATE



DSW 303370

				APPROVED BY					
				DATE					
REV. NO.	DATE	REVISION	BY						

DE-43 11/73



PROCESS
IN DETAIL

DSW 303372

WATER_PCB-SD0000050923

STANDARD MANUFACTURING PROCESS
FOR
AROCLOL

III. PROCESS IN DETAIL

A. Batch Liquid Chlorinators

Crude Liquid Aroclors are produced by batch chlorination in a vertical vessel. The reaction is cooled by external air cooled heat exchangers and agitated by a circulating pump. Biphenyl is chlorinated with chlorine gas added at the bottom of the vessel. HCl gas vented at the top of the chlorinator is a by-product which is recovered in the Muriatic Acid Department. Crude Aroclor produced in batch chlorinators is transferred to the blowing operation when the chlorination is completed. Chlorine gas is fed to the chlorinators from the Chlorine Plant and/or from tank cars.

A chlorinator charge (3,000 lbs.) of specification grade Biphenyl is pumped to the chlorinator charge tank from a Biphenyl storage tank. One and one-half (1-1/2) pounds of Ferric Chloride is slurried in Aroclor and dumped into the charge tank with each chlorinator charge. The Biphenyl and Ferric Chloride is agitated for a minimum of 15 minutes and then transferred to the chlorinator.

Aroclor 1142 can be used to charge a batch chlorinator for the production of 1148, 1154, 1160, or 1162. When 1142 is used for a chlorinator charge no Ferric Chloride is added.

After the chlorinator has been charged the circulating pump is started and then the chlorine feed is put on the chlorinator. As the chlorination progresses the specific gravity is checked at intervals until the desired gravity has been attained. When a chlorinator is finished it is transferred to a blowing tank.

B. Continuous Chlorinator Operation

The continuous chlorinator operation consists of four cascade flow chlorinators in series where specific levels of chlorination occur. Biphenyl, with Ferric Chloride added, is fed continuously into the first chlorinator where the material is chlorinated and overflows to the second chlorinator. The finished product comes off the final unit, flows into a catch tank and is pumped to a continuous blow tank. The material is pumped from this blow tank

DSW 303373

to another blow tank for additional blowing. Chlorine is fed to each chlorinator continuously. This chlorination system can be reduced to three (3) chlorinators in series in case problems occur with one unit. Normally 1142 is the only Aroclor produced on the continuous system, however, other Liquid Aroclors can also be produced with the chlorinators set for different levels of chlorination.

C. Biphenyl Mix Tank and Feed Tank

When Biphenyl is needed for the continuous chlorinator operation a charge is pumped from a Biphenyl storage tank to the mix tank. Three and one-half (3-1/2) pounds of Ferric Chloride is added to the tank. The agitator in the tank is started and the Biphenyl and Ferric Chloride is agitated until a charge is needed for the Biphenyl feed tank. The feed tank is located under the mix tank. The feed tank low level alarm sounds the annunciator system at 25% full, at which time the mix tank batch is dropped into the feed tank. This will allow a minimum of one hour agitation in the mix tank. The agitator in the feed tank runs continuously while Biphenyl is being fed to the first stage of the continuous chlorinator operation.

D. Batch Solid Chlorinator

Crude Solid Aroclors are produced by batch chlorination when a weighed quantity of Biphenyl, Santowax, or a mixture of the two is chlorinated in a vertical reaction vessel with Ferric Chloride as a catalyst. Chlorine gas is added at the bottom of the vessel.

By-product HCl gas vented at the top of the chlorinator is recovered in the Muriatic Acid Department. Crude Solid Aroclors produced in batch chlorination are transferred to the blowing tank operation when the chlorination is completed. (The only exception being 1168 which is not air blown but pumped directly from the chlorinator to the still). Chlorine gas is fed to the chlorinator from the Chlorine Plant and/or from tank cars.

E. Blowing Tank Operation

Crude Aroclors produced in the chlorinators are air blown for a minimum of one hour, for removal of residual HCl, before the material is fed to the stills. Steam is maintained on the coils

DSW 303374

of the Liquid blowing tanks during this operation. Therminol heat is maintained on the coil in the Solid blowing tank. Residual HCl blown out of the crude Aroclor is drowned in a water ejector which pulls a slight vacuum on the blowing tanks.

F. Liquid Stills

Aerated crude Liquid Aroclor from a blowing tank, or a roof tank containing aerated Aroclor, is fed to the still tank. Hydrated lime is added to the crude Aroclor.

Vacuum is applied to the still and the Aroclor is circulated through the furnace coil and heated to the vaporization temperature. Vapors pass up an open column to the condenser. Distilled Aroclor is collected in a receiver and then transferred into a blending tank.

Distilled Liquid Aroclors are treated with Attapulugus Earth in the blending tank, filtered, analyzed, and packaged or held in storage.

Still residue or bottoms are sold as Montars or discarded.

G. Earth Treatment and Filtration

All Liquid Aroclors are treated in the blending tank with Attapulugus Earth. This earth is dried in an oven for a minimum of four hours at a temperature of 210-250°C. The material, with the earth added, is agitated and then pumped through the filter press and back to the blend tank. When no earth is visible in the material it is then directed to a filtrate receiver. The filtered Aroclor is then ready for drumming or transfer to a storage tank.

H. Solid Still

Aerated Crude Solid Aroclor from a blowing tank is pumped to the No. 4 still pot. Hydrated lime is added to the crude Aroclor.

Vacuum is applied to the still and the Aroclor is circulated through the furnace coil and heated to the vaporization temperature.

Vapors pass up on open column to the condenser. Distilled Aroclor is collected in a receiver. Quality of the distilled Solid Aroclors is analyzed by the laboratory, and if in specification, it is drawn directly from the receiver into suitable containers or transferred to the flaker operation.

Still residue or bottoms are sold as Montars or discarded.

I. Packaging

Solid Aroclors are packaged in fibre drums, galvanized drums, galvanized cans and in paper bags.

Liquid Aroclors are packaged in steel drums and cans. These drums and cans are lacquer lined. Liquid Aroclors are also loaded into tank cars and trucks.

DSW 303376

COMMENTS
ON PROCESS

DSW 303377

STANDARD MANUFACTURING PROCESS
FOR
AROCLORS

IV. COMMENTS ON THE PROCESS

A. LIQUID AROCLORS

1. Chlorination

a. Temperature

Chlorination temperatures of 120°C or lower yield a product which will not pass the stability tests on electrical grade Aroclors. A minimum of 130°C has been set for the chlorinators. An upper temperature limit is not defined as critical to high quality Aroclor production. A practical limitation is governed by the increase of overhead product loss to the off-gas collection tank. The normal temperature range is between 140°C and 170°C.

b. Catalyst

The basket containing a bed of iron turnings in the chlorinator provides contact surface for the Chlorine and Biphenyl, and the iron reacts with the Chlorine forming Ferric Chloride which is the actual catalyst. Consumption of the turnings makes it necessary to inspect the chlorinator semi-annually and replenish the turnings if necessary. Ferric Chloride (anhydrous powder) in the amount of 0.05% is added to the chlorinator charge to provide assurance of adequate catalyst at all times during the chlorination. It is imperative that the iron turnings in each chlorinator be inspected semi-annually. Aroclor supplied to General Electric and Westinghouse must be produced in chlorinators containing the basket. (Prior to any shipment of Electrical Grade Aroclor being shipped to General Electric and Westinghouse from a process where the baskets are removed, Marketing will review the process change with both customers.)

The catalyst is added to the chlorinator charge tank or mix tank (continuous chlorinators) and agitated for at least 15 minutes but usually at least an hour before charging or feeding the chlorinators. EOL and WGL Aroclor Departments do not pre-mix the catalyst. Research has indicated the tendency to producing unstable Aroclors can be attributed to improper catalyst concentration which may be brought about by insufficient mixing prior to chlorination.

DSW 303378

Free Chlorine in the off-gas from an individual chlorinator or from the vent at the HCl absorber sewer box indicates problem operation in the Aroclor Department. It can be caused by inadequate reaction of the Biphenyl and might indicate formation of unstable Aroclor. If this situation occurs it should be corrected immediately. Take samples from the individual chlorinator off-gas and have them analyzed for free chlorine by the laboratory. Check grab samples from each chlorinator of the crude Aroclor being produced and observe it for unusual appearances. Inadequate amounts of catalyst present as Ferric Chloride or iron turnings can cause free chlorine in the off-gas line. In case this situation occurs and cannot be resolved, remove the chlorinator from operation. Contact department supervision immediately. A badly worn chlorinator distributor can also cause free chlorine in the off-gas line. *ee*

c. Chlorine

Although the effect of air in chlorine on Aroclor quality is unknown, the amount of air in chlorine does effect the chlorination rate and therefore effects the production rate.

Problems with the chlorine vaporizer can occur which would allow liquid chlorine to be fed to the chlorinators. In case this should happen, slugs of liquid chlorine would be fed into the chlorinator possibly causing excessive reaction or excessive loss and venting of chlorine to the atmosphere. Chlorine in the HCl off-gas pipe lines could cause extreme reactions which would generate excessive heat. This reaction can cause a pipe line to become red hot and melt the pipe line in two. This would result in chlorine gas escaping to the atmosphere. Every effort possible should be made to operate the vaporizer with the control system provided.

Entrainment of Biphenyl or crude Aroclor into the HCl lines can occur from excessive chlorine rates. This same situation can occur in case the Chlorine Plant is blowing too much air. This can be checked by observing the off-gas from the chlorinators in the sight glasses on top of the Brink Demister on each chlorinator. Normal operation of the chlorinator should be to minimize this carry-over. Abnormally high or low chlorine header pressures can cause operating problems. High chlorine header pressure causes problems within the Chlorine Plant to

DSW 303379

the department operating personnel. Chlorine fumes are forced back into the Chlorine Department and at times sulfuric acid can be blown from the Nash compressor system. Since this presents a safety hazard to the department personnel every effort should be made to eliminate high pressure problems. Low chlorine pressure can cause operating problems within the Aroclor Department. In cases of extreme low pressure, the crude Aroclor could back up in the chlorine line and begin blow-back into the chlorine header.

d. Chlorinator Off-Gas Collection

The demister on top of the liquids collection tank is used as a final knock-out for Aroclor and Biphenyl entrained droplets in the HCl gas system. Usually this system will be operating with the steam tracing around the outer shell shut off. When the steam tracing is turned on it will allow vapors to be carried through the demister to the Muriatic Acid absorber. The sight glass on the discharge of the demister can be checked to determine if entrained droplets are being carried over. The steam is installed for use to unplug the demister in case this should occur. The trappings from this clean-up system are used to produce Aroclor 1148 and 1162 only. The reason for this is to be sure that we do not produce any unstable electrical grade Aroclor.

It is extremely important that the tempered water system on the liquid HCl off-gas condenser be drained during excessive cold weather if it is shut down. The bottom head on the condenser has drain valves at two locations. Both of these drain valves must be open to remove water from the off-gas condenser when the system is shut down.

2. Air Blowing of Crude Aroclors

Air blowing of crude Aroclors takes place with compressed air from air blowers. The air is not dried by any means other than that caused by compression and cooling in the air blowers. Judging by the corrosion exhibited around nozzles atop the blowing tanks a good argument can be made for thoroughly dried air. Wet air may also be the cause of poor electrical properties that occasionally occur during period of rain and high humidity. Wet air can also carry in chlorides, SO₂, and H₂S from the atmosphere. An air dryer would tend to scrub out these undesirable gases.

DSW 303380

3. Distillation

Vacuum trouble on the stills is usually caused by flashing of product, plugged vacuum lines, or improper functioning of the vacuum jets. Flashing can usually be stopped by operating the still at a slower rate or increasing the flow of cooling water to the vapor condenser. Vacuum trouble can be traced systematically from the jets and usually isolated.

4. Earth Treatment

Exclusion of atmospheric moisture is naturally important where high electrical resistivity is required and of course the Attapulugus earth needs to be well dried. All liquid Aroclors are earth treated at the Anniston Plant to decrease the risk of contaminating electrical grade with non-electrical grade Aroclors. There are too many common lines downstream of the stills to chance non-treatment of non-electrical grade Aroclors.

5. Quality

Incomplete aeration in the blow tanks, or the addition of insufficient lime to the still pot, will cause high acidities. High inorganic chlorides either indicate contamination or the presence of attapulugus earth in the product caused by a poor filtration. Batches with high color usually indicate excessive gas heat on the heating coils or air leaks in the vacuum system. A distillation carried too far will also show a dark distillate from the distillation of colored material from the residues. Another reason for color can be the material formed when chlorinating at low temperatures or with insufficient catalyst. Batches with specific gravities, softening points and viscosities out of specifications can indicate over or under chlorinations, or incomplete draining of the process equipment when switching from one type of Aroclor to another. Iron has a deleterious effect on the electrical properties of the distilled Aroclors; therefore, tankage, process piping and pumps from the still receiver through the remainder of the process are either aluminum, tin and zinc sprayed or are constructed of monel, stainless steel or bronze.

6. Special Grades

a. Export Aroclor 1242

The export specs on 1242 call for a resistivity of 5,000 min. and a Biphenyl content of 0.05% max. Domestic specs call for 500 min. resistivity and no spec on Biphenyl content. The

DSW 303381

Biphenyl content of 1242 made on the continuous chlorinators averages below the 0.05% max. spec and therefore causes no production problems. To meet the 5,000 min. resistivity spec the 1242 is earth treated two to five times using normal procedures until the resistivity exceeds 7,000 (dept. spec) and then drummed out from the filtrate receiver.

b. Export Aroclor 1254

The export specs on 1254 call for a resistivity of 5,000 min. and a Biphenyl content of 0.025% max. Domestic specs call for 500 min. resistivity and no spec on Biphenyl content. To meet the 5,000 min. resistivity spec the 1254 is earth treated two to five times using normal procedures until the resistivity exceed 7,000 (dept. spec) and then drummed out from the filtrate receiver. Two methods have been tried to assure low Biphenyl content.

- (1) Batches three through eight in the normal ten batch distillation run have a greater tendency to meet the Biphenyl spec than the remaining four. Each batch is analyzed for Biphenyl and then earth treated when in spec. This method is hit and miss and has about a 50% reliability of success and causes still downtime.
- (2) Aroclor 1154 is chlorinated from low Biphenyl 1142 rather than Biphenyl. Chances of success are so great that all ten still batches have been earth treated and placed in an empty storage tank and then additional earth treatment is scheduled to avoid generating still downtime.

The second method is used whenever possible and has resulted in better inventory control of export 1254.

7. Safety

Biphenyl is flammable and burning should never be done in areas where Biphenyl or Biphenyl vapors are present.

Light Aroclors are also flammable. The viscous Aroclor oils and the resins do not support combustion when heated alone, even at their boiling points - temperatures above 350°C.

8. Dermatology and Toxicology

Aroclors do not produce primary irritancy or sensitization of the skin. At ordinary temperatures Aroclors have not presented industrial toxicological problems. If Aroclors are used at elevated temperatures such as 200° or 300°C in open systems, methods must be designed to exhaust any vapors arising from these open systems.

B. SOLID AROCLORS

1. Chlorination

Special flexible piping joints have been made to the weigh tank since this tank is placed on scales. Actuation of the scale system will require movement of this tank approximately 1/4 inch. These flexible piping joints should be carefully inspected to be sure there are no leaks since the material contained in them is extremely hot. Steam is provided on the transfer lines and the coil to the weigh tank. From the bottom piping on the weigh tank, all the transfer lines and coils in the Solid Aroclor process vessels have Therminol heat until the end of the process at the flaker hold tank.

The No. 9, Solid Aroclor, chlorinator does not have a basket of iron turnings. Due to this, it is absolutely necessary that five pounds of Ferric Chloride be added to each chlorinator charge in the weigh tank.

Plugging of the off-gas condenser in the Solids Clean-Up System can occur. To unplug the off-gas condenser the temperature of the Therminol system is elevated as required. During this time, the off-gas condenser can be by-passed.

The Solids knock-out tank is provided with a drain system to empty the tank completely when products are to be changed on the No. 9 chlorinator. This will eliminate contamination in the weigh tank.

2. Air Blowing of Crude Aroclor

The same comments on corrosion for Liquid Aroclors hold true for the Solid Aroclor blow tanks.

Undistilled Aroclors 2565 and 5060 are lime treated in the blow tank to give a neutral acid number. Flaked 5060 on the acid side will cause the bags to fail.

3. Distillation

Vacuum problems on the Aroclor stills are usually due to flashing of product over, plugged vacuum lines, leaks in the still system, leaks at pump packing, water adjustment on inter-condenser on jets, changes in steam pressure to the jets, or problems in the jet nozzles. Flashing can be stopped by adjusting the coolant flow on the condenser or operation of the still at a slower rate. Vacuum trouble must be traced systematically from the jet, to the receiver, and then to the still.

It is important to be sure the coil in the Therminol cooling system surge tank is covered when the system is in operation. This assures the circulating pump has a positive suction and prevents surging in the heater coil. This will allow more uniform control of the temperature in the Therminol cooling system.

Care must be taken not to fill the surge tank too full during normal operation. This can cause the tank to run over when the circulating pump is stopped. A normal level will be about 6" to 12" in the sight glass.

The flow measurement recording instrument for the still heater system is an extremely important portion of the controls for the No. 4 still system. A segmental orifice is used for obtaining differential pressure readings to indicate flow. This orifice system is in a specially designed Therminol jacketed pipe section. A specially designed d/p cell records the pressure readings and indicates flow. The orifice is designed to allow sucking the coil completely empty when a distillation is complete. The hole in the orifice plate is flush with the bottom of the pipe line. This flow recorder indicates low flow rates at the end of a distillation which aids the operator in completing a normal operation.

Lime addition to the Solid Aroclor still is very important. Cutting the amount of lime below 200 lbs. per batch results in occasional off-color batches.

The longer still bottoms are circulated in the heater the higher the formation of HCl and the greater the tendency to plug the heater coil. Every effort should be made to have Crude Aroclor available to charge the still at all times.

4. Solid Aroclor 1268

The bottleneck in 1268 production is the chlorinator. Increasing chlorine feed rates only increases Biphenyl and low chlorinated Aroclor carry-over to the off-gas collection system, and increases the tendency to over-chlorination and its ensuing problems with freeze-up of lines and equipment (hold points greater than 175°C).

The still has been found to run better when there is feed always available. Circulating of still bottoms while waiting on crude increases the breakdown of the Montar into more viscous material and hydrogen chloride. Draining the still between batches increases the tendency to plug the heater coil.

DSW 303384

Comments on the Process
Aroclors

8.

To move the bottleneck from the chlorinator to the still requires charging the chlorinator with 1142 and chlorinating at a rate high enough to insure finishing the batch before the still batch is completed. Other higher chlorinated crudes can be charged to the chlorinator but they will not be as readily available as 1142 from the continuous chlorinators.

DSW 303385

DSW 303386

WATER_PCB-SD0000050937

STANDARD MANUFACTURING PROCESS
FOR
AROCLOP

APPENDIX A - EQUIPMENT LIST

Area 200 - Bulk Storage

Area 500 - Chlorinator

Area 600 - Still

DSW 303387

APPENDIX A - EQUIPMENT LIST
AREA - 200

BULK STORAGE

<u>Item</u>	<u>Equipment</u>	<u>Equip. No.</u>	<u>Description</u>
201	No. 1 Aroclor Storage Tank	22-0077	10' x 26' metal sprayed with zinc and tin. Capacity 15,000 gallons.
202	No. 3 Aroclor Storage Tank	22-0089	Horizontal, cylindrical dia. 8'-0", length 30'-0". Metal sprayed with zinc and tin. Capacity 11,000 gals.
203	No. 2 Aroclor Storage Tank	22-0076-01	10' x 26' metal sprayed with zinc and tin. Capacity 15,000 gallons.
204	No. 4 Aroclor Storage Tank		10' x 26' metal sprayed with zinc and tin. Capacity 15,000 gallons.
205	South Pyranol Mix Tank	22-0108	14'-0" I.D. x 10'-0" high. Dwg. D-6455.
206	North Pyranol Mix Tank		6'-6" I.D. x 8'-2" jacketed steel tank.
208	No. 1 Storage Tank Pump		Taber vertical submerged pump. Type CL-4. Capacity 75 GPM.
209	No. 2 Storage Tank Pump		Taber vertical submerged pump. Type CL-4. Capacity 75 GPM.
210	No. 5 Aroclor Storage Tank	22-0663	12,000 gal. vertical aluminum tank. See 22-0662, 19-0397, 19-0398, 94-0179, 94-0180, 49-0777, 28-0532.
211	No. 6 Aroclor Storage Tank	22-0662	12,000 gal. vertical aluminum tank. See 22-0662, 19-0397, 19-0398, 94-0179, 94-0180, 49-0777, 28-0532.
212	No. 6 Storage Tank Pump	19-0397	20-A-30, 3 x 2 - 13/110 Mark 11 pump w/crane seal #9 outside QP-191, Base steel plate #53 7FL Falk Spacer coupling.

DSW 303388

APPENDIX A - EQUIPMENT LIST
AREA 200

2.

<u>Item</u>	<u>Equipment</u>	<u>Equip. No.</u>	<u>Description</u>
213	No. 5 Storage Tank Pump	19-0398	20-A-30, 3 x 2 - 13/110 Mark 11 Pump w/crane seal #9 outside QP-191 Base steel plate #53 7FL Falk Spacer coupling.
214	No. 3 Storage Tank Pump		Taber vertical submerged pump. Type CL-4. Capacity 75 GPM.
215	No. 4 Storage Tank Pump		Taber vertical submerged pump. Type CL-4. Capacity 75 GPM.

DSW 303389

APPENDIX A - EQUIPMENT LIST
AREA 500

3.

CHLORINATOR

<u>Item</u>	<u>Equipment</u>	<u>Equip. No.</u>	<u>Description</u>
505	No. 9 Chlorinator	22-0774	Vertical 4'-6" dia. x 20' straight side carbon steel vessel with dished heads (3" Sch. 80 coil 75' long). Capacity 2,390 gallons.
506	No. 9 Chlorinator Circulating Pump	19-0494	Self-priming, centrifugal pump (Pacific 2-1/2" SS type SVC, 10" impeller, 15 HP, 1750 RPM, 440 volt motor). Pump has a Crane No. 9 outside mechanical seal, Code QP-868. Capacity 250 GPM at 73' TDH.
507	Aroclor Cooler No. 9 Chlorinator		1'-8" dia. x 18'-6" long, carbon steel shell heat exchanger with 304-3/4" O.D. 304 stainless steel, 14 BWG x 16" long tubes. Four passes on Therminol side (tube side). Capacity 1,400,000 BTU/hour.
508	No. 4 Chlorinator Cooler	32-0260	An air-cooled 1" BWG 16.8' long steel tube with aluminum finned cooler for removing heat from crude Aroclor. 3' wide x 8' long x 6' high, equipped with an Annin pneumatic operator to adjust louvers (58 tubes). Fan has a 2 HP, 1800 RPM, 440 volt motor. Capacity 1,100,000 BTU/hour.
509	No. 3 Chlorinator Circulating Pump	19-0495	Self-priming, centrifugal pump (Pacific 1-1/2 L Type SVC, 1-7/8" impeller, 15 HP, 1800 RPM, 440 volt motor). Pump has a Crane No. 9 outside mechanical seal, code QP-868. Capacity 220 GPM at 81' TDH.
510	No. 3 Chlorinator Cooler	32-0261	Same as Item 508
511	No. 2 Chlorinator Circulating Pump	19-0496	Same as Item 509

DSW 303390

APPENDIX A - EQUIPMENT LIST

4.

AREA 500

<u>Item</u>	<u>Equipment</u>	<u>Equip. No.</u>	<u>Description</u>
512	No. 2 Chlorinator Cooler	32-0262	Same as Item 508.
513	No. 1 Chlorinator Circulating Pump	19-0497	Same as Item 509.
514	No. 1 Chlorinator Cooler	32-0263	Same as Item 508.
515	No. 3 Blow Tank	22-0596	7'-6" dia. x 7'-6" high Dwg. D-234-4
516	No. 3 Blow Tank Pump		Peerless Pump - Model 1-1/2 DSO.8, Serial No. 121336.
517	No. 4 Blow Tank	22-0595	Same as Item 515.
518	No. 4 Blow Tank Pump		Peerless Pump - Model 1-1/2 DSO.8.
520	Biphenyl Feed Tank	22-0788	5'-6" dia. x 5'-8" long, carbon steel tank with dished heads. Tank contains a 2" Sch. 80 coil approximately 90' long. Capacity 1,000 gallons.
520-A	Biphenyl Feed Tank B & C Agitator & Drive		6-blade, turbine type, carbon steel agitator with a 4 ELHZ Faulk motor reducer driven by a 10 HP, 1800 RPM, 440 volt motor. The agitator has a 2-3/4" dia. x 8'-8" long shaft. Capacity provides medium agitation.
521	Biphenyl Feed Pump	19-0498	Self-priming, centrifugal pump (Goulds Model 3196 AVS-260, 6-5/8" open impeller, 5 HP, 1800 RPM, 440 volt motor). The pump has a Crane No. 9 outside mechanical seal, Code QP-868. Capacity 60 GPM at 42' TDH.
522	No. 5 Chlorinator Cooler	32-0264	Same as Item 508.
523	No. 6 Chlorinator Cooler	32-0265	Same as Item 508.

DSW 303391

APPENDIX A - EQUIPMENT LIST
AREA 500

5.

<u>Item</u>	<u>Equipment</u>	<u>Equip. No.</u>	<u>Description</u>
524	No. 7 Chlorinator Cooler	32-0266	Same as Item 508.
525	No. 8 Chlorinator Cooler	32-0267	Same as Item 508.
526	Crude Aroclor Catch Tank	22-0789	5'-6" x 5'-8" steel vertical tank with flanged and dished top head. Capacity 1,000 gals.
527	Crude Aroclor Catch Tank Pump	19-0526	100 GPM @ 75' head - vertical heavy duty centrifugal Taber pump - CV-3 #393.
528	Continuous Air Blow Tank	22-0682	500 gallon steel tank. Purchased from Dixie Steel & Supply Co. See MCC Dwg. D-548-3-0.
529	Continuous Blow Tank Transfer Pump	19-0499	Self-priming, vertical centrifugal pump (Taber vertical submerged, CL2-382, 7-1/2" open impeller, 3 HP, 1800 RPM, 440 volt motor). Capacity 15 GPM at 45' TDH.
530	Air Blower North	11-0338	Model 7 H - Spec. SM-488-530. Purchased from Sutorbilt Corp.
531	Solids Off-Gas Eductor		S&K Fig. 4010 Scrubber and S&K 4040 Separator Box Haveg & P.E. Const. 6" Suct. 6" Discharge.
532	Liquids Off-Gas Eductor		S&K Fig. 4010 Scrubber - 8" Suct. 8" Discharge.
533	Therminol Surge Tank #9 Chlor.	22-0775	4'-0" O.D. x 5'-3". Steel const. Purchased from Opelika Welding, Machine & Supply, Inc.
534	No. 9 Surge Tank Pump	19-0500	Spec. SM-488-534. P.O. No. OED-7734. Purchased from Goulds Pumps, Inc.
535	Weigh Tank		6'-6" x 7'-9" C. steel equipped with heating coil. 5 turns 2" Sch. 80 C.S. pipe. Capacity 2,000 gallons.
536	Weigh Tank Pump	19-0501	Goulds Model 3196 AVS No. A 60 constructed of Ductile Iron. Capacity 125 GPM.

DSW 303392

APPENDIX A - EQUIPMENT LIST
AREA 500

6.

<u>Item</u>	<u>Equipment</u>	<u>Equip. No.</u>	<u>Description</u>
537	No. 9 Chlorinator Mist Eliminator		Monsanto Brink type demister in a 30" O.D. x 4' x 6' long carbon steel tank. Bottom nozzle 10" dia., 10" flange, with a 6" gas outlet. Designed for 30 psig at 450°F. Contains a Brink element No. 2448. Capacity 325 cfm at 8 psig and 250°C. Inlet mist loading 2,000 mg/cu.ft. Outlet mist loading 20 mg/cu.ft. Efficiency 99%.
538	No. 9 Chlorinator Cooler	32-0259	Carbon steel heat exchanger 8' long with 3/4" O.D. x 15 BWG tubes containing 184 tubes. Four pass heat exchanger on tube side. Special design Marquette expansion joint constructed of 304 SS on shell side of this heat exchanger. Capacity 1,400,000 BTU/hour.
539	No. 2 Blow Tank Pump	19-0502	Self-priming, centrifugal pump (Goulds Model 3196 AVS-A60, 7" open impeller, with a 5 HP, 1800 RPM, 440 volt motor, and a Crane No. 9 outside mechanical seal, Code QP-868). Capacity 125 GPM at 29' TDH.
540	Biphenyl Feed Pump		Self-priming, centrifugal pump (Goulds Model 3196 AVS-A60, 6-1/16" open type impeller, 3 HP, 1800 RPM, 440 volt motor, and a Crane No. 9 outside mechanical seal. Code No. QP-868). Capacity 75 GPM at 39' TDH.
541	No. 2 Blow Tank	22-0777	7'-6" O.D. x 7'-6" straight side vertical tank, flat top with a dished bottom. Tank contains a 2" Sch. 80 pipe coil approximately 75' long. Capacity 2,500 gallons.
542	No. 1 Blow Tank	22-0266-02	Same as Item 515.
543	No. 1 Blow Tank Pump	19-0355	Vertical, centrifugal, submerged Taber pump. 2-1/2" inlet - 1-1/2" outlet. Cast iron construction. Capacity 50 GPM.

DSW 303393

WATER_PCB-SD0000050944

APPENDIX A - EQUIPMENT LIST
AREA 500

7.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
544	Weigh Tank Scale	24-0194	Fairbanks-Morse, Inc. overhead suspension scale, Model CA, coaxial cabinet dial, having a 22" dia. chart graduated 0-24,000 pounds with a main tare bar graduated 20,000 pounds and 5,000 pounds. The dial has one adjustable Reed switch to operate at scale set point. Scale to be hung below 11' platform level built to contain the weigh tank. Capacity 0-24,000 pounds x 20 pounds increments for scale measurement. Overall capacity 0-50,000 pounds.
545	Crude Aroclor Catch Tank Transfer Pump		Self-priming, vertical centrifugal pump (Taber CV-3, #393, with a 2" suction and 1-1/2" discharge, 10" dia. x 3-1/2" I impeller, 5 HP, 1800 RPM, 440 volt motor). Capacity 80 GPM at 100' TDH.
546	Brink Demister No. 7 Chlorinator	41-0037	1'-8" dia. x 4'-2" straight side carbon steel shell with a Brink type mesh demister pad.
547	Brink Demister No. 6 Chlorinator	41-0038	Same as Item 546.
548	Brink Demister No. 5 Chlorinator	41-0039	Same as Item 546.
549	HCl Off-Gas Condenser (Liquid System)		20' I.D. x 8' long, standard Monsanto fixed tube sheet heat exchanger with 304 3/4" O.D. x 14 BWG tubes. Exchanger has four passes. Overall area 460 sq.ft. Capacity 132,000 BTU/hour heat removal.
550	Chlorine Vaporizer	32-0185	Armstrong Chlorine Vaporizer, vertical type, complete with super heater section Monel construction - purchased from R. M. Armstrong Co. 90 tons/day capacity.
551	No. 1 Chlorinator	22-0675	3'-0" dia. x 16'-0". See Drawing E-6209.
552	No. 3 Chlorinator	22-0673	See Item 551.

DSW 303354

APPENDIX A - EQUIPMENT LIST
AREA 500

8.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
553	No. 3 Chlorinator	22-0673	See Item 551.
554	No. 4 Chlorinator	22-0674	See Item 551.
555	No. 8 Chlorinator		See Item 551. ✓
556	No. 7 Chlorinator		See Item 551. ✓
557	No. 6 Chlorinator	22-0677	See Item 551. ✓
558	No. 5 Chlorinator	22-0676	See Item 551. ✓
559	No. 4 Chlorinator circ. pump	19-0414	Same as Item 509.
560	Brink Demister No. 1 Chlorinator	41-0040	Same as Item 546.
561	Brink Demister No. 2 Chlorinator	41-0041	Same as Item 546.
562	Brink Demister No. 3 Chlorinator	41-0042	Same as Item 546.
563	Brink Demister No. 4 Chlorinator	41-0043	Same as Item 546.
564	Brink Demister No. 8 Chlorinator	41-0044	Same as Item 546.
565	Instrument Air Compressor	37-0098	Joy Manufacturing Company No. WFPOS-9, Size 8" x 5" skid-mounted air compressor unit with after-cooler and air receiver. Operated by a 25 HP, 1800 RPM, 440 volt motor. Capacity 105 cfm at 100 psig.

DSW 303395

APPENDIX A - EQUIPMENT LIST
AREA 500

9.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
566	Control Room Sink and Hood		Fume Hood designed by Browne-Morse Co. #73 Green unit with a gray colorlith top. Fan is an American Blower 1-1/4 HP removing 782 cfm at 1/2 S.P., 1/3 HP, 110 volt, 1725 RPM motor. Capacity fume removal 782 cfm.
567	Control Room Air Conditioner	44-0098	Carrier Model 48 DA008-601. Roof mounted cooling and heating unit; 98,000 BTU/hour nominal cooling; 240,000 BTU/hour heating nominal direct fired natural gas heater.
569	No. 6 Chlorinator Circulating Pump	19-0414	Same as Item 509.
570	No. 7 Chlorinator Circulating Pump	19-0414	Same as Item 509.
571	No. 5 Chlorinator Circulating Pump	19-0394	Same as Item 509.
572	No. 8 Chlorinator Circulating Pump	19-0405	Same as Item 509.
573	HCl Off-Gas Catch Tank (Liquid Chlorinator System)	22-0811	8'-6" dia. x 11'-3" straight side, carbon steel tank, with dished heads and a 3" Sch. 80 carbon steel pipe coil approximately 120' long. Capacity 5,000 gallons.
574	Trappings Transfer Pump (Liquid)		Self-priming, horizontal centrifugal pump (Goulds Model 3196 AVS-A60, constructed of Ductile iron with a 7" dia. open type impeller, 5 HP, 1800 RPM, 440 volt motor, with a Crane No. 9 outside mechanical seal, Code QP-868). Capacity 125 GPM at 75' TDH.
576	Brink Mist Eliminator HCl Catch Tank		Monsanto Brink type unit, 2'-6" dia. x 11' straight side, containing a Brink element designed for 5 psig at 105°F. temperature. Capacity 635 cfm at 85% HCl gas. Designed to collect 99% of entrained droplets and return to the catch tank.

DSW 303396

APPENDIX A - EQUIPMENT LIST
AREA 500

10.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
577	Therminol Surge Tank for Solid HCl Clean-Up Condenser		3'-6" dia. x 4'-3" straight side, carbon steel tank, containing a 2" Sch. 80 carbon steel coil approximately 10' long. Capacity 300 gals.
578	Therminol Pump for Solids Knock-Out Condenser		Self-priming, horizontal centrifugal pump (Goulds Model 3196 AVS-A60, constructed of Ductile iron, with an 8" open type impeller, driven by a 7-1/2 HP, 1800 RPM, 440 volt motor, pump equipped with a Crane No. 9 outside mechanical seal, Code QP-868). Capacity 150 GPM at 50' TDH.
579	Therminol Cooler for Solids Knock-Out Condenser System		Disler Model No. 2 VH air-cooled aluminum finned tube cooler, 3' x 3'-6" x 5'-9" high. Contains a fan driven by a 1 HP, 1800 RPM, 440 volt motor. Capacity 550,000 BTU/hour.
580	HCl Off-Gas Condenser (Solids System)		2' dia. x 10'-6" long, carbon steel heat exchanger, with 80 1-1/2" O.D. x 14 BWG x 8' long tubes (six tube passes). Condenser contains 250 sq.ft. surface area. Designed for 150 psig and 160°C.
581	HCl Off-Gas Catch Tank for Solid System		4' dia. x 3'-9" straight side, carbon steel tank with dished heads. Tank is a jacketed carbon steel tank designed for 125 psig at 160°C. Jacket designed for 110 psig at 160°C. Capacity 300 gallons.
582	Water Pump for Liquid Aroclor Knock-Out Condenser		Self-priming, horizontal centrifugal pump (Goulds Model 3196 AVS-A60, 2 x 3 - 10, with an 8" open type impeller constructed of Ductile iron with a Crane No. 9 outside mechanical seal, Code QP-868, driven by a 7-1/2 HP, 1800 RPM, 440 volt motor). Capacity 200 GPM at 75' TDH.
584	Weigh Tank	22-0776	6'-6" O.D. x 7'-9". Purchased from Art Welding Co.

DSW 303397

APPENDIX A - EQUIPMENT LIST
AREA 500

11.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
585	Chlorinator Charge Tank	22-0787	1,000 gallon steel tank purchased from Missouri Boiler & Tank Co. P.O. NO. D-47909. Installed Nov. 1959.
586	Air Blower (South)	11-0323	Rotary positive with extended base and V-drive, intake filter muffler, exhaust muffler, V-belt guard, 1" pressure relief valve. Purchased from Sutorbilt Corp.
587	Air Blower (Middle)	11-0324	Same as Item 586.
588	Biphenyl Feed Tank Agitator	18-0106	Falk 4 ELHZ, Ratio 13.9 to 1. Spec. SM-488-520C. Dwg. D-488-M116.
589	Fume Jet No. 2 Blow Tank	35-0122	Spec. SM-488-531. S&K Figure 4014, Scrubber-Separator consisting of: 1-6" Fig. 4010 Haveg 61 Fume Scrubber. 1-6" Fig. 4040 Polyester Fiberglass Separator.
592	Instrument Air Compressor	37-0073	JOY WFPOS-9, Size 6½ x 4½, Capacity 50 SCFM at 100 psig output. Water cooled, single stage double acting, carbon ring, non-lubricated, with V-belt drive.
594	Instrument Air Dryer	14-0029	Electric regenerative desiccant dryer, manufactured by Trinity Equipment Corporation.

DSW 303398

APPENDIX A - EQUIPMENT LIST
AREA 600

12.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
601	No. 3 Blend Tank Pump	19-0293	1-1/2 DSO 8 - all bronze, SAE 63 Frame 213 - Serial No. 121340 - Peerless.
602	No. 3 Blend Tank	22-0087	7'-0" O.D. x approx. 7'-0" on shell. Dwg. 9-C-8247.
603	No. 2 Blend Tank Pump	19-0269	1-1/2 DSO 8 - all bronze, SAE 63 Frame 213 - Serial No. 121340-Peerless.
604	No. 2 Blend Tank	22-0270	Same as Item 602.
605	No. 1 Blend Tank Pump	19-0267	Same as Item 603.
606	No. 1 Blend Tank		Same as Item 602.
607	No. 1 Vacuum Receiver	22-017801	Same as Item 609.
608	No. 1 Vacuum Receiver Pump	19-0272	Fig. 3202 No. 90 VPR, all bronze. Purchased from Blackmer Pump Co.
609	No. 2 Vacuum Receiver	22-0026	5'-6" I.D. x 5'-6" Monel. Dwg. C-795.
610	No. 2 Vacuum Receiver Pump	19-0271	Same as Item 608
611	No. 3 Vacuum Receiver		6'-6" dia. x 7'-9" straight side, vertical tank, lined with Monel plating. Tank equipped with a Monel coil constructed of 1-1/2" Sch. 40 Monel pipe with approx. 48 linear feet of coil. Capacity 2,000 gallons.
612	Receiver Pump (No. 3 Still)	19-0244	Model GZ-CH NEMA 213. Fig. 3202, 90 VPR bronze, Blackmer Pump.
613	No. 2 Filtrate Receiver	22-0086	7'-0" O.D. x 7'-0". Dwg. 9C-8247. Purchased from Chattanooga Boiler and Tank Co.

DSW 303399

APPENDIX A - EQUIPMENT LIST
AREA 600

13.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
614	No. 1 Filtrate Receiver	22-0269	Same as Item 613.
615	No. 4 Still	22-0779	7'-6" dia. x 7'-0" straight side, vertical carbon steel tank with dished heads. Tank equipped with a heating coil constructed of 2" Sch. 80 carbon steel pipe with approx. 77 sq.ft. of surface area. Capacity 2,500 gallons.
616	No. 4 Still Circulating Pump	19-0505	Self-priming, centrifugal, vertical pump, (Taber vertical submerged No. 345 CC-5, 12-1/2" dia. open type impeller, 25 HP, 1800 RPM, 440 volt motor, with a jacketed stuffing box for water cooling). Pump has a special Zallee expansion joint with rods and discharge pipeline. Capacity 150 GPM at 120' TDH.
617	Freight Elevator	10-0200	1,000# all steel freight elevator with 3' x 4' platform. Purchased from Midvale Material Handling Equipment Co. Capacity 1,000#.
618	No. 4 Still Heater	32-0271	Direct fired natural gas heater with insulation and refractory lining constructed by Alcorn Combustion Company. 7'-11" dia. x 11'-8" high shell containing a 3" dia. Sch. 80 coil, 259 sq.ft. surface area in a 6' dia. circle. Heater also contains a John Zink VEM No. 10 gas burner. Capacity 355,000 BTU/hour heat absorption (heat input 550,000 BTU/hour).
619	No. 4 Still Condenser	32-0272	1'-2" dia. x 10' long, carbon steel heat exchanger with 1 1/2-3/4" x 14 BWG x 8' long tubes, designed for four passes on tube side. Designed for full vacuum. Condenser contains a special design Marquette expansion joint constructed of 304 stainless steel with Sch. 30 welding ends. Capacity 477,000 BTU/hour.

DSW 303400

APPENDIX A - EQUIPMENT LIST
AREA 600

14.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
620	No. 4 Still Vacuum Receiver	22-0780	6'-6" dia. x 7'-9" straight side, carbon steel vertical tank with dished heads. Contains a 2" Sch. 80 carbon steel coil with approximately 52 sq.ft. of surface area. Designed for full vacuum operation at 400°C. Capacity 2,000 gallons.
621	No. 4 Still Vacuum Receiver Transfer Pump	19-0506	Self-priming, centrifugal pump (Goulds Model 3196 AVS-A60, 7-7/8" dia. open type impeller, 10 HP, 1800 RPM, 440 volt motor, jacketed stuffing box for Therminol heat, Crane No. 9 outside mechanical seal, Code QP-868, casing of pump jacketed for Therminol heat). Capacity 125 GPM at 50' TDH.
622	Therminol Surge Tank	22-0781	3'-6" dia. x 4'-3" straight side, carbon steel vessel with dished heads, contains a 2" Sch. 80 carbon steel coil with 7 sq.ft. surface area. Capacity 325 gallons.
623	Therminol Cir. Pump for No. 4 Still Coolant	19-0507	Self-priming, centrifugal pump (Goulds Model 3196 AVS-A60, 8-1/16" dia. open type impeller, 7-1/2 HP, 1800 RPM, 440 volt motor, with a Crane No. 9 outside mechanical seal, Code QP-868, constructed of Ductile iron). Capacity 100 GPM at 65' TDH.
624	Therminol Cooler No. 4 Still System		Air-cooled heat exchanger, Model No. 58E5A48, 1" O.D. BWG 16 tubes, 4' long, with aluminum fins, 60 sq.ft. surface area. Total fin surface area 1,171 sq. ft. Overall dimensions 2' x 4' x 4'-6" high. Manufactured by Disler Engineering Company. Fan for cooler 1 HP, 1800 RPM, 440 volt motor, capacity 5,500,000 BTU/hour.
625	No. 4 Still Catch Tank	22-0782	3' O.D. x 3'-9" straight side, carbon steel tank with dished heads and a 1-1/2" Sch. 80 coil 6' long. Capacity 200 gallons.

DSW 303401

APPENDIX A - EQUIPMENT LIST
AREA 600

15.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
626	No. 4 Still Vacuum Jet	35-0123	Three-stage Worthington No. 2-1/2 JPD jet ejector. First stage special design; second stage No. 5; and third stage No. 6-A. Jet also includes a 10" I.D. porcelain barometric inter-condenser.
627	Therminol Surge Tank (No. 4 Still Cooling System)	22-0783	5'-0.D. x 12' straight side, horizontal, carbon steel tank with dished heads. Tank contains a 3" Sch. 80 coil 20' long. Capacity 1,900 gallons.
628	Therminol Heater Circulating Pump	19-0508	Self-priming, centrifugal pump (Taber vertical submerged No. 346-CC6, with a 12-1/2" open type impeller; 50 HP, 1800 RPM, 440 volt motor, jacketed stuffing box for water cooling). Capacity 350 GPM at 165' TDH.
629	Therminol Heater	32-0274	A helical tube heater with cross-flow connection of the four circulation type for heating Therminol FR-2 by direct firing with natural gas, complete with insulation and refractory lining and coil, 8'-6" dia. x 15'-6" high shell with walls lined with 5" castable refractory Lummite-Haydite vermiculite 124 mix, suitable for 2,000°F service. 4" Sch. 80 coil 350' long with a total surface area of 950 sq.ft. Furnace fabricated by Alcorn Combustion Co. Furnace provided with a North American 6" No. 214-88 dual fuel burner. Capacity rated at 8,150,000 BTU/hour for the burner. Total heat absorbed 6,000,000 BTU/hour.
631	Flaker Tank & Agitator	22-0700	6-blade turbine type impeller, 26" dia., constructed of carbon steel with a 3-1/2" dia. x 8 ft. long shaft. Motor reducer is a Falk 4 ELHZ speed reducer with a 13.9:1 ratio. Slow speed shaft, 125 RPM, driven by a 10 HP, 1800 RPM, 440 volt motor. Capacity, provide medium agitation, in 250 gal. steel jacketed tk.
632	Barometric Sump Tank		3' dia. x 3' long vertical tank, fabricated by Fiberglass. Capacity 150 gallons.

DSW 303402

APPENDIX A - EQUIPMENT LIST
AREA 600

16.

<u>Time</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
634	Lime Hopper No. 4 Still		2' x 2' x 1' high hopper, fabricated of carbon steel. To fit a 6" nozzle. Capacity 50 lbs. of lime.
635	Montar Conveyor	33-0055	2'-0" x 20'-0" steel roller type conveyor. Drawing D-488-M132.
636	Still Bottoms Blower	11-0339	Centrifugal fan per spec. SM-488-636. Blower for fume removal. V-belt drive and motor assembly for above fan. 5 HP motor, 440 volt, 60/CY, 3/Phase, for chemical duty.
638	Drum Conveyor		2'-6" x 10'-0" used at Liquid Aroclor drum-out station.
639	Scale (Toledo)	24-0182	Model 2881 - Serial No. 4159, used for weighing Liquid Aroclor drums.
640	Polishing Filter		Sparkler Filter - 1'-9" x 3'-6" OAH.
641	Entrainment Separator #4 Still	41-0045	Purchased from Wright Auston. P.O.No. OED-7952.
642	Earth Oven	14-0033	16 KW-2 door oven with controls. Standard electric oven. Purchased from Despatch Oven Company.
643	Pyranol Filter	15-0029	Model 18-S-23 horizontal plate type 304, st. w/scavenger plate feature on pipe legs. Purchased from Sparkler Manufacturing Co.
644	Filtrate Receiver Pump	19-0268	Pump 1-1/2 DSO 8 all bronze - SAE 63 Frame 213.
645	Floor Cleaning Machine	49-0537	Model No. 204325 - Serial No. EP 17646. 5 HP, 220 volt, 3 phase motor. Purchased from G.H. Tennant.
646	Micro Pulverizer	17-0027	2 DH Micro Pulverizer with discharge chute mounted on steel base. Main motor 10 HP, 3450 RPM, fan cooled. Feed motor 1/3 HP, 1725 RPM, Serial No. 12184. Purchased from Pulverizing Machinery.

DSW 303403

APPENDIX A - EQUIPMENT LIST
AREA 600

17.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
650	No. 1 Still Pot	20-0018-01	6'-0" dia. x 5'-6" high closed steel as per drawing C-809.
651	Circulating Pump No. 1 Still	19-0230-01	No. 345 semi-stl. CC-5, centrifugal vertical motor pump. 3" inlet, 2-1/2" outlet. Purchased from Taber Pump Co.
652	No. 1 Still Heater	42-0018	3'-9" I.D. x 4'-10-1/2" steel sheet made per Dwg. E-886.
653	No. 1 Still Condenser	32-0116	12" dia., 7'-6" long, 19-1-1/2" 1 PS Sch. 40 Monel tubes. Purchased from Dowington Iron Works.
654	No. 1 Still Ejector	35-0105	#9 N9BON001 Type G Carbate Ejector.
655	No. 2 Still Pot	20-0002-01	6'-0" dia. x 5'-6" high closed steel as per drawing C-809.
656	Circulating Pump No. 2 Still	19-0240-01	Same as Item 651.
657	No. 2 Still Heater	42-0050	No Description on File.
658	No. 2 Still Condenser	32-0141	Same as Item 653.
659	No. 2 Still Ejector	35-0106	Same as Item 654.
660	No. 3 Still Pot	20-0052	Same as Item 655.
661	Circulating Pump No. 3 Still	19-0248	Same as Item 651.
662	No. 3 Still Heater	42-0052	No description on file.
663	No. 3 Still Condenser		Same as Item 653.

DSW 303404

APPENDIX A - EQUIPMENT LIST
AREA 600

18.

<u>Item</u>	<u>Equipment</u>	<u>Equip.No.</u>	<u>Description</u>
664	No. 3 Still Ejector	35-0125	2-1/2 JPA 3rd Stage. Purchased from Worthington Corp. P.O. No. 82036. Installed January 1966.
669	Lime Slurry Tank	22-0790	Slurry Tank for No. 3 Still. 10" IRS steel Hyd. & flange assembly, steel internal parts.
670	No. 1 Roof Tank	22-0078-01	8' dia. x 30' long. Capacity 11,000 gallons.
674	Circulating Pump (Spare)	19-0238-01	Same as Item 651.
690	Aroclor Flaker	14-0031	Model VMG 5-56, size 24" x 48". Stainless steel drum with feed pans. Purchased from Goslin.

DSW 303405

STANDARD MANUFACTURING PROCESS
FOR
AROCLOL

APPENDIX B - MATERIAL SPECIFICATIONS

I - FINISHED GOODS

- a. Aroclor 1221
- b. Aroclor 1232*
- c. Aroclor 1242
- d. Aroclor 1248
- e. Aroclor 1254
- f. Aroclor 1260
- g. Aroclor 1262
- h. Aroclor 1268
- i. Aroclor 2565 (undistilled)
- j. Aroclor 4465
- k. Aroclor 5442
- l. Aroclor 5460

*Aroclor 1232 is a mixture of approximately 53.4% Aroclor 1221 and 46.6% Aroclor 1242 (by volume).

II - RAW MATERIALS

- a. Biphenyl
- b. Santowax C
- c. Santowax R
- d. Chlorine
- e. Hydrated Lime
- f. Attapulugus Earth
- g. Ferric Chloride

DSW 303406

] IN 54

MONSANTO CHEMICAL COMPANY
 ORGANIC CHEMICALS DIVISION

ANNISTON

**FINISHED PRODUCT
SPECIFICATION**

PRODUCT CODE

826.10 (27-L)

METHOD IDENTITY

SALES CODE

1040-225-04/11/16-09

PRODUCT (Trade name)

AROCOR 1221

TYPE

Distilled

PRODUCT (Chemical name)

TYPE

CHEMICAL FORMULA

Mixture

Issued 4/25/61

By:

W. W. Klemme
 W. W. Klemme
 Anniston Plant
 Standards Dept.

MOL. WT.

USE

Sales

SUPERSEDES SPECS OF

7/29/55

SPECIAL JOB NO.

- -

SAMPLE FOR ANALYSIS

2 x 1/2 gal. amber jugs

UNCLASSIFIED SPECIFICATION
Routine Tests

<u>Property</u>	<u>Specification</u>	<u>Method No.</u>
Color, APHA	100, Max.	14-44-54
Specific Gravity @ 25/15.5°C.	1.182-1.192	14-49-54
Acidity, mg KOH/g.	0.014, Max.	14-42-54
Viscosity @ 100°F., SUS	38-41	14-4-52

General Information

Chlorine, %	20.5-21.5	14-13-54
-------------	-----------	----------

DSW 303407

WATER_PCB-SD0000050958

Monsanto Chemical Company
ORGANIC CHEMICALS DIVISION
ANNISTON
FINISHED PRODUCT
SPECIFICATION

PRODUCT CODE
826.11 (27-L)
METHOD IDENTITY
--
SALES CODE
1040-230-04/11/16-09

Issued: 10/28/63

By: *O. E. Doli*
O. E. Dolin
Anniston Plant
Standards Dept

PRODUCT (Trade name) **AROCLOX 1232** TYPE **Distilled**
PRODUCT (Chemical name) TYPE

CHEMICAL FORMULA

Mixture

MOL WT.

USE

Sales

SUPERSEDES SPECS OF **4/25/61** SPECIAL JOB NO. **--** SAMPLE FOR ANALYSIS **2 x 1/2 gal. amber jugs**

UNCLASSIFIED SPECIFICATIONS

Routine Tests

Property	Specification	Method No.
Color, APHA	100, Max.	14-44-54
Acidity, mg KOH/g.	0.014, Max.	14-42-54
Specific Gravity @ 25/15.5°C	1.270-1.280	14-49-54
Viscosity @ 100°F., SUS	44-51	14- 4-52

General Information

Chlorine, %	31.5-32.5	14-13-54
-------------	-----------	----------

RESTRICTED SPECIFICATIONS

Routine Tests

Water, ppm	35 Max.	14-53-54
Refractive Index @ 25°C	1.6180-1.6200	14-34-54
Condition	Clear Liquid	10-084

General Information

Inorganic Chlorides, ppm	0.10 Max.	014
Resistivity, ohm-cm 500V DC, 0.1" gap, 100°C.	500 x 10 ⁹ Min.	14-35-54
Dielectric Constant @ 1 KC, 100°C.	4.3-4.5	14-36-54
Distillation Range, °C. corrected:		
10%, by wt.	293, Min.	14-31-54
50%, by wt.	310-320	
90%, by wt.	360, Max.	

DSW 303408

Anniston
Finished Product Specification
Product: Aroclor 1232 (Distilled)

Page 2

RESTRICTED SPECIFICATIONS

General Information - Continued

Four Point, °C.	Below -30	14-47-54
Corrosion Test		
Change in wt. aluminum, %	0.0	
Acid No. after test, mg KOH/g.	0.015, Max.	14-52-54
Inorganic Chlorides after test, ppm	0.10, Max.	014
Condition after test	Clear	G-1
Color after test, APHA	100, Max.	14-44-54

OED/gm
10/28/63

DSW 303409

WATER_PCB-SD0000050960

Monsanto
ORGANIC CHEMICALS DIVISION
ANNISTON AND KRUMMICH
FINISHED PRODUCT
SPECIFICATION

PRODUCT Aroclor 1242, Electrical Grade		DEPT. 246 27 L
PLANT WGK, Ann.	PRODUCT CODE 81101 - 82612	USE Sales
DATE EFFECTIVE 6/17/66	SALES CODE 1040-240-04/11/16-03/09	

APPROVALS	
PRODUCTION P. E. Heisler	RESEARCH R. W. Bucknell
SALES W. E. Ault	QUALITY CONTROL (Chief Chemist) O. E. Dolin W. J. Gresham
GENERAL DESCRIPTION	CHEMICAL FORMULA

Also known as Pyranol 1499

MOL WT. SUPERSEDES SPECS. OF 4-13-62	SAMPLE FOR ANALYSIS 3 x 5 pt. Amber Bottles	ISSUED BY J. P. Prader
CHARACTERISTIC	LIMITS	METHOD

UNRESTRICTED INFORMATION

		WGK	Anniston
(R) Color (APHA)	40, max.	10084	14-44-54
(R) Condition	Clear	10084	G-1
(R) Sp. Gr. @ 25/15.5°C.	1.381 - 1.392	10114	14-49-54
(R) Acidity (mg. KOH/g.)	0.010, max.	10087	14-42-54
(R) Moisture (H ₂ O)	35 ppm., max.	10620	14-53-54
(R) Refractive Index @25°C.	1.6240 - 1.6260	10125	14-34-54
(R) Inorganic (Free) Chlorides	0.05 ppm., max.	12741	014
(R) Pour Point	-14°C. or Lower	10115	14-47-54
(R) Dielectric Constant (1000 cycles @ 100°C.)	4.70 - 4.90	11608	14-36-54
(R) Resistivity (500 VDC @ 100°C. 0.1" Gap) As OHM-CM x 10 ⁹	500, min.	11607	14-35-57
(R) Hydrolysis Stability Test (As Chlorides)	0.5 ppm, max.	12672	010
(R) Monsanto Stability Test (As Chlorides)	0.4 ppm., max.	11503	14-61-57
(R) Distillation Range (ASTM D-20, Corrected)		10117	14-31-54
10% Distilled by wt.	325°C., min.		
90% Distilled by wt.	360°C., max.		
Sulfates	None	12169	-----
Dielectric Strength @ 25°C.	35 KV, min.	11605	043
Flash Point (°F)	338 - 392	10123	14-28-54
Fire Point	None	10123	14-28-54
Corrosion Test (6 hrs. @ 210°C. with bright aluminum foil)		10126	14-55-57
Change in wt. of AL %	0.0		
A.C. Color (APHA)	60, max.	10084	14-44-54
A.C. Condition	Clear	10084	G-1
A.C. Acidity (mg. KOH/g.)	0.010, max.	10087	14-42-54
A.C. Inorganic (Free) Chlorides	0.05 ppm., max.	12741	014
Visc. @ 100°F. (SUS)	82 - 92	10086	14- 4-52

(Over)

DSW 303410

(R) Routine Analysis. All others Analyzed by request Only

WATER_PCB-SD0000050961

GENERAL INFORMATION

Specific Heat @ 25°C.	0.30
Coefficient of Expansion (25 - 65°C)	0.00068 cc/°C.
Fixed Chlorine	42.5 - 53.5%
Power Factor @ 100°C.	(Typical Values)
60 cycles	5
1000 cycles	0.30

SPECIAL CUSTOMER REQUIREMENTS

General Electric

Thermal Chemical Chlorides	0.65 ppm., max.	12722
----------------------------	-----------------	-------

Westinghouse

Hydrolizable Chlorine (Westinghouse Stability Test)	1.0 ppm., max.	T-2219
--	----------------	--------

DSW 303411

MONSANTO CHEMICAL COMPANY
ORGANIC CHEMICALS DIVISION

**W.G.K. and Anniston
FINISHED PRODUCT
SPECIFICATION**

PRODUCT CODE 81102	DEPT 246 W.G.K. 27 Ann.
USE Sales	
SALES CODE 1040-260-04/11/16-03/09	

Issued 1/10/64

W. G. Krummrich Plant
By M. D. Dsch

PRODUCT (Trade Name) Aroclor 1248 Electrical Grade (Pyranol 1498)	TYPE
--	------

PRODUCT (Chemical Name)	TYPE
-------------------------	------

CHEMICAL FORMULA

Mixture

Corrected 2-16-64

MOL WT.

SUPERSEDES SPECS OF
2-12-60

SAMPLE FOR ANALYSIS
3 x 5 pt. Amber Bottles

UNRESTRICTED INFORMATION

<u>Routine Tests</u>	<u>Specification</u>	<u>Method No's.</u>	
		<u>WGK</u>	<u>Anniston</u>
Color, APHA	50, max.	10,084	14-44-54
Condition	Clear	10,084	G-1
Sp. Gr. @65/15.5°C	1.405 - 1.415	10,114	14-49-54
Acidity (mg. KOH/gm.)	0.010, max.	10,087	14-42-54
Moisture	35 ppm., max.	10,620	14-53-54
Refractive Index @25°C	1.6285 - 1.6305	10,125	14-34-54
Inorganics (Free) Chlorides	0.05 ppm., max.	T-2092	014
Pour Point	0°C, max.	10,115	14-27-54
Dielectric Constant (1000 cycles @100°C)	4.5 - 4.7	11,608	14-36-54
Resistivity (500 VDC @100°C 0.1" GAP) as OHM-CM x 10 ⁹	500, min.	11,607	14-35-57
Hydrolysis Stability Test (as Chlorides)	1.0 ppm., max.	T-2087	010
Monsanto Stability Test (as Chlorides)	0.5 ppm., max.	11,503	14-61-57
Distillation Range (ASTM D-20, Corrected)		10,117	14-31-54
10% Distilled by Wt.	345°C, min.		
90% Distilled by Wt.	385°C, max.		

General Tests (made only on request)

Sulfates	None	12,169	- - -
Dielectric Strength @25°C	35 KV, min.	11,605	043
Fire Point	None to boiling point	10,123	14-28-54
Viscosity @54.4°C (SUS)	73 - 80	10,086	14-4-52
Corrosion Test (6 hrs. @210°C with bright aluminum foil)		10,126	14-55-57
Change in wt. of Al (%)	0.0		

DSW 303412

(over)

A. C. Color, APHA 60, max.
A. C. Condition Clear
A. C. Acidity (mg. KOH/g.) 0.010, max.
A. C. Inorganic (Free) Chlorides 0.05 ppm., max.

10,084 14-44-54
10,084 G-1
10,087 14-42-54
10,118 010

GENERAL INFORMATION

Specific Heat @25°C 0.27
Fixed Chlorine 47.5 - 48.5%

DSW 303413

MONSANTO CHEMICAL COMPANY

ORGANIC CHEMICALS DIVISION

Krummrich and Anniston

**FINISHED PRODUCT
SPECIFICATION**

PRODUCT CODE

81102 - 82641

DEPT. OR JOB

WGK 246-Ann.27

USE

Sales

SALES CODE

1040-260-04/11/16-03/09

Issued JUL 15 1964

W. G. Krummrich Plant

 By M. D. D. J.

PRODUCT (Trade Name)

Aroclor 1248

TYPE

Non-Electrical Grade

PRODUCT (Chemical Name)

TYPE

CHEMICAL FORMULA

Mixture

MOL WT.

SUPERSEDES SPECS OF

5-4-61

SAMPLE FOR ANALYSIS

3 x 16 oz. W. M. Bottles

UNRESTRICTED INFORMATION
Routine Tests
Specification
WGK
Anniston
Method No.

Appearance

 Colorless to light
yellow-green liquid

10084

G-1

Condition

Clear

10084

G-1

Color, APHA

100 max.

10084

046

Sp. Gr. @65/15.5°C

1.405 - 1.415

10114

052

Acidity (mg. KOH/g.)

0.010, max.

10087

050

 Moisture (H₂O) (ppm.)

50, max.

10620

047

Viscosity @130°F (SUS)

73 - 80

10086

049

General Tests

Flash Point (°F)

370, min.

10123

14-28-54

DSW 303414

Monsanto

ORGANIC CHEMICALS DIVISION

FINISHED PRODUCT SPECIFICATION

PRODUCT	Aroclor 1254 Electrical Grade	DEPT.	246 - 27
PLANT	WGK and Ann.	PRODUCT CODE	81103 82613
DATE EFFECTIVE	JUN 28 1965	SALES CODE	1040-280-04/11/16-03/09
		USE	Sales

APPROVALS

PRODUCTION	P. E. Heisler	RESEARCH	R. W. Bucknell
SALES	W. E. Ault	QUALITY CONTROL (Chief Chemist)	W. J. Gresham. O. E. Dolin
GENERAL DESCRIPTION		CHEMICAL FORMULA	

Also called Pyranol 1476

MOL WT.

SUPERSEDES SPECS. OF	SAMPLE FOR ANALYSIS	ISSUED BY
4-13-62	3 x 5 pt. Amber Bottles	M. D. Dealy
CHARACTERISTIC	LIMITS	METHOD

UNRESTRICTED INFORMATION

		WGK	Ann.
(R) Color, APHA	50, max.	10084	14-44-54
(R) Condition	Clear	10084	G-1
(R) Sp. Gr. @65/15.5°C	1.495 - 1.505	10114	14-49-54
(R) Acidity (mg. KOH/gm.)	0.010, max.	10087	14-42-54
(R) Moisture (H ₂ O)	35 ppm, max.	10620	14-53-54
(R) Refractive Index @25°C	1.6370 - 1.6390	10125	14-34-54
(R) Inorganic (Free) Chlorides	0.05 ppm., max.	T-2092	014
(R) Pour Point	7 - 12°C	10115	14-27-54
(R) Dielectric Constant (1000 cycles @100°C)	4.15 - 4.35	11608	14-36-54
(R) Resistivity (500 VDC @100°C 0.1" GAP) as OHM-CM x 10 ⁹	500, min.	11607	14-35-57
(R) Hydrolysis Stability Test (as Chlorides)	1.0 ppm., max.	T-2087	010
(R) Monsanto Stability Test (as Chlorides)	0.5 ppm., max.	11503	14-61-57
(R) Distillation Range (ASTM D-20, Corrected)		10117	14-31-54
10% Distilled by Wt.	366 - 378°C		
50% Distilled by Wt.	371 - 383°C		
90% Distilled by Wt.	379 - 394°C		
Sulfates	None	12169	
Dielectric Strength @25°C	35KV, min.	11605	043
Flash Point	None	10123	14-28-54
Fire Point	None	10123	14-28-54
Visc. @210°F (SUS)	44-48	10086	14-4-52
Corrosion Test (6 hrs @210°C with bright aluminum foil)		10126	14-55-57
Change in wt. of Al (%)	0.0		
A. C. Color, APHA	60, max.	10084	14-44-54
A. C. Condition	Clear	10084	G-1
A. C. Acidity (mg. KOH/g.)	0.010, max.	10087	14-42-54
A. C. Inorganic (Free) Chlorides	0.05 ppm., max.	10118	010

DSW 303415

(R) Routine Analysis, All others Analyzed by request Only

WATER_PCB-SD0000050966

MONSANTO CHEMICAL COMPANY

ORGANIC CHEMICALS DIVISION

**Krummrich and Anniston
FINISHED PRODUCT
SPECIFICATION**

PRODUCT CODE

81103 - 82613

DEPT. OR JOB

WGK-246-Ann.27

JUL 15 1964

Issued

W. G. Krummrich Plant

By

W. D. D. S. S.

USE

Sales

SALES CODE

1040-280-04/11/16-03/09

PRODUCT (Trade Name)

Aroclor 1254

TYPE

Non-Electrical Grade

PRODUCT (Chemical Name)

TYPE

CHEMICAL FORMULA

Mixture

MOL WT.

SUPERSEDES SPECS OF

5-4-61

SAMPLE FOR ANALYSIS

3 x 16 oz. W. M. Bottles

UNRESTRICTED INFORMATION

Routine Tests

Specification

WGK

Anniston

Method No.

Color, APHA

100, max.

10084

046

Sp. Gr. @65/15.5°C

1.495 - 1.505

10114

052

Acidity (mg. KOH/g.)

0.010, max.

10087

050

Moisture (H₂O) (ppm.)

50, max.

10620

047

General Tests

Flash Point

None

10123

14-28-54

Viscosity @210°F (SUS)

44-48

10086

049

DSW 303416

FINISHED PRODUCT SPECIFICATION

PRODUCT CODE 82614 - Anniston
81105 - WGK
Department 27 - Anniston
246 - WGK
SALES CODE
1040-290-04/11/16/-03/09

Issued 2/7/62
W. G. Krumrich Plant
By E. O. Martin

PRODUCT (Trade name) **AROCLOR 1260** TYPE **Electrical Grade**
PRODUCT (Chemical name) TYPE
(Pyranol 1482)

CHEMICAL FORMULA

MIXTURE

MOL WT. ---

USE

Sales

SUPERSEDES SPECS OF

3/28/58

SPECIAL JOB NO.

SAMPLE FOR ANALYSIS

3 x 5 pt. Amber Bottles

UNRESTRICTED INFORMATION

Routine Tests:

Property	Specification	Method No.	
		WGK	Anniston
Color (APHA)	150, Max.	10,084	14-44-54
Condition	Clear	10,084	-----
Specific Gravity @ 90/15.5°C	1.555-1.566	10,114	14-49-54
Acidity (mg KOH/g.)	0.014, Max.	10,087	14-42-54
Moisture (H ₂ O) (ppm.)	35, Max.	10,620	14-53-54
Viscosity @ 210°F (SUS)	72-78	10,086	14- 4-52
Refractive Index @ 25°C	1.6455-1.6470	10,125	14-34-54
Inorganic (Free) Chlorides	No detectable amt.	10,118	14-48-57
Pour Point (°C)	25-34	10,115	14-47-54
Distillation Range		10,117	14-32-54
(ASTM D-20, Corrected)			
10% Distilled by wt.	385-398°C		
50% Distilled by wt.	390-404°C		
90% Distilled by wt.	400-420°C		
Corrosion Test		10,126	14-55-57
(6 hrs. @ 210°C with bright aluminum foil)			
Change in weight of Al (%)	0.0		
A.C. Color (APHA)	150, Max.	10,084	14-44-54
A.C. Condition	Clear	10,084	-----
A.C. Acidity (mg.KOH/g.)	0.014, Max.	10,087	14-42-54
A.C. Inorganic (Free) Chlorides	No detectable amt.	10,118	14-48-57

General Tests (made only on request)

Flash Point	None	10,123	14-28-54
Fire Point	None	10,123	14-28-54
Fixed Chlorine (%)	59.5-60.5	10,088	14-13-54
Dielectric Strength @ 50°C	30 K.V., Min.	11,605	-----
Resistivity	500, Min.	11,607	14-35-57
(500 VDC @ 100°C 0.1" gap as ohm-cm. x 10 ⁹)			
Dielectric Constant	3.6-3.8	11,608	14-36-45
1000 cycles, 100°C			

DSW 303417

UNRESTRICTED INFORMATION (continued)

<u>Property</u>	<u>Specification</u>	<u>Method No.</u>	
		<u>WCK</u>	<u>Anniston</u>
<u>Restricted Information</u>			
Monsanto Stability Test (As Chlorides in ppm.)	0.7, Max.	11,503	14-61-57
<u>General Information</u>			
Evaporation Loss after 6 hrs. @ 100°C (%)	0.2, Max.	10,116	14-32-54
Stability test - after 30 days at 100°C in glass sealed in air	No liberation of chlorine or chlorides	-----	-----

Retyped:
March 7, 1963

/gm

DSW 303418

MONSANTO CHEMICAL COMPANY

ORGANIC CHEMICALS DIVISION

Krummrich and Anniston

FINISHED PRODUCT
SPECIFICATION

PRODUCT CODE

81105 - 82614

DEPT. OR JOB

WGK 246 Ann. 27

USE

Sales

SALES CODE

1040-290-04/11/16-03/09

Issued

JUL 15 1964

W. G. Krummrich Plant

By

M. D. Dwyer

PRODUCT (Trade Name)

Aroclor 1260

TYPE

Non-Electric Grade

PRODUCT (Chemical Name)

TYPE

CHEMICAL FORMULA

Mixture

SOL. WT.

SUPERSEDES SPECS OF

5-4-61

SAMPLE FOR ANALYSIS

3 x 16 oz. W. M. Bottles

UNRESTRICTED INFORMATION

Routine TestsSpecification

WGK

Anniston

Method No.

Condition

Clear

10084

G-1

Color, APHA

150, max.

10084

046

Sp. Gr. @90/15.5°C

1.555 - 1.566

10114

052

Acidity (mg. KOH/g.)

0.014, max.

10087

050

Moisture (H₂O) (ppm.)

50, max.

10620

047

Viscosity @210°F (SUS)

72-78

10086

049

General Tests

Flash Point

None

10123

14-28-51

Fire Point

None

10123

14-28-51

DSW 303419

Monsanto

ORGANIC CHEMICALS DIVISION

ANNISTON AND KRUMMRICH FINISHED PRODUCT SPECIFICATION

PRODUCT

AROCLOR DISTILLED 1262

DEPT. Ann-2
WGK-24

PLANT

ANN. & WGK

PROJECT CODE

82615-Ann.

USE

Sales

DATE EFFECTIVE

MAR 02 1966

SALES CODE

1040-300-04/11/16-03-09

APPROVALS

PRODUCTION

P. E. Heisler

RESEARCH

W. D. Robinson

SALES

W. E. Ault

QUALITY CONTROL (Chief Chemist)

W. J. Gresham

O. E. Dolin

GENERAL DESCRIPTION

CHEMICAL FORMULA

MOL. WT.

SUPERSEDES SPECS. OF

1/24/62

SAMPLE FOR ANALYSIS

2 x 16 oz. W.M. Bottles

ISSUED BY

J. P. Prader

CHARACTERISTIC

LIMITS

METHOD

UNRESTRICTED INFORMATION

(R) Appearance	Viscous liquid	10084
(R) Color	APHA 150, max.	10084
(R) Specific Gravity @ 90/15.5°C.	1.572 - 1.583	10114
(R) Acid No. (mg. KOH/g.)	0.014, max.	10087
(R) Viscosity @ 210°F.	86 - 100 SUS	10086

GENERAL INFORMATION

Pour Point (ASTM)	35 - 38°C.
Distilling Range	390 - 425°C.
Total Chlorine	61.5 - 62.5%
Specific Gravity Temperature Coefficient (85 - 130°C.)	0.001/°C.

DSW 303420

(R) Routine Analysis, All others Analyzed by request Only

WATER_PCB-SD0000050971

Monsanto
ORGANIC CHEMICALS DIVISION

FINISHED PRODUCT
SPECIFICATION

PRODUCT Aroclor 1268		DEPT. 27-S
PLANT Anniston	PRODUCT CODE 826,16	USE Sales
DATE EFFECTIVE 3-11-66	SALES CODE 1040-340-04-11-16-009	

APPROVALS

PRODUCTION R. G. Moody /s/ 12-30-65	RESEARCH L. R. Stark /s/ 3-8-66
SALES W. E. Ault /s/ 3-2-66	QUALITY CONTROL (Chief Chemist) O. E. Dolin /s/ 12-29-65
GENERAL DESCRIPTION	CHEMICAL FORMULA

Mixture

MOL WT.		
SUPERSEDES SPECS. OF 4-25-61	SAMPLE FOR ANALYSIS 1 x 16 oz. Can or W.M. Bottle	ISSUED BY J. A. Liddle 11-30-65
CHARACTERISTIC	LIMITS	METHOD

UNRESTRICTED INFORMATION

(R) Appearance and color Color, Gardner	White to off-white powder 6 Max.	G-1
(R) Hold Point, °C	150 - 170	119
(R) Acid No., mg KOH	0.05 Max.	116

RESTRICTED INFORMATION

Screen Test, % retained on

No. 16 mesh	0.1 Max.	120
No. 20 mesh	10 Max.	
Specific Gravity @ 25/25°C., Pycnometer	1.804 - 1.811	
Loss @ 163°C., %	0.15 - 0.24	

DSW 303421

(R) Routine Analysis. All others Analyzed by request Only

WATER_PCB-SD0000050972

Monsanto
ORGANIC CHEMICALS DIVISION

**FINISHED PRODUCT
SPECIFICATION**

PRODUCT Aroclor 2565		DEPT. 27-S
PLANT Anniston	PRODUCT CODE 826.18	USE Sales
DATE EFFECTIVE 3-11-66	SALES CODE 1040-390-04-009	

APPROVALS

PRODUCTION R. G. Moody /s/ 12-29-65	RESEARCH L. R. Stark /s/ 3-8-66
SALES W. E. Ault /s/ 3-2-66	QUALITY CONTROL (Chief Chemist) O. E. Dolin /s/ 12-29-65
GENERAL DESCRIPTION	CHEMICAL FORMULA

Mixture

MOL WT.		
SUPERSEDES SPECS. OF 4-25-61	SAMPLE FOR ANALYSIS 1 x 16 oz. Can or W.M. Bottle	ISSUED BY J. A. Liddle 11-30-65
CHARACTERISTIC	LIMITS	METHOD

UNRESTRICTED INFORMATION

(R) Appearance and Color	Black, brittle resin	G-1
(R) Softening Point, °C	66 - 72	115
(R) Acid No., mg KOH/g	1.4 Max.	116

DSW 303422

(R) Routine Analysis, All others Analyzed by request Only

WATER_PCB-SD0000050973

Monsanto

ORGANIC CHEMICALS DIVISION

FINISHED PRODUCT SPECIFICATION

PRODUCT			DEPT.
Aroclor 4465			27-S
PLANT	PRODUCT CODE	USE	
Anniston	826.19	Sales	
DATE EFFECTIVE	SALES CODE		
3-11-66	1040-430-04-11-16-009		

APPROVALS

PRODUCTION	RESEARCH
R. G. Moody /s/ 2-29-66	L. R. Stark /s/ 3-8-66
SALES	QUALITY CONTROL (Chief Chemist)
W. E. Ault /s/ 3-2-66	O. E. Dolin /s/ 12-29-65
GENERAL DESCRIPTION	CHEMICAL FORMULA

Mixture

MOL WT.		
SUPERSEDES SPECS. OF	SAMPLE FOR ANALYSIS	ISSUED BY
4-25-61	1 x 16 oz. Can or W.M. Bottle	J. A. Liddle 11-30-65
CHARACTERISTIC	LIMITS	METHOD

UNRESTRICTED INFORMATION

(R) Appearance and Color	Clear, light yellow resin	G-1
(R) Color, NPA, Moiten	2 Max.	G-4
(R) Softening Point, °C	60 - 66	115
Acid No., mg KOH/g	0.05 Max.	116

DSW 303423

(R) Routine Analysis, All others Analyzed by request Only

WATER_PCB-SD0000050974

Monsanto
ORGANIC CHEMICALS DIVISION

**FINISHED PRODUCT
SPECIFICATION**

PRODUCT		DEPT.
Aroclor 5442		27-S
PLANT	PRODUCT CODE	USE
Anniston	826.21	Sales
DATE EFFECTIVE	SALES CODE	
3-15-66	1040-470-04-009	

APPROVALS

PRODUCTION	RESEARCH
R. G. Moody /s/ 12-29-65	L. R. Stark /s/ 3-11-66
SALES	QUALITY CONTROL (Chief Chemist)
W. E. Ault /s/ 3-2-66	O. E. Dolin /s/ 12-29-65
GENERAL DESCRIPTION	CHEMICAL FORMULA

Mixture

MOL WT.		SAMPLE FOR ANALYSIS		ISSUED BY
SUPERSEDES SPECS. OF		1 x 16 oz. Can or W.M. Bottle		J. A. Liddle 11-30-65
4-25-61				
CHARACTERISTIC	LIMITS	METHOD		

UNRESTRICTED INFORMATION

(R) Appearance and Color	Clear, yellow sticky resin	G-1
(R) Color, NPA, Molten	2 Max.	G-4
(R) Softening Point, °C	46 - 52	115
Acid No., mg. KOH/g	0.05 Max.	116

DSW 303424

Monsanto
ORGANIC CHEMICALS DIVISION

FINISHED PRODUCT
SPECIFICATION

PRODUCT Aroclor 5460 (High Soluble Grade)		DEPT. 27-S
PLANT Anniston	PRODUCT CODE 826.22	USE Sales
DATE EFFECTIVE 3-11-66	SALES CODE 1040-480-04-009	

APPROVALS	
PRODUCTION R. G. Moody /s/ 12-30-65	RESEARCH L. R. Stark /s/ 3-8-66
SALES W. E. Ault /s/ 3-2-66	QUALITY CONTROL (Chief Chemist) O. E. Dolin /s/ 12-29-65
GENERAL DESCRIPTION	CHEMICAL FORMULA

Mixture

MOL WT. 4-21-61	SAMPLE FOR ANALYSIS 1 x 16 oz. Can or W.M. Bottle	ISSUED BY J. A. Liddle 11-30-65
CHARACTERISTIC	LIMITS	METHOD

UNRESTRICTED INFORMATION

(R) Appearance and Color	Clear, yellow to amber brittle resin or flakes	G-1
(R) Color, NPA, Molten	2 Max.	G-4
(R) Softening Point, °C	98 - 105	115
(R) Solubility (25 g in 25 g. Acetone @ 25°C.)	3 hour, Min.	117
Acid No., Mg KOH/gm.	0.05 Max.	116
Chlorine, %	58.5 - 60.6	093

RESTRICTED INFORMATION

Hydrolysis Stability, ppm Cl	300 Max.	092
------------------------------	----------	-----

DSW 303425

(R) Routine Analysis, All others Analyzed by request Only

WATER_PCB-SD0000050976

FINISHED PRODUCT
SPECIFICATION

PRODUCT (Trade name)

Biphenyl

TYPE

Technical

PRODUCT (Chemical name)

Biphenyl

TYPE

Technical

CHEMICAL FORMULA

 $C_{12}H_{10}$

Issued 5/5/61
Anniston Plant
By W. W. Klemmer
Revised to add typ
values 11-18-65

MOL WT.

154.20

USE

Sales

SUPERSEDES SPECS OF

7/8/65

SPECIAL JOB NO.

SAMPLE FOR ANALYSIS

1 x 16 oz. can

CharacteristicsLimitsTypical ValuesRoutine tests

Appearance

White to pale
yellow crystals

White crystals

Crystallizing Point, °C

68.7 Min.

68.8

Color, Gardner

6 Max.

1.5

General Test (determined only on request)

Biphenyl, %

99.9

Naphthalene, %

0.001

3 - Methylbiphenyl, %

0.05

4 - Methylbiphenyl, %

0.02

Unknown, %

<0.03

Distillation Range; °C
1st drop - Dry PointMust include
255°C

DSW 303426

This specification for internal use only

Monsanto

ORGANIC CHEMICALS DIVISION

FINISHED PRODUCT SPECIFICATION

PRODUCT

Santowax C

DEPT.

32

PLANT

Anniston

PRODUCT CODE

829.10

USE

Internal

DATE EFFECTIVE

3-9-66

SALES CODE

7795-050-99-009

APPROVALS

PRODUCTION	RESEARCH
J. G. Moody /s/ 2-1-66	L. R. Stark /s/ 2-3-66 - M. McEwen /s/ 2-22-66
LES	QUALITY CONTROL (Chief Chemist)
J. E. Ault /s/ 2-22-66	O. E. Dolin /s/ 2-1-66
GENERAL DESCRIPTION	CHEMICAL FORMULA

ixture

PL WT.

PERSEDES SPECS. OF	SAMPLE FOR ANALYSIS	ISSUED BY
-23-57	2 - 1 Pint cans	J. A. Liddle 1-12-66
CHARACTERISTIC	LIMITS	METHOD

RESTRICTED INFORMATION

Appearance	Brown, waxy solid	G-1
Boiling Point, °C	350 Min.	
Ortho Terphenyl, %	**10	064
Meta Terphenyl, %	**50	064
Para Terphenyl, %	**26	064
Quaterphenyl, %	**14	072
Biphenyl, %	**0.1	064
Flash Point, °C	**375	114
Chlorine, ppm	2.0 (Typical)	
Sulfur, ppm	2.0 "	

Distribution of isomers and quaterphenyls vary with operating condition.

*Only limited data is available; therefore, the limits given are typical values of these characteristics.

This product is mainly used as a raw material for other finished products produced at Anniston.

This specification is for internal use only.

DSW 303427

Monsanto
ORGANIC CHEMICALS DIVISION

FINISHED PRODUCT
SPECIFICATION

PRODUCT Santowax R		DEPT. 32
PLANT Anniston	PRODUCT CODE 826.02	USE Sales
DATE EFFECTIVE 3-2-66	SALES CODE 7795-000-11-009	

APPROVALS

PRODUCTION R. G. Moody /s/ 1-31-66	RESEARCH L. R. Stark /s/ 2-3-66 - M. McEwen /s/ 2-22-66
SALES W. E. Ault /s/ 2-22-66	QUALITY CONTROL (Chief Chemist) O. E. Dolin /s/ 1-25-66
GENERAL DESCRIPTION	CHEMICAL FORMULA

Mixed Terphenyls

MOL WT. SUPERSEDES SPECS. OF 2-12-62	SAMPLE FOR ANALYSIS 2 x 1 pint cans	ISSUED BY J. A. Liddle 1-11-66
CHARACTERISTIC	LIMITS	METHOD

UNRESTRICTED INFORMATION

(R) Appearance	Yellow white noncrystalline flaked solid	G-1
(R) Color (liquid), NPA	4.0 Max.	128
(R) Upper Hold Point, °C	136 - 145	129
(R) Lower Hold, °C	48 - 63	129

RESTRICTED INFORMATION

Distillation Range, °C, 80% (Uncorrected)	350 - 400
Ortho terphenyl, %	10.0 - Typical Value
Meta terphenyl, %	54.0 - " "
Para terphenyl, %	27.0 - " "
Quaterphenyls, %	9.0 - " "

DSW 303428

(R) Routine Analysis, All others Analyzed by request Only

Monsanto

ORGANIC CHEMICALS DIVISION

RAW MATERIAL SPECIFICATION

MATERIAL CODE	MATERIAL
826.38	Chlorine Gas
PLANT	USING DEPTS.
Anniston	04, 05, 27
DATE EFFECTIVE	SUPPLIER
8-31-66	See below

APPROVALS

PRODUCTION	RESEARCH
R. G. Moody /s/R. G. Moody - 8/11/66	L. R. Stark /s/L. R. Stark - 8/29/66
PROCUREMENT	QUALITY CONTROL (Chief Chemist)
H. L. Phillips /s/H. L. Phillips - 8/15/66	O. E. Dolin /s/O. E. Dolin - 8/10/66
GENERAL DESCRIPTION	CHEMICAL FORMULA

Greenish yellow gas

Cl_2
Mol. wt. 70.91

SUPERSEDES SPECS OF	SAMPLE FOR ANALYSIS	ISSUED BY
Sept. 11, 1957	GLC sampling container	J. A. Liddle - 5/16/66

CHARACTERISTIC	LIMITS	Typical Values	METHOD
----------------	--------	-------------------	--------

UNRESTRICTED INFORMATION

Material Supplied by Diamond Alkali Company

Assay, (Vol. %)	99.4 Min.	113
Residual Gases (Vol. %)	0.5 Max.	113
Moisture, ppm	40 Max.	
Non-volatile Matter, ppm	40 Max.	

Material Supplied by Depts. 56, 57 (Anniston)

Assay (Vol. %)	85.5 - 93.5	113
H ₂ (Vol. %)	1.5 - 2.5 (3 Max.)	113
CO ₂ (Vol. %)	0.5 - 1.0	113
CO (Vol. %)	Nil	113
N ₂ (Vol. %)	3.0 - 8.0	113
O ₂ (Vol. %)	1.2 - 3.0	113

DSW 303429

WATER_PCB-SD0000050980

Monsanto

ORGANIC CHEMICALS DIVISION

RAW MATERIAL SPECIFICATION

MATERIAL CODE

38800

PLANT

Anniston

DATE EFFECTIVE

7-20-66

MATERIAL

Lime Hydrate

USING DEPTS.

27, 28

SUPPLIER

Alabaster Lime Co.

APPROVALS

PRODUCTION

R. G. Moody /s/R. G. Moody 6/9/66

PROCUREMENT

H. L. Phillips /s/H. L. Phillips 6/10/66

RESEARCH

N. F. Mueller /s/N. F. Mueller 7/13/66

QUALITY CONTROL (Chief Chemist)

O. E. Dolin /s/O. E. Dolin 6/8/66

GENERAL DESCRIPTION

White fine powder

CHEMICAL FORMULA

$\text{Ca}(\text{OH})_2$

Mol. wt. = 74.10

SUPERSEDES SPECS OF

Sept. 11, 1957

SAMPLE FOR ANALYSIS

2 - 1 lb. cans

ISSUED BY

J. A. Liddle - 5/18/66

CHARACTERISTIC

ANALYSIS

METHOD

Typical Analysis

UNRESTRICTED INFORMATION

CaO (%)	73.00
MgO (%)	1.20
SiO ₂ (%)	0.36
Al ₂ O ₃ (%)	0.60
Fe ₂ O ₃ (%)	0.30
Ignition Loss (%)	24.04
Ca(OH) ₂ (% Calculated)	97.50(1)

Screen Analysis

Through 200 mesh (%)	99.8	120
Through 325 mesh (%)	94.5	

Basis of Specifications: Suppliers typical analysis of product (5/23/66)

$$(1) \text{ Ca}(\text{OH})_2 \% = 100\% - [\text{MgO}(\%) + \text{SiO}_2(\%) + \text{Al}_2\text{O}_3(\%) + \text{Fe}_2\text{O}_3(\%)]$$

DSW 303430

Monsanto

ORGANIC CHEMICALS DIVISION

RAW MATERIAL SPECIFICATION

MATERIAL CODE	MATERIAL
19400	Attapulgus Earth - RVM Grade
PLANT	USING DEPTS.
Anniston	27-L
DATE EFFECTIVE	SUPPLIER
7-20-66	Minerals & Chemicals Philipp Corp.

APPROVALS

PRODUCTION	RESEARCH
R. G. Moody /s/R. G. Moody 6/9/66	N. F. Mueller /s/N. F. Mueller 7/13/66
PROCUREMENT	QUALITY CONTROL (Chief Chemist)
H. L. Phillips /s/H. L. Phillips 6/10/66	O. E. Dolin /s/O. E. Dolin 6/8/66
GENERAL DESCRIPTION	CHEMICAL FORMULA

Impure Hydrous Magnesium Aluminus Silicate

SUPERSEDES SPECS OF	SAMPLE FOR ANALYSIS	ISSUED BY
Sept. 9, 1957	1 lb. can	J. A. Liddle - 5/25/66
CHARACTERISTIC	LIMITS	METHOD

UNRESTRICTED INFORMATION

Screen Analysis 95 Min.
Through 200 Mesh (%)

Loss on Ignition 15 Max.
(% @ 950°C)

Typical Analysis from Supplier (May 23, 1966)

SiO ₂ (%)	67.5
Al ₂ O ₃ (%)	12.8
Fe ₂ O ₃ (%)	4.5
MgO (%)	11.1
CaO (%)	1.6
K ₂ O (%)	0.8
Na ₂ O (%)	0.1
TiO ₂ (%)	0.6
Other (%)	1.0

DSW 303431

MINERAL CHEMICALS
W. G. KRUMHOLTZ
RAW MATERIAL
SPECIFICATION

PRODUCT CODE	13599
REVISION HISTORY	
APPLICABLE	

ISSUED Feb. 26, 1951
STANDARDS DEPARTMENT
W. G. Krumholz
W. M. Battle

MATERIAL
FERRIC CHLORIDE ANHYDRIDE
 CHEMICAL FORMULA



NET WT.
 162.21

REVISIONS SPEC OF	REVISIONS FOR ANALYSIS
New	Issued W. M. Battle

UNCLASSIFIED INFORMATION

General Tests (made only on request)

<u>Property</u>	<u>Specification</u>	<u>J. F. Q. Method No.</u>
Appearance	Black powder or granules	16-A
Assay	97.0%, Min.	22-579
Sulfates	0.05%, Max.	120-570
Ferrous Salts	Trace, Max.	8-570-1
Solubility:		112-570
10:10 of water	Complete to slightly turbid, Max.	
10:10 of Formula 30 alcohol	Complete to slightly turbid, Max.	

DSW 303432

STANDARD MANUFACTURING PROCESS
FOR
AROCOR

APPENDIX C - PHYSICAL CONSTANTS OF FINISHED GOODS, INTERMEDIATES,
AND MAJOR RAW MATERIALS

FINISHED GOODS

Aroclor 1221 - Not Available
Aroclor 1232 - Not Available
Aroclor 1242
Aroclor 1248
Aroclor 1254
Aroclor 1260
Aroclor 1262
Aroclor 1268 - Not Available
Aroclor 2565 - Not Available
Aroclor 4465 - Not Available
Aroclor 5442 - Not Available
Aroclor 5460 - Not Available

RAW MATERIALS

Biphenyl
Santowax - Not Available
Santowax R-Not Available
Chlorine
Hydrated Lime
Attapulgu Earth-Not Avail.
Ferric Chloride

DSW 303437

WATER_PCB-SD0000050988

STANDARD MANUFACTURING PROCESS
FOR
AROCLOL

APPENDIX D - UTILITIES

The following is a list of utilities supplied to the Aroclor Department:

Steam

Steam is supplied to the department through a 6" main at 120 PSIG. Standard practice on steam is 137.5 lbs/cwt. Liquid Aroclors, 239.5 lbs/cwt. Solid Aroclors.

Recovered Water

Recovered water is supplied through a 6" main from a cooling tower. Standard practice is 625 gals/cwt. Liquid Aroclors, 1,197 gals/cwt. Solid Aroclors.

Electricity

Electrical power is supplied at 440 volts for power and 110 volts for lighting. Standard practice is 3.25 KWH/cwt. Liquid Aroclors, 7.19 KWH/cwt. Solid Aroclors.

Natural Gas

Natural gas is supplied through a 3" main at 40 psig. Standard practice is 0.1625 MCF/cwt. Liquid Aroclors, 0.5389 MCF/cwt. Solid Aroclors.

DSW 303438

STANDARD MANUFACTURING PROCESS
FOR
AROCLORS

APPENDIX E - PERSONNEL

The Aroclor operations require one Chief Operator, Classification A-2, and four Operators, Classification A-4, per shift. The chief operator is in charge of all Aroclor operations during his shift. He is also in charge of the Biphenyl group of products. A relief chief operator relieves a regular chief operator one shift per week, and an operator four shifts per week. He also works for the regular chief operator during vacations, sickness, and other absences. Each operator is assigned a specific job and is responsible for the operation of his job. There are ten (10) laborers, Classification A-8, employed in the Aroclor-Biphenyl group of departments. Each shift has a minimum of two laborers while some shifts have three laborers. Approximately 60% of all laborers' time is spent in Aroclor operations. The laborers' duties consists of filling drums, bags, crushing, screening, and milling of various products. They also clean tank cars.

The Aroclor Department is supervised by a Production Supervisor and a Production Foreman.

Attached is a Job Classification Schedule and Schedule of Hourly Wage Rates which also shows length of time required between automatic rate increases.

DSW 303439

WATER_PCB-SD0000050990

JOB CLASSIFICATION SCHEDULE AND SCHEDULE OF HOURLY WAGE RATES
AUTOMATIC RATE INCREASE GIVEN AT END OF:

Appendix A October 15, 1965 Department and Classification	Probationary Period		Max.										Job Shift Rate	
	Starting Rate	12 Weeks	24 Wks.	32 Wks.	40 Wks.	48 Wks.	60 Wks.	72 Wks.	108 Wks.	144 Wks.	180 Wks.	208 Wks.		Day Rate
A. <u>Production Department</u>														
2. Chief Operator	-----Filled by upgrading	--	--	--	--	--	--	--	--	3.12	3.17	3.22	3.22	7.5%
4. Operator	--	--	--	--	--	2.84	2.89	2.94	2.99	--	--	--	2.99	
7. Operator Trainee	2.48	2.61	2.66	2.72	2.77-(1)	--	--	--	--	--	--	--	2.77	
8. Operating Labor	2.40	2.46	2.49	2.52	--	--	--	--	--	--	--	--	2.52	

Day Rate Plus 7.5%

(1) May be employed at any of foregoing rates when having special qualifications.

STANDARD MANUFACTURING PROCESS
FOR
AROCLOL

APPENDIX F - PACKAGING AND SHIPPING

Solid Aroclors are packaged for shipment in 41 gallon Fibre drums, 40 gallon galvanized drums, 5 gallon galvanized cans, and in paper bags.

Liquid Aroclors are packaged in 55 gallon steel drums, and 5 gallon steel cans. These drums and cans are lined with one coat of A & W 4.A lacquer. Liquid Aroclors are loaded into tank cars and trucks. Tank cars used for electrical grade material are either aluminum or zinc and tin lined. All trucks are stainless steel. Occasionally non-electrical Aroclor is loaded into an iron car.

All tank cars and trucks are entered, cleaned, and inspected prior to loading.

DSW 303441

APPENDIX I
AROCOR FINISHED GOODS
CONTAINER WEIGHTS

<u>Material</u>	<u>Weights Lbs.</u>	<u>Container</u>
1221	50 or 520	5 Gal.-55 Gal. Steel Drum
1232	50 or 550	" " " "
1242	55 or 600	" " " "
1248	55 of 600	" " " "
1254	60 or 600	" " " "
1260	60 or 600	" " " "
1262	60 or 600	" " " "
1262 - 10% Toluene	60 or 600	" " " "
1262 - 10% Xylene	60 or 600	" " " "
1260 - 10% Toluene	60 or 600	" " " "
A13B3B	65 or 675	" " " "
5442	50 or 450	5 Gal.-40 Gal. Galv. Drum
4465	50 or 500	" " " "
2565	500	" " " "
Inerteen PPO	65, 360 or 674	5 Gal.-30 Gal.-55 Gal. Drums
Inerteen 54201-KA	674	" " "
Santicizer 794	600	" " "
MCS 295	635	" " "
5060	100	Paper Bag
5460	100	Paper Bag
1268	200	41 Gal. Fibre Drums

DSW 303442

WATER_PCB-SD0000050993

STANDARD MANUFACTURING PROCESS
FOR
AROCLORS

APPENDIX G - MATERIAL BALANCE

Standard and Theoretical usages of raw materials and hydrogen chlorine produced for each Aroclor is shown in Figure 3.

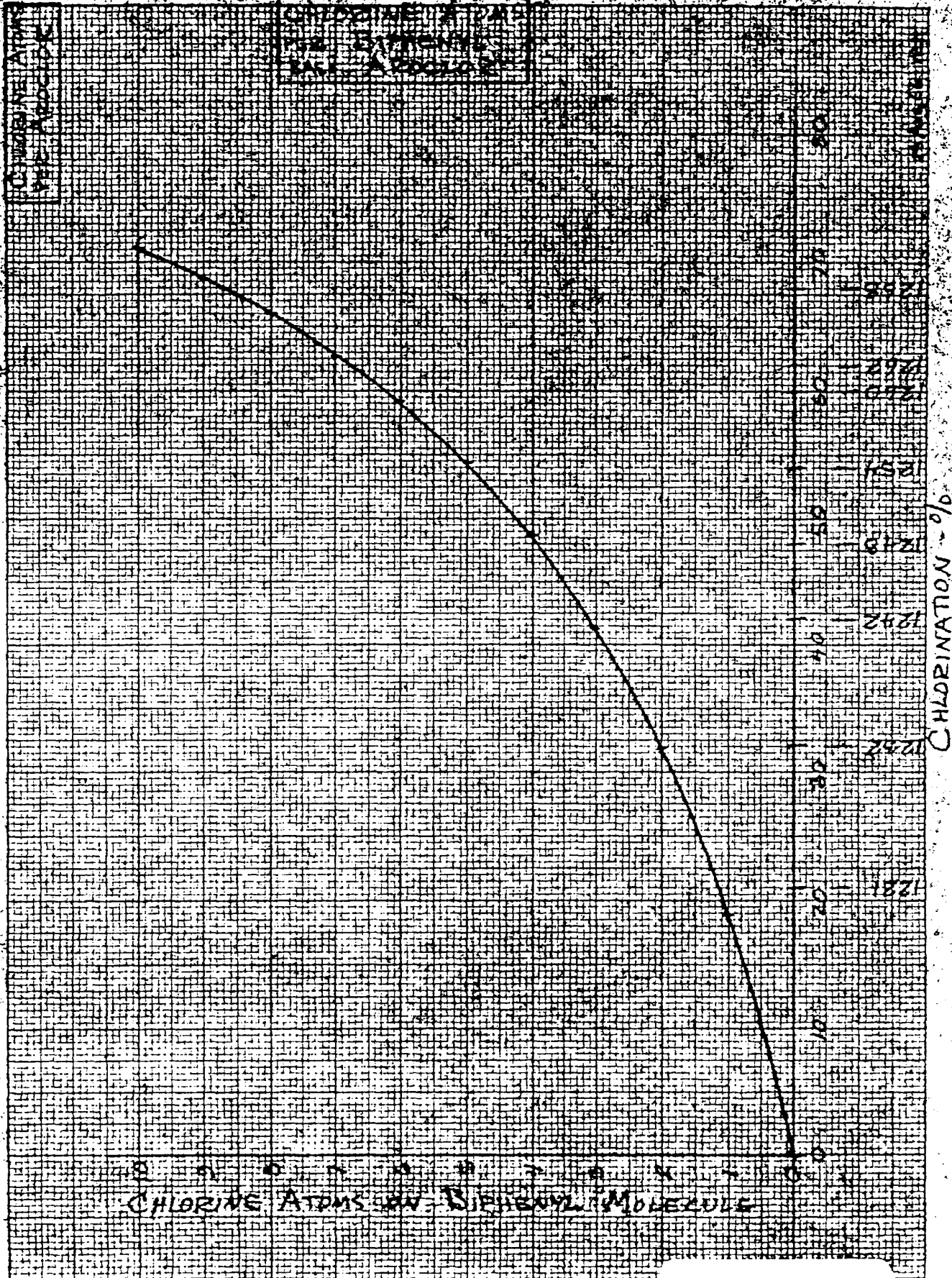
Figures 1 and 2 show the relationship between average number of chlorine atoms and per cent chlorine for Biphenyl and Santowax base Aroclors.

Figure 4 gives molecular weights of the various Biphenyl base Aroclors.

DSW 303443

FIGURE No. 1

CHLORINE ATOMS
IN BIPHENYL
AND POLYCHLORINATED
BIPHENYL MOLECULES

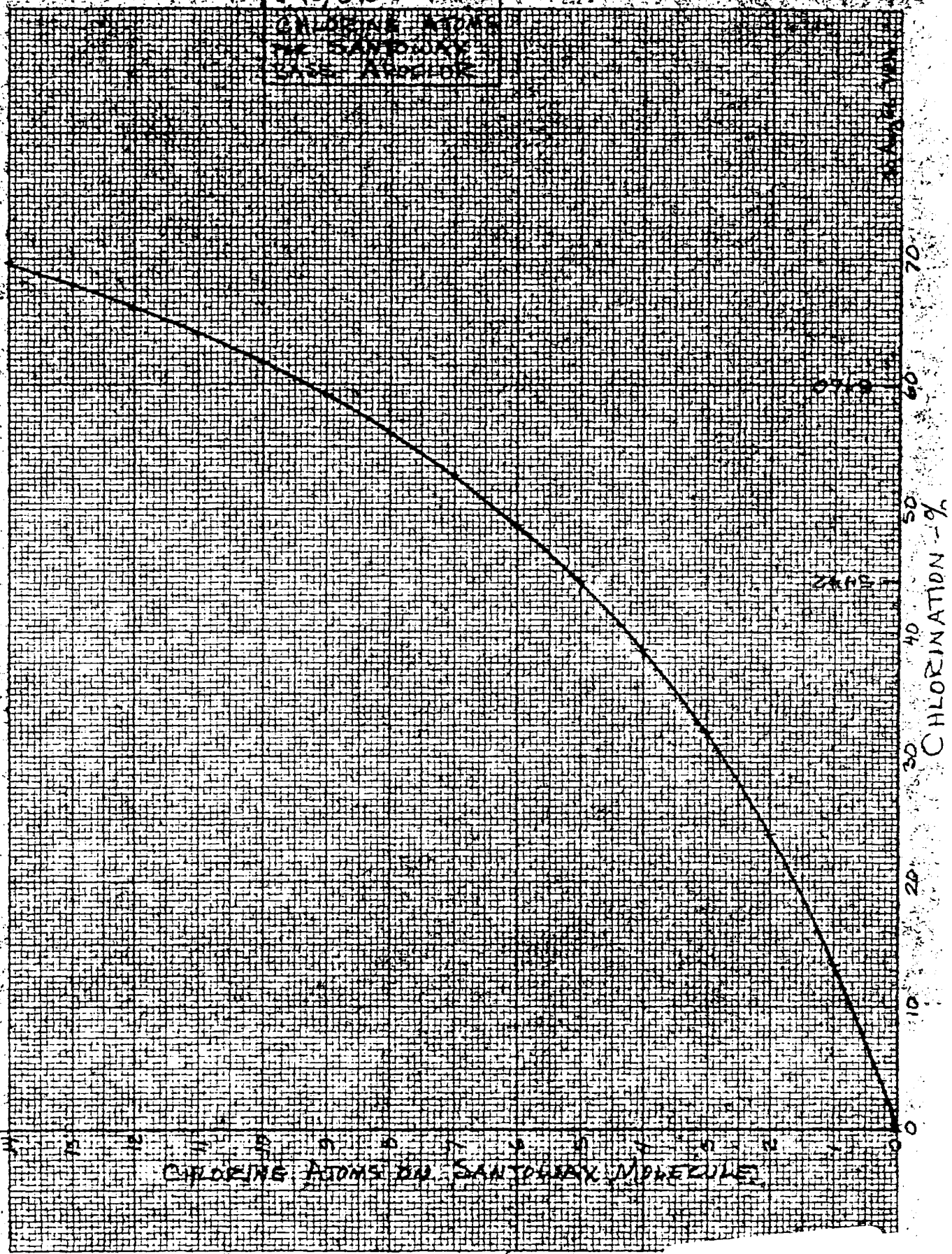


DSW 303444

WATER_PCB-SD0000050995

FIGURE No 2

CHLORINE ATOMS
ON SANTOXIN
MOLECULE

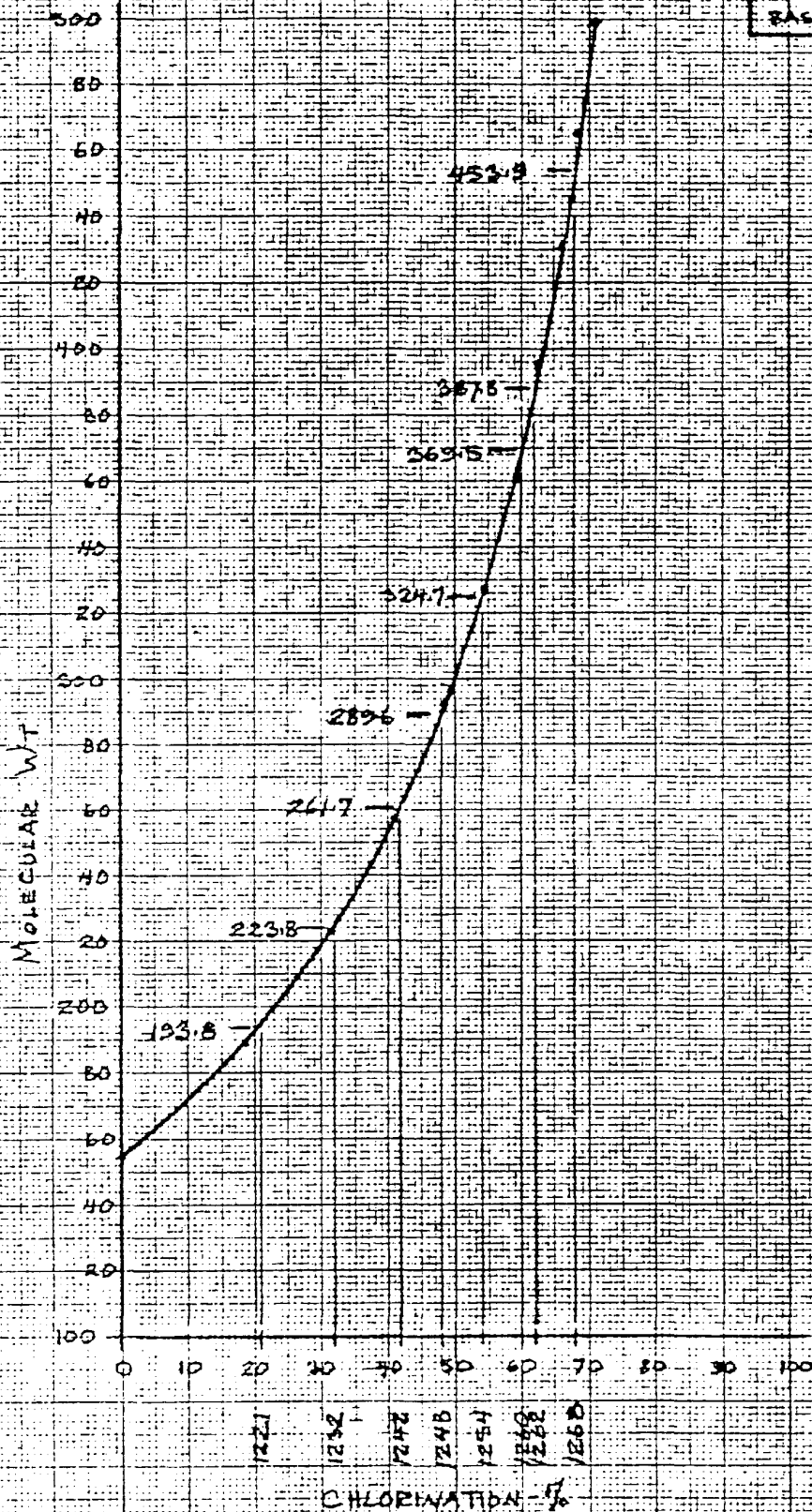


CHLORINE ATOMS ON SANTOXIN MOLECULE

FIGURE NO. 3
Theory and Standard Usage for Aroclors
 Lbs. per CWT Aroclor

Aroclor	Biphenyl		Santowax R		Santowax C		Chlorine		Hyd. Chloride	
	Theor.	Std.	Theor.	Std.	Theor.	Std.	Theor.	Std.	Theor.	Std.
1221	79.60	82.06	--	--	--	--	42.00	45.16	21.59	18.35
1242	59.19	61.02	--	--	--	--	84.00	90.32	32.90	36.70
1248	53.37	55.02	--	--	--	--	96.00	103.23	43.18	41.94
1254	47.53	49.00	--	--	--	--	108.00	116.13	49.34	47.19
1260	41.71	43.00	--	--	--	--	120.00	129.03	55.51	52.43
1262	39.76	40.99	--	--	--	--	124.00	133.33	61.68	54.18
1268	33.93	39.68	--	--	--	--	136.00	163.86	69.90	59.42
2565	27.63	28.19	--	--	9.21	9.40	130.00	132.65	66.82	56.80
4465	22.10	25.86	14.74	17.24	--	--	130.00	156.63	66.82	56.80
5442	--	--	59.19	69.23	--	--	84.00	101.20	43.18	36.70
5060	--	--	41.71	42.55	--	--	120.00	122.45	61.68	52.43
5460	--	--	41.71	48.78	--	--	120.00	144.58	61.68	52.43

FIGURE No 4
MOLECULAR WT
OF BIPHENYL
BASE AROCLORS



30 Aug 66 VEA

STANDARD MANUFACTURING PROCESS
FOR
AROCOLOR

APPENDIX H - ATMOSPHERIC & STREAM DISCHARGE & WASTE DISPOSAL

The following is an estimate of the atmospheric and stream discharge from the Aroclor operations at the maximum rate of 4,027,000 lbs/month.

	<u>Type</u>	<u>Amount</u>
Chlorine	Atmospheric	Negligible
HCL (Gas)	Atmospheric	Negligible
Montars	Plant Dump	Approximately 10% of Total Aroclor Production

DSW 303448

STANDARD MANUFACTURING PROCESS
FOR
AROCLOR

APPENDIX I - PROCESS CONTROL PROCEDURES

Control samples are taken from the chlorinator circulating line at intervals throughout the chlorination. The samples are checked, by the operator, for specific gravity, softening point or hold point dependent upon the Aroclor being produced. Temperature readings are taken every hour and recorded on data sheet.

A representative sample is taken of all distilled Aroclors and sent to the Analytical Laboratory for control tests prior to drumming, bagging, or loading into tank cars or trucks. The following control tests are performed by the Laboratory:

<u>Test (Liquid Aroclors)</u>	<u>Anniston Analytical Method No.</u>
Specific Gravity	052
Moisture	047
Resistivity	122

<u>Test (Solid Aroclors)</u>	<u>Anniston Analytical Method No.</u>
Softening Point	115
Hold Point	119
Color	128
Solubility	117

A complete analysis is run by the laboratory on all final lot samples.

DSW 303449

STANDARD MANUFACTURING PROCESS
FOR
AROCOLOR

APPENDIX J - SAFETY & TOXICITY DATA AND ACCIDENT REPORTS

Safety of operations in the Aroclor Plant depends on the individual operator. Materials being handled and operations must be observed closely to guard against the following:

Cause and Action to Prevent

1. Chlorine Gas

Chlorine gas is a very toxic corrosive gas. Leaks or high pressure can occur to liberate it.

Action to Prevent

- a. Report Chlorine gas exposure immediately for treatment with Antidote. Milk may be used in mild cases to relieve throat irritation. Oxygen from the resuscitator in the Dispensary is available.
- b. Keep gas mask in good condition. Keep a good canister in the mask. Use these type masks only in chlorine concentrations to 2-3% chlorine in air.
- c. Use a Scott Air Pak in concentrations of chlorine gas above 3% in air. Three units are available in the Aroclor-Biphenyl Department.
- d. If caught in chlorine gas without a mask, hold your breath and go cross wind until you are out of range of the gas. Place a handkerchief over your nose in small quantities of chlorine.
- e. Report all chlorine leaks for repair immediately. Make temporary repairs with tape if possible.
- f. In case of a major chlorine leak, the source of chlorine gas must be stopped.

Notify the Chlorine Plant immediately if the pressure on the chlorine manifold exceeds 30 psig. Cut off liquid chlorine from the car.

Cause and Action to Prevent

2. Thermal Burns

Contact with hot uninsulated product lines at the chlorinators and still areas.

Contact with uninsulated steam lines and open steam.

Contact with hot Aroclors from broken lines or leaks.

Contact with open flames at stills or from heating lines in the department.

Action to Prevent

- a. Know that contact with equipment and product will cause a thermal burn due to the high temperatures. Make changes in operations with this hazard in mind.
- b. Know the first aid treatment for burns. Cold Water Treatment attached.
- c. Wear gloves and long sleeve shirts with the sleeves rolled down when operating. Gloves and clothing can make the difference between first and second degree burns.
- d. Report uninsulated lines for repair of insulation.
- e. Use only steam lines with hose and hose clamps in good condition.
- f. The following operations are hazardous and should be done with care:
 - (1) Catching samples at the Aroclor chlorinator circulating pumps.
 - (2) Sampling the Solid Aroclor still receiver.
 - (3) Drawing bottoms on the Aroclor stills.

DSW 303451

Cause and Action to Prevent

3. Explosion

An explosion could occur in the chlorinators by slugging liquid chlorine to the distributor. Slugs of liquid chlorine could react fast enough to cause excessive pressure in the vessel. A rupture of the chlorinator could occur.

Action to Prevent

- a. Observe operation of the chlorine vaporizer and do not let liquid chlorine get into the rotameters. Pipe lines will be frosted.
- b. In case chlorine feed lines are frosted, close liquid chlorine feed valve to the vaporizer. Check steam addition to the vaporizer. Do not start up until source of trouble has been located.
- c. Observe temperature of the chlorinator periodically and be sure it is not too hot (180°C. maximum).

4. Falls

Slipping on cement floors, stairs, Aroclor on floor, ice conditions; tripping over objects (hose, pipe, valve handles, etc.); or losing footing on ladders, etc., can cause falls.

Action to Prevent

- a. The majority of minor, serious, and near-miss injuries in the Aroclor Dept. are the result of falls. Prevention is always within the ability of the operator by use of expert operating practices.

Cause and Action to Prevent

4. Falls (Contd.)

b. The following items should be done:

- (1) Return hose to rack after using.
- (2) Keep operating areas clean of all foreign objects.
- (3) Return ladders to racks when not in use.
- (4) Keep gloves on and hands out of pockets, especially going up stairs.
- (5) Place guards around any area that presents a hazard that should be guarded.

c. Poor lighting due to burned out bulbs should be corrected at once. Poor lighting due to inadequate facilities should be reported.

d. Spills must be cleaned up immediately following their occurrence. Report any spill that cannot be cleaned up within the shift to the Chief Operator. Spills are to be reported in the log book.

5. Fires

Fires can occur by:

Biphenyl, Santowax R, or Santowax C ignition.

Chlorine problems in the vaporizer allowing reaction with steam to release hydrogen.

Furnace burner backfire.

Wood, rags, or paper on hot lines.

Sparks from welding apparatus of open flames.

DSW 303453

Cause and Action to Prevent

5. Fires (Contd.)

Action to Prevent

- a. Area must be clear of trash or flammable materials. Use trash skips and cans.
- b. Smoke only in areas so designated.
- c. Use spark producing equipment only with a fire permit and proper inspection.
- d. Use care in lighting gas torch or the still furnaces.
- e. Fire fighting equipment includes:
 - (1) Two wheeled dry powder units.
 - (2) Hose house with 200 ft. of hose.
 - (3) Two dry powder hand extinguishers, One CO₂ hand extinguisher.
 - (4) Two 3/4" water hoses.
 - (5) Hose connection north of Santowax.
 - (6) Scott Air Paks.
 - (7) Fire truck in fire house.
- f. Observe visitors in the plant and be sure they observe safety rules.

DSW 303454

Cause and Action to Prevent

6. The following is a list of chemicals handled in the Aroclor Plant operations and the precautions to be taken during their handling:

a. Chlorine Gas

This is a corrosive gas which is toxic. Leaks or back pressure can occur and liberate chlorine gas in the area. Operators should have their cartridge type respirators handy at all times while in the area. Use of gas masks is explained in Section 1. Chlorine Gas.

If you are caught without a gas mask, and large amounts of chlorine gas are liberated, hold your breath and go cross wind until you are out or range of the gas. Once out of the area, obtain an emergency gas mask and return to the area and correct the trouble.

A victim of chlorine gas fumes must be removed as quickly as possible from the gaseous area. Place patient on his back with head and back elevated. Milk may be used in mild cases as a relief from throat irritation. If the patient has stopped breathing, start artificial respiration immediately.

b. Biphenyl

Biphenyl is a flammable material which will burn. Burning should not be done in the presence of Biphenyl or Biphenyl vapors. The melting point is 68.7°C., and the auto ignition temperature is 258°C. Inhalation of Biphenyl fumes is not recommended, since it can cause a drugged effect on the person.

c. Santowax R

Santowax R is a flammable material which will burn. It is relatively non-toxic to human beings. However, the vapors of Santowax R are irritating to the mucous membranes and bronichial tract. Prolonged breathing of the vapors should be avoided. The melting point is 140°C.

DSW 303455

Cause and Action to Prevent

6. (Continued)

- | | |
|-----------------------------|---|
| d. Santowax C | Same as Santowax R, since it is the crude material prior to distillation. |
| e. Ferric Chloride | Ferric Chloride is a black-brown solid. It causes only slightly irritating effects on the skin or mucous membranes. When heated it will decompose and release toxic fumes of Chlorine. It will react with water to produce toxic and corrosive HCl fumes. |
| f. Attapulugus Earth | This material is a dried earth. No danger in handling. Wash off with water to clean. |
| f. Hydrated Lime | No particular hazard in handling this material. The dust is irritating to the eyes and skin, and should be washed off with water. |
| h. Crude Aroclors | Crude Aroclors are hot, since all are handled in steam traced or jacketed lines. They can easily cause thermal burns in contact with the skin. Prior to air blowing, the material has HCl fumes entrained in it which are irritating to the skin or mucous membranes. |
| i. Hydrogen Chlorine Gas | This is a corrosive gas which is toxic. It is irritating to the mucous membranes, to the eyes, and to the respiratory tract. It will react with water to form Hydrochloric Acid. |
| j. Still Bottoms or Residue | Fumes from still bottoms are toxic. They are extremely irritating to the mucous membranes, to the eyes, and respiratory tract. |

DSW 303456

STANDARD MANUFACTURING PROCESS
FOR
AROCLORS

APPENDIX K - BACKGROUND INFORMATION

History of the Process in Monsanto

In the years 1927 - 1928, the Swann Chemical Company developed a process for making diphenyl, but as the expected demand for it did not continue, other outlets were sought for the diphenyl. Aroclors were produced in the laboratory in 1928, and as they showed good electrical properties, fire resistance, and general inertness, the General Electric Company became interested, and developed the use of Aroclors in transformers and capacitors. Biphenyl production began on February 28, 1928; Aroclor production (3,000 lbs/day) started on December 16, 1928.

In 1935 Monsanto acquired the Swann Company, so gaining access to their Aroclor patents. As the demand for Aroclors increased, additional production equipment was installed, located at the Monsanto Illinois factory (then Plant "E" and now called the Krummrich Plant). This equipment came on stream in September 1936.

The Krummrich plant site was chosen partly because of the availability of chlorine there, but when chlorine consumption overtook the production, and chlorine had to be purchased, proximity to the diphenyl production at Anniston had more weight, and additional capacity, largely designed on the plant "B" model, was installed at Anniston in 1941.

By 1946 both plants had doubled their capacity, to give a total production capacity approaching 2 million pounds a month. About 1954, chlorine cells were installed at Anniston.

The Newport plant was designed, mainly on Krummrich plant information, but to a less extent on Anniston information, under Newport Project No. 12 of the London Central Engineering Department.

Chlorination started at Newport in June 1951, Mr. Jack Clegorn from Anniston spending some time at Newport to assist in starting the diphenyl and Aroclor plants there.

The first production of Pyranol by MCC was in 1936, when Pyranol 1488 was made for the General Electric Company, the blending being done at first in rail tanks. Pyranol 1495 was made in July 1938.

Batch chlorinated Aroclors were made until November 1957 when continuous chlorinated Aroclors were first made in plant equipment.

DSW 303457

Several test runs were made and sold to customers for evaluation. On February 14, 1961, continuous chlorinated Aroclor 1242 was made on a full time basis. Today Aroclor is made by both batch and continuous chlorination of Biphenyl and/or Santowax.

In May 1965 an Aroclor expansion was completed to bring the capacity to 58 M pounds per year which includes 10 M pounds per year of Solid Aroclors.

Aroclor Uses

Monsanto Aroclors are used "as is" or in special formulations for the liquid coolant-insulation fluids in transformers and capacitors.

In capacitors, they provide for power factor correction in electric utility transmission lines. In fixed paper capacitors, Aroclors are the dielectric substance. These capacitors are used in home appliances such as air conditioners, furnaces, washers, and driers, in electric motors, and for ballast in fluorescent fixtures.

General Electric, Westinghouse, and other producers of electric capacitors use Monsanto Aroclors in special formulations for the dielectric substance.

Other important applications of both Liquid and Solid Aroclors in the electrical field are in wire and cable coatings and as impregnants for cotton and asbestos braided insulation. In addition, they are excellent dielectric sealants and are used to close the pores of carbon resistors and to seal electrical bushings and terminals.

Aroclors are outstanding for use as the heat transfer liquids in indirect heating systems. Aroclor systems can transfer closely controllable, uniform heat to chemical process vessels, food cookers, potato chip and frito fryers, drying ovens, to heat corrugating rolls, crimping presses, and calenders in material processing, to heat treating metals such as Titanium and Vanadium, and to efficiently indirect heat distillation equipment.

DSW 303458

It is interesting to note that as recently as 1941 only two methods of delivering process heat were known. Today non-pressurized, high temperature, fire resistant heat transfer systems based on Monsanto's Aroclor Liquids overcome practically all the deficiencies and drawbacks of Direct-Firing and Pressurized Vapor Heating.

Presently, there are numerous other uses for Liquid and Solid Aroclors and Aroclors in formulations. Only some of these uses are listed below:

As the actuating medium in bellows controls, thermostats, industrial temperature control regulators, and other kinds of automation equipment.

Liquid sealant for furnace roofs.

Vacuum diffusion pump oil-pumps used to pull high vacuum for metalizing plastics, dehydrating foods, and medicinals.

Dust entrapment-used to coat fibrous glass air filter pads, metal mesh, and other materials used for filtering air and gas streams.

As a vapor suppressant to increase the kill life of insecticides.

In waxes-waxes containing Aroclor are widely used in making dental castings, in the precision casting of aircraft parts, and for casting costume jewelry.

In specialized lubricants, adhesives, ink formulation, caulking compounds, plastics, rubbers, paints, varnishes, and lacquer formulations.

New uses for the various Aroclors are continually being developed. Continuous quality control, coupled with the unusual chemical and physical characteristics of the Aroclors, has resulted in widespread customer acceptance and a limitless expansion of uses for them.

STANDARD MANUFACTURING PROCESS
FOR
AROCLORS

APPENDIX K - BACKGROUND INFORMATION

The following is an up-to-date (June 1, 1966) bibliography
of Research reports on Aroclors:

- Ashworth, G. W., Keller R. E. Development of Analytical Methods - 1960
Part 9 - Evaluation of General Electric's Improved Thermal
Stability Test for Pyranols. Organic Research St. Louis
No. 3033, January 1964 (On Aroclor 1242).
- Ashworth G. W. et al Analytical Chemistry and Spectroscopy Investigations-
1962 Part 2 - Special Studies. Organic Research St. Louis
No. 3067 1/64 (On identification of Aroclors)
- Argo W. B., Knapp W. G., Gibbs M. E., Smull W. L. Heat Transfer Studies
in a Fifty Gallon Glass-lined Jacketed Reactor. Organic
Research St. Louis No. 2536 6/21/61 (On Aroclor 1248)
- Barkley L. B., Lippman A. E., Nufer H. L. O-Phenyl-O-14-Nitrophenyl
methylphosphonothioate. Ag. Research No. P2 6/15/61
(On Aroclor 1254 as a solvent)
- Bindbeutel A. J., Fowler L., Koerner W. E., Luebke H. W., Manno A. L.,
Trump W. N., Vogler C. E. Therminol Fluid Vapor Pressure
and Thermal Stability Instrument. Organic Research St. Louis
No. 2961 7/6/63 (Aroclors and Aroclor 1232)
- Brooks R. F., Koerner W. E., Spector M. B. IBM 704 Computer Program for
Therminol Fluid Heat Transfer Coefficients. Organic Research
St. Louis No. 2904 9/6/62 (On Aroclor 1242, 1248, 1254)
- Bucknell R. W., Campbell D. M. W. Building Manufacture of Santicizer 640.
Organic Research St. Louis No. 2440 2/8/60 (On Aroclor 1260)
- Close H. J., Knapp W. G. Aroclor HDK-1 Pilot Plant Preparation. Organic
Research St. Louis No. 3007 11/30/61 (On Aroclor HDK-1 and
Aroclor 1132)
- Cunningham E. P., Densmore D. W., Rinehart W. M. Evaluation of Chemical
Modifiers for Asphalts. Organic Research No. 2529 1/4/60
(On Aroclor 1232)

DSW 303460

- Darby J. R., Touchette N. W., Wangler L. R. Application Work on Present Plasticizers - 1959. Organic Research St. Louis No. 2488 3/21/60 (Aroclors mentioned)
- Ellenburg G. M. Assistance to Sales and Development Departments on Aroclors, Biphenyl and Related Products - 1954. Organic Research St. Louis No. 2420 1/8/60 (On Aroclor 1242, 1248, 1254, 1260, and 5460)
- Ellenburg A. M. 1960 Minor Plant Investigations - Suitability of Biphenyl from Union Carbide as Raw Material for Aroclor Production. Organic Research St. Louis No. 3133 9/29/64 (On Aroclor 1148 and 1248)
- Ellenburg A. M. New Dielectrics - 1959 and 1960. Organic Research St. Louis No. 2737 2/22/61 (On Aroclors)
- Ellenburg A. M., Busch W. A. High Dielectric Constant Aroclor. Organic Research St. Louis No. 2676 12/27/60 (On Aroclor 1242)
- Ellenburg A. M., Merten H. L. Exploratory Fluorination Studies of Biphenyl Derivatives. Organic Research St. Louis No. 2455 4/19/60 (On Aroclor 1260)
- Ellenburg A. M., McHugh K. L. Manufacture of Aroclor HDK-1. Organic Research St. Louis No. P1090 12/28/60 (On Aroclor HDK-1)
- Erickson F. B., Sakornbut S. S. Paper Transparentizers - A Literature Survey. Organic Research St. Louis No. 2486 4/4/60 (On Aroclor 1260)
- Fallwell W. F., Raether L. O. Final Report on Fire Retardents for Polyesters, Epoxies, Acrylics and Phenolics. Organic Research No. 2855 4/3/62 (On Aroclor 1260)
- Freeman R. C., Marco G. J., Speziale A. J. The Preparation of Radioactive CP 40294 (O-Phenyl-Cl₄-P-Nitrophenyl Methylphosphonothioate) Ag. Research No. 18 8/1/62 (On Aroclor 1242)
- Gaertner V. R., Schisla R. M. Chlorinated Diphenyl Ether Dielectrics. Central Research No. 1472 8/7/61 (On Aroclor 1242)

DSW 303461

- Gebura S. E. Glycidyl Ethers and Polyethers of Bisphenol-A, Tetra-chlorobisphenol-A and Hydrolyzed Aroclor 1262. Plastics Research Springfield No. 1413, 2/2/57 (On Aroclor 1262)
- Graham P. R., Darby J. R., Wangler L. R. Study of Plasticizers in Plastics Part II. Organic Research St. Louis No. 2648 11/14/60 (On Aroclor 1254)
- Hatton R. E., Cunningham E. P., Luebke H. W., Stark L. R. Functional Fluids Application Research. Organic Research St. Louis No. 2472 2/16/60 (On Aroclors)
- Hatton R. E., McEvan M., Sands T. P. Status Report on Synthetic Functional Fluids - 1950-66. Organic Development, Greve Coeur No. 1298 12/29/61 (On Aroclors)
- Knowles W. S., Smith T. C. Final Report on Preparation of Perfluorinated Compounds for Use as Functional Fluids Part I. Organic Research St. Louis No. 3176 3/25/65 (On Aroclor 1242 and 1268)
- Lancey R. C. Industrial Heat Transfer Fluids - A Market Study. Organic Development Greve Coeur No. 1266 3/10/61 (On Aroclor 1248)
- Lancey R. C. Utilization of By-product Aroclors from the HDK-1 Process. Organic Development Greve Coeur No. 1263 12/7/60 (On Aroclor 1242, 1248, 1254, 1260, HDK-1, 1221)
- McHugh K. L., Ellenburg A. M. Improved Aroclor 1242 Process Studies /Aroclor HDK-1/ Organic Research St. Louis No. 2755 3/6/61 (On Aroclor 1132, 1248, 1254, 1260, HDK-1)
- McHugh K. L., Sullivan J. D. MCS 177. Organic Research No. P1179 5/16/62 (On Aroclor 1134, 1234)
- McHugh K. L., Sullivan J. D. Tentative Process for MCS-177. Organic Research St. Louis No. P1234 3/12/63 (On Aroclor 1232)
- Mestdagh J. J. Flame Retardent Phenolic Phenolic Varnishes. Plastic Research Springfield No. 1581 10/14/60 (On Aroclor 5460)
- Parks R. M. Polymerite File Market Study - Special Report. Organic Development Greve Coeur No. 1277 7/26/61 (On Aroclors)

APPENDIX K

-4-

- Terpstra M. A. Hydrolyzed Aroclor 1268 /Alkaline - Methanol Hydrolysis/
Organic Research St. Louis No. 2585 9/7/60 (On Aroclor 1268)
- Ellenburg A. M. Assistance to the Aroclor Production. Phosphorous Research
SF 2084 Job No. 171-299 12/31/46
- Ellenburg A. M., Feltham J. E., Richards H.B., Jr. Aroclor 1142, Production,
Increase in Chlorination Rate. Phosphorous Research SF 2790
Job No. 171-1028 7/3/52
- Ellenburg A. M. Aroclor 1154, Removal of Hydrogen Chloride from Waste
Gases. Phosphorous Research An. SF 1978 Job No. 207
9/21/45
- Ellenburg A. M., Simone R. A. Aroclor 1142 and 1154 prepn. from biphenyl
by contin. chlorination. Organic Research SL Fin. 1530
Job No. 3040 6/4/56
- Metzner W. P., Becker T. H. Use of Standard Reactors for Making Aroclor
1160. Organic Research SL Fin. X-1491-IV Job No. 117-1491,
IV 4/21/42
- Ellenburg A. M., Cleveland T. H. Evaporation Loss of Aroclors. Phosphorous
Research An. SF 2234 Job No. 451 7/14/48
- Ellenburg A. M., Richards H. B., Sears J. K. Aroclor 1221 and 1242,
chlorination with DeNora CL-air mixt. Phosphorous Research
SF 2673 Job No. 171-790 12/10/51
- Durgin C. B., Tremontozzi Q. A. Specific Heats of Aroclors 1221 and 1232
Phosphorous Research SF 2083 1/17/47
- Ellenburg A. M., Forbus F. C. Aroclor 1232, Brominated, Prepn. Phosphorous
Research SF 2512 Job No. 171-628 9/13/50
- Ellenburg A. M., Good R. J., Wade W. W. The Effects of Prolonged Heat
on the Electrical Properties of Aroclors. Job No. 171-616
12/7/50
- Nellums R. O., Gruetzemacher E. C. Aroclors 1232 and 1248, as components
of hydraulic fluids. Organic Research SL Fin. 1753 Job No.
3388 2/14/57

DSW 303463

- Ellenburg A. M., Good R. J., Richards H. B., Jr. Aroclor 1236, Dielectric Properties Phosphorous Research SF 2612 Job No. 171-615 5/21/51
- Ellenburg A. M., Forbus F. C., Richards H. B., Jr. Aroclors 1236, 1238, and 1240 for General Electric Company. Phosphorous Research SF 2539 Job No. 171-609 12/11/50
- Ellenburg A. M., Klaus I. S. Aroclor 1242, as heat transfer medium, Properties. Organic Research SL Fin 2252 Job No. 3371 3/11/59
- Ellenburg A. M., Wade W. W. Aroclor 1242, Brominated, Prepn. Phosphorous Research SF 2419 Job No. 171-535 12/9/49
- Ellenburg A. M., Richards H. B., Jr. Containers for Aroclor 1242. Phosphorous Research SF 2523 Job No. 171-623 11/23/50
- Ellenburg A. M., Richards H. B., Jr. Containers for Aroclor 1242. Phosphorous Research SF 2772 Job No. 171-719
- Ellenburg A. M., Richards H. B., Jr. Aroclor 1242, 1254, and 1260, distillation at 20-30mm. Hg. Phosphorous Research SF 2694 Job No. 171-1028 5/15/52
- Ellenburg A. M. Aroclor 1242, Electric Properties, Under Vacuum storage at 75° and 100°C. Organic Research SL Fin. 1849 Job No. 2910 8/2/57
- Metzner W. P., Behr L. C., Tyler L. D. Aroclor 1242, Formulation, for use as potential hydraulic fluid. Organic Research SL Fin. X-1332 Job No. 117-1332 10/21/42
- Ruehrwein R. A., Pritchett T. R. Heat Content of Dowtherm A and the Aroclors. CR Sp. 687 - Job No. 4-8065, 4/18/52.
- Ellenburg A. M., Morgan J. F., Wade W. W. The Preparation of Aroclor 1244, 1242, 1248, 1254, and P1467 for the General Electric Company. Phosphorous Research SF 2362 Job No. 171-498 6/2/49
- Keller R. E., Ashworth G. W. Aroclor 1242, High Chloride content, analysis, Comparison of G. E., Westinghouse, and Monsanto Methods. Organic Research SL Fin. 2188 Job No. 3427 6/24/58

DSW 303464

- Ellenburg A. M., Good R. J., Wade W. W. Aroclor 1242 and 1254, Effect of u.v. light on Electric Properties. Phosphorous Research SF 2500 Job No. 171-623 6/23/50
- Ellenburg A. M. Aroclor 1242, 1248, and 1254, Density and Resistivity Studies. Organic Research SL Fin. 2420 Job No. 2681 1/8/60
- Gaertner V. R., Weingarten H. I. Dielectric Liquids: Formation, Composition and Improvement of Aroclor 1242 and 1232. RE. Res. Da. Int 1318 Job No. 9442 1/29/60
- Ellenburg A. M., Wade W. W. Aroclor-lithium stearate mixtures. Phosphorous Research SF 2496 Job No. 171-614 5/29/50
- Ellenburg A. M., Wade W. W. Aroclor 1242, Mixture with Santodex, as hydraulic fluid. Phosphorous Research SF 2497 Job No. 171-530 12/5/49
- Ellenburg A. M., Garrett L. C., Wade W. W. Aroclor 1242 and 1248 hydraulic fluid flammability tests for the Ford Motor Company. Phosphorous Research SF 2447 Job No. 171-614 2/4/50
- Keller R. E., Koerner W. E., Brackbill C. H., Cecil O. B., Livasy J., Williams W. D. Aroclor 1242, Mixts. with polymers, electrical properties of. Organic Research SL Fin. 2314 Job No. 3168 7/6/59
- Ward D. W. Physical properties of the aroclors 1242, 1248, 1254, and 1260. CR Research Sp 597 Job No. 4-3065 11/29/49
- Ellenburg A. M., Ashworth G. W., Merten H. L. Study of the Use of Ferric Chloride as Catalyst for the Production of Aroclor 1242. Organic Research SL Fin. 1842 Job No. 3368 7/12/57
- Ellenburg A. M., Good R. J., Morgan J. F., Richards H. B., Jr. Utilization of Airco process for Aroclor production. Phosphorous Research SF 2642 Job No. 171-781 9/28/51
- Ellenburg A. M., Wade W. W. Preliminary evaluation of anion exchange resins in the removal of inorganic chlorides from Aroclors. Phosphorous Research SF 2349 Job No. 171-535 4/21/49
- Cleveland T. H., Gemeninhardt P. G., Powers E. L., Sanguinetti P. A. Aroclor 1242, reaction medium for manufacturing of tolylene diisocyanate. Phosphorous Research Int. 2943 Job No. 171-1098 10/21/53

DSW 303465

APPENDIX K

-7-

- Cleveland T. H., Machwart G. M. Aroclor 1242, reaction medium for phenyl isocyanate production. Phosphorous Research SF 2816
Job No. 171-1013 11/25/52
- Saunders J. H., Gemeinhardt P. G., Liss R. L., Powers E. L. Aroclor 1242, reaction medium for tolylene diisocyanate.
Phosphorous Research SF 2699 Job No. 171-720 2/20/52
- Saunders J. H., Cleveland T. H. Aroclor 1242, reaction medium for tolylene diisocyanate. Phosphorous Research Fin. 2700
Job No. 171-743 2/22/52
- Ellenburg A. M., Simone R. A. Resistivity Improvement of Aroclor 1242.
Organic Research SL Fin. 2096 Job No. 3203 6/6/58
- Weddell D. S. Aroclor 1242, rubber swelling by. Phosphorous Research
Fin. 2082 Job No. 171-308 12/19/46
- Cleveland T. H., Davis B. R., Flores Hector, Herndon A. B., Saunders J. H. Aroclor 1242, solvent in prepn. of m-tolylene diisocyanate.
Phosphorous Research Fin. 2547 Job No. 171-603 12/22/50
- Ellenburg A. M., Merten H. Aroclor 1242 - Study of Unstable Product.
Organic Research SL Fin. 1640 Job No. 3269 10/22/56
- Ellenburg A. M., Good R. J., Richards H. B., Jr. Low Temperature dielectrics. Phosphorous Research Int. 2912 Job No. 171-1054 12/2/53
- Ellenburg A. M., Ashworth G. W., Merten H. L. 1. Aroclors (1242, 1254, 1260), chloride stability of, dependence on process variables. 2. Aroclor 1242, up-grading off-grade product. Organic Research SL Fin. 2072
Job No. 3428 5/4/58
- Chadwick D. H., Duncan J. M., Kaufman S., Koeple J. S., Llewellyn G. H., Reetz T. Aroclor 1242-tolylenediisocyanate, liquid-vapor equil. data. Organic Research An
Fin. 880 Job No. 171-1101, 117-2669, VII 2/26/54
- Ellenburg A. M. Aroclor 1248 as a Heat Transfer Medium in a Lamont Forced Circulation Heater. Organic Research SL
Fin. 1306 Job No. 2957 3/15/55

DSW 303466

- Ellenburg A. M., Richards H. B., Jr. Aroclor in gases. Phosphorous Research SF 2325 Job No. 171-1089 10/16/53
- Wallace P. P., Spring W. E. Insecticides plus Aroclors. Organic Research CC Fin. 965 Job No. 117-2542 4/14/54
- Howard R. E., Hulette V. H., Vandeven P. Design and testing of a portable Aroclor heating unit. Organic Research SL Fin. 238 Job No. 117-2060, II. 8/1/48
- Ellenburg A. M., Knight R. R. Treating contaminated Aroclor 1248. Phosphorous Research SF 2342 Job No. 171-451 3/25/49
- Ellenburg A. M., Forbus F. C. Aroclor 1248 and Santodex - hydraulic fluid. Phosphorous Research SF 2529 Job No. 171-614 9/7/50
- Ellenburg A. M., Morgan J. F. Pour point depressants for Aroclor 1248. Phosphorous Research SF 2396 Job No. 171-530 8/9/49
- Hatton R. E., Sullivan J. D. Functional Fluids - Application Research (1956-1957). Organic Research SL Fin. 2176 Job No. 3132 10/14/58
- Fordyce R. G. Aroclor 1254, as plasticizer for Cerex 226. CR Int. 397 Job No. 4004 10/15/45
- Ellenburg A. M., Good R. J. Aroclor 1254, comparison with Fenclor. Phosphorous Research SF 2887 Job No. 171-1075 6/11/53
- Ellenburg A. M., Good R. J. Aroclor 1254, comparison with Fenclor 54. Phosphorous Research SF 2527 Job No. 171-623 9/6/50
- Ellenburg A. M. Galvanized plate drums versus galvanized hot dipped drums for Aroclor 1254. Phosphorous Research SF 2055 Job No. 171-297 7/18/46
- Ellenburg A. M., Good R. J., Knight R. R., Wade W. W. Aroclor 1254 drum storage tests to check power factor change. Phosphorous Research SF 2421 Job No. 171-524 11/30/49
- Ellenburg A. M., Richards H. B., Jr. Effect of exposure to mild steel on electrical properties of Aroclor 1254. Phosphorous Research SF 2771 Job No. 171-718 5/28/52

DSW 303467

- Ellenburg A. M., Knight R. R. Complaint of high power factor of Aroclor 1254. Phosphorous Research SF 2355 Job No. 171-524 5/13/49
- Ellenburg A. M., Wade W. W. Corrosion inhibitor tests with Aroclor 1254. Phosphorous Research SF 2336 Job No. 171-530 3/17/49
- Ellenburg A. M., Richards H. B., Jr. Aroclor 1254, distillation at 20mm. Hg. Phosphorous Research SF 2689 Job NO. 171-1028 1/30/52
- Nellums R. O., Argo W. B. Aroclor 1254, heat transfer studies (Votator). Organic Research SL Fin. 1906 Job No. 2828 10/23/57
- Ash J. R., Daniloff K. B., Dahn W. M. Preparation and Testing of an Emulsion of Chlorinated Rubber and Aroclor 1254. West Research SF 912 Project No. 13012-49-1002 3/28/49
- Ellenburg A. M., Richards H. B., Jr. Phosphorous Research SF 2949 Job No. 171-1075 10/28/53
- Ellenburg A. M., Sanguinetti P. A. Aroclor 1254, mixt. with Aroclor 4465, as waterproofing agent for electrical cables. Phosphorous Research Fin. 2027 Job No. 171-206 3/8/46
- Ellenburg A. M., Richards H. B., Jr. Mixing Aroclor 1254 with secondary butyl acetate. Phosphorous Research SF 2788 Job No. 171-1029 6/25/52
- Ellenburg A. M., Sanguinetti P. A. Aroclor 1254, mixt. with dibenzothiophene sulfone, dielectric constant. Phosphorous Research SF 2020 Job No. 171-243 2/8/46
- Ellenburg A. M. Aroclor 1254, mixt. with trichlorobenzene, coefficient of expansion. Phosphorous Research SF 2964 Job No. 171-1075 1/18/54
- Ruehrwein R. A., White W. A. Studies on light stability of electrical resistivity of Aroclor 1254. CR Sp. 543 Job No. 4-8065 1/11/49
- Ellenburg A. M., Good R. J., Wade W. W. Power factor vs. resistivity of Aroclor 1254. Phosphorous Research SF 2458 Job No. 171-623 3/15/50

- Metzner W. P. Aroclor 1254, recovery from sample contaminated with mineral oil dielectric. Organic Research SL Fin. X-1334-II Job No. 117-1334, II 4/23/42
- Ellenburg A. M., Good R. J., Richards H. B., Jr. Line Material Company, Milwaukee, Wisconsin, complaint on Aroclor 1254 (Elemex). Phosphorous Research SF 2511 Job No. 171-614 9/12/50
- Hubbard H. L., Mitoray J. P. Aroclor 1260, color of, investign. Organic Research SL Fin. 1570 Job No. 2319 7/19/56
- Ellenburg A. M., Marten H. L. Aroclor 1260, fluorination of. Organic Research SL Fin 2455 Job No. 3199 4/19/60
- Ellenburg A. M., Knight R. R. Aroclor 1260, mixts. with tri-(2-ethyl-hexyl)- and trioctyl silicon fluorides, dielec, props. Phosphorous Research SF2376 Job No. 171-532 7/12/49
- Ellenburg A. M., Richards H. B., Jr. Aroclor investigation on difference between Anniston and Krummrich Aroclor 1260. Phosphorous Research SF 2881 Job No. 171-1088 5/6/53
- Hubbard H. L., Brooks R. F. Aroclor 1260, viscosity and Refr. Index studies, comparison between Anniston and W GK. Organic Research SL Fin. 2118 Job No. 2602 7/16/58
- Darby J. R., Cannon J. A. Aroclor 1262, as plasticizer for Saran, Evaln. Organic Research SL Fin. 2379 Job No. 3591, IV 10/26/59
- Drumm M. F., Gebura S. E. Aroclor 1262, hydrolyzed, glycidyl ethers of. Pl Res Spf Fin. 1413 Job No. 1845 2/2/57
- Nellums R. O., Terpstra M. A., Goss J. C., Lynch D. K. Pilot Plant Preparation of Hydrolyzed Aroclor No. 1262. Organic Research SL Fin. 1335 Job No. 3036 11/14/55
- Ellenburg A. M., Wade W. W. Aroclor 1268, hydrolysis to hexachlorodihydroxybiphenyl. Phosphorous Research SF 2317

APPENDIX K

-11-

- Sharp D. B., Dazzi J., Gabbert J. D., LeBlanc J. R. Hydrolysis of Aroclor 1268 in Methanolic Sodium Hydroxide. RE Research Da Fin. 1061 Job No. 9411 7/11/57
- Terpstra M. A. Hydrolyzed Aroclor 1268 (Alkaline-Methanol Hydrolysis). Organic Research SL Fin. 2585 Job No. 3502 9/7/60
- Knapp W. G., Cooper R. L., Vandeven P. Pilot Plant Preparation of Hydrolyzed Aroclor 1268. Organic Research SL Fin. 2334 Job No. 3899 8/17/59
- Nellums R. O., Terpstra M. A., Goss J. C., Lynch D. K. Pilot Plant Preparation of Hydrolyzed Aroclor No. 1268. Organic Research SL Fin. 1528 Job No. 3093 5/29/56
- Adams J. F., Knapp W. G., Matthews R. L., Vandeven P. Pilot Plant Preparation of Hydrolyzed Aroclor 1268. Organic Research SL Fin. 2124 Job No. 3665 7/17/58
- Gable C. L. Aroclor 1268, thiolysis product, as peptizing agent for rubber. Organic Research Ni SF 7649 Job No. 1550 7/8/57
- Jenkins R. L., Metzner W. P. Aroclor 1269, solubility tests in various organic solvents. Organic Research SL Fin. X-1249-I & II Job Nos. 117-1249, I 1/30/40 (Pt. 1) 3/20/40 (Pt. II)
- Metzner W. P., Dobratz E. H. Aroclor 1270 from B-221-o. Organic Research SL Fin. 104 Job No. 117-1533 2/20/46
- Lefforge J. W., Smith R. E. Impregnation of wood with Aroclor "4465" and "330". Phosphorous Research SF 2101 Job No. 171-398 3/17/47
- Bogaard T. D., Saunders J. H., Slocombe R. J. Aroclor 5442, as plasticizer in residue from distn. of TDT. Phosphorous Research SF 2403 Job No. 171-523 9/21/49
- Ellenburg A. M., Rueckert T. C. Muriatic Acid Tower Repairs. 5442. Phosphorous Research An SF 2140 Job No. 376 10/17/47
- Gittings L. D., Moose J. E. Aroclor 5460, adhesives from. Phosphorous Research Fin. 2054 Job No. 171-203, 171-298 7/11/46

DSW 303470

- Crenshaw R. L., Ellenburg A. M., Morgan J. F., Wade W. W. Investigation of precipitate formation in organic solutions containing Aroclor 5460. Phosphorous Research SF 2492 Job No. 171-623 5/22/50
- Ellenburg A. M., Richards H. B., Jr. Aroclor 5460 from Dow Tarophen D-25. Phosphorous Research SF 2883 Job No. 171-1077 5/12/53
- Ellenburg A. M., Wade W. W. Aroclor 5460, mixt. with Li stearate. Phosphorous Research SF 2496 Job No. 171-614 5/29/50
- Ellenburg A. M., Richards H. B., Jr. Aroclor 5460, soly, in mineral oil. Phosphorous Research SF 2949 Job No. 171-1075 10/28/53
- Ellenburg A. M. Solubility of Aroclor 5460. Phosphorous Research SF 2088 Job No. 171-397 2/6/47
- Ellenburg A. M., Morgan J. F., Wade W. W. The Solubility of Aroclor 5460 in Various Solvents. Phosphorous Research An SF 2272 Job No. 451 12/9/48
- Ellenburg A. M., Richards H. B., Jr. Preparation of a concentrated solution of Aroclor 5460 in xylene. Phosphorous Research SF 2960 Job No. 171-1075 10/28/53
- Ellenburg A. M. Investigation of decomposition in Aroclor still. Phosphorous Research SF 2046 Job No. 171-299 5/16/46
- Metzner W. P., Durland J. E. Aroclors, (1254 and 1260), odor improvement, study of. Organic Research SL Fin. X-1402-III Job No. 117-1402, III 12/5/40
- Darby J. R., Allen W. G., Cannon J. A., Cowell E. E., Touchette N. W., Wangler L. E. Aroclors 1262 and 1232, migration studies (1956). Organic Research SL Fin. 1730 Job No. 3129, III 1/22/57
- Terpstra M. A., Lynch D. K. Hydrolyzed Aroclors 1262 and 1268 Sample Preparation. Organic Research SL Fin. 1706 Job No. 3038 1/8/57

DSW 303471

- Bennett G. E., Sharp D. B., Dazzi Joachim. Exploratory Aroclor Chemistry. Part I. Ethers by Alcoholysis. Hydrolyzed Aroclor Derivatives. RE Research Da Int. 1181 Job Nos. 4-9261, 4-9310, 9378 8/1/58
- Chadwick D. H., Ellenburg A. M., Marshall W. W. Alkylation of Aroclors for Evaluation as Extender Plasticizers. Organic Research SL Fin. 1184 Job No. 2682 (SL) 171-1042 (An) 1/17/55
- Darby J. R., Cowell E. E. Application Work on Present Plasticizers 1953-54 - Flammability Studies with the SPI Tester Organic Research SL Fin. 1649 Job No. 2616, IV 11/20/56
- Nielsen M. L. Aroclors, as glowproofers. CR Fin 572 Job No. 5-8064 7/19/49
- Hatton R. E., Cunningham E. P., Luebke H. W., Stark L. E. Aroclors, as industrial fluid base stocks, evaln. Organic Research SL Fin. 2472 Job No. 3505, II 2/16/60
- Metzner W. P., Durland J. R., Powers E., Lyles J. M. General Electric Stability Test for Aroclors and trichlorobenzene. Organic Research SL Fin. X-1436 Job No. 117-1436 4/17/42
- Kyrides L. P., Craver J. K. Compatibilities of Aroclors with Resins. Organic Research SL Fin. X-1362 Job No. 117-1362 6/4/40
- Ellenburg A. M., Sears J. K. Exploratory Research Cyanogenation of Aroclors. Organic Research SL Fin. 1677 Job No. 2678 11/18/55
- Ruehrwein R. A., Kenyon A. S., White W. A. Degradation of Aroclors. RE CR Da Int. 833 Job No. 4-4042 7/29/54
- Ellenburg A. M., Richards H. B., Jr. Aroclor in Gases. Phosphorous Research Int. 2892 Job No. 171-1089 6/17/53

DSW 303472

- Ellenburg, A. M., R. A. Simone - Aroclors, dielectric properties, Organic Research SL Fin. 2305 - New Dielectrics - 1957 and 1958. Job No. 3316, 6/16/59.
- Sharp, D. B., Joachim Dazzi - Exploratory Aroclor Chemistry Part III. Glycidylation of Hydrolyzed Aroclors. Research RE Da Int 1183, Job No. 9411, 10/9/58.
- Hubbard, E. M., F. R. Short - Determination of the oxirane (o-epoxide) group by hydrochlorination in dioxane. RE CR Da Fin. 943. Job No. 9378, 1/30/56.
- Darby, J. R., R. M. Parks - Preliminary Evaluation of Plasticizers - 1954. Part I: The Intensive Evaluation of Secondary Plasticizers Including the Aroclors. Org Res SL Fin. 1281. Job No. 2821, I, 5/16/55.
- Darby, J. R., W. G. Allen, J. A. Cannon, A. Y. Coran, P. R. Graham, L. R. Wangler. Org Res SL Fin. 1666. Job No. 2892, II, 10/28/56.
- Craver, J. K., J. Dazzi, J. A. Herbig, J. R. LeBlanc, J. Schwendemann. Synthesis and Application Studies on Hydrolyzed Aroclors and Derivatives. RE Res Da Fin. 1115. Job Nos. 9411, 9411-03-3, 9416-03-3, 12/30/57.
- Sharp, D. B., Joachim Dazzi, J. E. Katon. Exploratory Alcoholysis and Hydrolysis of Chlorinated Aromatics. RE Res Da, SF 1175. Job No. 9378, 9/9/55.
- Sharp, D. B., Joachim Dazzi, Exploratory Aroclor Chemistry, Part II. Diphenols by Hydrolysis. RE Res Da, Int. 1182. Job Nos. 9310, 9378, 9411, 5/13/58.
- DeGarmo, O, W. S. Dempsey, G. F. Ludvik, W. E. Spring. Insecticide Special Problems - Comparative Laboratory Evaluation of Niran, Niran and Benzil, and Niran and Aroclor 5460 Formulations (July 14 to October 12, 1955). Org Res SL Fin. 1382. Job No. 2760, 12/55.
- Metzner, W.P., J. R. Durland. Moisture in TCB and Pyranol Mixes. Org Res SL Fin. X-1508. Job No. 117-1508, 3/21/41.
- Sharp, D. B., Joachim Dazzi. Exploratory Aroclor Chemistry Part V. Nitrogen Derivatives by Ammonolysis. RE Res Da Int. 1177, Job No. 9411, 8/4/58.

Ellenburg, A. M., P. A. Sanguinetti. Aroclors, operation of hydrogen Chloride absorpition tower. Phos Res An Fin. 1979. Job No. 230, 6/14/45.

Jenkins, R. L., W. P. Metzner. Aroclors, prepn. by chlorination of mixtures of diphenyl and high boilers. Org Res SL Fin X-1168-IV. Job No. 117-1168, IV, 10/7/40.

Ellenburg, A. M., H. L. Merten. Continuous Chlorination of Aroclors - Pilot Plant Demonstration. Org. Res. SL Fin. 2036. Job No. 3192, 5/7/58.

Ellenburg, A. M., J. K. Sears. Aroclors, pyrolysis with HCN, lit. survey. Org Res SL Lit. 1246. Job No. 2680, 4/26/55.

Darby, J. R., E. E. Cowell, P. R. Graham, R. M. Parks, P. W. Spink, L. R. Wangler. Org Res SL Fin 1523. Job No. 2595, 2563, 5/24/56. Aroclors, stabilizn.

DSW 303474

STANDARD MANUFACTURING PROCESS
FOR
AROCLOR

APPENDIX L - TIME CYCLES

The continuous chlorinator operation, used for production of Crude Liquid Aroclors, consists of four cascade flow chlorinators in series. The maximum daily production from the continuous system is 90,000 pounds.

There are four batch operated liquid chlorinators and one batch operated solid chlorinator. See following table for time cycles and batch weights.

The bottleneck in the production of Aroclors is the still system.

AROCLOR TIME CYCLES AND BATCH WEIGHTS

AROCLORS	1121	1142	1148	1154	1160	1162	1168*	2565*	4065*	5042*	5060*
Chlorinator Charge: Lbs. Biphenyl	3,000	3,000	3,000	3,000	3,000	3,000	9,500	7,550	5,850		
Santowax C								2,560			
Santowax R									3,920	9,800	9,500
Approximate Lbs. Chlorine Required to Produce Batch	1,600	4,330	5,560	7,040	9,030	9,600	40,375	37,750	34,620	14,875	28,500
Approximate Weight of Crude Batch	3,750	5,080	5,660	6,350	7,160	7,500	22,420	27,000	25,000	16,000	22,900
Approximate Time Required to Produce Batch	6 hrs.	9 hrs.	10 hrs.	13 hrs.	17 hrs.	18 hrs.	20 hrs.	17 hrs.	16 hrs.	9 hrs.	13 hrs.

*Solid Aroclors

DSW 303476

STANDARD MANUFACTURING PROCESS
FOR
AROCLORS

APPENDIX M - QUALITY CONSIDERATIONS

Aroclors for electrical equipment uses must be highly pure with only slight traces of contaminants. Two to three drops of sweat contain enough salt to reject a drum of Aroclor on chlorides. They must be chemically stable and non-corrosive to a wide variety of structural materials.

Obviously, continuous quality control must be maintained during manufacture, packaging, and shipping to assure that the customer is always satisfied with the performance of our Aroclors.

Normal power factor correcting capacitors used in air conditioners contain about ten cents worth of Aroclor. If the capacitor should explode and cause a fire, damage as high as \$20,000 could result. These capacitors are designed for 12 to 20 year's life. Quality ingredients, such as Anniston Aroclor, must be used to guarantee this life as there is no non-destructive procedure for pre-determining this life. This vividly illustrates the "mush-rooming" damage and the major reason for such intense desire for quality products by our customers.

Each sales lot of Aroclor is analyzed prior to shipment. Preliminary samples are analyzed prior to loading in cars, trucks, and/or drums. The specs to be met are given in greater detail in Appendix B.

Each Aroclor tank car is thoroughly cleaned before filling to prevent contamination by material left in the car by the customer. Examples of the quality control charts kept by the Laboratory are included in this appendix. The charts are for Aroclors 1242 and 5460 (our heaviest demand products). All electric grade Aroclors have control charts that are monitored by the Laboratory, Quality Control Board, and Production. These charts are used to spot trends which would eventually lead to poor quality Aroclor. Besides showing maximum and/or minimum specs on each property, the averages and two sigma values are shown which make it relatively easy to spot both good and bad trends. Production is responsible for knowing the cause and corrective measures to be taken for any trends toward poorer quality Aroclors. Export lots of Aroclors 1242 and 1254 must meet additional specs of higher resistivities and lower biphenyl content.

<u>Aroclor</u>	<u>Property</u>	<u>Spec.</u>
1242	Biphenyl Resistivity	0.05 % max. 5,000 giga ohm cm min.
1254	Biphenyl Resistivity	0.025% max. 5,000 giga ohm cm min.

DSW 303477

MONSANTO CHEMICAL COMPANY - CONTROL CHART

APPENDIX C

• 64 questions

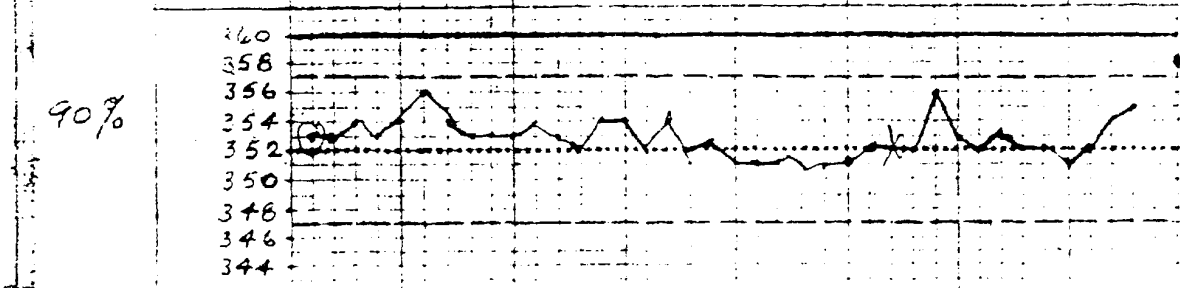
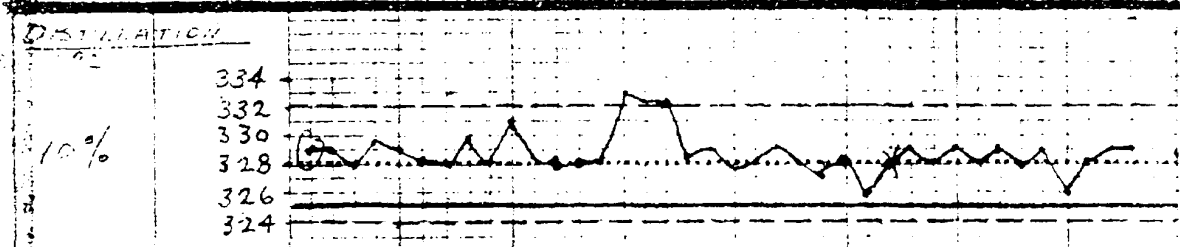
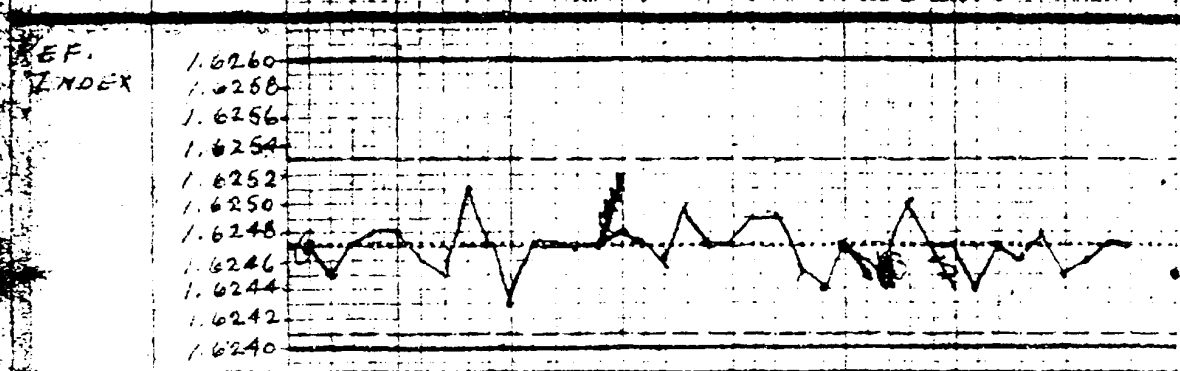
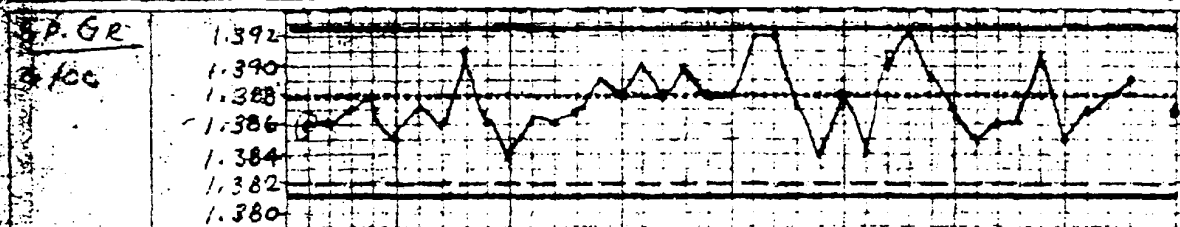
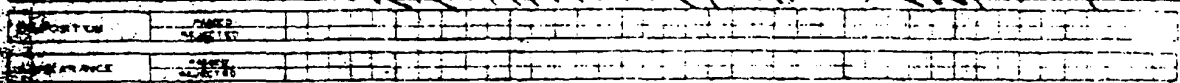
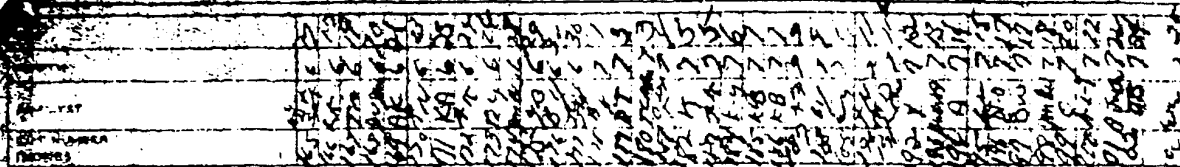
de plus de 2000

1994

ARCADE 1242

ELECTRICAL GRADE

Key 12/11/20



DSW 303478

MONSANTO CHEMICAL COMPANY - CONTROL CHART

Haupt

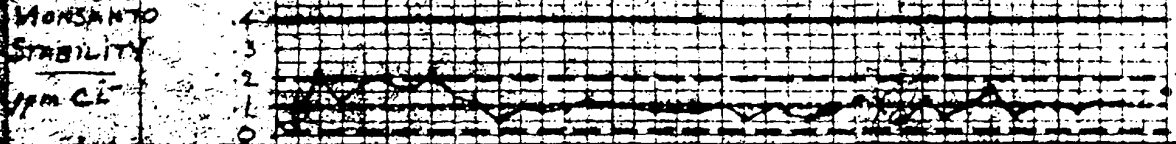
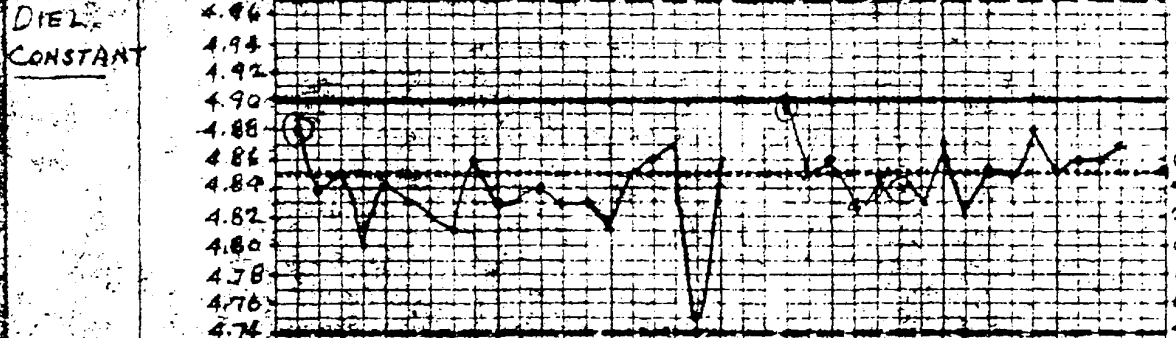
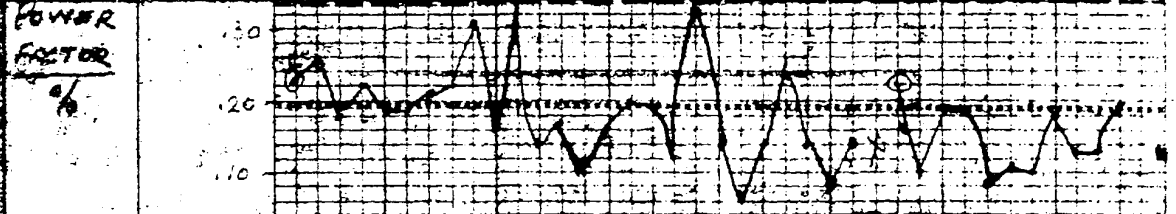
AROSLER 1242

ELECTRICAL GRADE

DATE	10/18/78
MONTH	10
ANALYST	W. J. ...
LOT NUMBER	...
OTHER	...

DISPOSITION	...
-------------	-----

APPEARANCE	...
------------	-----



DSW 303479

378

4.90

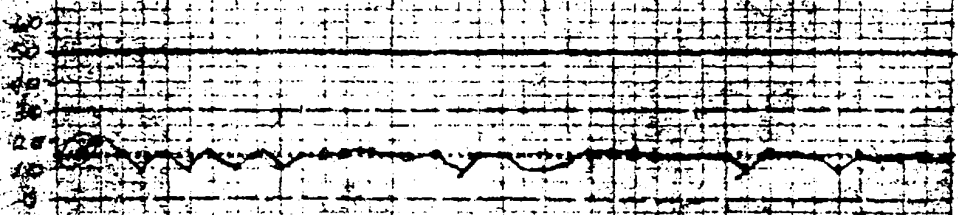
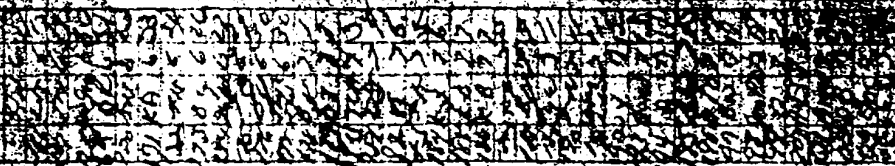
SPEC. 4 max.

OPERATING SPEC. 4 max.

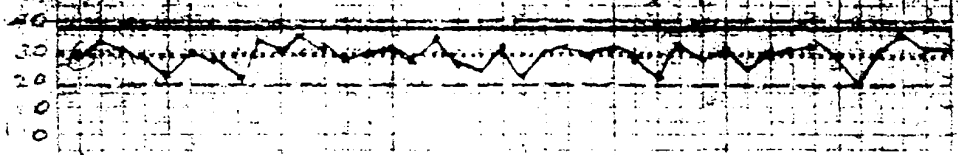
1.3

MONSIEUR'S COMPANY CONTROL CHART

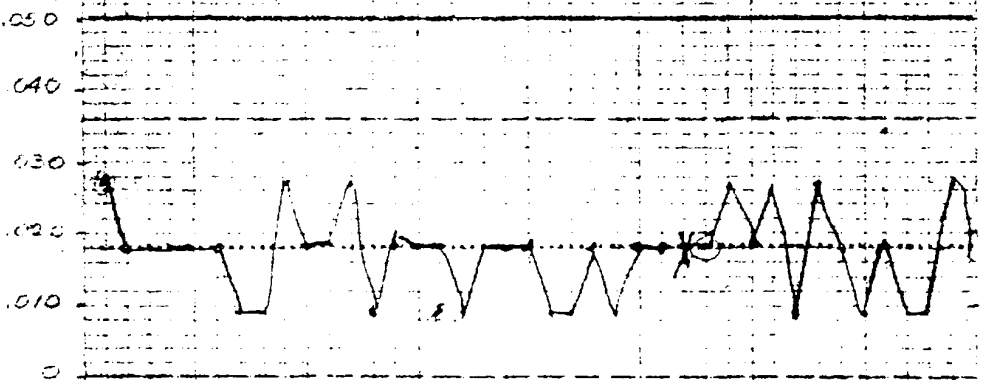
LOCATION _____
 ADDRESS _____
 PHONE NUMBER _____
 SPECIFICATION _____



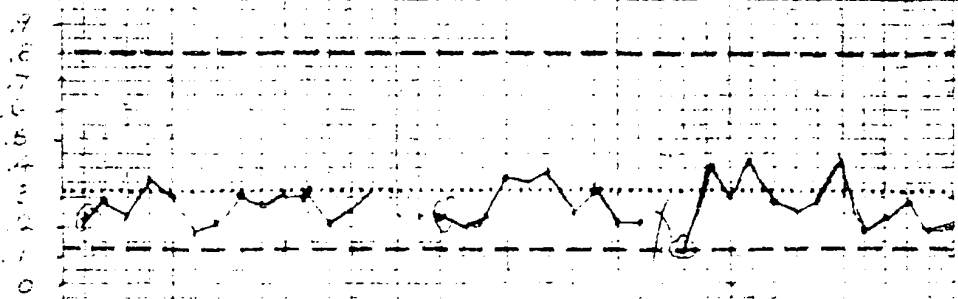
SPEC
 50 max



SPEC
 25 max



SPEC
 .05 max



Handwritten signature: *Handwritten signature*

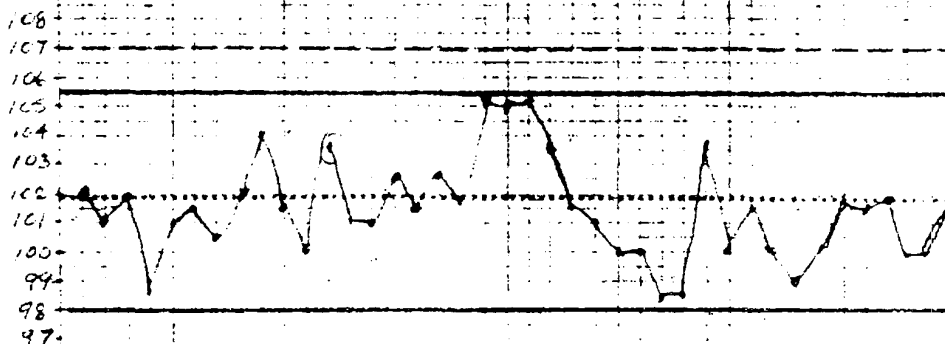
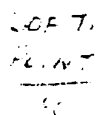
— 266 —

Arax: or 5460

© 1999 John Wiley & Sons, Inc.

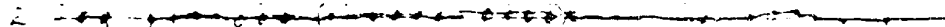
Rev 12.0

HH 219



45

Page 2



— 222 —



25. 11. 1944

DSW 303481

STANDARD MANUFACTURING PROCESS
FOR
AROCLORS

APPENDIX N - MISCELLANEOUS

DSW 303482