



3

1340-13

908

W. R. Kieplein



Here's a Good One on Polar Discoveries!

IT was a dark night. A man was riding a bicycle with no lamp. He came to a cross-roads where there was nothing but a sign-post. He knew not which way to turn. He felt in his pocket for a match. He found but one. Climbing to the top of the pole, he lit the match carefully, and in the ensuing glimmer read:

"WET PAINT."

The Chamele



A brief history of the OLD DUTCH PROCESS of corroding lead. by L.G. Burr



[This is the first of a series of articles to be prepared by Mr. Burr, the Manager of our White Lead Department at Pullman. The series will cover, in addition to this article, "White Lead Processes," "Our Facilities," "Our Motto and Our Brand," and "Looking Forward." We commend the series most heartily to the careful reading of every Sherwin-Williams representative and employee, as it will give the complete details of our new proposition.—Ed.]



Mr. L. G. Burr.

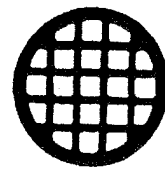
THE method employed in England and in the United States in the production of white lead, except for a limited amount, is that known as the "Dutch Process." In other words the Dutch are generally credited with the discovery of this successful method, though this contention is disputed by many. For instance, in an old Roman work written about 300 B. C., we find the following: "Lead is placed in earthen vessels over sharp vinegar, and after it has acquired some thickness of a sort of rust, which it does in about ten days, they open the vessel and scrape it off." There are many references which could be quoted to show that lead was manufactured in many other countries of Europe prior to the period usually referred to for the establishment of the industry in

Holland, i. e., the sixteenth century.

When metallic lead is exposed to the action of the air and moisture it becomes covered with a white substance composed of carbonate and hydrated protoxide of lead, which is the white lead of commerce. By the Dutch process white lead is produced by artificially creating these conditions and submitting metallic lead to the corroding effects of a warm atmosphere loaded with the vapors of carbon dioxide (carbonic acid gas) of acetic acid and of steam. The indications are that the chemical process in the corrosion of lead by the Dutch process involves at the start the oxidation of the lead and the formation of basic lead acetate; that is, basic lead acetate is decomposed by the carbon dioxide present, the basic lead being precipitated as white lead, and the neutral acetate again acting on more metallic lead is changed into the basic form ready to be acted upon again by the carbon dioxide, and so continuing as long as all the conditions requisite are present. Let this suffice for the chemical process which is undergone by the metallic lead and pass on to the method of handling the lead in the corroding stacks.

Quoting from an old work, we have the following used in Rotterdam and Amsterdam: "The lead sheets, which were rolled in a spiral and placed to eight inches high and four to five inches in at the top than at the bottom. To prevent this at the bottom they placed inside the pot, and at a depth, a piece of wood cut the length of the diameter was the Rotterdam method. At Amsterdam they moulded in the inside of the pot, and at about three little points which served instead of the wood. The stacks were built in one range of four, each square. After the pots had been filled up to an inch with vinegar, and the spiral lead placed in position in rows in the stack upon a bed of dung, the pots were placed together as closely as possible, and covered with the pots, plates of lead were laid over the whole layer covered with boards. These boards were covered with dung and another tier of pots placed as before, filling the stack and covered in the same manner. This was repeated until the tiers or layers were built up. The lead was left in the pots for four to five weeks, 'according,' Jars says, 'to the quantity of the dung.'" This method, in so far as corrosion is virtually the same as is in use today, except that the dung which has been used in tanning operations) is used instead of a flat lead plate (called a buckle) instead of a further fact that the thermometer and ventilating part in the handling of the stacks of today.

The process of today is as follows: A bed of tanbark of say fourteen inches depth is spread over what is called a stack. Upon this bed of bark are set in even rows hundreds of corroding pots. In the bottom of each pot is placed about a pint of 2½% acetic acid. The pot is then filled to the top with



A Lead Buckle.

lead buckles, each of which weigh about three-quarters of a pound. When the entire space is filled except for a tan wall on four sides a ventilating pipe with a two-inch hole is placed in position at or about the center of the surface is covered with a layer of boards. Upon the boards is placed another layer of tanbark, then more pots, acid and lead, and so on until the stack is filled to its limit, and the stack is then finished and ready to be manipulated by the corroder. The hole in the top of each vent pipe is provided with a small pear-shaped button, which is used to regulate the flow of the vapor from each bed, as the temperature

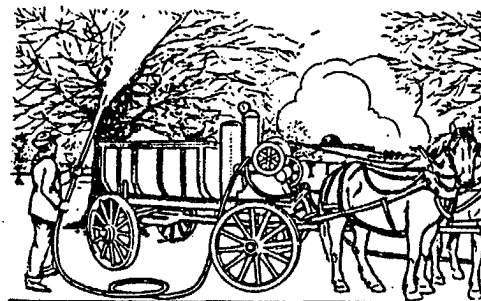
indicated by the thermometer and as the judgment of the corroder indicates. When the beds are closed, heat is generated by slow combustion in the tanbark, and at the same time oxidation takes place. The heat also evaporates the acetic acid, which attacks the lead oxide and forms a basic acetate of lead, which, when combined with the carbonic acid gas generated by the combustion in the tanbark, forms what is known as the white lead of commerce. The process requires approximately one hundred and twenty days from start to finish, or from the time of starting to put the lead in until it has all been taken out.

The results as to the percentage of lead corroded and the percentage of increase in weight vary in nearly every stack. The former is likely to be anywhere from 70 to 100 per cent, and I venture to say that year in and year out the average of all factories will not show over an 80 per cent corrosion. The increase does not vary greatly and it will be found that for every one hundred pounds of metal corroded there will be taken out from one hundred and twenty-four to one hundred and twenty-six pounds, or an increase of 24 to 26 per cent.

This ends the first step in the process, and all that remains is to put the lead in proper shape for the market. To do this the lead is first passed through a system of screens and rolls so that all metallic lead may be taken out; it then passes over to where it is put into water in order to pass it on to the mills to be ground to a proper degree of fineness and then pumped to a washing and floating system. The lead must be thoroughly washed to take out the acetic acid which remains in the lead from the process of corrosion. There then remains only to take out any tanbark which may have dropped in with the lead, settle out the coarse parts to be reground and float on the fine particles to the settling system, from whence it is pumped on to dry pans and the water evaporated. We then have our strictly pure dry white lead, ready to be pulverized, packed in barrels and sent to our various plants or ground in oil and sent into the market.



A Pig of Lead.



An Address on Insecticides Delivered at Richland County, Ill., Horticultural Convention

PREPARED BY H. M. ASHBY.



THE two most important spray materials today are Lime-Sulfur and Arsenate of Lead. The former is steadily gaining ground that with the latter the orchardist can combat any insect or fungus with two possible exceptions.

Since the Illinois Horticultural Society campaign I have had the pleasure of attending many society meetings in which it was brought out very prominently that the most advanced orchardists had experimented during the summer with Lime-Sulfur as a summer spray in place of Bordeaux. They had entire satisfaction. It is also being definitely recommended at the meetings that Lime-Sulfur as a winter spray for the next year will do a great deal in controlling subsequent development of the summer spray very materially.

Sherwin-Williams Lime-Sulfur Solution is a stable material. It is composed of Calcium Pentasulfide—that is to say, it is a very stable material composed of calcium combined with all the sulfur in the world and still stay in solution. This material tests not only as a summer spray and is less liable to crystallize than those solutions of the old combination of lime and sulfur salts containing a very apt to crystallize out in a hard, granular crystalline mass clogging up the spray nozzle.

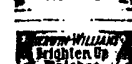
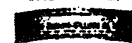
The use of Lime-Sulfur as a summer spray has been the result of experimentation behind it to warrant any careful, conscientious grower using it, at least to a limited extent on his orchard, and I am heartily recommending any man to try it on a portion of his orchard with great care in so doing. We do not feel, however, sufficiently far advanced for us to go out and make recommendations that Lime-Sulfur mixed with Arsenate of Lead

*G*EE, but these birthdays
and short months have
put a crimp in me for fair.
Guess I'll have to stay out of
Saint Patrick's parade on the
17th of March to make up for
lost time.

The Little Paint Man



Brighten Up
SHERWIN-WILLIAMS
PAINTS & VARNISHES



The Chamele

MARCH 1910



The Chameleon

Published on the Twentieth day of each month.
Canal Road, N. W., Cleveland, for the interest
benefit of THE SHERWIN-WILLIAMS Co.'s loyal



VOL. XIII.

MARCH, 1910

S W P vs. LEAD AND OIL

BY J. K. KENNEDY, OF THE MISSISSIPPI VALLEY



MUCH has been written
said concerning the re-
paint and lead and oil, and
toward showing why mixed
nomical to the house owner
to the dealer than white
mixed by hand.

It seems to me that
viz., the painter, has been
due mainly perhaps to the
in most cases would use white
sired in order to get the
dealer handled or recommended
quality.

Mr. J. K. Kennedy.

Again, the painter, the
building mechanics—the
for all the other mechanics

to be the very last one to be consulted as to what
is due largely to the erroneous idea, perhaps, that a
beat and will dope a job whenever they get a chance.

My experience is that if folks would trust the
much as they do their contractor, and put him on his
get even more from him than he agrees to give them.

But the point I wish to call to your attention in
the practical painter should use S W P."

The idea that a painter will be ridiculed, con-
not knowing his business because he uses S W P is
great delusion that has kept white lead and oil as the
idolized by painters.

And, again, some cling to lead and oil because
of mystery around their manipulation of it and
of their superior knowledge. This is another fallacy.
painter tells his intelligent customer that white lead
paint to use on his building he is literally insulting

client by practically telling him that all trades and sciences have advanced during the past twenty-five years, but that the paint business has stood still and remains just where it was twenty-five years ago.

Now you can't cram that statement down any thinking man's throat, for every well posted house owner today knows that the advances in good paint making have kept pace with the triumphs of modern engineering skill. The durability and economy of S W P have been proven by the most searching investigations and tests, and the unswerving conclusion is that (by reason of its admitted superiority under tests) the S W P of today is the correct paint for general use. In other words S W P is your old-fashioned lead and oil prepared in an up-to-date manner.

Does the painter *know* his lead is strictly pure? Is he certain it does not carry 5%, 10% or even 15% adulteration? Does he know that all proper care and precaution have been taken in its manufacture? How about his oil? Is he absolutely certain that it is not adulterated? And his turpentine—does he *know* it is chemically pure?

Then there is that probability of trouble with the drier. What is it made of? Does the painter know? And his tinting colors—are they strictly pure and does he know what kind of oil they are ground in?

How is the painter to know that the above ingredients are all pure and just what they should be? Does he test each and every one of them out himself or have them analyzed by an expert paint chemist to be sure they are right?

Is the painter certain that his japan will not burn the very life out of his oil? Does he put just exactly the same amount of oil, turpentine and japan to each 100 lbs. of white lead for second coating on outside work or does he *guess* at the amount each time he mixes up 100 lbs. of white lead?

There is not one painter in a hundred who is positive of these conditions, and there is not one in twenty that can tell you offhand how much oil, turpentine and japan are required to correctly reduce 100 lbs. of white lead to the proper consistency for second coating; and, furthermore, there is not one in twenty that can tell you offhand how many feet of surface 100 lbs. of white lead properly mixed for second coating will cover.

The majority buy whatever is offered as pure and go ahead *guessing* as painters have done from time immemorial. Sometimes they hit it right and get a fairly good job, and again it goes all to pieces in a short time, and they wonder why. Their customers also wonder why, and blame or praise them accordingly.

And this is the real reason that painters are largely looked upon as dopesters. It is their idolized lead which is causing them all their trouble. But they close their eyes to the real cause and blame something else, refusing to see the clay feet of their idol.

Does the painter know what chemical changes will take place when he mixes together all this lead, oil, turpentine, japan and coloring matter that he doesn't know is pure? And after he has his paint all stirred up with a paddle, what has he? The result is a mixture of paint that is coarsely ground with one grinding in oil, while the lead is still wet from

its washing; a mixture which upon examination would reveal undissolved lumps of lead; one which would cover only 450 square feet one coat to the gallon, and plenty of elbow grease will have to be used to

When all this is done the painter must confess what he has done or how long it will last after he has done it. On the other hand when he opens up a gallon can of SWP, a standard U. S. gallon of pure paint weighing 19 lbs., he gets the last degree of impalpable fineness. It is a pure white, blue or black added to kill the color of the oil, for the oil is extracted from the lead and zinc, and the oil clarified that will make a white mark on any other white mark. It is one that will work better and cover better than any other paint you can make or buy, one that carries more raw linseed oil than any other paint (all paint) than any other paint he could buy or possess.

In other words, he has a strictly pure paint composed of the best skill known to science—by many of the best men in their business over 40 years ago upon the rock of experience known today as the largest paint and varnish man.

Sherwin-Williams Paint, Prepared (S W P), upon the label. This enables the painter to know a great deal about the quality of the paint.

Now let's see what the comparative costs of the two methods are.

100 lbs. white lead costs	\$7.50—makes 25
4 gallons oil	3.28— " 4
3/4 gal. turpentine	.20— " 1
3/4 gal. japan	.15— " 1
Labor mixing	.50
	\$11.63

7 gallons lead and oil covers 3,150 square feet, one coat
4 1/2 gallons S W P covers 3,150 square feet, one coat

Clear saving first cost of material

In using lead and oil one is required to dip 7 gallons of paint to cover 3,150 square feet of surface. In using S W P one is required to dip up only 4 1/2 gallons to cover the same surface. This means that on the given surface one is required to use more of the lead and oil than S W P, to say nothing of the extra labor in getting down off the ladder or stage to get this extra 2 1/2 gallons. In other words, one is expending 30% more labor in applying the lead and oil than would be required to apply the S W P. You stop to pick up a nail that he had dropped, for his time is wasted.

For illustration, let us say a painter spreads sev

NOTE—Mr. Kennedy's figures in the above are very conservative, existing a few weeks ago before they were as high as they are now.—E

oil in a day; his wages are \$4.00 per day; 30% of this makes an additional \$1.20 for labor above what SWP would require. This \$1.20 lost labor plus \$2.95 saving in the first cost of material plus 50c for mixing the lead equals \$4.65, which is the actual loss on every 100 lbs. of lead he uses or \$93.00 loss on every ton of white lead. Now, then, if he uses 10 tons per year, this would equal \$930.00 per year, which is a pretty fair salary in itself. If he had been in business for 20 years this would amount to the handsome sum of \$18,600.00.

Look at every one of our prosperous and successful mechanics of today. They are the men who yesterday paid close attention to the details of their business—men who did not stop to pick up the nail, because, if they had, they never would have been successful.

Our representatives should call the painter's attention to the details of his business.

If the writer is right, it will be to your advantage to tell the painter to lay aside his old-time prejudice against everything but lead and oil. Tell him to try it out for himself—to paint two houses just alike in size and conditions, one with hand-mixed lead and oil and the other with SWP—and ask him to watch the results. He will be convinced that you are right and will make more money in his business this year than he ever made before.

My contention is based on 24 years of experience with SWP in many sections of the United States and the Philippine Islands at prices ranging from \$1.75 per gallon in the United States to \$2.50 per gallon in the Philippines.

The first argument a practical painter will usually advance is, "Mixed paint is no good; I use pure lead and oil and then I know what I have got." Well, let's pause a moment and see what he's got. The so-called strictly pure white lead is made by treating metallic lead with weak acetic acid (venegar), this gives Lead Acetate, which is treated with a carbon dioxide and water, leaving the white lead in a powdered form to be washed and ground in oil. If not sufficiently treated with carbon dioxide and water, the lead, instead of being washed, remains in the form of lead acetate, which will burn the life out of the oil, and, furthermore, if not sufficiently washed, free acetic acid remains, which also attacks the oil, killing its life and the durability of the pigments with which it is mixed. Thus the life of the paint is impaired.

When the dry white lead is being ground in oil it is possible to add 10% of an adulterant (usually Barium Sulphate) (Barytes).

Mr. Fallis Gives an Address.

Mr. W. S. Fallis, Sales Manager of the Winnipeg Sales Division, delivered an interesting address on "The Evolution of Paint and Varnish Manufacturing, and Application to Standardizing in Railway Finishes" before the Western Canada Railway Club recently.

The Sherwin-Williams Order of to December 1st, 190

Top-Notcher Manager,
Mr. J. A. Bowman, Savannah Sales Division.

Top-Notcher Salesman,
Mr. H. McCallum, Cleveland Sales Division.

Top-Notcher Superintendent,
.....

Top-Notcher Pro
Mr. J. R. Stout,

Top-Notcher Ca
Mr. J. S. Covert,

Top-Notcher Ra
Mr. W. B. Albri

Top-Notcher Mfg. Sales Representative
Mr. F. E. Draw, Eastern District.

General Standing Managers' Top-Notcher Competition, De

DIVISION	SALES MANAGER	PLACE	DIVISION
Savannah,	Mr. J. A. Bowman, T. N.,	1	Kansas City,
Los Angeles,	Mr. J. B. Kutz, T. N.,	2	Chicago,
Philadelphia,	Mr. C. P. Jarden, T. N. 3,	3	Dallas,
Cleveland,	Mr. A. E. Kydd, T. N.,	4	St. Louis,
Newark,	Mr. R. L. Graves, T. N.,	5	Winnipeg,
Cincinnati,	Mr. W. W. Haynes, T. N.,	6	Omaha,
Montreal,	Mr. R. C. Hison,	7	Portland,
San Francisco,	Mr. L. W. Wolcott, T. N.,	8	Boston,

DIVISION	SALES MANAGER	PL
Minneapolis,	Mr. J. P. Coleman, T. N.,	

Top-Notcher Travelling Representatives, December

TRADE SALES DEPARTMENT	TR
Central District,	Pacific Coast
1—H. McCallum,	1—G.
2—B. L. Reynolds,	2—W.
3—F. B. Kellogg,	3—S. C.
Western District,	Atlantic Coast
1—F. A. Bush,	1—A.
2—R. Clunton,	2—M.
3—G. J. Manassa,	3—F. I.
Mississippi Valley District,	New England
1—W. V. Hammond,	1—E. T.
2—I. A. Steger,	2—J. H.
3—O. K. Lois,	
Southwestern District,	Canadian Dis
1—K. Miller,	1—J. E.
2—W. F. Hall,	2—H. E.
3—F. C. Kenower,	3—J. W.

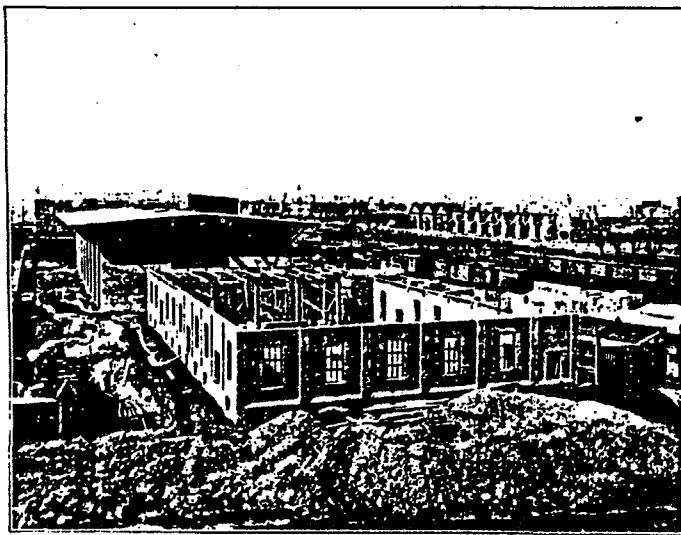
Our Modern White Lead Facilities at Chicago.



THE Sherwin-Williams Company has for a long time produced the most important raw materials that enter into its manufactured products. That white lead has been left until nearly the last is due to the fact that it has required immense property, large investment and considerable preparation to install a corroding plant. But now, after our recent purchase of twelve acres of land adjoining our present plant property at Pullman (Chicago), which gives us a total of about thirty acres, we are able to set aside space for a lead plant and still leave room enough to enlarge it, as we expect will be necessary after we are once well under way in the business.

In starting a line of manufacture, the first thing to do is to have the right man in charge. We have secured as superintendent of the plant a man who has spent 26 years in lead manufacturing under the Old Dutch process method. This man is Mr. L. G. Burr, whose photograph was shown in the last number of THE CHAMELEON, and who at this time has plans well under way for making our new venture a distinct success in every particular. The department foremen are all practical men of many years of experience, and as a consequence we are well equipped in this particular line.

Our consumption of carbonate of lead is so great that we can maintain in constant operation a lead manufacturing plant of considerable size.



The New Mill Building in the foreground to be four stories high.

But as it is the policy of the company to make, sell complete line of everything in the paint and varnish line, in addition to manufacturing lead used as an ingredient in our products, we are called upon to offer to our trade white lead of such quality as will be recognized throughout the country. For this reason we have adapted the Old Dutch process, explained in detail in the last CHAMELEON.

Our new lead plant is well under way and is being pushed vigorously, so that we expect to have all equipment installed and in active commission in fact, we expect to be in a position to ship Sherwin-Williams Process (O D P) Strictly Pure White Lead not later than the end of the year and we are prepared to book small orders now for the same.

In our illustrations we show two exterior views of the new plant. The building which is practically complete is 213 feet long and 84 feet wide and contains a capacity of 10,000 pots—each pot containing a capacity of 25 lbs. of lead.

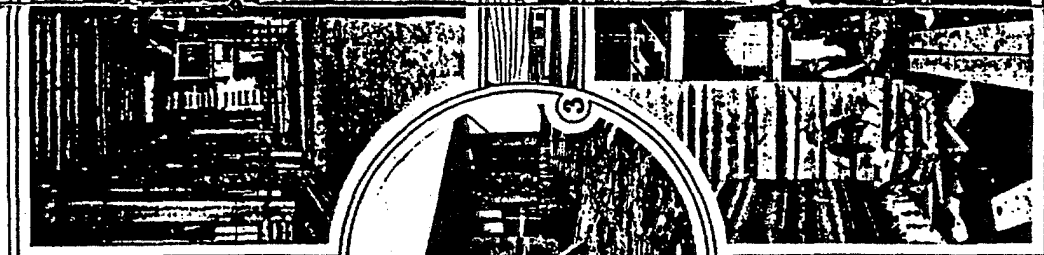
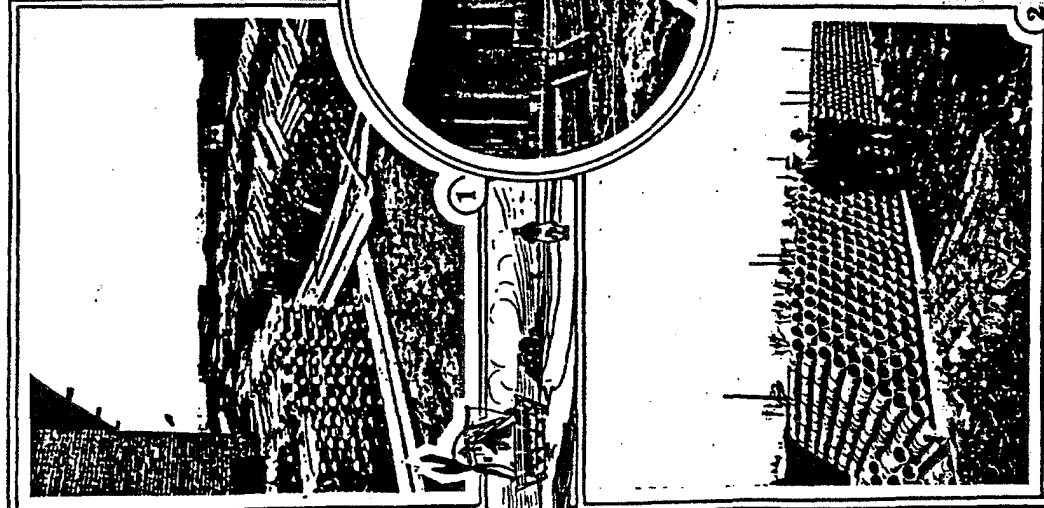
The mill building, the walls of which are only one story high, is 150 feet long and 75 feet wide, and where the pots are placed. This building will contain the grinders, kettles, the buckle casting machines, etc.

In the group illustration shown on our two covers we show a number of views which have been taken around the new plant and in order to give our readers a better understanding of the same we have numbered them and the following is a description of each.

Of course, the basis of all white lead is pig lead which is brought to the plant and is put into the stacks. As you can be seen in this picture there is a total of about 3,200 tons of lead, including labor, approximately \$150,000.00 worth of lead is the best double refined, silverized lead that has ever been analyzed which we have received from our Chemical Department. The analysis shows an infinitesimal amount of impurity. The significance of this is when you consider the fact that only the purest of lead can be used in the Old Dutch process, whereas, in all other processes, the lead will answer the purpose. The conclusion naturally is that the white lead in the market is therefore made by the Old Dutch process.

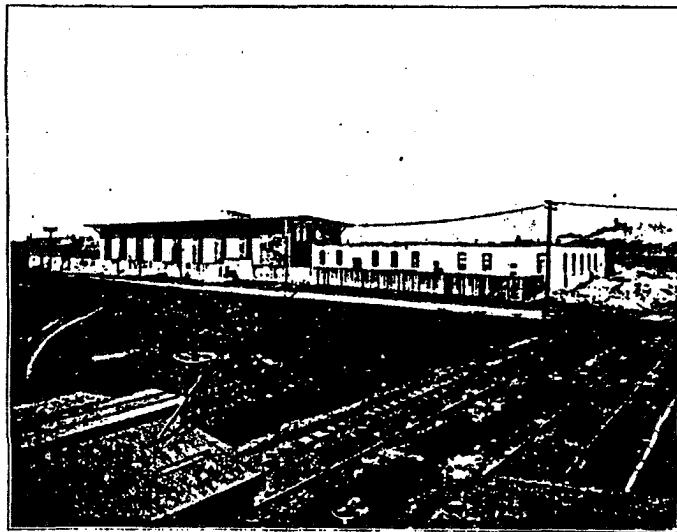
In view No. 2 is shown a pile containing 15,000 pots of lead ware corroding pots and is only a small part of the total. The pots are rapidly being set into the stacks at this time. When the pots are completely filled we will have used about 160,000 of lead. Each pot will contain about 25 lbs. of lead so you can see that we have over 4,000,000 lbs. of lead in the stack at this time. While there is some breakage each time that these pots are used, quite small, and they are used over and over many times.

View No. 3 does not require any special mention. It shows a portion of the end of one side of the corroding building. There are 25,000 or 30,000 more of the corroding pots shown in this view.



View taken in and about the new corroding building where we are corroding lead by the Old Dutch Process.

Views No. 4 and No. 6 show the upper deck between the two rows of corroding stacks, and the reader will notice the rail-guard for the stairways which lead to the decks below. There are three of these decks and their purpose will be better explained in connection with view No. 8. The upper deck is used only by the foreman or superintendent to enable him to readily direct the operations going on in various stacks on each side, and also to enable him to conveniently reach the top of each stack for the purpose of taking the temperatures. These two pictures show the two traveling cranes, which not only operate backwards and forwards through the entire space shown, but which also are so arranged as to leave the bridge behind to travel out over any individual stack. The work of these cranes is very important, as it makes the handling of material both into and out of the stacks very economical. When our mill building is completed, the buckle machine and a casting kettle will be placed in that building, and these cranes, leaving the corroding building and passing around the curve, will take the buckles from the mill building and deposit them in quantities of 3,000 lbs. at a time into any stack which we are filling. At the same time they will take the corroded buckles from the stack and deposit them conveniently in the separating house of the mill building. With these cranes we will be able to put lead into the stack and also take it out as economically, if not more so, than any concern in the country in this line of business.



Corroding Building, on left, containing 16 stacks.

View No. 5 shows the interior of the stack, which the pots are set, some of the pots in position and three of the decks which are used for storing. It shows the crane in the operation of lowering a weight down into the stack for use in filling these pots. It shows the act of putting buckles into the pot, but this is the bed of bark shown is entirely covered with pots. A layer of boards is placed over the top of the pots, and so continuing until the entire stack is covered with successive layers of bark, pots and lumber.

View No. 7 shows the casting, kettle and bucket in a temporary location, and through the doorway can be seen the supply of pig lead, explained in description of view No. 6. It also shows the hoisting hook of one of the traveling cranes attached to a wheelbarrow loaded with buckles and ready to take them out of the stacks. The buckle machine is simply an engine upon which we will cast approximately 75,000 tons of lead. You will also notice a pig of lead being fed into the machine. Chased our pure pig lead, it is only necessary to run it into these buckles in order to put it into convenient shape.

View No. 8 was taken on one of the lower decks of the stack. It shows the store empty corroding pots. It also shows the machinery which is shown in view No. 7 being only the temporary drive for the buckle machine. The stack is gone, instead of going to the expense of taking the pots outside and filling them, they will be rapidly reset on the decks and stored there until the stack is reset, when they will be handily packed back into the stack. The saving is considerable.

The above descriptions will enable our readers to see the operations used in the manufacture of our white lead, and to see with more accuracy than could have been possible without them.

We hope to show in the next CHAMELEON our packages, labels, and to tell how these goods are packed for the trade. These matters have not yet been definitely being discussed at the present time.

The Little Paint Man

I'm ready any time now to write those full pages for our agents. I enclosed an order blank with the letter, but if your agents haven't got it handy, tell 'em to just write them the ad. But be dead sure to give me the number of columns. It can be either 7 columns wide, 6 columns wide or 5 columns wide.