



U. S. DEPARTMENT
R. P. LAMONT,
BUREAU OF
SCOTT TURNER

MINERAL RESOURCES
OF THE
UNITED STATES
1922

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PART I—M



IMPORTS

Imports since 1920 are shown in the following table:

Tantalum ores imported into the United States, 1920-1929

Year	Pounds	Value	Year	Pounds	Value
1920.....	160	\$54	1925.....	5,022	\$1,247
1921.....	1,229	788	1926.....	350	149
1922.....	6,785	1,327	1927.....	18,119	20,012
1923.....			1928.....		
1924.....	5,600	3,031	1929.....	15,280	19,418

¹ Classified as "Ferro-alloys (tantalum)."

² Classified as "Steel-hardening ores (tantalum)."

³ Classified as "Ores of ferro-alloying metals (tantalum)."

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- HEINRICH, FR., and PETZOLD, F. Tantalum as a material for making chemical apparatus. *Chem. Fabrik*, 1928, pp. 689-91. *Chem. Abs.*, vol. 23, March 10, 1929, p. 1015.
- "A compilation of data on the physical and chemical properties of Ta, supplemented by the authors' experiments, indicates that although it may serve well for the manufacture of chemical apparatus because of its great resistance to the attack of acids, yet limitations as to the temperature at which it may be used, especially in the presence of various gases, may prevent it from becoming wholly acceptable for this purpose."
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RARE METALS

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"Details of procedure are given."

YNTEMA, L. F. The separation of columbium and tantalum by hydrolysis. *Trans. Am. Electrochem. Soc.*, vol. 55, 1929, p. 3179. *Chem. Abs.*, vol. 23, July 10, 1929, p. 3179.

"An incomplete separation of Cb and Ta is effected by the mixed salts the electrolyte in the anode compartment. The separation depends on the precipitation of the tantalum acids at the different H-ion concentrations produced at the electrodes."

PATENTS

ARMSTRONG, PERCY A. E. Iron alloy. U. S. Patent 1688481, May 7, 1929. *Chem. Abs.*, vol. 23, January 10, 1929, p. 88.

"An alloy resistant to scaling at elevated temperatures. Si 3.5 and Ta 3 per cent; the remainder is principally Fe for valves of internal-combustion engines."

— Ferrous alloys containing chromium and tantalum. U. S. Patent 1711484, May 7, 1929. *Chem. Abs.*, vol. 23, July 10, 1929, p. 3179.

"A rust, heat, and acid resisting alloy suitable for cut-off valves. Fe together with Cr 7-25, Ta about 0.2-3.0 per cent or less. Mn and Si also may be added in small proportions."

AUSTIN, MINER M., and BALKE, CLARENCE W. (to Fansteel Metallurgical Corp.) Apparatus for removing carbon and gases from molten metal. U. S. Patent 1688481, October 23, 1928. *Chem. Abs.*, vol. 23, March 10, 1929, p. 89.

ENGLE, EDGAR W., and AUSTIN, MINER M. (to Fansteel Metallurgical Corp.) Tantalum-alloy pen. U. S. Patent 1701299, February 20, 1929. *Chem. Abs.*, vol. 23, March 20, 1929, p. 1383.

"Pens are made with nibs of Ta 90 and W 10 per cent. MARDEN, JOHN W., and RICH, MALCOLM N. (to Westinghouse Electric & Mfg. Co.) Tantalum alloy. U. S. Patent 1728941, September 24, 1929. *Chem. Abs.*, vol. 23, November 10, 1929, p. 5150.

"In order to obtain Ta, V and Cb powder free from moisture, the metal is heated with Ca and CaCl₂ in a sealed or evacuated container."

ARMY-STÄHLWERKE STAHL SCHMIDT & Co. Ges. Alloy tools. British Patent 289990, February 28, 1927. *Chem. Abs.*, vol. 23, February 20, 1929, p. 813.

"An alloy steel is specified containing W 17, Cr 5, V 1, and Mo 2 per cent."

ARMY-STÄHLWERKE RÖCHLING BRÜDERER A. G. and KROPP material. German Patent 469433, December 19, 1928. *Chem. Abs.*, vol. 23, April 10, 1929, p. 1732.

"Ta powder, which may contain up to 10 percent of graphite in the proportions 3:2 with the use of a binder; the mixture is molded under pressure, dried at 300-500° C., and sintered at 1000-1300° C. The molded piece may be given a coating of graphite. The products may be used as electrical-resistance elements, furnace linings, etc."

TITANIUM

DOMESTIC PRODUCTION

During 1929 the American Rutile Co. (a subsidiary of the Hermit Corporation) shipped ilmenite and rutile from Roseland, Nelson County, Va. The production of minerals during the last 10 years, so far as data is available, is shown in the following table. The quantities are largely for shipments rather than for production.

Titanium minerals produced in the United States, 1920-1929

Year	Rutile concentrates			Ilmenite concentrates		
	Short tons	Percentage of TiO_2	Value	Short tons	Percentage of TiO_2	Value
1920	277	94-96	(¹)	268	50	(¹)
1921						
1922	310	95	(¹)	5,181	52-54	(¹)
1923	270	95	(¹)	6,270	52	(¹)
1924				4,769	52	\$75,300
1925	46	95	\$11,000	5,558	52	\$89,000
1926	30	95	\$7,500	4,813	52	\$77,000
1927	524	95	129,861	3,500	50	44,000
1928	(¹)	(¹)	(¹)	(¹)	(¹)	(¹)
1929	(¹)	(¹)	(¹)	(¹)	(¹)	(¹)

¹ Bureau of Mines not at liberty to publish figures.² Arbitrary value.³ Figures for Florida only. Bureau of Mines not at liberty to publish figures for Virginia.

During 1928 the Vanadium Corporation of America acquired ilmenite properties on the border of Nelson and Amherst Counties, Va., and did considerable drilling during 1929. At the end of the year its subsidiary, the Southern Mineral Products Corporation, awarded a contract for designing and constructing a mining, milling, and chemical plant at its Virginia properties. The Vanadium Corporation has been engaged in developing a concentrating process for separating the ilmenite from the gangue, and according to common report is contemplating adding titanium pigments to its list of products. The Titanium Pigments Co. also did some drilling on its properties in the vicinity of Roseland, Va. A shaft has been sunk in the ilmenite deposit at Fort Edwards, N. Y. The Buckman & Pritchard operation near Jacksonville Beach, Fla., was discontinued. Some further work was done on the beach deposits in California, but the Bureau of Mines has no record of shipments being made.

GENERAL CONDITIONS

Domestic sales of titanium pigments were said to have been one-third larger in 1929 than in 1928. Titanium white is marketed in the form of the nearly pure oxide and also blended with blanc fixe and zinc oxide. A new product, titanated lithopone, was also marketed in 1929. The desirable properties of titanium pigments have been summarized by Hixson and Plechner,⁵ as follows:

Paints made with titanium pigments have shown no tendency to crack or peel, but have worn down evenly and uniformly, the surface finally becoming powdery ("chalking") and remaining in excellent condition for repainting. Since a pure titanium oxide paint is inclined to be too elastic the addition of zinc oxide as a hardener gives an even more durable paint. Owing to the fact that titanium oxide is a neutral pigment it has no action whatsoever on the vehicle, making a very slow-drying "tacky" paint. This condition is overcome by the addition of zinc oxide or driers. Since titanium oxide, like the similar oxide of silicon, is the most stable compound of the metal, the pigment is extremely resistant to attack by any of the destructive agencies to which it is likely to be exposed. That it will resist the attack of sulphuric acid, which is notoriously the chief destructive agent in the air of cities, that it is not fully oxidized and therefore is not attacked by sea air or salt water, make it an excellent pigment.

⁵ Hixson, A. W., and Plechner, W. W., Titanium Offers New Possibilities as a Pigment: Chem. and Met. Eng., vol. 36, February, 1929, pp. 76-78.

Titanium pigments are manufactured in the United States by the Titanium Pigments Co., a subsidiary of the Commercial Solvents Co., with plants at St. Louis and Niagara Falls, and by the Titanium Pigments Co., a subsidiary of the Commercial Solvents Co., with a plant at Baltimore. These pigments are produced from ilmenite. Ilmenite is also used in the manufacture of titanium, produced by the Titanium Alloys Manufacturing Co., which produces carbide-free ferrotitanium, produced by the Metal Industries Co. Rutile is used chiefly in the ceramic industry and in dinner ware, to add color and strength.

The United States probably consumes more titanium than all the rest of the world; Europe, however, uses as much as compared to the total amount of all kinds of pigments due in part to restrictive legislation affecting the use of pigments. European sales are reported to have doubled in 1929. The plant of the Titan A/S. company at Fredrikstad, Norway, to produce titanium pigments, has been closed, but has been completed by the same interests at Leoben, near Cologne. The Titan concern (now affiliated with the Lead Co.) is the largest European producer, but there is also a factory in Germany, a factory in England, and one in Italy. The present no titanium industry, but a Canadian company (Ltd.), has secured control of the Ivry ilmenite deposits with a view of utilizing material from them in the manufacture of white; however, no plant has been erected yet. Partly as a result of a reduction in the price of ferrocenol, cents a pound in the case of the 15 per cent ferrocenol, sales of titanium ferro-alloys increased in 1929.

IMPORTS

Ilmenite sand is imported extensively from Travancore, India, from Senegal. From time to time shipments have also come from Brazil. The imports of ilmenite as recorded by the Foreign and Domestic Commerce amounted to 1,000 tons, valued at \$104,887, in 1929, as compared with 1,000 tons, valued at \$79,640, in 1928. A study of the customhouse records shows that in 1928 there was also imported 5,000 tons of ilmenite, valued at around \$4 a ton, from these ilmenite-pigment deposits and entered by the importer and hence classified in the statistics as "sand, other than glass sand." Much of this ilmenite is imported from beach deposits in which it occurs with monazite and zircon, and in former years undoubtedly was included in the statistics for sand and other minerals. Ilmenite are exempt from duty. A study of the statistics, however, indicates that this situation is much in doubt, even now some of the ilmenite imported probably will be recorded under its special category. In Quebec, for example, massive ilmenite are worked at Ivry and at Baie St. Paul, which contains up to 38 per cent TiO_2 , is shipped to the United States, but this movement does not appear in the statistics. Rutile is imported from Norway, but only in small quantities. A mineral is also associated with the ilmenite produced near Baie St. Paul, in Quebec, and shipped to the United States. Co.'s works at West Lynn, Mass.

WORLD PRODUCTION

The world production of titanium minerals in 1928 and 1929, as gathered from Government reports and other available sources, is shown in the following table:

World production of titanium minerals, 1928 and 1929

Mineral and country	1928			1929		
	Ore produced	Content of TiO_2	Value ¹	Ore produced	Content of TiO_2	Value ¹
	Metric tons	Per cent		Metric tons	Per cent	
Ilmenite:						
Brazil	1,498	(²)	(³)	6,381	(⁴)	\$75,100
Canada (Quebec)	2,038	(⁵)	\$6,728	2,493	(⁶)	7,304
India (Travancore)	25,718	(⁷)	202,228	(⁸)	(⁹)	(¹⁰)
Norway	7,948	42.6	61,915	(¹¹)	(¹²)	(¹³)
Portugal	7,703	(¹⁴)	8,140	(¹⁵)	(¹⁶)	(¹⁷)
Senegal	2,110	50	6,623	17,240	(¹⁸)	24,290
United States	(¹⁹)	(²⁰)	(²¹)	(²²)	(²³)	(²⁴)
Rutile:						
Norway	50	90-93	18,414	(²⁵)	(²⁶)	(²⁷)
United States	(²⁸)	(²⁹)	(³⁰)	(³¹)	(³²)	(³³)

¹ Values as officially reported, converted to United States currency at the annual average rate of exchange as published by the Federal Reserve Board.

² Exports.

³ Data not available.

⁴ Estimated.

⁵ Bureau of Mines not at liberty to publish figures.

⁶ Concentrates.

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- "A discussion of the effect of additions of Ni, Cr, W, Mo, V, Ti, Cu, and Al."
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- "Preliminary to Ti oxide manufacture FeSO_4 is removed from a solution containing Ti, ferrous Fe, and H_2SO_4 by cooling the solution to about 5° or lower and separating the FeSO_4 crystals which form."
- Preparation of titanium and like compounds (oxides and hydroxides). British Patent 276672, August 9, 1927. Chemistry and Industry (London), vol. 48, March 15, 1929, p. 207.
- "Solutions obtained by the decomposition of ilmenite, etc., with sulphuric acid containing, e. g., 200 g. TiO_2 and 500 g. H_2SO_4 (free and combined) per liter, are introduced at 100-110° into about one-quarter of their volume of boiling water at a uniform rate during about 10 minutes, the mixture being well stirred during the addition and then boiled for 2-3 hours, to precipitate colloidal titanium hydroxide. The product may be dispersed as described in British Patent 247296. Compounds of the rare-earth metals may be similarly prepared."
- BRITISH ENGINEERING STANDARD ASSOCIATION. British specification for white titanium dioxide pigment for paints. No. 392, 1929, 7 pp. Chem. Abs., vol. 23, July 20, 1929, p. 3585.

BROWN, GEORGE H. Notes on production of Age, vol. 11, 1928, pp. 77-79. Chem. Ab. 253.

"Several series of clay mixtures of rutile and iron contents were prepared and burned. Cream color was obtained, but with increasing to greenish gray. Conclusion: (1) Rutile is a pigment of cream-colored vitreous floor-tile bodies in bodies which are free from ball clay and lo from TiO_2 or rutile is largely dependent upon Fe_2O_3 . (3) The use of TiO_2 as a coloring agent, abilities, and satisfactory colors can be secured if TiO_2 is utilized instead of rutile. TiO_2 is very crystalline on a clay body, and has an unexpected variable color effects owing entirely to vit burning."

DEUTSCHE GASOLUHLICHT-AUER-GES. Treating 309597, June 6, 1929. Jour. Am. Ceram. Soc. p. 691.

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DITMAR, RUDOLPH. Titanium dioxide in the 5, 1929, pp. 79-81. Chem. Abs., vol. 23,

"The characteristics and behavior of TiO_2 as are described. Among the properties which make to soften rubber notably during milling and the erator to obtain a strictly white vulcanizate pigment."

FARBENIND, I. G. Manufacture of titanium 290174, May 7, 1928. Chem. and Ind., vol.

"The mixture obtained by treating ilmenite, etc. hot with sufficient quantities of an alkali salt (pr water to yield a saturated potassium titanium salt and then cooled to crystallize out the double salt purified by washing with alkali sulphate solution dilute sulphuric acid, and worked up to give the pigment."

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"A discussion of the history and occurrence of pigments and of the methods of preparing Ti pig outlined."

HEATON, NOEL. The development of titanium (London), vol. 21, November 9, 1929, p. 423.

HIXSON, A. W. and PLECHNER, W. W. Titanium pigment. Chem. and Met. Eng., vol. 36, vol. 23, April 10, 1929, p. 1761.

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HYPER, J. E. Vanadium and titanium in cast iron. vol. 41, 1929, pp. 173-4, 176. Chem. Ab. p. 5456.

Previous investigations studying the influence of cast Fe are reviewed and valuable possibilities development of cast Fe containing V and Ti."

JACK, R. LOCKHART. Pigment minerals of South S. Australia, No. 12, 1928, 70 pp. Chem. Ab. p. 5446.

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- MILLOT, MARIUS. Apparatus for washing titaniferous sand. French Patent 651838, March 28, 1928. Chem. Abs., vol. 23, July 20, 1929, p. 3432.
- MONK, REGINALD H. Titanium oxide. U. S. Patent 1695341, December 18, 1929. Chem. Abs., vol. 23, February 20, 1929, p. 939.
- "Refractory Ti compounds such as those from rutile or ilmenite after elimination of other impurities and a large portion of the Fe content are converted into water-soluble sulphates by mixing with alkali metal sulphate and an acid solvent such as H_2SO_4 , heating the mixture at 120–300° until a solid cake is formed, cooling and crushing the cake."
- NATIONAL METAL & CHEMICAL BANK, LTD. Treatment for titaniferous ores. German Patent 478136, May 18, 1926. Chem. Abs., vol. 23, September 10, 1929, p. 4181.
- "The powdered ore with its natural moisture content or after addition of moisture is treated with oleum and the reaction is allowed to proceed without the application of heat. TiO_2 may be obtained from the product by dilution with water."
- OSBORNE, F. F. Certain magmatic titaniferous iron ores and their origin. Econ. Geol., vol. 23, 1928, pp. 724–761, 895–922.
- Soc. MINIERE "LA BARYTINE" and RAFFIN, MAURICE. Paint. French Patent 655085, October 14, 1927. Chem. Abs., vol. 23, August 20, 1929, p. 4088.
- "Ti white is prepared from Fe-Ti ores, by heating the ores with Na_2CO_3 and coal to reduce the Fe to the metal, which is removed by treatment with dilute H_2SO_4 . The Ti is converted to the sulphate by treatment with stronger acid and to TiO_2 by addition of BaO ."
- STEPHENS, F. G. C., ANDERSON, L. J., and CASH, W. A. Titanium pigments. British Patent 299835, August 2, 1927. Chem. Abs., vol. 23, July 20, 1929, p. 3587.
- "A Ti sulphate solution is added gradually to a prepared bulk of a solution of a soluble alkali earth metal salt such as $BaCl_2$, and a temperature of 30–35° is suitable during the precipitation of the alkali earth metal sulphate, with a subsequent heating at about 100° to precipitate substantially all the Ti from the solution. Various details are given."
- TIMES TRADE AND ENGINEERING SUPPLEMENT (LONDON). Titanium salts and pigments. Principal outlets. Vol. 23, 1928, p. 33. Chem. Abs., vol. 23, March 20, 1929, p. 1476.
- "A review of the main processes. Much interest is now being shown in the Monk-Irwin process involving hitherto unused Canadian ores."
- TITANIUM PIGMENT CO. Recovery of titanium compounds. British Patent 288569, March 15, 1928. Chem. and Ind. (London), vol. 48, August 2, 1929, p. 597.
- "Finely ground ilmenite is mixed with 72–92 per cent sulphuric acid and steam is injected into the lower part of the reaction vessel until the temperature reaches 120° and the mass thickens; a small quantity of water is then forced into the lower part of this thick pulp, followed by vigorous streams of air to keep the mass in agitation and to render it porous, and finally the cooled solid mass is treated with just sufficient water to dissolve the iron and titanium sulphates. The clear solution is agitated with scrap iron or zinc to reduce ferric to ferrous sulphates, and the titanium is recovered by hydrolysis in the usual way."
- TRAILL, R. J., and MCCLELLAND, W. R. Process for the treatment of ilmenite for the recovery of electrolytic iron and titanium oxide concentrate for pigment and other purposes. Canada Dept. Mines, Mines Branch No. 689, 1928, pp. 96–101. Chem. Abs., vol. 23, January 20, 1929, p. 341.
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- ULRICH, FRANTISEK. Occurrence of rutile in the siderite veins of Rožnava, Slovakia and its position in the vein paragenesis. Rozprawy České Akad. Cl. 2, vol. 37, 1928, 15 pp. Chem. Abs., vol. 23, May 20, 1929, p. 5441.
- "Rutile is found with albite in a vein cavity. The wall rock is an aplite composed of albite, quartz, tourmaline, apatite, rutile, and siderite. The aplite intrusions are younger than the siderite veins and are closely associated with sulphide ores."

- VAN HOEK, C. P. Titanium dioxide and titanium. 1929, pp. 2828–32. Chem. Abs., vol. 23, Nov.
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- VEREIN FÜR CHEMISCHE UND METALLURGISCHE PRODUKTE and pigments. British Patent 302659, December 10, 1923, September 10, 1929, p. 4356.
- "Ti paints and pigments comprise TiO_2 , separately dehydrated by heat, and preferably 10–30 per cent diluting substances."
- VILLA, A. Titanium oxide pigments. Tech. mod. 402. Chem. Abs., vol. 23, September 10, 1929.
- "A review of the history, sources, and manufacture of white titanium oxide pigments. Tech. mod. 434–9. Chem. Abs., vol. 23, November 10, 1929.
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- WASSMAN, S. Problemas metalurgicos sobre el tratamiento de la costa atlantica de Buenos Aires. (N) treatment of the ferruginous sands of the Atlantic. Argentina Dirección Gen. de Minas, Geol. 6 H. 1929, 7 pp.
- WHITEMORE, CARL RAYMOND. Titanium oxide. Canadian Patent 1699173, January 15, 1929. Feb. 23, 1929, p. 320.
- "This process of preparing titanium oxide from titanium ore involves reducing the iron content to sponge iron without fusion in a nonoxidizing atmosphere, and separating the sponge iron from the nonmagnetic portion. The treated ore is then heated with an acid to 250° C. in a digestion vessel. After having been disintegrated, the mass is lixiviated with water to precipitate titanium hydroxide, which is then calcined."
- WURM, ERICH. Titanium white or lithopone? G. 2752. Chem. Abs., vol. 23, November 10, 1929.
- "Experiments during the past 3 years by W. do not show that TiO_2 is more economical than lithopone in rubber vulcanizates containing lithopone age considerably better than vulcanizates containing TiO_2 , particularly with p. Examination of TiO_2 from different sources shows: It is considered that lithopone is much preferable for rubber goods."

TUNGSTEN

UNITED STATES

Contrary to general expectations, the sales of tungsten in the United States dropped in 1929 to only 8,000 tons carrying 60 per cent of tungsten trioxide. However, due to the rise in price, the decrease in value was relatively small, from \$654,000 in 1929. Tungsten ore from domestic sources was sold in 1929 by 14 companies.

The tungsten production of the United States in 1929 has been as follows:

LEAD ¹

By ELMER W. PEHRSON ²

GENERAL SUMMARY

Salient figures of the lead industry in the United States

Production of refined primary lead:	
From domestic ores.....	short tons
From foreign ores and base bullion.....	
Recovery of secondary lead:	
As pig lead.....	
In alloys.....	
Total production of pig lead (primary and secondary).....	
Imports (general):	
Lead in base bullion.....	
Lead in ore.....	
Exports:	
Refined pig lead from domestic ores.....	
Refined pig lead from foreign materials.....	
Apparent consumption of refined primary lead.....	
Estimated consumption of primary and secondary lead.....	
Prices per pound of refined lead at New York:	
Highest monthly average.....	
Lowest monthly average.....	
Average for year.....	
Quotation at end of year.....	
Mine production of recoverable lead.....	short tons
Southeastern Missouri district.....	per cent of
Utah.....	
Coeur d'Alene region.....	
Joplin (tri-State) region.....	
All other.....	
World smelter production of lead.....	short tons
United States.....	per cent of
Mexico.....	
Australia.....	
Spain.....	
Canada.....	
All other.....	

¹ Exclusive of the output from Virginia, figures for which the Bureau of Mines has no record.

Conditions in the lead industry were generally out most of 1929. Domestic production and price were above the 1928 averages. The production of lead from domestic ores was 7 per cent more than that of 1928, but this was due to decreased imports of Mexican base bullion and to refined lead from foreign ores and base bullion. The total primary lead output from all sources was 10 per cent less than that in 1928. Production in the United States was stimulated in March and April by the rapid increase in the threatened curtailment of supply from Mexico. This increase in production, together

² Work on manuscript completed November, 1930.
³ The statistics in this report were prepared (unless otherwise indicated) on imports and exports were compiled from records of the Bureau of Foreign and Domestic Commerce, and the production statistics from the Bureau of Mines. J. A. Dorey, both of the Bureau of Mines.

off in domestic deliveries in April, May, and June, caused a sharp increase in stock, which reacted unfavorably on prices. From April to August production and price had declined steadily, whereas stocks had increased steadily, but by October consumption and production had returned to the March level, prices had recovered somewhat, and stocks were down. This rally, however, was terminated by the collapse of the stock market, and in November and December production, consumption, and prices declined sharply.

The New York quotation for lead increased steadily from 6.65 cents on January 2 to a high for the year of 7.80 cents during the latter part of March. Thereafter it declined steadily to an average of 6.75 cents in August. In October the average had increased to 6.87 cents, but in November and December there were sharp declines and on December 31 the quotation was 6.25 cents per pound. The average for the year was 6.83 cents, as compared to 6.31 cents in 1928.

The recovery of secondary lead in pig lead and alloys in 1929 again broke all previous records; it was equivalent to 46 per cent of the primary lead produced from domestic ores.

The total consumption of lead in the United States in 1929, as estimated by the American Bureau of Metal Statistics, surpassed all previous records, amounting to 945,200 tons, or an increase of 2 per cent over 1928 and 12 per cent over 1927. Storage batteries ranked first among the consumers of lead in 1929, taking 22.2 per cent of the total; cable covering was second, with 21.8 per cent; white lead third, with 12.5 per cent; and building fourth, with 10.2 per cent. There were decreases of 5 and 4 per cent, respectively, in the quantities of lead used in storage batteries and in white lead but an increase of 14 per cent in that used for cable covering.

Mine production of lead showed a 3 per cent gain in 1929. This rate of increase was fairly general for all areas, production in the Western States having increased 4 per cent and that in the Mississippi Valley States 3 per cent. Utah, Idaho, New Mexico, and Missouri showed the largest tonnage increases, while Colorado was the only State among the 10 largest producers to show a decrease.

General imports of lead in base bullion decreased 35 per cent as a result of a 47 per cent decrease in shipments from Mexico. Imports from Peru increased 130 per cent. Receipts of lead in ore and matte increased 21 per cent owing to increased shipments from Mexico. Exports of refined pig lead declined from 116,269 tons in 1928 to 73,251 tons in 1929 as a result of the decreased shipments of base bullion from Mexico which is refined in the United States under bond and exported from warehouse.

The Lead Industries Association, an organization of American producers and consumers of lead, expanded its statistical service during the year to include periodic reports on sales of domestic lead, manufacturer's stocks of pig lead, and the production, sales, and stocks of antimonial lead, mixed metals, white lead, red lead, litharge, sublimed lead, and orange mineral. It also distributed the monthly reports on production, shipments, and stocks of pig lead compiled by the Lead Producers Reporting Association of England.

World smelter production of lead in 1929 amounted to 1,971,000 short tons, an increase of 6 per cent over 1928. Production outside the United States increased 6 per cent, and the output of all but two

LEAD

of the 13 countries that produced 1 per cent or more of the total lead was greater in 1929 than in 1928. The United States, Spain, Germany, Mexico, Russia, and Peru showed percentage increases.

The London price of lead in 1929 averaged £23.10 or 5.04 cents per pound; this is 10 per cent above the 1928 average, while the 1929 New York quotation is 8 per cent above the 1928 average.

The European lead cartel was reported to have a working agreement between leading Mexican, Canadian, Indian, and continental producers. No announcement of fixing or production curtailment has been noted in 1929, so it is presumed that the cartel was engaged chiefly in the collection of statistics. Producers of lead in Mexico (Burma), and Australia cooperated in selling their output through a central London agency.

Development during 1929 at Mount Isa, in New South Wales, and at Buchans, in Newfoundland, resulted in greatly increased production. Among the new deposits that appear to be promising are the Lake George mine in New South Wales, which has a possible reserve of 5,000,000 tons of ore containing 10 per cent of lead in addition to other metals, and the deposit at Lawn Hills, Queensland. Several other promising deposits are being actively developed in Yugoslavia in the Stantrg mine, which has a possible reserve of 2,000,000 tons of ore containing 11.5 per cent of lead, 3 per cent of zinc, and 3 ounces of silver. In Canada the reserve of lead in the British Columbia mine, and the Mayo district, Yukon Territory, increased by development work in 1929. Extensive zinc ores were reported in the Great Slave Lake area, Northwest Territories. No extensive discovery of lead ore was reported in the United States in 1929.

PRODUCTION

GENERAL

Pig lead produced in the United States for domestic consumption and export is derived from three main sources—domestic ores, foreign ores, and base bullion, and secondary materials. The following table shows the quantity of pig lead produced from each of these sources in 1920 to 1929.

Pig lead produced in the United States, 1920-1929, in short tons

Year	From domestic ores and base bullion	From foreign ores and base bullion
1920	476,849	52,808
1921	398,222	50,397
1922	468,746	63,916
1923	543,841	74,481
1924	566,407	124,086
1925	654,921	112,048
1926	680,685	118,256
1927	668,320	128,210
1928	626,202	154,869
1929	672,498	104,136

The total production of pig lead in the United States in 1929 decreased 5,938 tons, or less than 1 per cent.

domestic ores and base bullion increased 7 per cent, whereas that from foreign ore and base bullion decreased 34 per cent. Production of pig lead from secondary materials increased slightly.

PRIMARY LEAD

REFINERY PRODUCTION

Pig lead is classified by the Bureau of Mines as primary or secondary, according to whether it is derived directly from the smelting of ores and from base bullion or is reclaimed from industrial lead scrap. Some plants that ordinarily smelt ore frequently add lead scrap to the charge, so that the resulting product is an inseparable mixture of lead from both sources and is all marketed as refined pig lead. Nevertheless, the smelter calculates the quantity of secondary lead so produced, thus enabling the Bureau of Mines to obtain a consecutive record of the amount of primary lead being supplied to the trade each year.

Primary lead may be produced from domestic ore or foreign ore and from base bullion. The smelting of lead-bearing ores from the Western States yields an impure pig lead (base bullion) that contains valuable amounts of silver and gold, in addition to objectionable impurities such as copper, arsenic, and antimony. This base bullion is given further treatment for the removal of the impurities and finally emerges for the market as refined pig lead. The smelting of lead-bearing ores from the Mississippi Valley and Southern States yields soft lead pure enough for marketing, but some of it is desilverized and is designated in this report as desilverized soft lead. Foreign ore is first smelted into base bullion and then refined for market. Foreign base bullion imported into the United States is also refined before marketing. All of these products for market are referred to as refined lead. The following table gives a summary of the refined primary lead produced in the United States from 1920 to 1929.

Refined primary lead produced in the United States, 1920-1929

Year	Short tons						Value	
	Desilverized lead ¹	Desilverized soft lead	Soft lead ²	Total production ¹	From domestic ores and base bullion	From foreign ores	Average per pound	Total
1920-1912.....	7, 176, 497	326, 091	2, 029, 180	10, 432, 668	8, 545, 383	1, 887, 285		\$924, 600, 000
1913.....	301, 160	29, 432	131, 867	462, 460	411, 878	13, 223	37, 359	\$0.044
1914.....	340, 397	43, 506	168, 219	542, 122	512, 794	7, 639	21, 689	.039
1915.....	344, 593	44, 001	161, 461	550, 055	507, 026	9, 581	33, 448	.047
1916.....	335, 376	68, 244	167, 616	571, 234	562, 228	6, 085	12, 821	.069
1917.....	365, 998	56, 268	188, 503	610, 769	548, 460	15, 872	46, 447	.086
1918.....	332, 314	47, 418	210, 463	590, 195	539, 965	15, 334	84, 956	.071
1919.....	266, 538	67, 938	147, 744	482, 220	424, 433	7, 655	50, 131	.053
1920.....	273, 185	66, 668	189, 854	529, 657	478, 849	8, 414	44, 394	.080
1921.....	238, 329	52, 747	157, 613	448, 689	398, 222	7, 120	43, 247	.045
1922.....	249, 107	74, 305	209, 250	532, 662	468, 746	13, 898	50, 018	.055
1923.....	366, 241	61, 332	190, 749	618, 322	543, 841	24, 288	50, 193	.070
1924.....	423, 429	63, 449	203, 615	690, 493	566, 407	46, 385	77, 701	.080
1925.....	457, 477	48, 632	290, 860	796, 969	654, 921	47, 024	65, 024	.087
1926.....	484, 669	55, 707	258, 566	798, 941	680, 685	45, 210	70, 046	.080
1927.....	507, 099	65, 487	238, 944	796, 530	668, 320	32, 369	65, 841	.063
1928.....	506, 603	49, 465	225, 003	781, 071	626, 202	28, 632	128, 237	.058
1929.....	463, 622	55, 666	236, 245	774, 633	672, 498	29, 675	72, 460	.063
	13, 502, 583	1, 267, 557	6, 269, 350	21, 039, 490	17, 798, 788	8, 230, 702		2, 315, 789, 000

¹ The lead content of antimonial lead is excluded (see p. 238). ² Desilverized soft lead is excluded.

The total production of refined primary lead in the United States in 1929 was nearly 1 per cent less than that in 1928, but 1 per cent less than the 1928 output, the maximum annual production. The total value of the 1929 output was 8 per cent greater than that of 1928, due to an increase of one-half cent per pound in the price of lead, but amounted to only 73 per cent of the peak annual value up to the end of 1929.

Two outstanding features in the production of primary lead in the United States in 1929 were the decreased output of refined lead from domestic ores and the increased production from domestic base bullion. The total production of refined lead from domestic ores in 1929 was 17,798,788 tons, a decrease of 128,237 tons in 1928 to 72,460 tons in 1929, a decrease of 13 per cent of the total primary lead output in 1928. Refined lead produced from foreign ores and base bullion amounted to 8,230,702 tons in 1929, an increase of 16 per cent in 1928 and 20 per cent in 1927. The total refined lead from domestic ores in 1929 was 46,296 tons, greater than in 1928, and was the second largest amount recorded. The following table shows the quantity of refined lead produced in the United States from 1920 to 1929:

Refined lead produced in the United States from domestic ores, 1920-1929

Year	Short tons	Value	Year
1920.....	476, 849	\$78, 296, 000	1925.....
1921.....	398, 222	35, 840, 000	1926.....
1922.....	468, 746	61, 682, 000	1927.....
1923.....	543, 841	76, 138, 000	1928.....
1924.....	666, 407	90, 628, 000	1929.....

GEOGRAPHICAL SOURCE OF LEAD RAW MATERIALS REFINED IN THE UNITED STATES

The following table shows the source of the ore and base bullion from which refined lead was produced in the United States from 1920 to 1929. A large part of the base bullion from the Western States is shipped to the Mississippi Valley States for refining. In reporting the production of refined lead from domestic ores, most of the refiners make the distribution on the basis of the total lead given in this table as from domestic ores, so that the total lead given in this table as from domestic ores, both base bullion and refined lead and thus containing the antimonial lead produced from domestic ores, exceeds the total production of refined lead from domestic ores, as shown in the preceding table. The production of antimonial lead from domestic ores amounted to 17,062 tons.

domestic ores and base bullion increased 7 per cent, whereas that from foreign ore and base bullion decreased 34 per cent. Production of pig lead from secondary materials increased slightly.

PRIMARY LEAD

REFINERY PRODUCTION

Pig lead is classified by the Bureau of Mines as primary or secondary, according to whether it is derived directly from the smelting of ores and from base bullion or is reclaimed from industrial lead scrap. Some plants that ordinarily smelt ore frequently add lead scrap to the charge, so that the resulting product is an inseparable mixture of lead from both sources and is all marketed as refined pig lead. Nevertheless, the smelter calculates the quantity of secondary lead so produced, thus enabling the Bureau of Mines to obtain a consecutive record of the amount of primary lead being supplied to the trade each year.

Primary lead may be produced from domestic ore or foreign ore and from base bullion. The smelting of lead-bearing ores from the Western States yields an impure pig lead (base bullion) that contains valuable amounts of silver and gold, in addition to objectionable impurities such as copper, arsenic, and antimony. This base bullion is given further treatment for the removal of the impurities and finally emerges for the market as refined pig lead. The smelting of lead-bearing ores from the Mississippi Valley and Southern States yields soft lead pure enough for marketing, but some of it is desilverized and is designated in this report as desilverized soft lead. Foreign ore is first smelted into base bullion and then refined for market. Foreign base bullion imported into the United States is also refined before marketing. All of these products for market are referred to as refined lead. The following table gives a summary of the refined primary lead produced in the United States from 1720 to 1929.

Refined primary lead produced in the United States, 1720-1929

Year	Short tons						Value	
	Desilverized lead ¹	Desilverized soft lead	Soft lead ¹	Total production ¹	From domestic ores and base bullion	From foreign ores	Average per pound	Total
1720-1912.....	7, 176, 497	326, 991	2, 920, 180	10, 432, 668	8, 545, 383	1, 887, 285		\$924, 600, 000
1913.....	801, 180	20, 433	131, 807	462, 460	411, 878	13, 223	37, 359	\$0.044
1914.....	340, 397	43, 606	188, 219	542, 122	512, 794	7, 639	21, 689	.039
1915.....	844, 593	44, 001	181, 481	850, 055	807, 026	9, 581	33, 448	.047
1916.....	355, 375	68, 244	167, 615	571, 134	552, 228	6, 085	12, 821	.069
1917.....	355, 998	55, 288	188, 503	610, 789	548, 450	15, 872	46, 447	.068
1918.....	382, 314	47, 418	210, 463	640, 195	539, 905	15, 334	84, 056	.071
1919.....	266, 538	67, 938	147, 744	482, 220	424, 433	7, 666	50, 131	.053
1920.....	273, 135	66, 668	189, 854	529, 657	476, 849	8, 414	44, 394	.080
1921.....	238, 329	52, 747	157, 618	448, 699	398, 222	7, 120	43, 247	.045
1922.....	249, 107	74, 305	200, 250	523, 662	498, 746	13, 898	50, 018	.055
1923.....	368, 241	61, 332	190, 749	618, 322	543, 841	24, 288	50, 193	.070
1924.....	423, 429	63, 449	203, 615	690, 493	595, 407	46, 385	77, 701	.080
1925.....	457, 477	48, 922	230, 560	736, 959	654, 921	47, 024	65, 024	.087
1926.....	484, 669	55, 707	258, 565	798, 941	680, 685	48, 210	70, 046	.080
1927.....	507, 099	55, 487	233, 944	796, 530	668, 320	32, 369	55, 841	.083
1928.....	506, 603	49, 465	225, 003	781, 071	628, 202	26, 632	128, 237	.058
1929.....	483, 622	55, 666	235, 345	774, 633	672, 498	29, 675	72, 460	.063
18, 602, 583	1, 267, 557	5, 259, 350	21, 029, 490	17, 798, 788	8, 230, 702		2, 815, 789, 000	

¹ The lead content of antimonial lead is excluded (see p. 238). ² Desilverized soft lead is excluded.

The total production of refined primary lead in 1929 was nearly 1 per cent less than that in 1928, but 1 per cent less than the maximum of 1926. The total value of the 1929 output was 8 per cent of 1928, due to an increase of one-half cent per price of lead, but amounted to only 73 per cent of the peak annual value up to the end of 1929.

Two outstanding features in the production of refined lead in the United States in 1929 were the decreased output of base bullion and the increased production from domestic ores. The production of refined lead from foreign base bullion decreased from 128,237 tons in 1928 to 72,460 tons in 1929, a decrease of 43 per cent. Refined lead produced from foreign ores and base bullion decreased to 13 per cent of the total primary lead output in 1929, compared to 20 per cent in 1928 and 16 per cent in 1927. Refined lead from domestic ores in 1929 was 4 per cent greater than in 1928, and was the second largest recorded. The following table shows the refined lead produced in the United States from 1920 to 1929:

Refined lead produced in the United States from 1920 to 1929

Year	Short tons	Value	Year
1920.....	476, 849	\$76, 208, 000	1925.....
1921.....	398, 222	35, 840, 000	1926.....
1922.....	468, 746	61, 562, 000	1927.....
1923.....	543, 841	76, 138, 000	1928.....
1924.....	566, 407	90, 625, 000	1929.....

GEOGRAPHICAL SOURCE OF LEAD RAW MATERIALS REFINED IN THE UNITED STATES

The following table shows the source of refined lead from which refined lead was produced from 1920 to 1929. A large part of the base bullion from the Western States is shipped to the Mississippi Valley for refining. In reporting the production of refined lead, most of the refiners make the distribution on a basis so that the total lead given in this table as refined lead from both base bullion and refined lead and thus the antimonial lead produced from domestic ores and base bullion total given here for 1929, exceeds the total refined lead from domestic ores, as shown in the preceding table. The production of antimonial lead from domestic ores amounted to 17,062 tons.

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Primary lead smelted or refined in the United States, 1920-1929, by sources, in short tons

Source of ore	1920	1921	1922	1923	1924	1925	1926	1927	1928	1929
Domestic ore:										
Alaska.....	591	773	324	400	582	699	1,018	1,010	962	1,266
Arizona.....	5,987	3,313	7,218	8,828	9,372	10,281	10,646	7,918	8,144	8,153
California.....	2,260	614	3,018	5,168	2,305	4,148	4,044	1,232	637	1,459
Colorado.....	17,752	12,104	11,108	23,885	25,491	31,855	37,553	37,040	28,683	23,678
Idaho.....	117,191	99,707	91,487	127,797	123,709	123,363	132,564	150,518	141,748	147,593
Illinois.....	948	271	780	1,285	1,089	474	387	222	309	777
Kansas.....	8,421	10,930	10,900	20,207	12,895	15,001	20,884	25,493	27,014	26,104
Kentucky.....	114	41	73	66	201	96	27	139	268	508
Missouri.....	171,999	161,028	202,245	169,323	191,501	208,547	218,082	210,364	192,789	211,545
Montana.....	13,231	11,565	14,551	18,345	21,226	22,008	26,350	23,041	20,470	26,791
Nevada.....	8,650	3,553	4,284	8,044	8,070	10,978	11,379	8,680	6,823	12,256
New Mexico.....	1,123	384	1,230	1,638	2,263	3,782	4,383	8,377	8,469	12,249
Oklahoma.....	58,494	46,902	67,436	59,602	56,017	78,487	70,004	47,857	46,692	46,633
Oregon.....			37	47		6	5	2	7	18
South Dakota.....	6	2			2			1	79	19
Tennessee.....	2,705		781	1,020	985	436	770	19		559
Texas.....	1	1	5	40	27	31	102	275	437	157,085
Utah.....	64,006	51,872	63,130	104,678	119,318	165,844	151,127	155,199	151,724	(?)
Virginia.....	2,460	325	478	2,008	1,582	2,179	1,714	2,214	4,944	447
Washington.....	3,841	1,079	1,323	691	1,973	2,889	2,052	2,067	1,812	1,725
Wisconsin.....	201	101	75	691	15	2				5,424
Undistributed.....	4,366	713	1,284	362	1,320	1,541	505	1,905	102	1,903
Zinc residues.....	494,347	395,287	481,689	554,035	582,000	686,451	695,830	684,165	642,697	685,992
Foreign ore:										
Africa.....	1,171			99	6		6	12		5
Australia.....	450								2	
Canada.....	1,188	2,877	3,604	3,032	4,633	10,676	5,385	4,138	10,123	9,499
Central America.....	22	6	14	18	30	3		18	3	
Europe.....	19	17	519	139	328	470	233	19	67	28
Mexico.....	2,729	1,912	7,260	18,867	33,328	31,107	32,693	22,844	14,839	15,807
South America.....	2,747	2,235	2,458	1,461	7,663	4,496	9,893	5,506	1,568	3,284
Other foreign.....	90	73	43	72	397	269			30	591
Foreign base bullion:										
Canada.....	1,919	1,599	2,900							
Mexico.....	42,421	41,048	47,118	50,193	77,701	65,024	70,046	95,841	117,193	51,253
South America.....	54								11,044	21,165
	52,808	50,367	53,916	74,481	124,086	112,048	118,256	128,380	154,869	102,155
Grand total.....	647,155	445,654	545,605	628,517	706,086	798,499	814,085	812,545	797,566	788,127

¹ The lead produced by these States is nonargentiferous or soft lead. A small quantity of the lead produced from Colorado and New Mexico ores is also nonargentiferous.

² The smelter output of lead in Kansas and Oklahoma is separated on the basis of mine output as the Bureau of Mines has been unable to obtain an accurate separation of smelter output.

³ Included under "Undistributed." Bureau of Mines not at liberty to publish figures.

The table shows that in 1929, 87 per cent of the primary lead smelted or refined in the United States was derived from domestic ores, 4 per cent from the smelting of foreign ores, and 9 per cent from the refining of foreign base bullion. There were substantial increases in lead production in 1929 in most of the important lead-producing States. Decreased production was recorded for Colorado, Kansas, Oklahoma, Washington, and Wisconsin. Foreign ore smelted in the United States in 1929 came largely from Mexico, Canada, and South America (chiefly Chile). Refined lead produced in the United States from Mexican base bullion decreased 56 per cent in 1929 owing to the completion of new refineries in Mexico which treated a larger part of the Mexican production of base bullion. There was an increase of 92 per cent in the production of refined lead from South American (chiefly Peruvian) base bullion.

SPECIFICATIONS FOR PIG LEAD

The American Society for Testing Materials classifies pig lead in three grades—corroding, chemical, and common. The requirements as to chemical composition are shown in the following table.

Allowable chemical composition of the various grades of pig lead
American Society for Testing Materials

	Corroding lead, not over—	Chemical lead, not over—
	Per cent	Per cent
Silver.....	0.0015	0.005-0.0
Copper.....	0.0015	0.04-0.0
Copper and silver together.....	0.0025	(?)
Arsenic.....	0.0015	(?)
Antimony and tin together.....	0.0095	(?)
Zinc.....	0.0015	(?)
Iron.....	0.02	(?)
Bismuth.....	0.05	(?)
Lead, by difference.....	99.9300	(?)

¹ The maximum limits for bismuth, copper, and silver have been given, but delivery will contain the maximum of all three.

² Chemical lead is a designation that has been used for many years in the silverized lead produced from southeastern Missouri ores.

³ Not given.

⁴ Less than 0.005 per cent.

⁵ Not specified.

SOFT LEAD

A large part (43 per cent in 1929) of the lead produced in the United States from domestic ores is derived from the lead ores of the Mississippi River basin. The two principal producing areas are the southeastern Missouri district (Joplin) at the junction of Missouri and Oklahoma. The former district produced about 60 per cent of the total soft lead in ore mined in the United States. Eighty-one per cent of the soft pig lead produced in the United States is marketed without desilverizing, and 19 per cent is desilverized. The former, which is known in the trade as chemical lead, is used chiefly in the manufacture of cable, pipe, and other products. The latter 25 per cent of the total soft lead in ore mined is desilverized soft lead is used largely in the manufacture of where a very pure product is required.

A substantial quantity of high-grade lead concentrate from the Mississippi Valley is used each year in the manufacture of pigments, principally basic lead sulphate (white lead) and leaded zinc oxide. In 1929 nearly 9,500 tons of soft lead were produced in this manner.

The following table summarizes the production of soft lead from 1922 to 1929:

Soft lead produced in the United States from domestic ores, 1922 to 1929

Year	Soft pig lead			Soft lead recovered in pigments
	Undesilverized	Desilverized	Total	
1922.....	209,250	74,305	283,555	18,892
1923.....	190,749	61,833	252,582	18,612
1924.....	203,615	63,449	267,064	15,338
1925.....	260,560	48,932	309,492	17,547
1926.....	288,565	55,707	344,272	13,685
1927.....	233,944	55,487	289,431	13,135
1928.....	225,008	49,456	274,464	10,465
1929.....	285,345	55,666	341,011	9,429

¹ Includes domestic refined lead, domestic lead in antimonial lead, and domestic lead in other forms.

ANTIMONIAL LEAD

Antimonial lead is an important by-product of the refining of base bullion. In 1929, 25,669 tons of primary antimonial lead was produced, which contained 22,617 tons of lead, a decrease of 22 per cent from the 1928 output of antimonial lead and equivalent to 5.3 per cent of the total refined lead produced from base bullion.

The decreased importation of base bullion from Mexico had a marked effect on the production of primary antimonial lead. In 1928 more than 15,000 tons was produced from the smelting and refining of foreign ores and base bullion, whereas in 1929 only 8,600 tons was derived from the same source. Foreign ores also yield a higher percentage of antimonial lead. The production of antimonial lead from foreign ores was equivalent to 9.8 per cent of the refined lead produced from those ores in 1928 and to 8.4 per cent in 1929, while the yield from domestic ores was 5.1 and 4.5 per cent, respectively.

The principal uses of antimonial lead are in the manufacture of storage batteries, bearing metal, and type metal. The consumption of antimonial lead in storage batteries decreased about 3 per cent in 1929. The following table shows the production of primary antimonial lead from 1920 to 1929:

By-product primary antimonial lead produced in the United States, 1920-1929

Year	Total lead refined from base bullion (short tons)	By-product antimonial lead (percent- age of total)	Antimonial lead						
			From domestic ore (short tons)	From foreign ore (short tons)	Total				Lead con- tent by difference (short tons)
					Short tons	Value	Antimony content		
							Short tons	Percent- age	
1920	273, 135	4.6	9, 604	2, 931	12, 535	\$1, 963, 255	2, 033	16.2	10, 502
1921	238, 329	4.2	7, 881	2, 183	10, 064	870, 059	1, 589	15.8	8, 475
1922	249, 107	3.2	7, 068	1, 007	8, 075	844, 993	1, 441	17.8	6, 634
1923	366, 241	3.9	11, 635	2, 555	14, 190	1, 950, 370	2, 170	15.3	12, 020
1924	423, 429	4.9	13, 593	7, 194	20, 787	3, 376, 713	2, 763	13.3	18, 024
1925	457, 477	4.3	14, 472	5, 195	19, 667	3, 785, 547	2, 024	13.8	17, 643
1926	484, 669	4.6	18, 909	3, 615	22, 524	3, 910, 714	2, 693	12.0	19, 831
1927	507, 099	4.8	16, 040	8, 307	24, 347	3, 277, 043	2, 736	11.2	21, 611
1928	600, 603	6.5	17, 930	15, 123	33, 053	3, 975, 218	3, 432	10.4	29, 621
1929	483, 622	5.3	17, 002	8, 607	25, 609	3, 267, 095	3, 052	11.9	22, 557
Average		4.8						12.9	

SECONDARY LEAD

Much of the lead consumed in the United States is used in the manufacture of articles from which there is a large return of secondary metal. This is particularly true of the storage-battery industry, which is the leading single consumer of lead and in which there is a large and rapid return of secondary metal. This industry is responsible for much of the increase in the ratio of secondary lead production to domestic primary lead production. In 1914 the secondary output was equivalent to only 12 per cent of the primary output from domestic ores; in 1929 the ratio was 46 per cent, a decrease from 49 per cent in 1928. Other important lead-consuming industries in which

there is a large return of secondary metal include building, bearing-metal manufacture, and type-metal.

Fifty-five per cent of the secondary lead output is recovered in the form of alloys, principally alloys of lead with antimony, and 45 per cent was recovered as pig lead.

The following tables show the recovery of secondary lead to the refined primary lead produced in the United States from domestic ores. Further details of the secondary lead are found in the Mineral Resources chapter on secondary lead.

Secondary lead recovered in the United States, 1922-1929

Year	Pig lead	Lead alloy
1922	75, 480	84, 480
1923	90, 480	98, 480
1924	90, 400	114, 480
1925	112, 420	114, 480
1926	125, 000	182, 480
1927	119, 000	167, 480
1928	138, 000	170, 480
1929	138, 500	172, 480

Ratio of secondary lead recovered to refined primary lead produced in the United States from domestic ores, 1913-1929

Year	Refined primary lead	
	Short tons	Average price per pound (cents)
1913	411, 878	
1914	512, 794	
1915	507, 026	
1916	552, 228	
1917	548, 450	
1918	539, 905	
1919	424, 433	
1920	476, 849	
1921	398, 222	
1922	468, 746	
1923	543, 841	
1924	566, 407	
1925	654, 921	
1926	680, 685	
1927	668, 320	
1928	628, 202	
1929	672, 498	

PIGMENTS

Lead pigments manufactured in 1929 contain lead derived from the sources shown in the following table. Pig lead is used in the manufacture of white lead, sublimed lead, and orange mineral. In 1929 about 48.3 per cent of the lead contained in lead pigments was derived from white lead; white lead accounting for 48.3 per cent, litharge 15.8 per cent, and sublimed lead and orange mineral 35.9 per cent. Sublimed lead and leaded zinc oxide are the only pigments in which the lead content is derived from ores.

Further details of the production and value of lead pigments in the United States are given in the Mineral Resources chapter on lead and zinc pigments and salts.

Lead in pigments,¹ 1922-1929, by sources, in short tons

Year	Lead in pigments from—				
	Domestic ore	Foreign ore	Metal	Scrap	Total
1922	13,892		241,098	111	255,099
1923	13,612		236,641		250,253
1924	15,838	846	230,930		247,614
1925	17,647		258,064		275,711
1926	13,686		240,932	1,029	255,647
1927	13,136		242,713	1,246	257,095
1928	10,465		242,914	654	243,033
1929	9,429		248,657	2,427	250,513

¹ Includes also lead recovered in zinc oxide and leaded zinc oxide.

MINE PRODUCTION

BY STATES

The following table shows the mine production of recoverable lead by States from 1920 to 1929:

Mine production of recoverable lead, 1920-1929, by States, in short tons

State	1920	1921	1922	1923	1924	1925	1926	1927	1928	1929
Alaska	875	759	377	410	631	789	778	1,008	1,019	1,318
Arizona	6,936	3,271	7,635	8,146	9,321	11,938	11,629	9,933	7,190	8,027
Arkansas	8	21	42	25	20	55	18	23	38	61
California	2,407	562	3,156	4,775	2,382	3,283	4,047	1,359	946	715
Colorado	23,315	9,830	11,739	22,849	23,779	31,483	34,494	33,386	26,761	24,448
Georgia	2	4								
Idaho	118,665	99,235	97,917	121,178	124,476	128,521	136,490	151,019	145,323	148,658
Illinois	1,708	672	1,325	1,381	1,464	1,601	655	277	385	448
Kansas	16,452	18,679	23,271	17,598	18,560	22,775	28,463	27,497	25,276	26,596
Kentucky	120	69	92	53	175	152	58	139	39	297
Massachusetts	14				(¹)					
Missouri	161,812	180,085	178,412	109,743	189,929	211,566	207,012	198,780	195,393	198,469
Montana	14,846	10,183	14,884	17,973	19,738	18,765	21,153	17,949	10,880	10,807
Nevada	10,100	3,594	4,695	9,078	10,030	12,238	11,184	7,592	7,874	9,848
New Mexico	1,435	339	1,506	1,910	1,817	3,210	3,480	8,025	7,805	11,190
Oklahoma	64,080	41,552	62,855	66,904	71,358	79,946	89,704	51,680	43,687	46,818
Oregon			38	14	(¹)	3		6	3	7
Pennsylvania	7									
South Dakota	6	2						69	37	
Tennessee	1,880		732	1,250	937	448	800	42	244	348
Texas			4	13		8				
Utah	66,898	44,594	67,606	101,724	116,955	153,335	147,635	151,285	145,915	149,877
Virginia					1,275	2,229	2,450	2,393	(¹)	(¹)
Washington	2,749	72	691	1,453	1,998	2,814	2,273	478	842	805
Wisconsin	2,594	972	691	734	1,254	1,877	1,546	2,069	1,098	1,384
	496,814	414,491	477,633	547,217	596,068	684,439	683,917	665,489	627,153	647,994

¹ It is said that a few tons of lead ore was mined at the old Chipman mine in Massachusetts and was treated at a small smelter owned by the mine operators.

² 513 pounds.

³ Bureau of Mines not at liberty to publish figures.

⁴ Exclusive of the output from Virginia, figures for which the Bureau of Mines is not at liberty to publish.

BY DISTRICTS

The following table gives the lead production in the principal lead-producing districts.

Mine production of recoverable lead in the principal lead-producing districts of the United States, 1921-1929, in short tons

District	State	1921	1922	1923	1924	1925
Southeastern Missouri region	Missouri	178,736	176,708	167,458	187,737	208,915
Coeur d'Alene region	Idaho	94,543	91,216	114,426	118,327	120,856
Joplin region	Kansas, Missouri	61,681	87,771	86,787	92,110	105,372
Bingham	Utah	12,288	19,348	26,353	24,786	54,982
Tintic	do	18,188	23,046	38,900	40,674	45,889
Park City region	do	8,840	16,129	26,220	33,346	40,656
San Juan Mountains	Colorado	5,323	5,332	13,167	13,148	17,092
Rush Valley	Utah	607	588	1,284	926	2,465
Butte	Montana	6,614	10,070	12,745	13,648	12,091
Barker	do	850	971	846	1,272	1,420
Willow Creek	New Mexico					
Leadville	Colorado	1,762	2,742	2,804	4,795	6,931
Central	New Mexico	2	176	656	592	1,182
Pioche	Nevada	666	932	2,697	2,162	2,806
Banner	Arizona	10	2,140	1,731	160	49
Jack Rabbit	Nevada	868	791	636	120	224
Tybo	do				1	3
Dome	Idaho	24			276	602
Upper Mississippi Valley	Iowa, Northern Illinois, Wisconsin	1,506	1,259	948	1,039	2,365
Warm Springs	Idaho	2,374	1,759	925	357	354
Cedar Plains	Montana	140	23	274	230	175
Blasée (Warren)	Arizona	2,335	4,187	4,221	5,719	6,657
Pend d'Oreille	Idaho	1	239	304	67	113
Vulture	Arizona		3		4	28
Kentucky-Southern Illinois	Kentucky-Illinois	207	849	1,220	1,254	675
Aspen	Colorado	1,198	1,778	1,486	455	1,820
San Francisco	Utah	442	1,119	1,459	1,069	1,990
Inyo County	California	633	3,136	4,744	2,806	3,098
Yellow Pine	Nevada	93	158	786	3,855	2,444
Montana	Montana	749	776	912	1,241	1,427
Magdalena	New Mexico	30	462	753	818	1,217
Spruce Mountain	Nevada	458	237	261	506	647
Eureka	do	388	251	673	888	2,163
Summit County	Colorado	262	280	1,946	1,105	1,707
Bryant	Montana	83	188	998	1,743	1,150
Eagle County	Colorado	6	161	230	1,145	2,041
Texas	Idaho	971	2,168	2,167	2,167	1,714
Big and Little Cottonwood	Utah	2,230	2,754	8,851	3,126	3,374
Northport	Washington	50	534	1,062	1,232	2,218
Troy	Montana	377	1,628	1,058	630	259
Opbri	Utah	480	2,921	3,443	1,414	2,945
Port Hill	Idaho	750	1,440	2,116	2,317	1,506
New Placers	New Mexico				3	125
Mammoth	Nevada	3	5	2	21	331
Bell	do		612	1,160		128
Embrae	Tennessee		732	1,260	937	448
Austinville	Virginia				1,276	782
Boulder	Idaho				(¹)	(¹)
Oro Blanco	Arizona				1	4

¹ Not listed according to rank.

² Bureau of Mines not at liberty to publish figures.

LEAD BALANCE SHEET OF THE UNITED STATES

Most of the lead ore mined in the United States has its lead content emerges in marketable form as primary lead and antimonial lead. Some of the ore is used in the manufacture of pigments and some may be exported, but for the amount exported has been negligible. The total lead in the various products in which it is produced is compared in the following table with the estimates of the lead in the ore mined. The table begins with 1907, the first year for which the mine production of the entire country was determined. The quantities compared were determined by independent producers and consumers of lead-bearing ore, the table is based on the accuracy of the figures.

Comparison of mine production of lead and primary domestic lead accounted for in lead products manufactured in the United States, 1907-1929, in short tons

Year	Mine production	Primary lead accounted for in—				Primary excess (+) or mine excess (—)
		Refined lead	Antimonial lead	Pigments made from ore	Total	
1907-1918.....	5,653,706	5,249,081	124,168	121,760	5,503,330	-150,376
1919.....	429,589	424,433	8,339	11,179	443,954	+14,565
1920.....	496,814	476,849	8,032	15,212	500,093	+3,279
1921.....	414,491	398,222	6,675	9,989	414,886	+395
1922.....	477,633	468,746	5,810	13,892	488,448	+10,815
1923.....	547,217	543,841	9,794	13,612	567,247	+20,030
1924.....	566,068	566,407	11,741	15,338	593,486	+27,418
1925.....	684,439	654,921	12,500	17,047	684,468	+29
1926.....	683,917	680,686	16,820	13,685	710,990	+27,073
1927.....	665,489	668,320	14,044	13,136	695,500	+29,011
1928.....	627,153	626,202	15,825	10,465	652,492	+25,339
1929.....	647,965	672,498	14,751	9,429	696,678	+48,713
1919-1929.....	6,270,805	6,181,124	124,131	143,484	6,448,742	+177,927
1907-1929.....	11,924,511	11,430,205	248,299	265,240	11,952,072	+27,561

¹ Includes 8,325 tons of lead produced from domestic ores smelted outside the country.

² Includes 8 tons of lead produced from domestic ores smelted outside the country.

³ Exclusive of the output from Virginia, figures for which the Bureau of Mines is not at liberty to publish.

The table shows substantial discrepancies for individual years, probably due largely to variations in mine and smelter stocks of ore, but in none of the last 11 years has the discrepancy exceeded 7 per cent of the total lead produced. From 1907 to 1918, inclusive, mine production exceeded the primary lead accounted for in lead products by 150,000 tons, or less than 3 per cent. From 1919 to 1929 there was an excess of nearly 178,000 tons, or less than 3 per cent, in favor of the lead contained in the lead products. For the total period of 23 years, 1907 to 1929, the primary lead content of lead products exceeded mine production by only 27,561 tons, or about one-fifth of 1 per cent.

LEAD SMELTERS AND REFINERIES

A list of lead smelters and refineries in the United States, Canada, and Mexico follows.

UNITED STATES

Arizona: Douglas—Phelps Dodge Corporation. (Smelter.)
 California: Selby—Selby plant, American Smelting & Refining Co. (Smelter and refinery.)
 Colorado:
 Durango—Durango plant, American Smelting & Refining Co. (Smelter.)
 Leadville—Arkansas Valley plant, American Smelting & Refining Co. (Smelter.)
 Idaho: Bradley—Bunker Hill & Sullivan smelter. (Smelter and refinery.)
 Illinois:
 Collinsville—St. Louis Smelting & Refining Works of National Lead Co. (Smelter and refinery.)
 Federal—Federal plant, American Smelting & Refining Co. (Smelter.)
 Indiana:
 East Chicago—International Lead Refining Co.¹ (Refinery only.)
 East Chicago—U. S. S. Lead Refinery (Inc.)² (Refinery only. Betts electrolytic process.)
 Kansas: Galena—Galena plant, Eagle-Picher Lead Co. (Smelter.)
 Missouri:
 Granby—American Zinc, Lead & Smelting Co.³ (Smelter.)
 Hercules—St. Joseph Lead Co. (Smelter and refinery.)
 Joplin—Eagle-Picher Lead Co. (Smelter.)
 Montana: East Helena—East Helena plant, American Smelting & Refining Co. (Smelter.)

¹ Subsidiary of Anaconda Copper Mining Co.

² Subsidiary of the United States Smelting, Refining & Mining Co.

³ Plant idle in 1929.

Nebraska: Omaha—Omaha plant, American Smelting and refinery.)

New Jersey:

Newark—Balbach plant, U. S. Metals Refining Co.
 Perth Amboy—Perth Amboy plant, American Smelting & Refining Co. (Smelter and refinery.)

Oklahoma: St. Louis—Ontario smelter, Eagle-Picher Lead Co.

Pennsylvania: Carnegie—Pennsylvania Smelting & Refining Co.

Texas: El Paso—El Paso plant, American Smelting & Refining Co.

Utah:

International—International Smelting Co.¹ (Smelter and refinery.)

Midvale—United States Smelting, Refining & Smelting Co.

Murray—Murray plant, American Smelting & Refining Co.

CANADA

British Columbia: Trail—Consolidated Mining & Smelting Co. (Ltd.). (Smelter and refinery. Betts electrolytic process.)

Ontario: Galetta—Kingdon Mining, Smelting & Refining Co. (Smelter.)

MEXICO

Chihuahua: Chihuahua—American Smelting & Refining Co.

Coahuila: Torreon—American Metal Co. (Ltd.). (Smelter.)

Hidalgo: Zimapan—Cia. Min. y Fundidora de Zimapan. (Smelter.)

Nuevo Leon:
 Monterrey—American Metal Co. (Ltd.). (Smelter and refinery.)

Monterrey—American Smelting & Refining Co. (Smelter.)

San Luis Potosi: San Luis Potosi—American Smelting & Refining Co. (Smelter.)

Minera Asafco. (Smelter.)

Zacatecas: La Fé—American Metal Co. (Ltd.). (Smelter.)

TARIFF

The tariff act of 1930 became effective January 1, 1930. It changed the lead schedules from the 1922 act. The chief value of lead, but not plated with platinum, or colored with gold lacquer, whether partly or wholly manufactured in the United States. The import duty on articles coming under the tariff act of 1930 increased from 40 per cent to 45 per cent ad valorem. The table gives the duties on the principal lead imports.

Import duties on lead commodities under the tariff act of 1930

Paragraph	Item
391	Lead-bearing ores, flue dust, and matte of all kinds. (Duty on lead content of copper, gold, or silver ores, or copper matte, 10 per cent.)
392	Lead bullion or base bullion, lead in pigs and bars, lead scrap, scrap lead, antimonial lead, antimonial scrap lead, Babbitt metal, solder, all alloys or combinations of lead provided for.
393	Lead sheets, pipe, glazier's lead, and lead wire.
397	Articles or wares n. s. p. l., if composed wholly or in chief value of lead, but not plated with platinum, gold, or silver, or colored with whether partly or wholly manufactured.
72	Lead pigments:
	Litharge.....
	Orange mineral.....
	Red lead.....
	White lead.....
	All pigments containing lead, dry or in pulp, or ground in oil or water, n. s. p. l.
40	Lead chemicals:
	Acetate, white.....
	Acetate, brown, gray, or yellow.....
	Nitrate, arsenate and resinate.....
	All other lead compounds, n. s. p. l.

¹ Subsidiary of Anaconda Copper Mining Co.

zinc concentrates and 6,335 tons of 74.47 per cent lead concentrates were produced. The lead concentrates contained 58.35 ounces of silver per ton. Operations were conducted at a loss, but profitable operations were expected to result from the completion of additional mill capacity and a lead smelter in 1930.

SPAIN

Spain produced nearly 143,000 metric tons of pig lead in 1929, an increase of 9 per cent over the 1928 output and equal to 8 per cent of the world total. Exports amounted to 99,600 tons, of which 2,400 tons was argentiferous lead. France took 33 per cent of the total, England 17 per cent, Russia 16 per cent, and Italy 13 per cent. The mine production of Spain in 1929 amounted to 180,890 tons of ore containing from 25 to 70 per cent of lead, as compared with 177,059 tons in 1928. Fifty-three per cent of the 1929 output came from the Province of Jaen, 18 per cent from Cordoba, 13 per cent from Murcia, and 5 per cent from Ciudad Real. The low price of lead prevailing during the latter part of 1929 and the early part of 1930 is reported to have resulted in the closing down of many of the smaller properties. Only the richest of the mines, or those where lead is found in large quantities and near the surface, are able to operate.

TUNISIA

Tunisia produced 18,850 metric tons of pig lead in 1929, as compared with 17,606 tons in 1928. Exports of pig lead in 1929 amounted to 18,296 tons, 88 per cent of which went to France. Production of lead ore was reported at 32,000 tons in 1929 and 34,000 tons in 1928.

YUGOSLAVIA

Developments in the lead industry of Yugoslavia in 1929 have been summarized by the Mining Journal (London) as follows:

The country's output in lead ores advanced in 1929 to 26,250 tons, against 15,886 tons in the preceding year; the bulk of the production emanated from the Slovenian properties of the Central European Mines (Ltd.). This company treats imported parcels of ore, which swell the figures of the lead output from their smelter. Official returns appear to indicate that roughly some 9,460 tons of pig lead were produced in Yugoslavia during the year. Toward the close of the year lower metal prices began to adversely affect the production of lead ores by small producers. The foregoing figures solely refer to lead ores mined, smelted or sold, and do not take into account the considerable tonnage of lead-zinc ores which are being accumulated at surface on some of the Southern Serbia and Macedonia properties and await treatment. In July, 1929, the issue in London and oversubscription of 1,600,000 shares of Trepcia Mines (Ltd.) attracted attention to this very promising enterprise, which has been hard at work since 1926 exploring and developing the Stantrg ore deposits near Kosovska Mitrovica. These deposits have now been sufficiently opened up for the estimate to be made that there are close upon three million tons of good-grade lead-zinc ore above the level of the lowest adit. Since last summer secondary development has been proceeding apace, and a very substantial tonnage is now blocked out; an aerial ropeway, a power plant, and a fine modern 500-ton mill are approaching completion. It is hoped that toward the autumn the producing stage will be reached, and even with to-day's depressed metal prices this concern should be able to give a very good account of itself. The company's concession and prospecting rights cover roughly 400 square miles, and there are indications of ancient mining operations in several localities, notably at Meljenica and

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Trepca Hill, where recent exploration has shown ore beneath the old surface and underground workings. The original prospectors of this concession located some fields in the Government laboratories, showed values of over 100 ounces of silver per ton, and it seems possible that systematic work will reveal the source of this eroded material. To the south of the concession are the extensive areas held by Novo Brdo Mines, which amalgamated a few months ago with a sister concern, the merger being the normal outcome of a situation in which the concession was completely surrounded by the big holdings of the Novo Brdo Co. The joint capital is now £300,000 sterling, which will continue the development of the lead-zinc ore bodies to further examine the extensive tract of ancient workings in the Novo Brdo areas. More recently Selection Trust Ltd. and London two other companies to take over concessions in the Trepcia areas and in the vicinity of Mount Kopaonik. The Trepcia concessions extending northward from this mountain to the Penarroya Co., have been for some time past exploring and developing workings known as the Plana Mine. The American company, which acquired a year or two ago the Srebrenitsa Concession, is again an extensive concession comprising an ancient mine, which was also a scene of mining activity during the war, but however, the opening up of the property has not been possible.

IMPORTS

The United States imports much lead in various forms, in base bullion and ore. Some of it is entered for consumption; most of it, however, is entered for export, and refined in bond, and is then reexported as manufactured products.

GENERAL IMPORTS

The following table shows the general imports of lead into the United States, by classes, including both imports for consumption and imports for warehouse. Of the total imports in 1929, as shown in the table, 71.6 per cent was in ore, and 27.0 per cent in ore, and 1.4 per cent in pigs, etc. Imports in base bullion decreased 35 per cent in 1929, and in pigs, bars, etc., increased 21 per cent and 25 per cent respectively. Total imports of lead decreased 25 per cent.

Lead imported into the United States, 1920-1929, by classes

(General imports)

Year	Lead in ore	Lead in bullion
1920	29,488,816	96,102
1921	14,234,039	71,732
1922	24,474,481	124,074
1923	60,882,850	161,496
1924	95,320,168	155,501
1925	88,962,602	141,253
1926	116,232,669	150,570
1927	81,276,126	236,587
1928	81,829,630	266,936
1929	62,661,871	106,141

General imports of lead by countries are shown in the following table. Mexico supplied nearly 76 per cent of the total imports (principally Peru) 20 per cent, Canada 4 per cent.

¹¹ Mining Journal (London): Vol. 170, July 26, 1930, p. 603.

other countries only small amounts. Imports from Mexico and Canada decreased 36 per cent and 37 per cent, respectively, while imports from South America increased 131 per cent. The decrease from Mexico was due to a large decline in imports of base bullion, while the decrease from Canada was due to lower imports of lead in ore. The increased shipments of base bullion from Peru and lead ore from Chile accounted for the increase of lead imports from South America.

Lead imported into the United States, in ore, base bullion, and refined, 1920-1929, by sources, in pounds

[General imports]

Year	Canada	Mexico	South America	Europe	Other countries	Total
1920	8,199,234	131,815,278	7,085,351	38,768,504	11,163,100	197,029,527
1921	8,127,158	103,408,238	2,381,365	30,224,701	1,430,949	145,970,411
1922	8,663,236	141,216,136	3,803,478	1,454,384	513,335	155,650,569
1923	11,399,671	242,362,788	6,300,161	11,094,938	520,635	270,678,193
1924	11,022,091	243,161,069	16,820,867	4,857,805	453,432	276,315,264
1925	11,646,820	218,873,800	10,621,737	1,594,963	1,621,110	244,258,430
1926	12,316,740	264,203,852	24,790,075	3,224,504	572,856	295,108,027
1927	6,214,867	300,634,815	14,583,513	1,226,049	118,883	322,778,127
1928	14,382,907	275,184,287	20,408,574	30,400	85,915	310,088,149
1929	9,023,527	175,872,302	47,062,018	29,441	141,480	232,118,768

The following tables show the imports of lead in ore, matte, and base bullion, by countries.

Lead imported into the United States, in ore and matte, 1925-1929, by countries, in short tons

[General imports]

Country	1925	1926	1927	1928	1929
Mexico	32,971	40,239	30,615	17,055	23,415
Canada	4,978	5,524	2,558	7,131	3,953
Peru	1,887	3,642	3,415	1,170	1,601
Chile	3,216	7,506	3,877	504	2,295
Other countries	1,429	1,115	173	55	67
	44,481	58,116	40,638	25,915	31,331

Imports of lead in ore from Mexico increased 37 per cent in 1929, amounting to 75 per cent of the total. Imports from Chile and Peru increased considerably, while imports from Canada decreased 45 per cent.

Lead imported into the United States, in base bullion, 1925-1929, by countries, in short tons

[General imports]

Country	1925	1926	1927	1928	1929
Mexico	70,605	77,063	117,038	110,935	63,438
Peru		436		8,525	10,605
Spain			330		
Other countries	122	187	10	8	8
	70,627	78,285	118,284	128,468	83,071

Imports of base bullion from Mexico in 1929, although 47 per cent less than in 1928, amounted to 76 per cent of the total. The decrease was due to the completion of additional refinery capacity during the year, which permitted an increase in the quantity of

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refined lead produced in Mexico. Imports from Peru decreased 10 per cent owing to increased production in that country by the de Pasco Copper Corporation.

IMPORTS FOR CONSUMPTION

Statistics on imports for consumption, as shown in the following, include imports entered for immediate consumption, withdrawals from warehouse for domestic consumption, and lead imported is smelted and refined in bond, and lead imported for consumption is considerably less. In 1929 imports for consumption were equivalent to 100 per cent of the general imports of lead.

Lead remaining in warehouse in the United States, December 31, 1929, in pounds

[Stated in the form in which the material was entered]

Year	Ore (lead content)	Base bullion (lead content)	Pigs, bars, and old	Year	Ore (lead content)
1920	32,023,299	69,517,026	2,644,576	1925	105,114
1921	13,679,465	24,692,350	693,440	1926	99,000
1922	10,470,144	7,076,544		1927	97,000
1923	76,031,259	80,306,855	5,234,475	1928	120,000
1924	103,474,370	95,007,016	10,090,668	1929	

¹ Includes lead in reclaimed scrap, as follows: 1925, 241,078 pounds; 1926, 241,078 pounds. Corresponding figures for earlier years and for 1928 and 1929 are not available.

Lead imported for consumption in the United States, 1925-1929, in short tons

Year	Lead in ore ¹		Lead in base bullion		Pigs, bars, and old ²	
	Pounds	Value	Pounds	Value	Pounds	Value
1920	13,700,499	\$908,619.66	816,820	\$3,340,600.68	901,959	\$3,887,821.10
1921	6,842,807	327,046.56	819,980	1,668,410.61	909,094	1,972,715.10
1922	23,649,851	812,603.65	859,422	1,815,142.8	596,110	335,924.10
1923	22,420,046	1,190,414.11	124,151	626,201.42	925,265	2,510,466.10
1924	65,147,976	3,145,791.28	796,402	1,542,167.27	362,392	1,850,862.10
1925	12,706,508	672,112.24	747,665	1,823,566.15	464,205	1,038,850.10
1926	68,762,911	3,393,039.13	341,038	972,758.24	365,570	2,191,658.10
1927	70,200,478	4,203,897.14	629,000	925,328.9	933,679	635,807.10
1928	46,511,608	2,626,231.13	597,455	850,983.20	488,373	808,359.10
1929	21,645,699	1,160,633.12	395,696	627,455.20	178,922	1,052,037.10

¹ Includes also lead in matte beginning Sept. 22, 1922.
² Includes also imports of antimonial lead, antimonial scrap lead, babingtonite, and lead with tariff of Sept. 22, 1922, except in 1925, 1926, 1927, 1928, and 1929, when not available.

Type metal¹ imported for consumption in the United States, 1925-1929, in short tons

Year	Type metal	
	Pounds	Value
1920	29,887,353	\$1,340,851.10
1921	38,971,851	1,110,112.10
1922	44,895,120	1,112,433.10
1923	(?)	(?)
1924	(?)	(?)
1925	6,009,161	1,460,161.10
1926	8,621,175	2,191,658.10
1927	878,346	635,807.10
1928	2,990,548	808,359.10
1929	4,440,857	1,052,037.10

¹ Classification for 1925 to 1929 is "Type metal and antimonial lead"; see page 257.
² Figures not available.

EXPORTS

The chief lead export of the United States is refined pig lead, derived from the treatment of foreign ores and base bullion. Exports decreased 37 per cent in 1929 owing primarily to a 35 per cent decrease in imports of base bullion. Europe took 69 per cent of the total, Asia 25 per cent, South America 5 per cent, and North America and Africa 1 per cent. In Europe United Kingdom, Germany, and Sweden, were the largest importers, while in Asia and South America Japan and Brazil ranked first. United Kingdom took 32 per cent and Japan 22 per cent of the total United States exports.

Refined pig lead exported from the United States, 1928 and 1929, by destinations

Destination	1928		1929	
	Short tons	Value	Short tons	Value
North America:				
Canada.....	132	\$18,755	141	\$24,662
Cuba.....	152	18,249	248	27,899
Mexico.....	79	10,816	83	16,666
Panama.....	19	2,655	52	6,981
Other.....	90	11,567	109	23,107
	472	62,431	593	105,345
South America:				
Argentina.....	1,243	129,705	599	57,845
Brazil.....	2,439	242,303	1,638	164,185
Chile.....	404	56,284	989	112,220
Colombia.....	87	11,906	93	13,481
Ecuador.....	18	2,392	33	4,465
Peru.....	78	7,308	88	8,693
Uruguay.....	354	26,640	448	43,248
Other.....	91	9,181	64	7,318
	4,834	495,619	3,852	401,405
Europe:				
Belgium.....	861	74,649	806	92,065
Denmark.....	3,673	321,700	4,322	419,031
France.....	4,828	397,123	2,202	205,009
Germany.....	23,552	2,063,281	9,745	945,460
Italy.....	336	29,440	103	9,700
Netherlands.....	5,088	445,294	1,522	149,812
Sweden.....	5,316	465,050	7,255	708,216
United Kingdom.....	43,428	3,817,353	23,732	2,295,238
Other.....	279	25,418	872	84,945
	87,161	7,637,208	50,649	4,909,532
Asia:				
China.....	2,131	190,306	813	78,439
Japan.....	21,216	1,927,144	16,416	1,594,706
Philippine Islands.....	210	22,118	111	14,417
Other.....	245	24,296	715	74,184
	23,802	2,183,864	18,055	1,761,716
Africa.....	(1)	57	(1)	219
Oceania, French.....	(1)	42	(1)	121
	116,269	10,359,221	73,251	7,178,337

¹ Less than 1 ton.

The following table shows the exports of lead from the United States from 1920 to 1929. Of the lead exported with benefit of drawback in 1929 miscellaneous lead manufactures (principally lead foil) accounted for 55.1 per cent, white lead in oil 25.1 per cent, red lead and litharge 9.9 per cent, babbitt 2.7 per cent, lead-covered cable 2.0 per cent, type metal 1.8 per cent, shot shells 1.6 per cent, and others 1.8 per cent.

LEAD

Refined lead exported from the United States

Year	Pigs, bars, and old		Foreign lead exported in manufactures with benefit of drawback (short tons)	Year
	Short tons	Value		
1920.....	20,093	\$3,032,758	6,175	1925.....
1921.....	28,624	2,462,633	9,369	1926.....
1922.....	38,178	2,781,965	5,677	1927.....
1923.....	50,735	5,455,743	12,526	1928.....
1924.....	82,090	10,396,766	15,276	1929.....

CONSUMPTION

REFINED PRIMARY PIG LEAD AVAILABLE FOR CONSUMPTION

The manufacturing industry of the United States consumes refined primary pig lead in four main forms; namely, refined primary pig lead, primary lead in the form of monmetal lead, primary lead in the form of alloy lead. The following table shows the refined primary pig lead available for consumption from 1924 to 1929.

Refined primary pig lead available for consumption in short tons

	1924	1925
Supply:		
Stock in bonded warehouse Jan. 1.....	2,617	5,041
Imports of pigs, bars, and old.....	12,247	7,021
Production.....	680,493	766,965
	705,357	779,027
Withdrawn:		
Exports—		
Pig lead.....	82,090	103,611
In manufactures, with benefit of drawback.....	15,276	11,699
Stock in bonded warehouse Dec. 31.....	5,045	5,161
	102,410	120,470
Supply available for consumption.....	602,947	658,557

¹ Includes 622 tons of "old, reclaimed, and scrap" in 1925 and 1926 for other years not separately recorded.

The foregoing table is intended to indicate the amount of refined primary pig lead smelted directly from ore. There are discrepancies which should be pointed out. The production from secondary materials some from the primary lead in the final product reported above may therefore contain some errors. There are also some inaccuracies in the statistics of warehouse transactions. Nevertheless, available, the table gives approximately the amount of lead available to American industry each year. It does not consider fluctuations in producers' prices, but indicates the trend of consumption only in

CONSUMPTION BY USES

The quantity of lead in all forms consumed by American industry is estimated each year by the American Bureau of Metal Statistics. The following table is quoted in full from the 1929 yearbook of that bureau.

Lead consumed in the United States,¹ 1921-1929, in short tons

Purpose	1921	1922	1923	1924	1925	1926	1927	1928	1929
White lead ²	136,000	156,000	130,000	150,000	131,000	120,000	126,000	123,000	118,000
Red lead and litharge ³	23,500	30,000	46,000	34,000	42,000	36,000	38,000	31,000	30,000
Storage batteries ⁴	87,000	130,000	143,000	170,000	180,000	190,000	176,000	220,000	210,000
Cable covering ⁵	67,000	93,000	131,000	138,000	156,000	185,000	161,000	180,000	204,000
Building ⁶	48,000	71,000	75,000	83,100	88,400	93,700	88,000	96,000	96,000
Automobiles ⁷	4,800	7,200	10,400	10,700	12,800	16,700	12,000	17,000	18,000
Locomotives ⁸	1,000	850	2,100	1,100	750	1,000	650	500	800
Railway cars ⁹	1,800	2,500	6,000	4,500	4,450	3,800	2,700	2,000	3,500
Shipbuilding ¹⁰	850	100	100	200	100	100	200	100	100
Ammunition ¹¹	20,500	38,000	40,000	27,000	31,500	32,000	34,000	39,000	41,100
Terneplate ¹²	3,000	4,700	4,600	4,050	4,500	4,100	3,900	4,400	4,200
Foil ¹³	23,800	25,800	33,300	35,000	32,500	35,000	30,000	35,000	30,000
Bearing metal ¹⁴	21,000	26,000	28,000	32,000	34,000	35,000	31,000	32,000	33,000
Solder ¹⁵	12,000	20,000	30,000	30,000	35,000	37,000	35,000	37,000	37,000
Type metal ¹⁶	7,500	9,500	12,000	13,000	15,000	16,000	16,000	17,000	18,000
Calking ¹⁷	20,000	20,000	25,000	26,500	30,000	32,000	30,000	32,000	31,500
Castings ¹⁸	12,000	15,000	15,000	16,000	18,000	18,000	17,000	18,000	18,000
Other uses ¹⁹	25,000	32,500	36,500	38,000	40,500	44,500	40,500	46,000	50,000
Total	520,550	683,150	768,000	812,150	856,500	900,900	840,950	930,600	945,200

¹ Source: American Bureau of Metal Statistics. These estimates are for the total consumption of lead, irrespective of whether its origin be primary or secondary. Antimonial lead is included and the manufacture of lead for export under drawback provisions is included.

² From data of sales reported by U. S. Bureau of Mines up to 1929. The 1929 figure has been determined by an improved method.

³ Exclusive of oxides for storage batteries; based on data of sales reported by U. S. Bureau of Mines.

⁴ Based on reports from a large proportion of manufacturers in each industry.

⁵ Based upon estimates of manufacture and the quantity of lead used per unit of manufacture. Under the heading of building is included the lead used in chemical construction.

⁶ Based on the estimated consumption of lead for this purpose by the railways.

⁷ Estimates of persons engaged in the trades.

⁸ Conjectural.

According to the foregoing table the lead consumed in the United States in 1929 amounted to 945,200 tons, which surpassed all previous records. It exceeded 1928, the previous peak, by 2 per cent and 1927 by 12 per cent. In the following table the individual uses of lead have been arranged according to their percentage relation to the total lead consumed in 1929.

Percentage of total lead consumed in the United States, 1925-1929, by uses

Use	1925	1926	1927	1928	1929
Storage batteries	21.02	21.09	20.81	23.64	22.22
Cable coverings	18.21	20.53	19.16	19.34	21.70
White lead	15.29	13.32	14.98	13.22	12.48
Building	10.32	10.40	10.46	10.32	10.16
Ammunition	3.68	3.55	4.04	4.26	4.26
Solder	4.09	4.11	4.18	3.98	3.92
Bearing metal	3.97	4.00	3.69	3.44	3.49
Calking	3.50	3.55	3.57	3.44	3.33
Litharge and red lead	4.90	4.00	4.52	3.33	3.17
Foil	3.80	3.89	3.57	3.76	3.17
Castings	2.10	2.00	2.02	1.93	1.91
Type metal	1.75	1.78	1.00	1.83	1.91
Automobiles	1.49	1.85	1.43	1.83	1.91
Terneplate	.63	.46	.46	.47	.44
Railway cars	.62	.42	.32	.21	.37
Locomotives	.09	.11	.08	.05	.08
Shipbuilding	.01	.01	.02	.01	.01
Miscellaneous	4.78	4.94	4.82	4.94	5.29

LEAD

For several years storage batteries have been of lead, having taken first place from white lead. Batteries are used chiefly in motor vehicles; a 23 per cent increase in the number of passenger automobiles manufactured in 1929 there was a decrease of 5 per cent of lead used in storage batteries. The decrease is partly due to the larger percentage of light-weight storage batteries in 1929 which operate with the smaller-size storage batteries. The American Bureau of Metal Statistics has estimated the use of automobile storage batteries at 16,000,000 in 1928. Exports of storage batteries have a total of about 207,000 in 1925 to 353,000 in 1929, a 70 per cent increase. The expansion of the automobile trade has more than offset the decreased use of storage batteries which resulted from the introduction of radios. The American Bureau of Metal Statistics has estimated that 170,000 radio storage batteries were manufactured in 1929 compared to 400,000 in 1928.

The second largest use of lead is as a protective covering for wire cables used chiefly in telephone and underground lines. The quantity of lead so used in 1929 was 180,000 tons, being 14 per cent greater than in 1928 and 23 per cent greater than in 1926, the previous peak. This use is expected to increase. General expansion of the use of electric power in large electrification programs announced by some of the power companies and the development of the Boulder Dam power projects, will require large amounts of lead. Increasing civic consciousness will undoubtedly result in many of the present unsightly overhead telephone lines being run underground, as has been the custom in many cities for many years. Research work now in progress to develop remedies of certain defects in lead-covered wire and its use in some industries should yield profitable results in the market for lead-covered cable.

The manufacture of white lead is the third largest use of lead and in 1922 totaled 136,000 tons or 23 per cent of the total estimated lead consumed in 1922. In 1929 it was 118,000 tons (12.5 per cent of the total) was substituted by the use of white lead is due to the substitution of white lead pigments, principally lithopone and zinc oxide, in the use of nitrocellulose paints in which white lead is used. F. E. Wormser, secretary of the Lead Industries Association, summarized developments in the uses of lead as follows:

A lead-calcium alloy which is finding wide use as a substitute for lead in the manufacture of lead pipe has been developed. The addition of a very small amount of calcium to lead produces an alloy of high-fatigue strength which is resistant to corrosion and is much as are often encountered in cables due to expansion and contraction. Lead linings are now being extensively used in the manufacture of lead pipe principally because of developments which have produced secrets of the various processes of manufacturers making lead pipe usually carefully guarded, but the fact remains that

proved. Could production costs be lowered, these products would find even a wider field of application. Large amounts of antimonial-lead sheet are used as tank linings in the chromium-plating industry, and it has been found that if these tanks are put into use immediately or if they are given anodic treatment at regular weekly intervals the life of the lining can be extended.

Lead-coated copper is finding growing favor for architectural purposes such as roofing, leaders, and gutters. It has the advantage of the strength and durability of copper plus the protection and durability of lead. Moreover, it furnishes a pleasing gray color, one which will not discolor, and is less expensive than antimonial or soft lead. This gives it a wide field of application on moderate-priced dwellings where lead is considered too expensive. Here it is interesting to note that results of tests on the corrosion of metals by city air, conducted in England and reported by J. Newton Friend, place antimonial lead at the top, followed by the highest grade of tin and then soft lead. Other metals, common tin, stainless steel, aluminum, copper, and so on, follow.

The Bureau of Standards is doing considerable research work on lead-bearing metals at the request of the Assistant Secretary of War with the idea of replacing tin-base bearings with lead base for the purpose of conserving tin. E. J. Hall, of the Hall-Scott Motor Car Co., reports that they have found that a bearing composed of 25 per cent lead and 75 per cent copper, having no lining or backing but a uniform structure throughout the shell, will give from 100,000 to 150,000 miles of intercity motor-coach service without attention.

The use of a baked red-lead priming coat under asphaltic or bituminous protective coatings on underground pipe lines has only received general attention in the last six months, although such a coat has been successfully used in isolated instances for several years. This red-lead coat has been entered in the tests of coating materials being conducted by the American Gas Association and the Bureau of Standards and should give a good account of itself judging by past performances. This may open another large market for red lead.

PRICES

The two major markets for lead in the United States are at New York and St. Louis, and a large part of the lead produced in the United States is sold at prices based on the quotations in these two markets. The New York quotations are influenced to some extent by the lower prices usually prevailing on the London market, so that the New York price seldom exceeds the St. Louis price by as much as the freight differential, 0.35 cent a pound. The following tables show the average monthly quoted prices of lead at St. Louis, New York, and London for 1927, 1928, and 1929 and the average yearly prices from 1910 to 1929, inclusive.

Average monthly quoted prices of lead at St. Louis, New York, and London, 1927-1929, in cents per pound¹

Month	1927			1928			1929		
	St. Louis	New York	London	St. Louis	New York	London	St. Louis	New York	London
January.....	7.41	7.68	8.95	6.29	6.50	4.74	6.50	6.65	4.79
February.....	7.29	7.48	8.92	6.08	6.34	4.41	6.73	6.86	5.01
March.....	7.34	7.61	8.03	5.82	6.03	4.34	7.38	7.50	5.60
April.....	8.84	7.11	8.76	6.00	6.11	4.42	7.03	7.20	5.37
May.....	6.27	6.60	5.43	6.02	6.13	4.46	6.77	7.00	5.19
June.....	6.16	6.43	5.30	6.16	6.30	4.57	6.80	7.00	5.13
July.....	6.13	6.36	5.09	6.05	6.22	4.47	6.68	6.80	4.94
August.....	6.39	6.68	5.02	6.05	6.23	4.68	6.69	6.89	5.10
September.....	6.05	6.31	4.66	6.29	6.46	4.77	6.55	6.76	5.02
October.....	6.01	6.26	4.45	6.32	6.50	4.78	6.67	6.87	5.08
November.....	6.04	6.28	4.55	6.23	6.39	4.60	6.13	6.29	4.71
December.....	6.33	6.51	4.83	6.24	6.50	4.62	6.10	6.25	4.68

¹ St. Louis: Metal Statistics, 1930, p. 389. Average daily quotations of soft Missouri lead, f. o. b. St. Louis (open market), as reported daily in the American Metal Market.

New York: American Metal Market, daily issues. Pig lead, New York (outside market) prompt shipment from west.

London: Metal Statistics, 1930, p. 393. Average price of foreign lead. Price per long ton, as published in Metal Statistics, converted to cents per pound at average exchange rate reported by the Federal Reserve Board.

LEAD

Average yearly prices of lead at St. Louis, New York, and London, 1910-1929, in cents per pound¹

Year	St. Louis (per pound)	New York (per pound)
1910.....	4.23	4.41
1911.....	4.31	4.41
1912.....	4.38	4.46
1913.....	4.20	4.40
1914.....	3.74	3.87
1915.....	4.67	4.67
1916.....	6.80	6.83
1917.....	8.92	8.77
1918.....	7.25	7.41
1919.....	5.55	5.83
1920.....	7.93	8.00
1921.....	4.80	4.80
1922.....	5.52	5.77
1923.....	7.35	7.21
1924.....	8.10	8.00
1925.....	8.92	9.07
1926.....	8.25	8.47
1927.....	6.52	6.77
1928.....	6.14	6.33
1929.....	6.66	6.85

¹ Compiled from daily quotations in open market at St. Louis, as published in Metal Statistics, 1930, p. 387.

² Average prices based on quotations of American Smelting & Refining Co., as published in Metal Statistics, 1930, p. 381.

³ Prices of foreign lead in London, as published in Metal Statistics, 1930, p. 393. Prices of average exchange rates published in the reports of the Federal Reserve Board.

⁴ Import duties on refined pig lead have been as follows: Aug. 6, 1908, to Oct. 4, 1913, 25 per centum ad valorem; average tariff 25 per centum; Sept. 22, 1922, to date (1929)—2½ cents per pound.

To facilitate the settlement of its buying and selling of lead the American Smelting & Refining Co., which produces about 50 per cent of the United States lead production, at frequent intervals publishes an official contract price based on the New York quotations. The following are the company's quotations for 50-pound lead, in 1928 and 1929, in cents per pound:

1928			
Feb. 10.....	6.35	Feb. 11.....	6.35
Feb. 20.....	6.25	Feb. 19.....	6.15
Feb. 27.....	6.15	Feb. 27.....	6.15
Mar. 1.....	6.00	Mar. 6.....	6.00
Apr. 2.....	6.10	Mar. 19.....	6.20
May 25.....	6.20	Mar. 20.....	6.30
May 29.....	6.30	Apr. 4.....	6.20
July 9.....	6.20	Apr. 8.....	6.30
Aug. 23.....	6.30	Apr. 10.....	6.40
Aug. 27.....	6.40	Apr. 15.....	6.50
Sept. 17.....	6.50	July 5.....	6.60
Nov. 9.....	6.35	July 10.....	6.50
Dec. 3.....	6.50	Sept. 5.....	6.50
1929			
Jan. 2.....	6.65	Oct. 29.....	6.75
Feb. 6.....	6.75	Oct. 31.....	6.75
		Nov. 6.....	6.75
		Nov. 7.....	6.75

The gradual upward trend in the price of lead continued in the early part of 1929. The New York price for Jan. 2 and the average for that month was 6.65 cents per pound.

sharp increase from 6.50 cents on December 31, 1928. During February there was a further increase to an average of 6.86 cents. The threatened curtailment of supplies from Mexico, due to the revolution, sent the price to 7.80 cents toward the latter part of March, a high for the year, but following the collapse of the revolution the price quickly declined to 7 cents by April 15. Meanwhile rising prices had stimulated production, and when domestic consumption declined sharply in May and June stocks increased rapidly and the price of lead decreased to an average of 6.75 cents in August. In September and October the price increased slightly as a result of increased demand, but in November and December the price dropped off sharply following the stock-market crash. The New York quotation was 6.25 cents on December 31, 1929.

The average New York price for 1929 was 6.83 cents—8 per cent over the 1927 average and 24 per cent under the record 1925 price. The New York market averaged 0.17 cent per pound over the St. Louis market in 1928, as compared with 0.17 cent in 1928 and 0.23 cent in 1927.

The average price, including domestic and foreign sales, received by lead producers in the United States in 1929, according to reports of producers to the Bureau of Mines, was 6.3 cents a pound; in 1928 it was 5.8 cents.

The London quotation reached the high monthly average in March (equivalent to 5.5 cents per pound), declined to approximately 5 cents during July, August, September, and October, then declined further to 4.68 cents in December. The yearly average was 10 per cent greater in 1929 than in 1928; it was 1.79 cents per pound under the New York price in 1929, as compared to 1.73 cents in 1928.

GOLD, SILVER, AND LEAD

(MINE REPORT)

BY CHAS. W. HEN

SUMMARY

Metal mining in South Dakota is confined to Custer, Lawrence, and Pennington counties—the Black Hills. The Homestake gold mine in Custer County, has been producing almost continuously since 1891 and is now the largest producing gold mine in the world. This mine to the company's mills amounted to an increase of 20,986 tons over 1928, but the value of gold-silver bullion bars amounted to \$48,946.74 less than in 1928. The total production of gold and silver in South Dakota in 1929 was a lode mine in Pennington County. The following tables show the production of gold, silver, and lead in South Dakota from 1920 to 1929.

Mine production of gold, silver, and lead in South Dakota, 1920-1929, recovered metal

Year	Mines producing			Ore (short tons)	Gold (lode and placer)	
	Lode	Placer	Total		Fine ounces	Value
1920	3	4	7	1,356,189	226,224.23	\$4,876,470
1921	3	6	9	1,725,685	319,524.95	6,005,167
1922	3	3	6	1,751,454	315,207.99	6,817,788
1923	5	—	5	1,605,439	309,908.05	6,405,809
1924	4	—	4	1,674,720	298,930.24	6,117,421
1925	3	—	3	1,590,079	289,747.10	6,389,804
1926	2	2	4	1,415,755	276,528.93	6,778,876
1927	2	2	4	1,376,583	322,031.74	6,656,987
1928	5	4	9	1,422,233	317,378.94	6,580,805
1929	2	—	2	1,463,159	316,836.85	6,549,599

For total production of gold and silver in South Dakota, see Mineral Resources, 1929, pt. 1, p. 194.

Work on manuscript completed February, 1931.

For information on the geology of the metal resources of the Black Hills, see Geol. Atlas, Central Black Hills Folio 219, 1925.

the deposits are situated in the Atlas Mountains, and it is estimated from top layers alone 20,000,000 tons of bauxite can be obtained. The mineral has a large alumina content, estimated at 70 per cent, and numerous water falls in the vicinity can furnish water power for the mining operations.

GOLD COAST

Bauxite, or aluminum hydrate, with certain impurities is found over portions of Gold Coast Colony and Ashanti, as a cap from a few feet to more than 50 feet in thickness on the summits of several mountains. The ores are of fairly high grade. The most important deposits are those of the Mount Ejuanema Kwahu district, estimated to have about 4,000,000 tons, and the Sefwi Bekwai district, having about 33,300,000 tons.³⁹ The Ejuanema deposit is 1½ miles in a direct line from Nkawkaw railway station, which is 111 miles from the port of Accra or 249 miles from the port of Sekondi, via Kumasi. The Sefwi Bekwai deposits are about 55 miles northwest of Dunkwa, a station on the Sekondi-Kumasi Railway 100 miles from Sekondi. Average analyses of 198 specimens from Mount Ejuanema and of 51 specimens from Mount Supirri, Sefwi, are reported as follows:

	Mount Ejuanema	Mount Supirri
SiO ₂	0.87	1.88
Al ₂ O ₃	60.66	55.77
Fe ₂ O ₃	9.27	10.30
TiO ₂	1.94	1.12
H ₂ O.....	26.80	30.69

During 1928 a large deposit of bauxite was discovered⁴⁰ 4 to 6 miles west of Kumasi, 18 miles northeast of Bibiani, at the end of a proposed Western Province Railway in the Yenahin district; the deposit is alleged to contain 180,000,000 tons of good ore. Bauxite deposits have been located in the Pusu Pusu River, Kibi district, in Akim, and small deposits at Atiwa Range, Sajumase.

Although the economic value of these Gold Coast deposits has not been demonstrated and the quantity of bauxite reported for the Yenahin district seems almost fantastic, it seems probable that the Gold Coast may be counted upon to provide reserves of substantial character for the future.

OCEANIA

AUSTRALIA

Bauxite deposits have been found 60 miles southwest of Gladstone, at Crows Nest near Toowoomba, and at Cooranga station in the Gaynah district, Queensland, and numerous high-iron bauxite deposits are known in New South Wales. Aluminous clays, alunite, and bauxitic clays have been found in other sections of Australia. Production has been small and has come chiefly from Victoria which produced 555 metric tons of bauxite in 1929, compared with 196 tons in 1928 and 936 tons in 1927.

³⁹ American Metal Market, Large Bauxite Deposits of Gold Coast: Dec. 9, 1927.
⁴⁰ Mining Journal, London, vol. 104, No. 4852, Mar. 10, 1929, p. 200.

LEAD AND ZINC PIGMENTS

By ELMER W. PER

SUMMARY

Lead pigments accounted for 55.6 per cent of the total value of the lead and zinc pigments produced in 1929, which amounted to an increase of 2 per cent over 1928. Lead pigments sold increased 2 per cent in total value. The quantity of lead pigments sold by producers increased to 295,460 short tons, a decrease of nearly 1 per cent over 1928. Large, red lead, and orange mineral increased 1 per cent, white lead in oil and white sublimed lead in oil were the lowest since 1918 and decreased 1 per cent of the 1922 sales. Sales of dry lead were practically unchanged in 1929. The quantity of zinc pigments sold by producers increased to 394,075 short tons, an increase of 2 per cent over 1928. Lithopone increased by 6,000 tons, or 3 per cent, zinc oxide increased by 3,000 tons, or 12 per cent, and decreased slightly. Sales of zinc chloride increased 2 per cent. Quoted prices of lead pigments in 1929 were higher than in 1928; the average price of lead pigments sold by producers, as reported to the Bureau of Mines, decreased only 3 per cent, whereas the average price of lead increased 8 per cent. Quotations for zinc pigments changed throughout the year at levels estimated. The average price received by producers per ton of lead pigments reported to the Bureau of Mines, decreased 3 per cent, the average quoted price for slab zinc increased 1 per cent. Imports of lead pigments in 1929 were 1,000 tons, an increase of 3 per cent in quantity and amounted to 3 per cent of the total quantity of lead pigments produced. The total quantity of white lead exported decreased 9 per cent, whereas the quantity of red lead exported increased 39 per cent. Imports of zinc pigments in 1929 decreased 5 per cent in value and were equivalent to 3 per cent of the sales by domestic producers. Lead and zinc oxide decreased, whereas lithopone doubled. Exports of zinc pigments in 1929 amounted to 1,000 tons, a quantity and 26 per cent in value and amounted to 3 per cent of the total domestic output.

⁴¹ Work on manuscript completed December, 1930.
⁴² The statistics in this report were prepared (unless otherwise noted) from records of the Bureau of Mines.
 J. A. Dorsey, both of the Bureau of Mines.

An outstanding feature of the lead and zinc pigment industry during the last few years has been the severe competition between white lead and other pigments for the paint trade. The effect of this is indicated roughly in the comparative sales of the pigments concerned. Sales of white lead in oil decreased from 153,000 tons in 1922 to 105,000 tons in 1929. During the same period the combined sales of zinc oxide and lithopone increased from 212,000 to 367,000 tons. As three-fourths of the lithopone and about one-third of the zinc oxide produced in the United States are used in the manufacture of paints, these figures indicate the marked degree to which the zinc pigments have invaded the paint industry at the expense of white lead.

The rapid expansion in the use of nitrocellulose paints in recent years has not altered materially the quantity of pigments consumed. The average nitrocellulose paint contains considerably less pigment per gallon than oil paints, but a heavy pigment priming coat and more finishing coats are used to offset the low pigment content. White lead is seldom used in these paints owing to its chemical instability in contact with nitrocellulose products. As it is desirable to have a low pigment content in nitrocellulose paints there has been an increased use of high hiding-power pigments such as high-strength lithopone and titanium pigments.

PRODUCTION

In this chapter the pigments and salts sold are considered as representing the production, no account being taken of the stocks on hand at the beginning and end of the year. The amounts used by producers at their own plants are included under sales. The total value of the lead and zinc pigments sold by domestic producers was approximately \$98,045,000 in 1929 as compared with \$96,136,000 in 1928, the increase being shared about equally by the lead and the zinc pigments. The quantity of lead pigments was slightly less in 1929 than in 1928, but the average value per ton was 3 per cent more; the average quoted price for pig lead was 8 per cent higher than in 1928. There was a slight increase in the quantity of zinc pigments in 1929, but the average value per ton was 0.3 per cent less than in 1928; the average quoted price for slab zinc increased 8 per cent over 1928.

The following tables show the production of lead and zinc pigments and zinc salts as reported to the Bureau of Mines.

Lead pigments sold by domestic manufacturers in the United States, 1920-1929, in short tons

Year	White lead		Basic lead sulphate or sublimed lead		Red lead	Orange mineral	Litharge
	Dry	In oil	White	Blue			
1920	83,678	112,017	12,412	928	134,431	(1)	62,829
1921	26,788	143,845	11,558	463	21,806	381	41,009
1922	41,598	153,393	13,765	972	30,509	379	58,281
1923	37,786	126,087	11,949	800	38,037	646	75,107
1924	42,622	144,872	14,572	1,088	36,813	331	74,724
1925	43,426	120,479	14,906	1,090	41,669	840	86,546
1926	37,968	111,845	12,271	1,236	42,550	813	82,540
1927	38,069	119,026	13,482	1,061	39,073	709	81,555
1928	42,049	111,923	16,002	1,234	40,497	459	85,670
1929	42,150	104,872	15,580	1,234	43,021	678	87,916

¹ Figures for red lead include orange mineral prior to 1921.

LEAD AND ZINC PIGMENTS AND SALTS

Lead pigments sold by domestic manufacturers in the United States, 1920-1929, in short tons

	Short tons	1928	
		Total	Average
Basic lead sulphate or sublimed lead:			
White	16,002	\$2,211,470	
Blue	1,234	175,508	
Red lead	40,497	6,810,870	
Orange mineral	459	112,321	
Litharge	85,670	12,872,730	
White lead:			
Dry	42,049	6,683,115	
In oil	111,923	24,653,989	

Zinc pigments and salts sold by domestic manufacturers in the United States, 1920-1929, in short tons

Year	Zinc oxide	Leaded zinc oxide
1920	99,444	30,460
1921	74,329	16,103
1922	128,465	19,813
1923	126,987	23,804
1924	131,470	29,729
1925	151,354	31,750
1926	146,923	23,859
1927	151,245	26,064
1928	160,904	24,223
1929	160,611	27,149

¹ Figures cover production, not sales.

² Figures published in earlier reports of this series have been revised to publish the revised figures.

³ Bureau of Mines not at liberty to publish figures.

Zinc pigments and salts sold by domestic manufacturers in the United States, 1920-1929, in short tons

	Short tons	1928	
		Total	Average
Zinc oxide	160,904	\$20,655,442	\$127.12
Leaded zinc oxide	24,223	2,945,323	121.12
Lithopone	200,468	19,073,914	95.19
Zinc chloride, 50° Baumé	88,837	1,706,222	19.10
Zinc sulphate	(1)	(1)	(1)

¹ Zinc oxide containing 5 per cent or more of lead is classed as leaded.

² Figures published in the report of this series for 1928 have been revised to publish the revised figures.

³ Bureau of Mines not at liberty to publish figures.

Lead acetate, lead arsenate, and other lead salts (exclusive of lead pigments) produced in the United States, 1919, 1921, 1923, 1925, 1927, and 1929

Source	1928	
	Lead in pigments	Zinc pigments
Domestic ore.....	10,485	1
Metal.....	242,914	
Secondary material.....	664	
	254,033	

* Includes also lead recovered in zinc oxide and leaded zinc oxide. The and lead arsenate is not available, as no canvass of their production is. Moreover, these salts are derived from pig lead, and their metal content statistical account in the figures covering lead production.

WHITE LEAD

White lead is the basic carbonate of lead and is the most important of the white pigments used in paint. The properties of white lead that contribute to its usefulness are its very great hiding power and the facility with which it can be worked with a brush. Owing to its relative softness and the fact that white-lead paints are blackened by containing sulphur, the use of less-expensive zinc oxide and other paint pigments has increased during recent years, thus retarded expansion of the white-lead industry.

The largest part of the white lead manufacture in the United States is produced by the "stack" or old Dutch method. The process consists in placing perforated disks of high-grade metallic lead, each 12 inches in diameter, are placed in small clay pots, one on top of another, in a large room, well in the bottom containing dilute acetic acid. The pots are arranged in tiers within a large room. Each tier is covered with boards and a layer of spent tanbark. The lead is corroded by the action of acetic acid, CO_2 , and the heat necessary to vaporize the acetic acid by the fermentation of the tanbark. The process takes four months for completion. The corroded lead is then crushed, screened, and ground in water, and then dried to form dry white lead or ground with oil with the oil pigment. In the United States about 90 per cent of the white lead product is ground in oil.

Some white lead is also made by the Carter electrolytic process. In the former metallic lead is first melted in steam and is then treated with dilute acetic acid in slowly rotating wooden cylinders. The process takes about 12 days.

In the electrolytic process, a lead anode and used, each of which has an independent container, electrolyte kept separate by a porous diaphragm are solutions of sodium acetate containing sodium anolyte contains the exact amount of carbonate precipitate the basic lead carbonate, whereas the

¹ The statistics for 1927 and 1929 are not comparable with those for preceding years as the figures prior to 1927 do not include data for the production reported by establishments engaged primarily in the manufacture of patent and proprietary medicines and compounds.

Zinc stearate and other salts and compounds of zinc (exclusive of zinc pigments, zinc chloride, and zinc sulphate) produced in the United States, 1923, 1925, 1927, and 1929

: Not given.

The following table gives the quantity and source of the lead and zinc contained in the pigments and salts produced in the United States in 1928 and 1929. It shows that in 1929, 95 per cent of the lead in lead pigments was derived from pig lead, 4 per cent from ore, and 1 per cent from secondary material. For zinc pigments and salts, the proportions were 68 per cent from ore, 19 per cent from metal, and 13 per cent from secondary material. The zinc in zinc salts is derived largely from secondary material.

relatively large amount of carbonate. During electrolysis the lead dissolves at the anode surface and is immediately precipitated as the basic carbonate in suspension in the anolyte. At the same time CO_2 is generated at the cathode, passes through the diaphragm, and replaces the CO_2 precipitated with the lead, thus maintaining the anolyte at a constant concentration. The catholyte passes through a carbon dioxide absorption tower where it is restored to its original strength. The basic lead carbonate is filtered from the anolyte, washed, dried, pulverized, and barreled.

During 1929 it was announced¹ that a new process had been developed for producing white lead and other lead pigments directly from galena ore, in which the ore is first treated with nitric acid. The resultant solution is then purified by a secret process which yields an extremely pure solution of lead from which pigments of high purity can be produced. White lead is precipitated from this solution by the addition of sodium carbonate (soda ash or sal soda), chrome yellow by the addition of sodium bichromate, and basic lead sulphate by the addition of sodium sulphate. The precipitates are very finely divided and are collected in filter presses, washed, dried, and ground. The process involves the regeneration of nitric acid and the recovery of by-products, including zinc oxide, silica, and elemental sulphur. The General Reduction & Chemical Co. was contemplating the erection of three plants, each to cost \$1,500,000.

The principal use of white lead is in the manufacture of paint. It is the only important white pigment that is used unmixed with other pigments in oil paints; it is also used in varying proportions in many mixed pigments. Dry white lead is used in paints, as a constituent of the glaze used in the finer grades of pottery, and to some extent in the manufacture of rubber.

The following table shows the sales of white lead (dry and in oil) to various industries in 1929, as reported to the Bureau of Mines by the producers.

Distribution of white lead (dry and in oil) sales in 1929, by industries

Industry	Short tons	Per cent of total
Paint.....	136,526	92.8
Ceramics.....	4,246	2.9
Other.....	6,269	4.3
Total.....	147,031	100.0

The United States produces an exportable surplus of about 6,000 tons of white lead annually, most of which is manufactured in bond from foreign lead.

BASIC LEAD SULPHATE

Basic lead sulphate is also known as sublimed lead and is marketed in two forms, "white" and "blue." The white variety is usually made by vaporizing powdered galena at a high temperature in the presence of air, which oxidizes it to the basic sulphate. The resultant fume is collected in bags, and owing to the zinc content of the ore used it usually contains about 5 per cent of zinc oxide. This product is usually not as white as basic carbonate white lead, but has a more

stable color in the presence of sulphur gases. It can also be made directly from pig lead. The molten lead is atomized into a heated furnace, and oxygen and sulphur dioxide are introduced. The lead is oxidized, and converted into basic lead sulphate in a quantity proportionate to the oxygen and sulphur introduced into the furnace. The resultant fume is collected. It is claimed that this product is the whitest of white lead, and that it has a promising future in the manufacture of storage batteries where it is used with lead oxides to produce battery plates. In 1929 the development of a process for basic lead sulphate directly from ore, using sulphuric acid, was reported (see discussion under "Sulphur"). White sublimed lead is used chiefly in the manufacture of storage batteries and rubber. The blue variety is generally a by-product of the manufacture of lead and is collected from the fumes leaving the furnace. Its color is due to small amounts of carbon and iron. It is made from blue lead, and is especially useful as a pigment for metallic surfaces. Some blue lead is also used in the manufacture of rubber.

The following table shows the sales of basic lead sulphate to various industries in 1929, as reported to the Bureau of Mines by the producers.

Distribution of basic lead sulphate sales in 1929, by industries

Industry	Short tons	Per cent of total
Paints.....	136,526	92.8
Storage batteries.....	4,246	2.9
Rubber.....	6,269	4.3
Other.....	147,031	100.0

LITHARGE

Litharge is the monoxide of lead and is produced by heating molten lead at a moderate temperature in a furnace. The oxide forms on the surface of the molten lead and is skimmed off into iron pots. The texture and color of the litharge, and its use in the various industries, are controlled by the temperature of the furnace and the method of cooling. The litharge is used in the manufacture of storage-battery plates, in the manufacture of straight litharge, whereas the white lead is a mixture of red lead and litharge. The next largest use of litharge is in the oil-refining industry, where it is used to remove sulphur compounds from gasoline. Litharge is also used as a cure accelerator in the manufacture of mechanical rubber clothing. It is also used in the manufacture of color pigments, insecticides, pottery, enamel, and in the manufacture of varnish and linoleum.

The following table shows the sales of litharge to various industries in 1929, as reported to the Bureau of Mines by the producers.

¹ Oil, Paint and Drug Reporter, vol. 125, June 3, 1929, p. 29.

Distribution of litharge sales in 1929, by industries

Industry	Short tons	Per cent of total	Industry	Short tons	Per cent of total
Storage batteries.....	37,180	42.2	Varnish.....	3,124	3.5
Oil refining.....	18,615	15.5	Linoleum.....	322	0.4
Ceramics.....	8,693	9.8	Other.....	10,269	11.7
Chrome pigments.....	8,112	9.2			
Rubber.....	6,651	7.6		87,916	100.0

RED LEAD

Red lead is the tetroxide of lead and is made by heating powdered litharge at a moderate temperature in a "coloring oven." The marketed product varies in composition, averaging about 85 per cent pure red lead and 15 per cent litharge; the latter is unchanged in the process of manufacture. The principal use of red lead is in the manufacture of storage batteries where, mixed with litharge, it usually forms the positive plate of the battery. It is an important pigment in paint, used as a protective coating on metal surfaces and accounts for the familiar orange color seen on practically all steel construction. During 1929 the American Gas Association and the Bureau of Standards began tests for protective coatings on underground pipe lines. The tests were to include the use of a baked red-lead priming coat under asphaltic or bituminous protective coatings. Red lead is the most important of the lead compounds used in the production of flint and optical glass. It is also used in potter's glaze and in enamel ware and as a drier in the manufacture of varnish.

The following table shows the sales of red lead to various industries in 1929, as reported to the Bureau of Mines by the producers.

Distribution of red lead sales in 1929, by industries

Industry	Short tons	Per cent of total
Storage batteries.....	25,889	59.7
Paints.....	11,855	27.6
Ceramics.....	903	2.1
Other.....	4,574	10.6
	43,021	100.0

ORANGE MINERAL

Orange mineral is a form of red lead; it has the same chemical composition but differs in physical properties and has a lower specific gravity and lighter color. It is usually made by roasting white lead by a process that requires great care and skill. Consequently, the price is higher than that for any of the other lead pigments. Orange mineral is used chiefly in the manufacture of an artificial vermilion color pigment known as eosin lake; it is also used in the production of printer's inks and enamels.

The following table shows the sales of orange mineral to various industries in 1929, as reported to the Bureau of Mines by the producers.

Distribution of orange mineral sales in 1929, by industries

Industry	Short tons	Per cent of total
Color pigments.....		
Ink manufacture.....		
Other.....		

ZINC OXIDE

Zinc oxide, the monoxide of zinc, is made in two processes—the French and the American. In the French process zinc is heated in a current of air, and the residue is collected in long chambers. The product is greenish and is labeled "white seal," "green seal," "white seal," "green seal." The white-seal variety is the finest and whitest, and its specific gravity is much lower than those of the coarse grades, which are chiefly in high-grade enamel paints. Green seal is intermediate and heavy grades, respectively.

American process zinc oxide is made by heating zinc ore and anthracite to produce a metallic zinc vapor, which is treated with air to form the oxide. The oxide is collected in bag houses. As lead is usually present in the ore, the oxide contains some lead compounds. It is usually called "French oxide" but in other respects is very similar to the French oxide. The process of making zinc oxide predominates in the United States during the twentieth century has been the growth of the rubber industry. In an estimate for 1926, the use of zinc oxide in the various industries was as follows:

	Per cent	
Rubber.....	53	Ceramics.....
Paints and enamels.....	38	Miscellaneous.....
Oilcloth and linoleum.....	5	
Exports.....	4	

Zinc oxide is used in the manufacture of rubber because of the special properties it imparts to the rubber. Its strength and the resistance to abrasion and tearing qualities by counteracting the tendency of rubber to deteriorate under action of light and heat. All of these are essential in making automobile tires, tubes, and other products. Zinc oxide is seldom used alone as a paint pigment, but a hard, smooth surface which on aging tends to leave an unsatisfactory surface for repainting. It has a stable color and costs less than white lead, has good covering properties, and is durable. The use of zinc oxide as a useful constituent of mixed pigments finds a large usage. It is generally mixed with white lead, and inert pigments.

Zinc oxide is also used in the manufacture of oilcloth and linoleum, pottery glazes and glass, celluloid, druggists' supplies, printer's inks, and leather finishes, and for many other purposes.

The use of zinc oxide as a lubricant was reported ⁵ in 1929. A mixture of zinc oxide and oil in equal parts is said to prevent corrosion of bearings and to improve the bearing surface by depositing thereon a thin film of zinc.

Exports of zinc oxide in 1929 amounted to 11 per cent of the total production.

LEADED ZINC OXIDE

Leaded zinc oxide is made by a method similar to the American process of making zinc oxide. The ore used contains lead in varying proportions, and the finished product may contain from 5 to 35 per cent of basic lead sulphate. The properties of this pigment are similar to those of zinc oxide and basic lead sulphate. It has fair hiding power but is not so white as American process zinc oxide. Leaded zinc oxide is used chiefly in the manufacture of paint.

LITHOPONE

Lithopone is the newest of the important white pigments, and its introduction into the United States dates back to the close of the nineteenth century. It contains about 70 per cent of barium sulphate and about 30 per cent of zinc sulphide. The latter, a powerful pigment, is responsible for the useful pigment qualities of lithopone. The process of manufacture consists in mixing proper proportions of solutions of barium sulphide and zinc sulphate. The resultant precipitate is heated to a red heat and plunged into cold water, after which it is ground, dried, and powdered.

The use of high-strength lithopone, containing above 30 per cent of zinc sulphide, has been increasing in recent years. This product can be made directly or by mixing zinc sulphide with ordinary lithopone. The increase in imports of zinc sulphide from less than 4 short tons in 1926 to more than 315 tons in 1929 has probably been due to the increased manufacture of high-strength lithopone. This pigment is said to possess very high hiding power and nonsettling properties which make it particularly suitable for the manufacture of shelf goods.

The use of lithopone in the United States has increased rapidly and has spread to many industries. Its chief use is in paint manufacture, in which it is favored on account of its low cost and many good qualities. Formerly the use of lithopone was confined to interior paint, as the pigment would discolor in sunlight. In recent years, however, the refining process has been changed, and it is claimed that lithopone with a stable color suitable for exterior paints is now being produced. It is generally mixed with other pigments, notably zinc oxide.

A large quantity of lithopone is used in the manufacture of oilcloth and linoleum, artificial leather finishes, window shades, and printer's inks.

Imports of lithopone into the United States in 1929 amounted to 8,400 short tons, and exports amounted to 4,600 tons.

⁵ Chemistry and Industry, vol. 48, Nov. 22, 1929, p. 1136.

LEAD AND ZINC PIGMENTS

The following table shows the sales of lead and zinc pigments in various industries in 1928 and 1929, as reported by the producers.

Distribution of lithopone sales, 1928 and 1929

Industry	Short tons
Paints, varnish, and lacquers.....	1
Floor coverings and textiles.....	
Rubber.....	
Other.....	2

ZINC SALTS

Zinc sulphate, sometimes known as white vitriol, is used in treating secondary zinc material with which upon evaporation yields the white crystalline product. The principal uses of zinc sulphate is as a depressant in the treatment of lead-zinc ores. It has recently been used in nurseries of the United States Forestry Service for the solutions of electro-galvanizing cells, in the treatment of leather, and in dyeing, as a drier for linseed oil, as an astringent, and as a sizing on cement products. For these uses, several thousand tons is used annually. Zinc sulphate is not separately recorded by the Bureau of Mines.

Zinc chloride may be prepared by treating zinc metal with hydrochloric acid, the latter being in excess. Practically all zinc chloride produced in the United States is from scrap metal or secondary materials, such as zinc dross, zinc shavings, and galvanizer's scrap. Owing to its corrosive nature, it is usually marketed in solution, although it is also obtained in air-tight containers. Zinc chloride is used as a preservative of timber, but it is also used in the manufacture of insectant, in the refining of petroleum, in the manufacture of hard fiber, for medical purposes, and in the manufacture of pigments.

PRODUCERS AND

The following companies reported the production of lead and zinc pigments and zinc salts of their own manufacture in 1929:

White lead:	
Anaconda Lead Products Co. (electrolytic process)	
Carter White Lead Co.	
Eagle-Picher Lead Co.	
Do.	
Euston Lead Co.	
W. P. Fuller & Co.	
John T. Lewis & Bros. Co.	
National Lead Co. of California	
National Lead Co.	
Do.	
Do.	

White lead—Continued.

National Lead Co.	Brooklyn, N. Y.
Do.	Port Richmond, N. Y.
Do.	Cincinnati, Ohio.
National Lead & Oil Co. of Pennsylvania	New Kensington, Pa.
Sherwin-Williams Co.	Chicago, Ill.
Wetherill & Bro.	Philadelphia, Pa.
Basic lead sulphate, or sublimed lead:	
Eagle-Picher Lead Co.	Galena, Kans.
Do.	Joplin, Mo.
St. Louis Smelting & Refining Works of	
National Lead Co.	Collinsville, Ill.
Red lead, orange mineral, and litharge:	
Eagle-Picher Lead Co.	East St. Louis, Ill.
Do.	Joplin, Mo.
Do.	Newark, N. J.
Do.	Cincinnati, Ohio.
Evans-Walloway Lead Co.	Charleston, W. Va.
Franklin Lead Oxide Co.	Franklin, Pa.
W. P. Fuller & Co.	South San Francisco, Calif.
Grasselli Chemical Co.	East Chicago, Ind.
Do.	Newark, N. J.
John T. Lewis & Bros. Co.	Philadelphia, Pa.
Metals Refining Co.	Hammond, Ind.
National Lead Co. of California	Melrose, Calif.
National Lead Co.	Chicago, Ill.
Do.	St. Louis, Mo.
Do.	Brooklyn, N. Y.
National Lead & Oil Co. of Pennsylvania	New Kensington, Pa.
Sherwin-Williams Co.	Chicago, Ill.
United Color & Pigment Co.	Newark, N. J.
Wetherill & Bro.	Philadelphia, Pa.
Zinc oxide and leaded zinc oxide:	
American Zinc Co. of Illinois	Hillsboro, Ill.
American Zinc Oxide Co.	Columbus, Ohio.
Eagle-Picher Lead Co.	Hillsboro, Ill.
Empire Zinc Co.	Canon City, Colo.
Grasselli Chemical Co.	East Chicago, Ind.
International Lead Refining Co. (French	
process)	East Chicago, Ind.
Do.	Akron, Ohio.
Mineral Point Zinc Co.	Mineral Point, Wis.
New Jersey Zinc Co. (of Pennsylvania)	Palmerton, Pa.
Do. (French process)	Freemansburg, Pa.
Ozark Smelting & Mining Co.	Coffeyville, Kans.
River Smelting & Refining Co.	Florence, Colo.
Lithopone:	
Chemical & Pigment Co. (Inc.)	Oakland, Calif.
Do.	Collinsville, Ill.
Do.	St. Helena, Md.
Eagle-Picher Lead Co.	Argo, Ill.
Grasselli Chemical Co.	Grasselli, N. J.
Do.	Newark, N. J.
Do.	Philadelphia, Pa.
Krebs Pigment & Chemical Co.	Newport, Del.
Mineral Point Zinc Co.	Depue, Ill.
New Jersey Zinc Co. (of Pennsylvania)	Palmerton, Pa.
Sherwin-Williams Co.	Chicago, Ill.
United Color & Pigment Co.	Newark, N. J.
Zinc salts:	
American Smelting & Refining Co.	Selby, Calif.
Do.	Durango, Colo.
Do.	Omaha, Nebr.
Do.	Perth Amboy, N. J.
Arkansas Preservative Co.	North Little Rock, Ark.
Charles Cooper & Co.	Newark, N. J.
General Chemical Co.	Chicago, Ill.

LEAD AND ZINC PIGMENTS AND SALTS

Zinc salts—Continued.

Grasselli Chemical Co.	
Do.	
Do.	
Charles Lennig & Co. (Inc.)	
Merrimac Chemical Co.	
Monsanto Chemical Works	
Ozark Smelting & Mining Co.	

TARIFF

The tariff act of 1930 became effective. It changes in the schedules on lead and zinc pigments an increase from 1½ cents to 3 cents per pound to 1½ cent increase from 1¼ cents per pound to 1½ cent per cent ad valorem on lithopone containing by more of zinc sulphide. The rates on lead and zinc provided in the 1930 act are shown in the following table:

Import duties on lead and zinc pigments and salts

Para-graph	Item
72	Lead pigments:
5	Litharge
	Orange mineral
	Red lead
	White lead
	All pigments containing lead, dry or in pulp, or ground mixed with oil or water, not specially provided for
46	Lead chemicals:
	Acetate, white
	Acetate, brown, gray, or yellow
	Nitrate, arsenate, and resinate
	All other lead compounds, n. s. p. f.
77	Zinc pigments:
	Zinc oxide and leaded zinc oxide containing not more than 5 per cent of lead—
	Dry powder
	Ground in or mixed with oil or water
	Lithopone, and other combinations or mixtures of zinc sulphide and barium sulphate containing by weight—
	Less than 30 per cent of zinc sulphide
	30 per cent or more of zinc sulphide
93	Zinc chemicals:
	Zinc chloride
	Zinc sulphate
	Zinc sulphide

IMPORTS AND EXPORTS

LEAD PIGMENTS

Imports of lead pigments into the United States were unimportant. Exports of the white lead pigment in 1929 and amounted to 4 per cent of the total output of white lead; United Kingdom took 57 per cent in 1929, compared with 66 per cent in 1928. Exports of lead, litharge, and orange mineral increased in 1929, amounted to 2 per cent of the total domestic production of three pigments.

Basic carbonate white lead, red lead, litharge, and orange mineral imported for consumption in the United States, 1920-1929

Year	Basic carbonate white lead		Red lead		Litharge		Orange mineral	
	Pounds	Value	Pounds	Value	Pounds	Value	Pounds	Value
1920	169,977	\$19,449	23,076	\$1,789	21,002	\$2,647	52,094	\$5,352
1921	124,176	12,598	12,236	689	100	7	77,937	7,830
1922	222,465	21,335	1,406	94	1,770	162	111,419	9,729
1923	339,892	31,386	3,600	220	22,825	1,314	116,293	9,031
1924	141,186	13,476	1,874	158	1,638	193	99,241	8,047
1925	213,028	21,535	3,452	288	2,417	330	58,522	6,278
1926	297,867	31,690	9,591	817	122	13	58,104	5,143
1927	330,421	29,901	51,688	4,403	1,182	133	44,866	3,816
1928	166,236	15,889	24,676	1,510	2,100	153	62,409	5,227
1929	166,478	19,086	10,197	745	3,783	270	52,122	3,889

White lead and red lead exported from the United States, 1920-1929

Year	White lead ¹		Red lead ¹	
	Short tons	Value	Short tons	Value
1920	14,901	\$3,195,101	1,532	\$365,575
1921	5,161	993,130	548	115,500
1922	4,698	710,765	1,683	294,183
1923	15,172	836,201	1,940	272,828
1924	15,055	853,444	1,940	210,598
1925	10,832	1,291,108	802	183,591
1926	16,239	1,119,350	775	176,273
1927	16,047	947,226	1,429	271,376
1928	16,476	952,316	2,084	364,995
1929	15,908	921,509	2,890	503,243

¹ Exports of white lead and basic lead sulphate or sublimed lead not classified separately after 1922; the figures for 1923-1929 may include an unknown quantity of basic lead sulphate or sublimed lead.

¹ Exports of red lead and litharge not classified separately for 1923-1924 and 1927-1929; the figures for those years may include an unknown quantity of litharge; the figures for 1923 and 1929 include also an unknown quantity of orange mineral. Figures for other years are for red lead only.

White lead¹ exported from the United States, 1928 and 1929, by destinations

Destination	1928		1929	
	Short tons	Value	Short tons	Value
North America:				
Canada	58	\$9,285	122	\$19,807
Cuba	80	14,750	58	10,207
Mexico	149	32,040	127	28,776
Panama	284	67,022	568	117,691
Other	84	17,517	65	13,538
	655	131,214	940	189,939
South America:				
Brazil	45	7,043	12	2,231
Colombia	13	2,599	18	4,206
Uruguay	41	5,857	32	5,089
Venezuela	7	1,566	19	4,213
Other	402	59,370	415	68,061
	508	76,435	496	83,800

¹ Exports of white lead and basic lead sulphate or sublimed lead not classified separately after 1922; the figures may include an unknown quantity of basic lead sulphate or sublimed lead.

White lead exported from the United States, 1928 and 1929

Destination	1928		1929	
	Short tons	Value	Short tons	Value
Europe:				
Netherlands	557			
United Kingdom	4,274			
Other	64			
	4,995			
Asia:				
Philippine Islands	214			
Other	99			
	313			
Africa	1			
Oceania, including Australia	4			
	6,478			

ZINC PIGMENTS AND SA

Imports of lithopone, the principal zinc pigment, decreased 15 per cent in 1929 from 1928, or 4 per cent of the domestic output. Imports of lithopone in 1929 and were equivalent to less than 5 per cent of domestic production. There has been a large increase of zinc sulphide imported into the United States in recent years. This product is used principally in the manufacture of strength lithopone. Belgium supplied 60 per cent of the zinc sulphide imported in 1929, Netherlands 82 per cent of the zinc sulphide imported in 1928, and the United States chiefly to Canada (44 per cent) and the United States (25 per cent). Canada also took 87 per cent of the zinc sulphide imported in 1929.

Zinc oxide, lithopone, zinc sulphide (white), zinc sulphate, and zinc carbonate imported for consumption in the United States, 1928 and 1929

Year	Zinc oxide		Lithopone		Zinc sulphide (white)		Zinc sulphate		Zinc carbonate	
	Short tons	Value	Short tons	Value	Short tons	Value	Short tons	Value	Short tons	Value
1920	1,431	\$289,890								
1921	1,286	145,071								
1922	2,712	861,599								
1923	1,036	169,415								
1924	1,703	229,620								
1925	1,274	205,582								
1926	1,127	198,897								
1927	1,899	220,427								
1928	1,348	208,557								
1929	1,267	203,080								

Zinc oxide, lithopone, zinc sulphide (white), zinc chloride, and zinc sulphate imported for consumption in the United States, 1920-1929—Contd.

Year	Zinc sulphide ¹		Zinc chloride		Zinc sulphate	
	Pounds ¹	Value	Short tons	Value	Short tons	Value
1920.....	45,482	\$15,027	585	\$76,495	50	\$3,038
1921.....	93,537	8,330	2,199	213,985	17	820
1922.....	27,159	8,453	1,374	87,509	17	1,064
1923.....	11,844	5,403	551	44,164	20	3,466
1924.....	9,678	5,727	310	28,911	13	1,265
1925.....	14,604	5,589	469	50,712	17	1,368
1926.....	7,800	4,125	508	51,005	60	4,590
1927.....	132,223	22,984	470	40,069	205	9,390
1928.....	338,030	47,235	502	47,185	683	28,866
1929.....	630,565	84,260	638	50,324	909	32,127

¹ Zinc sulphide is stated in pounds rather than tons because of its high value and smaller quantities.

Zinc oxide and lithopone exported from the United States, 1920-1929

Year	Zinc oxide		Lithopone	
	Short tons	Value	Short tons	Value
1920.....	11,164	\$2,151,544	(¹)	(¹)
1921.....	2,511	426,253	(¹)	(¹)
1922.....	3,977	593,128	1,610	\$169,982
1923.....	5,024	743,577	1,485	176,624
1924.....	3,927	805,630	922	104,783
1925.....	10,865	1,503,561	1,287	132,771
1926.....	14,661	1,917,430	1,941	200,412
1927.....	14,494	1,821,756	2,110	222,585
1928.....	14,799	1,849,839	3,326	337,866
1929.....	17,638	2,301,643	4,556	463,234

¹ Not classified separately prior to 1922.

Zinc oxide exported from the United States, 1928 and 1929, by destinations

Destination	1928		1929	
	Short tons	Value	Short tons	Value
North America:				
Canada.....	6,404	\$882,130	7,702	\$1,062,618
Cuba.....	135	20,288	117	17,508
Mexico.....	102	19,112	73	13,448
Panama.....	94	17,291	87	16,486
Other.....	30	5,557	18	4,483
	6,765	944,528	7,997	1,113,543
South America:				
Brazil.....	8	1,683	5	1,129
Other.....	120	22,011	64	13,730
	128	23,674	69	14,859
Europe:				
France.....	(¹)	93	452	50,836
Italy.....	247	29,270	69	11,361
United Kingdom.....	4,315	491,668	6,860	762,010
Other.....	1,859	186,144	1,144	125,964
	6,421	707,165	8,025	960,171
Asia:				
Philippine Islands.....	93	11,111	88	12,264
Other.....	1,130	129,375	619	110,415
	1,223	140,486	907	122,679
Africa.....	4	724	3	594
Oceania, including Australia.....	258	33,312	637	99,797
	14,799	1,849,839	17,638	2,301,643

¹ Less than 1 ton.

PRICES AND VALUE

The total value and the average price received for sales of lead and zinc pigments and salts are given in this chapter. The range of market quotations for Oil, Paint and Drug Reporter is shown in the

Range of quotations on lead pigments and zinc pigments, 1927-1929, in cents a pound

Product	1927
Basic lead sulphate, or sublimed lead, dry, in casks.....	7.75-
White lead, or basic lead carbonate:	
Dry, in casks.....	8.50-
In oil, carload lots.....	10.78-
Litharge, American, commercial, powdered, casks.....	8.75-
Red lead, dry, in casks.....	9.75-
Orange mineral, American, casks.....	11.75-
Lead acetate, brown, broken, barrels.....	12.00-
Lead arsenate, powdered, drums.....	13.50-
Zinc oxide:	
American process, lead-free, bags, car lots.....	6.75-
American process, leaded, barrels, car lots.....	
French process, red seal, bags, car lots.....	
French process, green seal, bags, car lots.....	
French process, white seal, barrels, car lots.....	
Lithopone, American, bags, car lots.....	
Zinc chloride:	
Solution, ¹ drums.....	2.75-
Fused.....	
Zinc sulphate, barrels.....	8.00-

¹ At works.

The quotations for lead pigments are usually based on 25-point fluctuations in the price of pig lead. The price trend of lead pigments in 1929 was upward until October, after which there was a downward thereafter in sympathy with the quotations on zinc pigments, however, were year at levels established in 1927, notwithstanding an increase in the price of zinc up to October, 1929, by a sharp decrease in November and December. The strength lithopone, containing 50 per cent zinc, was reduced to 7½ cents per pound in December. The cuts for titanium pigments.