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REPORT NO.: MDA-651

JOB/PROJECT NO.: 643

DATE: 8 August 1985

PERIOD COVERED:

TITLE: ANALYSIS OF OIL FOR AROCLOR FORMULATIONS

AUTHORS: B. M. Hughes

ABSTRACT: Oil samples were analyzed for possible concentrations of Aroclor formulations. The results of these analyses are shown in this report.

TECHNICAL APPROVAL: P. L. Sherman
Sr. Research Group Leader

APPROVED BY: R. F. Ivory
Environmental Sciences Center
Service Project Manager

APPROVAL DATE: 8-8-85

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REPORT NO.: MDA-651 AUTHORS: B. M. Hughes
COPY NO.: TITLE: Analysis of oil for Aroclor Formulations.

MONS 023608

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ABSTRACT ONLY

H. A. Woltermann - U4E

Analysis:

B. M. Hughes
D. McKenzie
B. Williams

MONS 023609

Time Stamp - <850802.1100>

ANALYSIS REPORT

DATE: August 2, 1985

LOG NUMBER: 1-85-07-03-01 PROJECT NO: 760.18-643

TO: Ivory/PSC

FROM:

Environmental Sciences Center
Monsanto Company/Dayton Laboratory
1515 Nicholas Road
Dayton, Ohio 45418

Phone No: (513) 268-3411

Title: Analysis of Aroclor Formulations in 12 Oil Samples

Standard Sampling Method No(s): Samples supplied by PSC.

Standard Extraction Method No(s): See Analytical Request File for sample dilution information.

Standard Analysis Method No(s): MSAM03

Final Report File No(s): A64318

Table File No(s): B64318

Twelve oil samples were received for Aroclor formulation analysis. Table 1 summarizes these results. Note that three of the twelve samples contain Aroclor formulations greater than 50 ug/g. Therefore these three oils must be treated as "PCB-containing" materials.

Also included in this report are:

1) Capillary GC/MS chromatograms obtained for this project.
2) QA/QC documentation of analysis of spiked samples.
3) A copy of Standard Analysis Method MSAM03 used for this project.
4) Analytical Request Information giving sample dilution information.
5) A copy of the ESC/UAG Good Laboratory Practices Manual.

ANALYST(S): B. Mason Hughes, David McKenzie and Barbara Williams

ANALYST APPROVAL: B. Mason Hughes

Time Stamp - (850002.1021)

Table 1. Results of Aroclor Concentrations Detected.

Project No: 760.18-643 Date: August 1, 1985

Sample Identification		Extract ID	Formulation	Concentration (ug/g)
75AD6070	A7112A25A11	PCB-1381	-	ND
75AA06073	A7112A25AAK	PCB-1382	-	ND
5075AB595	A7352A50ARE	PCB-1383	-	ND
5075AB592	A7352A50ARE	PCB-1384	-	ND
77A200014-2095	A6323A51VA	PCB-1385	-	ND
5005611	61695946201	PCB-1386	1262	50.3
"	" (DUP)	PCB-1386(DUP)	"	65.1
7AA6769	A7A12A258	PCB-1387	-	ND
79ZH36A003	CBVR119124150K2	PCB-1388	-	ND
35348	TTCL-CSP	PCB-1389	1262	80.5
"	" (DUP)	PCB-1389(DUP)	"	62.4
69J7526-1295	A631DA76BA1	PCB-1390	1248	99.7
			1262	74.0
			total =	172.7
74AB12874-965	A631DA45BAE	PCB-1391	-	ND
5078A212509-728	A7552A50AA	PCB-1392	-	ND
SOLVENT BLANK		PCB-1393	-	ND

QA/QC Information: 1381-LS, 1382-LS, 1381-HS, 1382-HS are samples 1381 and 1382 which have been spiked at levels of Aroclor 1254 which correspond to ~5 and ~50 ug/g. Benzene-LS and benzene-HS are similar spikes of the benzene diluting solvent.

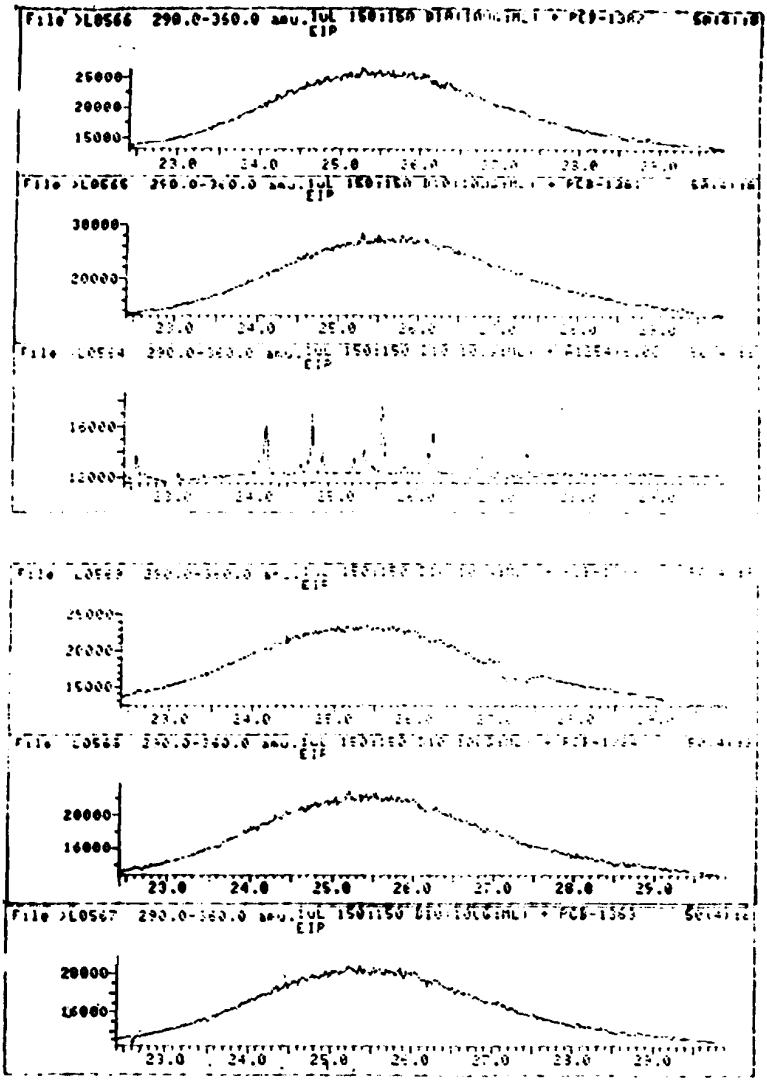
Footnotes: (a) ND - None Detected (<5 ug/g)

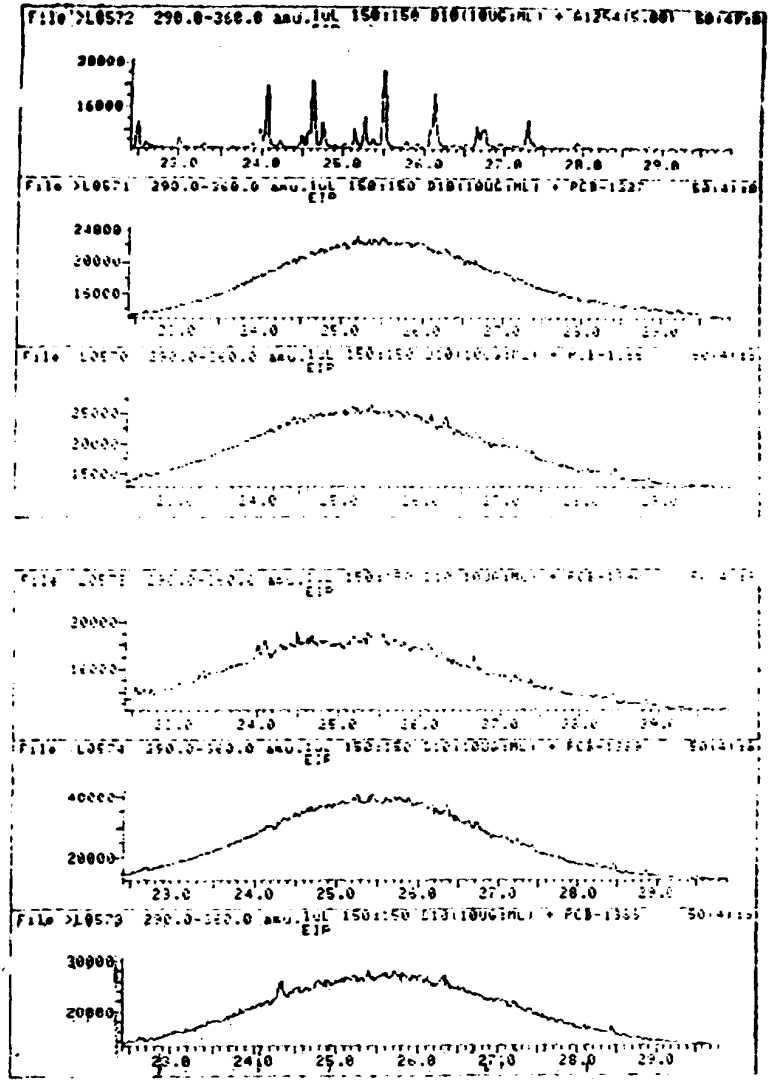
ANALYST(S): B. Mason Hughes, David McKenzie, and Barbara Williams

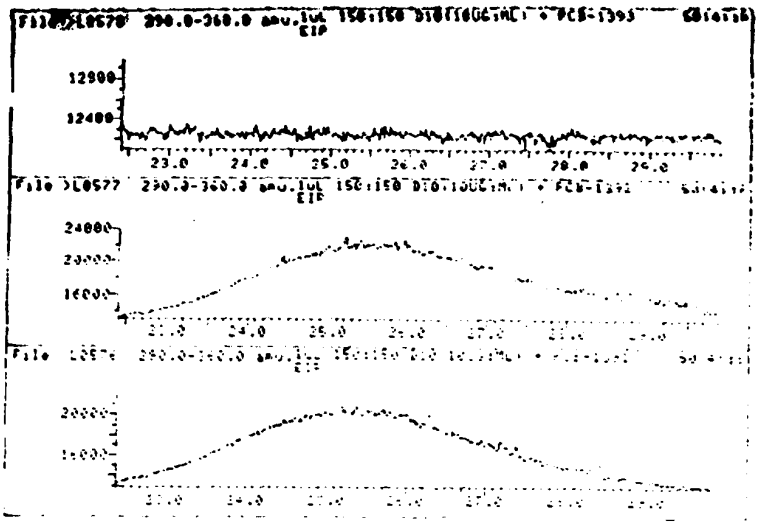
ANALYST APPROVAL: B. Mason Hughes

MONS 023611

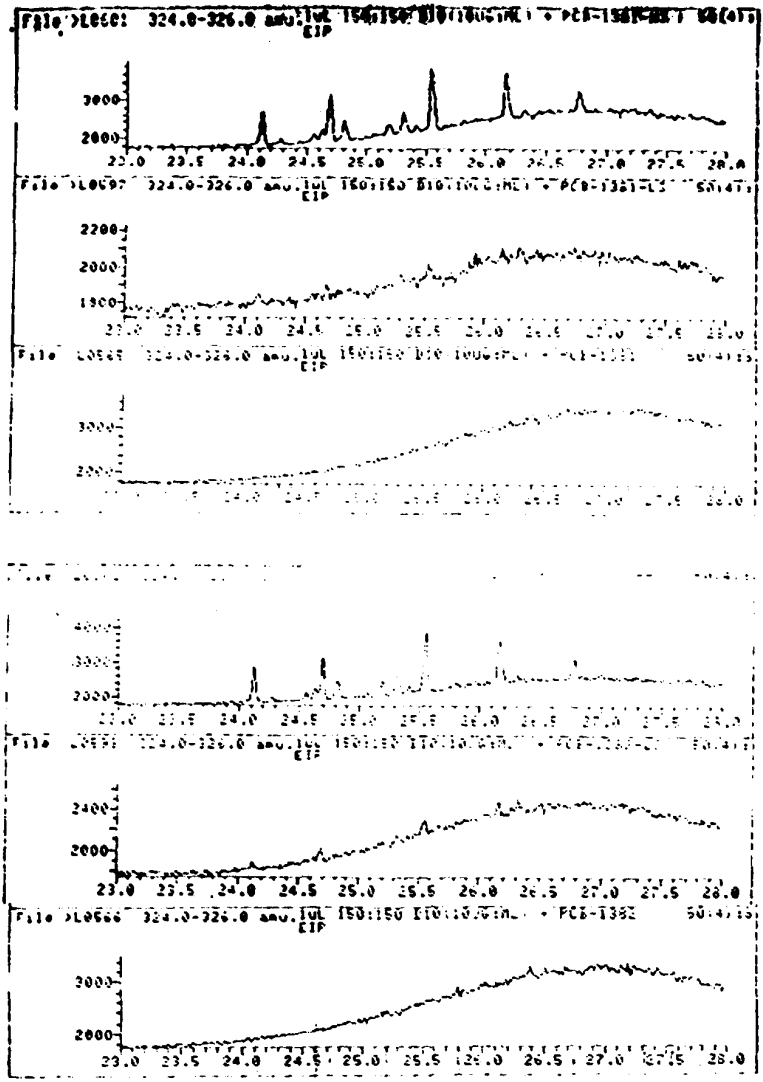
The following pages show calculations or raw data documentation of reported results.


File >L0566 290.0-360.0 amu.
EIP
25000
20000
15000
23.0
24.0
25.0
26.0
27.0
28.0
29.0
File >L0566 290.0-360.0 amu.
EIP
30000
20000
File >L0564 290.0-360.0 amu.
EIP
16000
12000
File >L0565 290.0-360.0 amu.
EIP
25000
20000
15000
File >L0565 290.0-360.0 amu.
EIP
20000
16000
File >L0567 290.0-360.0 amu.
EIP
20000
16000


File >L0572 290.0-360.0 amu.1uL 150:150 D10(100UG/ML) + A1254(5.00)
30000
16000
23.0
24.0
25.0
26.0
27.0
28.0
29.0
File >L0571 290.0-360.0 amu.1uL 150:150 D10(100UG/ML) + PCB-1254
EIP
24000
20000
15000
File L0570 290.0-360.0 amu.1uL 150:150 D10(100UG/ML) + PCB-1254
EIP
25000
20000
15000
File L0573 290.0-360.0 amu.1uL 150:150 D10(100UG/ML) + PCB-1254
EIP
20000
16000
File L0574 290.0-360.0 amu.1uL 150:150 D10(100UG/ML) + PCB-1254
EIP
40000
28000
File >L0575 290.0-360.0 amu.1uL 150:150 D10(100UG/ML) + PCB-1254
EIP
30000
20000


EIP
12800
12400
23.0
24.0
25.0
26.0
27.0
28.0
29.0
EIP
24000
20000
16000
EIP
20000
16000

The following pages show QA/QC data for selected samples.


EIP
3000
2000
22.0 23.5 24.0 24.5 25.0 25.5 26.0 26.5 27.0 27.5 28.0
EIP
2200
2000
1800
EIP
3000
2000
EIP
4000
3000
2000
EIP
2400
2000
23.0 23.5 24.0 24.5 25.0 25.5 26.0 26.5 27.0 27.5 28.0
EIP
3000
2000
23.0 23.5 24.0 24.5 25.0 25.5 26.0 26.5 27.0 27.5 28.0

The following pages describe Standard Extraction Methods and Standard Analysis Methods used for this project.

ESC Standard Analysis Method (SAM) ESC Standard Analysis Method No:
TITLE: Organic Compounds in Solvents MSAM03

ANALYTICAL INSTRUMENT AND MODE OF OPERATION: HP-5985/5987 Capillary GC/MS-Selected Ion Monitoring with RTE-VI operating system.

Typical Run Time:	3 - 40 min.
Typical Multiplier Voltage:	1800 - 2800 eV
Typical Dwell Times:	200 - 500 msec/mass
Maximum # of Masses:	20 masses x 5 ranges = 100 masses
Electron Energy:	70 eV

TYPICAL ANALYTE	TYPICAL INSTRUMENT LEVEL OF DETECTION (LOD) (ug/ml)	TYPICAL PRECISION @ 10X LOD (%)	TYPICAL ACCURACY @ 10X LOD (%)
Organic Compounds eluting between toluene and n-C34-ane	0.01 - 0.1	30	30

INTERFERENCES AND LIMITATIONS: Analyte characteristic mass must be chromatographically resolved from other eluting compounds which produce the same mass.

CHROMATOGRAPHIC COLUMN: 30-meter fused silica J&W DB-5 (0.25 uM) with wide bore (0.32 mm)

COLUMN TEMPERATURE PROGRAM: 50(4)/8/300 INJECTION TYPE: Splitless

CALIBRATION AND STANDARDIZATION:

Quantitative: Use internal standard quantitation technique and calibrate instrument response factors from standard solutions of analytes of interest.

Qualitative: When multiple characteristic masses are monitored for a given analyte, the ratio of characteristic mass responses must be within 20% of those for the authentic standard. For PCB analyses, the characteristic masses which are monitored are the molecular ion containing all 35-CL isotopes and the molecular ion containing one 37-CL isotope. The following table summarizes the masses and approximate retention time regions monitored for the analysis of internal standard, surrogate standards, and chlorobiphenyl (CL-BP) through decachlorobiphenyl (CL10-BP) PCB congeners.

Analyte	Masses	Retention Times (minutes)	Relative Retention Times (a)
CL-BP isomers, anthracene-d10	188,190	16 - 22.5	0.60 - 0.84
CL2-BP isomers	222,224	16 - 22.5	0.60 - 0.84
Pentachlorobenzene surrogate	247.9,249.9	16 - 22.5	0.60 - 0.84
CL3-BP isomers	255.9,257.9	16 - 22.5	0.60 - 0.84
CL4-BP isomers	289.8,291.8	22.5 - 30	0.84 - 1.12
3,4,3',4'-CL4-BP-d6 surrogate	296,298	22.5 - 30	0.84 - 1.12
CL5-BP isomers	323.9,325.9	22.5 - 30	0.84 - 1.12
CL6-BP isomers	357.8,359.8	22.5 - 30	0.84 - 1.12
CL7-BP isomers (b)	323.9,325.9	22.5 - 30	0.84 - 1.12
	391.8,393.8	30 - 50	1.12 - 1.87
CL8-BP isomers	427.7,429.7	30 - 50	1.12 - 1.87
CL9-BP isomers	461.6,463.6	30 - 50	1.12 - 1.87
CL10-BP	497.6,499.6	30 - 50	1.12 - 1.87

(a) Relative Retention Times calculated using a retention time of 26.78 minutes for 3,4,3',4'-CL4-BP-d6.

(b) CL7-BP isomers are analyzed using either CL2-loss characteristic ions (m/e 323.9 and 325.9) or molecular ions (m/e 391.8 and 393.8).

CALCULATIONS: Analyte concentration = (Analyte area)/[(Analyte RRF)(IS area)]

where: Analyte area is the characteristic mass area of the analyte.
RRF is the relative response factor for the analyte.
IS is the internal standard and is usually anthracene-d10.

RRF = (analyte area in standard)/[(analyte conc.)(IS area)]

PCB concentrations are calculated in one of two ways:

1) If an Aroclor formulation can be identified from the patterns of PCB isomers, selected PCB isomers or classes of isomers (i. e. all CL4-biphenyl isomers) are used to quantify the Aroclor concentration. Relative Response Factors are calculated using the concentration of the total Aroclor formulation in the above RRF equation.

2) If no unique Aroclor formulation is present, then the total PCB concentration is calculated by using standard solutions containing known amounts of chlorobiphenyl through decachlorobiphenyl isomers to calculate Relative Response Factors for each chlorine class in the above RRF equation. The RRFs for the corresponding PCB isomer classes are used to calculate total PCB isomer concentrations of each class in unknown samples.

SAFETY PRECAUTIONS: See PCB handling protocols in laboratory safety manual.

DATE OF METHOD CREATION: 12/9/83 Time Stamp - <850719.0936>

The following pages describe Analytical Request Information generated for this project.

Time Stamp: (850001.1047)

MONSANTO COMPANY/DAYTON LABORATORY

ENVIRONMENTAL SCIENCES CENTER

ANALYSIS REQUEST

DATE: JULY 3, 1985 LOG NO: 1-85-07-03-01

REQUESTER: PSC PROJECT NO: 760.81-643

REPORT RESULTS TO: FILES/MEES, IVORY, SHERMAN

ANALYSIS REQUESTS SENT TO: FILES, IVORY, SHERMAN, HUGHES

SAMPLE PICKUP BY: BARBARA WILLIAMS

SAMPLE LOCATION: BUILDING 2, LAB 28

SAMPLE DESCRIPTION FOR 12 SAMPLES

SAMPLE IDENTIFICATION		NUMBER OF BOTTLES		ESC NUMBER(S)
75AD6070	A7112A25A11	1	(0.9489 g. + 9 mls benzene)	PCB-1381
75AAD6073	A7112A25AAK	1	(0.9382 g. + 9 mls benzene)	PCB-1382
5075AB595	A7352A50ARE	1	(0.9146 g. + 9 mls benzene)	PCB-1383
5075AB592	A7352A50ARE	1	(0.9404 g. + 9 mls benzene)	PCB-1384
77A20G014-2095	A6323A51VA	1	(0.9449 g. + 9 mls benzene)	PCB-1385
5805611	61695946201	1	(0.9479 g. + 9 mls benzene)	PCB-1386
7AA6769	A7A12A258	1	(0.9593 g. + 9 mls benzene)	PCB-1387
792M36A003	CBVR119124150K2	1	(0.9555 g. + 9 mls benzene)	PCB-1388
35348	:TCL-CSP	1	(0.9306 g. + 9 mls benzene)	PCB-1389
69J7526-1295	A631DA76BA1	1	(0.9528 g. + 9 mls benzene)	PCB-1390
74AB12874-965	A631DA45BAE	1	(0.9564 g. + 9 mls benzene)	PCB-1391
5078A212509-728	A7552A50AA	1	(0.9698 g. + 9 mls benzene)	PCB-1392
SOLVENT BLANK			(benzene)	PCB-1393

QA/QC Information: 1381-LS, 1382-LS, 1381-HS, 1382-HS are samples 1381 and 1382 which have been spiked at levels of Aroclor 1254 which correspond to ~5 and ~50 ug/g. Benzene LS and benzene-HS are similar spikes of the benzene diluting solvent.

RECORDED BY: B. WILLIAMS

REQUIRED ANALYSES: Aroclor Formulations

MONS 023622

ANALYTICAL TECHNIQUES:

TECHNIQUE (EM)	DATE COMPLETED	TECHNIQUE(EM)	DATE COMPLETED
		x GC/MS (10)	8/1/85

QUALITATIVE: SEMIQUANTITATIVE: QUANTITATIVE: x

REQUIRED QA/QC: SEE GOOD LABORATORY PRACTICES MANUAL

COMPLETION DATE REQUESTED:

COMMENTS:

SAMPLE DISPOSITION: -- DISCARD -- RETURN -X HOLD FOR 30 DAYS -- OTHER- --

SAFETY PRECAUTIONS: SEE SAFE LABORATORY PRACTICES MANUAL

ANALYSIS REQUEST FILE NAME: AA643:LI:S9

The following is a copy of the ESC/UAC Good Laboratory Practices Manual.

Copy Number: _____

GOOD LABORATORY PRACTICES MANUAL

A Guide to Quality Control/Quality Assurance

Ultratrace Analysis Group

March 1984

Joseph J. Brooks
Group Leader, Ultratrace Analysis

H. A. Waltermann
Manager, Environmental Sciences Center

Issued to: _____

Date: _____

Environmental Sciences Center
Dayton Laboratory
Dayton, Ohio 45407

JR3/JOB A

MONS 023625

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I. Objectives

The main objectives of the Ultratrace Analysis Group quality assurance/quality control program are: 1) to assure that our laboratory generates high quality results; and 2) to maintain the necessary records that document laboratory performance.

II. Scope

The scope of this manual includes areas of GLP not specifically addressed in the ESC section manual and expands upon certain areas where QA/QC practices differ from those described in the section manual. It is intended to be used along with the section manual to provide a complete picture of the GLP within the Ultratrace Analysis Group.

III. Analysis

The generation of analytical results within the Ultratrace Analysis Group routinely involves two processes: sample preparation and instrumental analysis. Sample preparation involves one or more of a variety of techniques including extraction, dilution, acid/base washes, column chromatography, concentration and numerous other techniques to render the sample in a suitable condition for instrumental analysis. Instrumental analysis takes the suitably prepared sample and introduces it into an analytical instrument to measure the desired parameter(s). Both of these processes can be described by clearly defined and separable methods called sample preparation methods (SPMs) and instrumental analysis methods (IAMs). SPMs and IAMs may be combined in numerous ways (depending upon the nature of the samples and the needs of the customer) to produce a successful analytical protocol. For example, several different SPMs may be used to produce samples that are suitable for analysis by a single IAM. The measures incorporated into each of these experimental processes to ensure that an assessment of the quality of the results can be made constitute the QA/QC program for a project. The following subsections describe the elements of the analysis process:

A. Methods

Methods are the written documentation that describe the analytical process. They contain sufficient detail such that a competent scientist or technician can readily perform the required analyses. The usual practice is to have methods that describe the sample preparation process (SPMs) and methods that describe the instrumental analysis process (IAMs). Each method developed within the Ultratrace Analysis Groups is assigned a method number as described in the ESC Section Manual.

Methods are obtained from various sources including in-house development, modification/adaptation of literature methods, or specific protocol methods dictated by government agencies. Methods may be formally validated by a series of experiments that describe a classical validation protocol or may be less vigorously proven owing to factors such as frequency or use, customer needs and

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cost/timing restrictions. In any case a QA/QC program is applied to the analysis process to provide a measure of the quality of the results. Unless a QA/QC program is prescribed by a required protocol (e.g., EPA Dioxin in Soil protocol), certain minimum QA/QC measures (depending upon the analysis type) are incorporated into all Ultratrace Analysis projects. The following subsection describes the Ultratrace QA/QC program:

B. QA/QC Program

An effective QA/QC program incorporates experiments that provide information relative to each of the following questions:

(1) Are there any impurities, background or contamination that will influence the results?

(2) How reproducible are the results?

(3) How well can the analyte be recovered from the sample matrix?

(4) What is the limit of detection for the analysis?

(5) What is the dynamic range of the analysis?

(6) What is the accuracy of the analysis?

All of these questions are addressed by properly selecting/applying the use of blanks, replicates, spikes and standards in a QA/QC program. The discussion of the use of these QA/QC techniques as they apply to Ultratrace Analysis Group activities is best done by considering the sample preparation and instrumental analysis processes separately.

1. Sample Preparation QA/QC

a. Method blanks - Representative glassware, solvents and reagents are used with sample matrices which are similar to those for which results are being reported, yet do not contain the analytes of interest. Method blanks are analyzed at a frequency equal to 10% of the analyzed samples for sample sets which are multiples of 10. At least one method blank is analyzed for sample sets containing between 1 and 9 samples.

b. Replicates - If sufficient sample amounts have been submitted for analysis, replicate frequency is the same as the method blank frequency. If at all possible, replicates should be obtained from the same sample bottle in order to minimize sample inhomogeneity problems due to samples being taken at slightly different times.

c. Spikes - If sufficient sample is available for each sample type, the sample is split and spiked at two levels with

MONS 023628

the native (unlabeled) analyte(s). The levels are typically 2x (low) and 20x (high) the anticipated detection limit and serve as a technique for estimating the limit of detection for each matrix. These low and high spikes are conducted at a frequency equal to the Method Blank frequency. Obviously spiking with the native analyte cannot be done for widescan analysis since this presupposes knowledge of the identity of the analyte(s) in the sample.

Other spiking techniques are used to evaluate the extraction/recovery efficiencies for the analyte(s) of interest. Ideally extraction/recovery efficiencies are determined for each sample, since, in principle, each sample may represent a different matrix. This can be done very easily when the detection technique is mass spectrometry if an isotopically labeled version of the analyte(s) is available. Documentation of extraction/recovery efficiencies is not required for each sample if the sample matrix for a set of samples is constant. However, for a sample set for which the similarity of samples is not known, and for which adequate sample is available, each sample must be split and spiked with a known amount of the analyte(s) of interest to determine extraction efficiency. If adequate sample size is not available or the identity of the analyte(s) is not known before the fact (i.e., widescan analysis), surrogate spiking compounds may be used to assess extraction efficiencies in each sample without actually spiking each sample with the analyte(s) of interest. The choice of the surrogate spiking compound is tailored for each analyte. This surrogate must be chosen so that it is either chemically identical to the analyte of interest (for example using isotopically labeled analogues of the analytes), or is closely similar in terms of liquid/liquid partition properties, chemical reaction properties, column elution properties or other matrix isolation properties which are used to isolate the analyte(s) from the sample matrix. Another property of the surrogate spiking compound is that it not be present in the sample being analyzed.

2. Instrumental Analysis QA/QC

The final step in obtaining analytical results involves the use of an appropriate instrumental technique to produce the analytical data. Several QA/QC practices must be incorporated in this process to ensure validity of the data.

a. Instrumentation

Careful documentation of instrument usage, calibration, and maintenance is essential for the generation of high quality results. Each major instrument within the Ultratrace Analysis Group will be the assigned responsibility of a professional employee designated as the Instrument Steward. The Instrument Steward is responsible for making sure that logbooks are available and properly completed by instrument users. It is the

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responsibility of each individual using the equipment to record all the pertinent data required in the logbook. An example of a page from a GC/MS logbook is given in Figure 1 and that of a GC/EC logbook is given in Figure 2.

Instruments will be calibrated according to the instrument manufacturers' specifications or according to the method being used with the instrument. A log of instrument calibration will be maintained with the instrument.

The Instrument Steward will be responsible for ensuring that preventative and corrective maintenance programs are carried out on all instruments under his/her responsibility. If a specific problem is noted by an instrument user, it should be noted in the logbook and the Instrument Steward notified so that appropriate corrective action can be taken.

b. Blanks

A solvent blank will be analyzed at a frequency equal to the method blank. The method blank can serve as the solvent blank provided it is clean in the area of the analyte(s) of interest. This practice eliminates the possibility of background instrumental contamination.

c. Quantitation Internal Standard

A quantitation internal standard will be added to each sample following sample preparation but prior to instrumental analysis. This compound will serve for determining relative response factors and for correcting for variations in sample injection volumes and evaporation of solvents.

d. Standards

Standards are solutions of the analyte(s) of interest used to determine the response of the detector over the range of analyte concentrations found in the samples. For Quantitative analyses, sufficient standards and replication of standards will be run to demonstrate linearity and precision of the detector response over the range of analyte concentrations. Should sample concentrations fall outside the demonstrated linear range, either additional standards are prepared to expand the range, or appropriate dilution/concentration of the sample is done to place it in the proper range.

C. Degrees of Quantitation

The information provided for any particular analysis is tailored to the needs of the customer. In actual practice these needs vary across the broad continuum from the grossly qualitative to the precisely quantitative. It is informative to examine what is involved in providing the varying degrees of quantitation required by our customers.

MONS 023630

HP 5985A

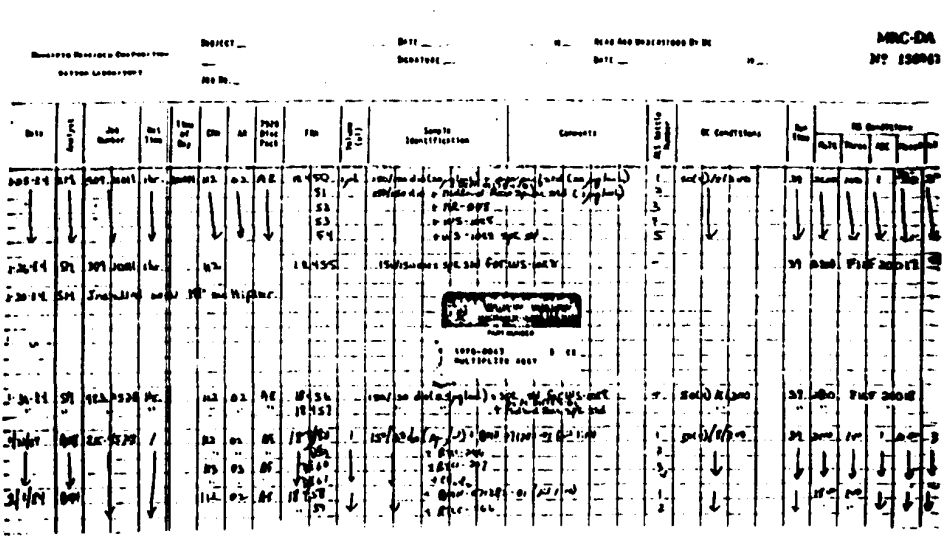


FIGURE 1. Example of GC/MS Logbook page.

MONS 023631

HP 5880 GC/EC

MRC-DA
N° 156363

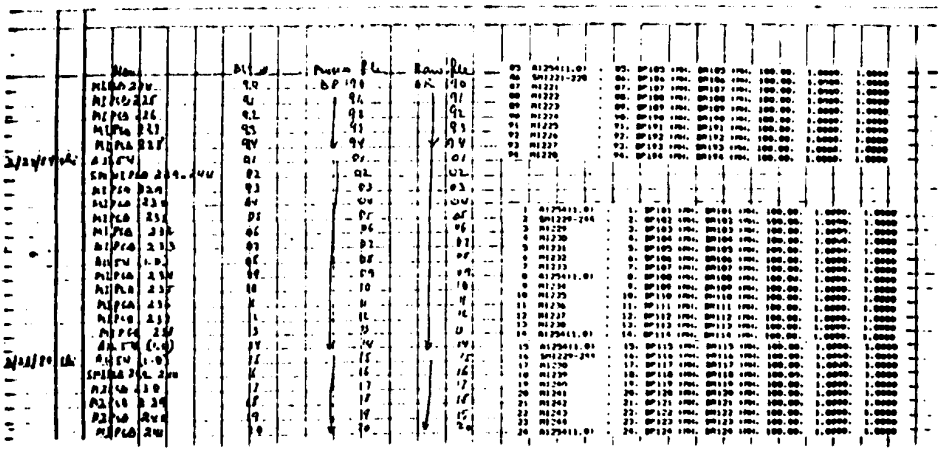


FIGURE 2. Example of GC/EC Logbook page.

1. Qualitative Analysis

Qualitative analysis implies no quantitation at all but rather seeks only the identity of the compound(s) present. GC/MS is particularly applicable to this type of analysis problem. Two types of identifications can be made:

a. Tentative

A tentative identification is made on the basis of the analyst's judgement of a match between the mass spectrum from the sample and a mass spectrum from a reference library. This is often done for the large number of compounds found in a widescan organic analysis.

b. Confirmed

A confirmed identification requires that the chromatographic retention time and the mass spectrum of the unknown compound match those of a known standard of the tentatively identified compound. This is often done in widescan analyses when standards can be obtained for tentatively identified compounds that are of concern.

The normal QA/QC requirements for qualitative analysis are (1) analysis of a blank, (2) analysis of duplicate samples and (3) use of a calibration compound (usually Decafluorotriphenylphosphine, DFTPP) to assure that the mass spectrometer will produce mass spectra that can be compared with library spectra.

2. Semi-Quantitative Analysis

Semi-Quantitative Analysis extends the information provided by a qualitative analysis to include some estimate of the amount of analyte(s) present. The method of semi-quantitation involves the comparison of detector response of the analyte(s) with that of an internal standard. The estimated concentration is obtained by assuming a unit response factor between the analyte(s) and the internal standard(s). An indication of recovery from the sample matrix is obtained from the recovery of the surrogate internal standard(s).

3. Quantitative Analysis

Quantitative Analysis provides for the accurate assessment of the amount of analyte(s) present in a sample. To be able to do quantitative analysis one must know the identity of the analyte so that standards can be prepared for spiking and for the generation of calibration curves. The method of quantitation is based on a three-point calibration curve using standards of the analyte(s) of interest over the range of concentrations in the samples. A one-point calibration method can be used (as with the laboratory data

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system) provided that linearity has been demonstrated over the concentration range of the samples using standards of the analyte(s) at three concentrations in the range. This calibration curve is then used to provide concentrations of the analyte(s) in the samples via an external standards method or more preferably via an internal standards method. For the internal standards method, the quantitation internal standard is added to each of the standards (just as with the samples) prior to instrumental analysis and relative response factors are determined.

For quantitative analysis the mass spectrometer is operated as a gas chromatographic detector. The MS is therefore tuned and operated in a manner specified in the approved method rather than using a calibration compound as required for qualitative analyses.

D. Quality Criteria and Remedial Action

The inclusion of QA/QC measures into an analysis program has little meaning unless the results are interpreted and appropriate action is taken when necessary. The ability to do this assumes some set of criteria that will trigger a response or remedial action. The following subsections describe the criteria that are used in the Ultratrace Analysis Group to evaluate when quality control data indicate a need for corrective measures and what remedial action will be taken.

1. Quality Criteria

The overriding quality criterion above all others is: Do the results meet the needs of the customer. This determination will be made jointly between the customer and the analyst. The analyst must exercise caution in not allowing the customer to overstate, misinterpret or unintentionally misrepresent the data. In addition to this prime quality criterion, the following conditions must be met:

a. Blanks - In the event that a blank produces a positive value, it may not be used as a valid blank if it exceeds 10% of any reported value for a sample.

b. Recoveries - Recoveries for all surrogate and native spikes must lie in the range of 50%<Recovery<150%.

c. Precision - Precision will be calculated based on the recoveries of surrogate internal standards. Average recoveries will be calculated with ± standard deviation for each sample set. Replicate analyses and low and high native spike recoveries may also be used to provide additional information/insight into the precision of the analysis.

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d. Limit of Detection - The limit of detection (LOD) will normally be stated as ½ the value of the low level native spike if it is recovered and quantifiable. Should the low level native spike not be recovered, the LOD will be stated as ½ the value of the high level native spike if it is recovered and quantifiable. Should neither spike be recovered, a decision must be made regarding the advisability of reprocessing the sample. Should all samples contain such large quantities of the analyte(s) that the levels of the low and high spikes are insignificant (<10%) by comparison, the value of the LOD becomes less important and is simply stated as the LOD for the instrumental analysis process determined by the analysis of standards and the ability to recover the surrogate spikes.

e. Accuracy - The accuracy of the analyses is best determined by analyzing some sample which contains the analyte at some predetermined level (e.g., an NBS standardized sample). If this type of standard is not available then the low level and high level spikes can be used as a measure of the accuracy of the analyses. Results must fall within a range of 75% to 120% of the excepted value for the analyses to be considered valid.

If the above quality criteria are not met the analyst will immediately assess the need for and extent of remedial action that may be necessary.

2. Remedial Action

The first course of action will be for the analyst and/or group leader to contact the customer and discuss the results pointing out the limitations of the data indicated by the quality control results. At this point the overriding quality criterion of customer needs will be addressed first. It could be that the customer's needs are met without meeting one or more of the other criteria. In such a case no remedial action is necessary. It could also be that the customer has such a demand for tightness in the results that his needs are not met even though the other quality criteria are. This would trigger further analyses with tighter quality control designed around the results obtained (e.g., replications of positives to better establish precision or spikes of non-detected samples to better establish the limit of detection).

In consultation with the customer additional experiments will be designed including the appropriate quality control to ultimately provide the results of sufficient quality to meet the customer's needs.

IV. Data

This section deals with the subject of data and how they are obtained, recorded and reduced so as to be useable in a report and yet readily available for referral and inspection.

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The samples will be labeled with a log number at sample login according to the procedure described in the ESC Section CLP Manual. When delivered to the responsible technical person, the samples may then receive additional labeling as deemed necessary. This additional labeling will be documented in a laboratory logbook and/or notebook. These logbooks and notebooks will be used by the analysts to record all pertinent work and results obtained from the samples. The laboratory notebooks will be filled out in accordance with Monsanto Uniform Notebook Procedures.

All raw data generated for a given project, which cannot be conveniently placed in the laboratory notebook (e.g., chromatograms, computer output, etc.), will be filed under some prescribed order (i.e., log number) in a central file under the supervisor of the Project Leader and/or QA Specialist.

Most of the data generated by the Ultratrace Analysis Group are computer data stored on magnetic discs. In order to archive these data the specific file name and disc number associated with the analysis will be written down in the laboratory notebook and/or instrument logbook. When the disc is completely full of data an entire unedited disc image or copy of the files on the disc will be transferred to magnetic tape. A logbook will be kept for recording the disc number, name and prefix associated with it, along with the associated tape and file number to which the disc was transferred. An example of such a logbook is given in Figure 3. When the data are transferred a verification program is used to check all transferred records for correctness. When tapes are filled (they typically hold seven to ten disc images) they should be stored in a fireproof vault.

All calculations of the results of the data will be specified in the method used. These calculations will be entered in the analysts laboratory notebook unless they are computer generated. In this case, a copy of the computer output will be filed with the raw data. The Project Leader will be responsible for checking the correctness of the data.

V. Sample Disposal

Samples will be disposed of in accordance with the instructions on the Analysis Request form. The only exception to this would result when the analysis reveals a problem which requires sample disposal in a manner other than that originally specified. If the Project Leader notes from the results of the analysis that a potentially toxic product was present in the samples then the samples will be disposed of in a manner deemed safe by the Safety Department. The date of sample disposal and the manner should be noted on the request form stored with the raw data completing the chain of custody of the sample.

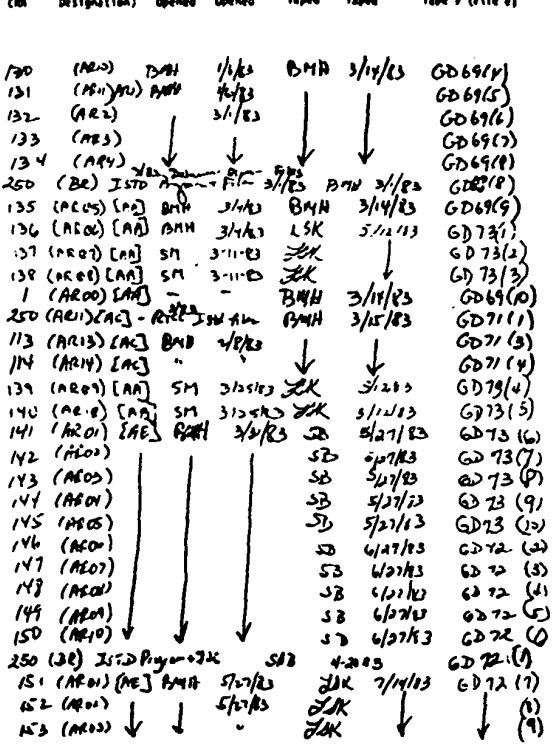


FIGURE 3. Example of DISC/Tape Logbook page.

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