

HOECHST CELANESE CORPORATION
TECHNICAL CENTER
Corpus Christi, Texas

To: All Lab Personnel

February 15, 1988

From: O. S. Fruchey

OSF-212-88

Information Required for Lab Safety
Committee Unit Review

ABSTRACT

This report contains a list of information required for Lab Safety Committee (LSC) unit review. This information (14 copies) must be given to the Lab Safety Committee Chairman at least three working days prior to any scheduled unit review meeting.

Keywords

Lab Safety Committee
Unit Review

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QUALITY AND TEAMWORK - A WINNING COMBINATION

DISCUSSION

Certain minimum safety-related information is required by the LSC to enable the committee to properly evaluate the unit's potential hazard. Any individual who requests LSC review of a new or existing unit must provide 14 copies of this unit information (shown below) to the LSC Chairman at least three working days prior to the scheduled review meeting. The purpose of this request is two-fold. First, it guarantees that the individual in charge of the unit has addressed these various safety points. Second, it should shorten the length of the review meeting since the LSC members can examine the information before the meeting.

LSC review of a lab unit is required whenever the unit involves:

- Pressure operation
- Carcenogenic, corrosive, or hazardous material
- Large quantities of chemicals

The review information format is shown below:

General Information

- Unit Supervisor's Name
- Unit Operator's Name
- Unit Location
Identify the lab where the unit will be located and also indicate if the unit will be in a hood.
- Attended or Unattended
Indicate whether the unit will be fully attended or automated to run unattended.

Chemical Considerations

- Chemical Reactions
Provide equations for all chemical reactions involved in the unit. The heats of reaction should also be provided.
- Chemical Quantities
Identify all the chemicals and their amounts used in your runs.

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- Physical Properties of Chemicals
Include boiling points, melting points, critical temperatures, critical pressures, flammable limits, and any other physical property that may be of importance.
- Chemical Hazards
Attach either an MSDS or Celmis forms for all chemicals involved in your runs (both reactants and products).
- Personal Protective Equipment
Include items such as gloves, respirators, face shields, etc.
- Protection of Maintenance Personnel
Identify procedures to decontaminate equipment before repair, which have been in contact with hazardous chemicals.
- Spill Procedures
Identify spill clean-up procedures and address any possible conditions which might require building evacuation.
- Disposal Procedures
Indicate how you plan to dispose of all chemicals involved in your runs (both reactants and products). These procedures should be consistent with those identified in the Celmis system. Also include venting procedures (i.e. type of scrubber or diluent used).

Mechanical Considerations

- Unit Drawing
Include a detailed diagram of the unit showing valves, vessels, gauges, etc (see Figure 1).
- Equipment Rating
Include pressure and temperature ratings for all parts of the unit. This can be incorporated in the unit drawing (see Figure 1). Indicate date and pressure of hydrostatic test.
- Pressure check
Indicate N₂ pressure value that will be used to pressure test system prior to operation.

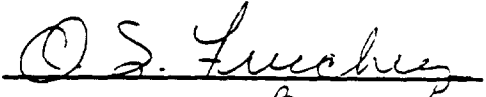
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- Safety Systems
Include all safety shutdown systems, alarms, rupture disc, pressure relief valves, etc.
- Operating Conditions
Include operating temperatures, pressures, etc.
- Operating Procedure
Include start-up, shutdown and run procedures. Also emergency shutdown procedures should be included.
- Worst Case Pressure Rise Calculation
This is a thermodynamic calculation assuming an adiabatic instantaneous pressure rise (see Attachment I). This calculation assumes an instantaneous release of the heat reaction.
- Materials of Construction
Metallurgy of all components of the unit must be identified. Evaluate compatibility of metallurgy with planned unit operations prior to the LSC meeting.

CONCLUSIONS

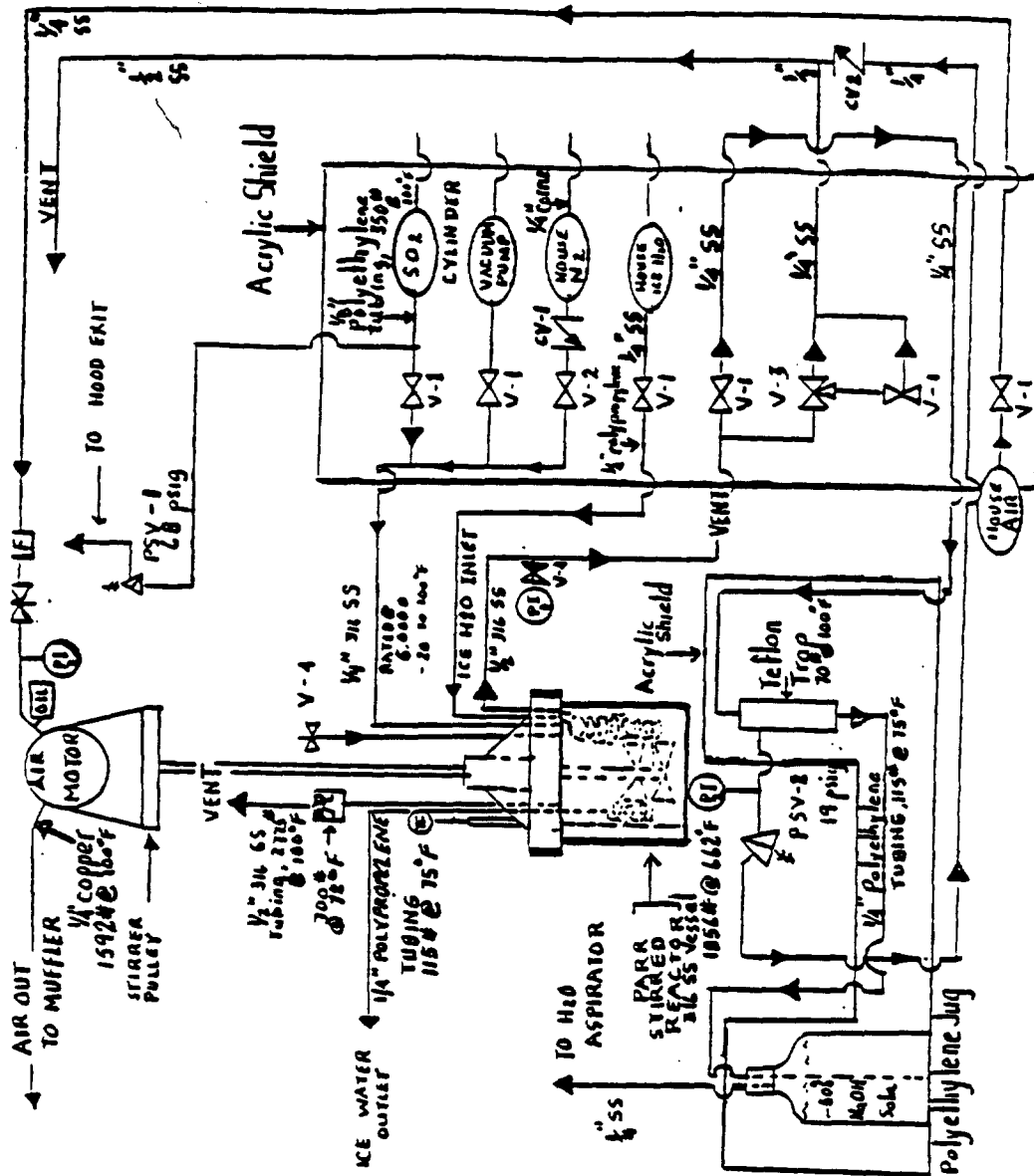
The LSC request that the information identified above be provided to the LSC members at least three working days prior to LSC review meetings.


O. S. Fruchey *By m*

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Figure 1



Beckmann Reaction Unit

V1	Whitey, SS-2RS4, 5000# @ 10°F
V2	Whitey, SS-1KS4, 5000# @ 10°F
V3	Whitey, SS-42X54, 3-Way, 55-2500# @ 10°F
V4	Whitey, 1/4 354, 3000# @ 10°F
PI1	Ashcroft Gauge, 30 1/4, 21219-S, 2300# @ 10°F
PI2	US Gauge, SS 316, 3-40, 2000# @ 10°F
PI3	US Gauge, 0-100, 2519, 2000# @ 10°F
CV-1	Muro Bros, 1/2, 4CI-1, 3000# @ 10°F
CV-2	Muro Bros, 1/2, 4CI-1, 3000# @ 10°F
PSV-1	Safety Relief Valve, Muro Bros, 316 SS, 55-2500# @ 10°F
PSV-2	Safety Relief Valve, Muro Bros, 316 SS, 55-2500# @ 10°F

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Attachment I

Thermodynamic Considerations

Worst Case Pressure Rise

Heat release if instantaneous reaction

$$Q = (90 \text{ g oxime}) \left(\frac{1 \text{ mol oxime}}{151 \text{ g oxime}} \right) \left(\frac{42 \text{ Kcal}}{1 \text{ mol oxime}} \right) = 25 \text{ Kcal}$$

cal req'd to raise 420 g SO₂ to critical Temp

$$Q = C_{p,m} \Delta T$$

$$Q = (.125 \text{ cal/g} \cdot ^\circ\text{C}) (420 \text{ g SO}_2) (127^\circ) = 6668 \text{ cal}$$

cal left for vaporization

$$25,000 - 6668 = 18332 \text{ cal}$$

mole SO₂ that can be vaporized at 157°C

$$m = Q / \Delta H_{\text{vap}}$$

$$m = 18332 \text{ cal} / (85 \text{ cal/g}) = 215 \text{ g SO}_2$$

$$\text{mol} = 215 / 64 = 3.36 \text{ mol SO}_2$$

Pressure of 3.36 mol SO₂ at 157°C

$$PV = nRT$$

$$P = (3.36 \text{ mol}) (0.0821 \text{ l-atm/K-mol}) (430 \text{ K}) / (1 \text{ l}) = 118 \text{ atm}$$

$$P_{\text{max}} = 1720 \text{ psig}$$

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