

TABLE 1. THERMODYNAMIC PROPERTIES OF MOIST AIR^a (STANDARD ATMOSPHERIC PRESSURE, 29.921 in. Hg) (Concluded)

FAHR. TEMP. t (°F)	HUMIDITY RATIO W _s	VOLUME CU FT/LB DRY AIR			ENTHALPY BTU/LB DRY AIR			ENTROPY BTU PER (°F) (LB DRY AIR)			CONDENSED WATER			FAHR. TEMP. t (°F)
		v _a	v _{as}	v _s	h _a	h _{as}	h _s	s _a	s _{as}	s _s	Enthalpy Btu/Lb h _{sw}	Entropy Btu/(Lb) s _{sw}	Vap. Press In. Hg p _s	
175	0.5292	16.001	13.475	29.476	42.087	601.1	643.2	0.07756	1.005	1.083	143.02	0.2553	13.675	175
176	0.5510	16.026	14.074	30.100	42.583	627.1	66.94	0.07794	1.047	1.125	144.02	0.2568	13.987	176
177	0.5760	16.051	14.710	30.761	43.136	654.1	697.3	0.07832	1.091	1.169	145.02	0.2584	14.304	177
178	0.6016	16.076	15.386	31.462	43.745	681.1	729.5	0.07870	1.137	1.216	146.03	0.2600	14.628	178
179	0.6288	16.102	16.104	32.206	44.401	708.1	763.3	0.07908	1.187	1.266	147.03	0.2616	14.958	179
180	0.6578	16.127	16.870	32.997	45.102	735.5	797.8	0.07946	1.240	1.319	148.03	0.2631	15.294	180
181	0.6887	16.152	17.675	33.841	45.847	763.9	832.4	0.07984	1.296	1.376	149.03	0.2647	15.636	181
182	0.7218	16.177	18.518	34.742	46.636	792.9	867.7	0.08021	1.357	1.437	150.04	0.2663	15.986	182
183	0.7572	16.203	19.504	35.702	47.469	822.9	903.5	0.08059	1.421	1.502	151.04	0.2678	16.340	183
184	0.7953	16.228	20.513	36.741	48.347	853.2	940.2	0.08096	1.490	1.571	152.04	0.2693	16.702	184
185	0.8363	16.253	21.601	37.854	49.269	883.2	977.7	0.08134	1.565	1.647	153.05	0.2709	17.071	185
186	0.8805	16.278	22.775	39.053	50.234	913.2	1016.0	0.08171	1.645	1.724	154.05	0.2724	17.446	186
187	0.9280	16.304	24.047	40.351	51.243	943.2	1055.9	0.08208	1.731	1.813	155.05	0.2740	17.828	187
188	0.9802	16.330	25.427	41.756	52.297	973.2	1096.6	0.08245	1.825	1.907	156.05	0.2755	18.217	188
189	1.037	16.355	26.934	43.288	53.398	1003.2	1138.1	0.08283	1.928	2.011	157.06	0.2771	18.614	189
190	1.099	16.379	28.580	44.959	54.544	1030.2	1180.6	0.08320	2.039	2.122	158.07	0.2786	19.017	190
191	1.166	16.405	30.385	46.760	55.734	1057.2	1224.1	0.08357	2.161	2.245	159.07	0.2802	19.427	191
192	1.241	16.430	32.375	48.805	56.967	1084.2	1268.6	0.08394	2.291	2.380	160.07	0.2817	19.845	192
193	1.324	16.455	34.581	51.038	58.243	1111.2	1314.1	0.08431	2.444	2.534	161.08	0.2833	20.271	193
194	1.416	16.480	37.036	53.518	59.563	1138.2	1360.6	0.08468	2.609	2.694	162.08	0.2848	20.704	194
195	1.519	16.508	39.785	56.291	60.911	1165.2	1408.1	0.08505	2.794	2.879	163.09	0.2864	21.145	195
196	1.635	16.531	42.885	59.416	62.291	1192.2	1456.6	0.08542	3.002	3.087	164.09	0.2879	21.594	196
197	1.767	16.556	46.402	62.958	63.743	1219.2	1506.1	0.08579	3.235	3.224	165.10	0.2895	22.050	197
198	1.917	16.581	50.428	67.007	65.243	1246.2	1556.6	0.08616	3.491	3.483	166.10	0.2910	22.514	198
199	2.091	16.607	54.974	71.581	66.783	1273.2	1608.1	0.08653	3.817	3.811	167.11	0.2925	22.987	199
200	2.295	16.632	60.510	77.142	68.363	1300.2	1660.6	0.08690	4.179	4.268	168.11	0.2940	23.468	200

^aCompiled by John A. Goff and S. Gratch.

given temperature (below the critical temperature), though this requires that it have a definite pressure p_s and that the coexisting condensed phase have the same temperature and pressure. The values of saturation pressure listed in Table 1 have been computed from the formulae of Goff and Gratch².

Thermodynamic Properties of Water at Saturation

Table 2 offers revisions to existing steam table data³ with extension downward to -160 F. These revisions and extension were a necessary preliminary to the construction of Table 1. A detailed explanation of the methods employed in constructing Table 2 is given in a paper² by John A. Goff and S. Gratch. As in Table 1 the temperature scale used as argument in Table 2 is the Fahrenheit scale defined in terms of absolute temperature T by Equation 1 whereas the Fahrenheit scale used as argument in existing steam tables is that derived from the International Centigrade scale by means of Equation 2. The symbols used as column headings in Table 2 are the same as those used in the steam tables and have the same meanings; therefore, a detailed explanation seems unnecessary.

DEGREE OF SATURATION

At given values of temperature and pressure the humidity ratio W of moist air can have any value between zero (dry air) and W_s (moist air at saturation). For convenience a parameter μ called alternatively *degree of saturation* or *percent saturation* is introduced through the definition,

$$W = \mu W_s \quad (3)$$

Obviously the degree of saturation μ can have any value from zero (dry air) to unity (moist air at saturation).

To a degree of approximation within the estimated uncertainty of the data in Table 1 at temperatures below about 150 F, the volume v of moist air per pound of dry air at any degree of saturation μ may be computed from the simple relation,

$$v = v_a + \mu v_{as} \quad (4)$$

To obtain comparable accuracy at temperatures above about 150 F it is necessary to add a correction term $\bar{v}$ as follows,

$$\bar{v} = \frac{\mu(1 - \mu)A}{1 + aW_s\mu} \quad (4a)$$

where a denotes the ratio of the apparent molecular weight of dry air (28.966) to the molecular weight of water (18.016), namely, 1.6078. In Table 3 are given, for each of several higher temperatures, the corresponding value of the coefficient A , the value of μ at which the correction term $\bar{v}$ attains its maximum value, and the maximum value of the correction term there attained.

At temperatures below about 150 F the enthalpy h of moist air per pound of dry air at any degree of saturation μ may be computed from the simple relation,

$$h = h_a + \mu h_{as} \quad (5)$$

To obtain comparable accuracy at temperatures above about 150 F it is