

WORLD HEALTH ORGANIZATION
ORGANISATION MONDIALE DE LA SANTÉ

JOINT FAO/WHO EXPERT COMMITTEE ON FOOD ADDITIVES
IDENTIFICATION OF A IDENTIFY AND TUNING OF FOOD
ADDITIVES AND THEIR ENVIRONMENTAL AND HEALTH
SOME ADDITIVES AND THEIR ENVIRONMENTAL AND HEALTH

AND

GENERAL CONSIDERATIONS ON IDENTIFYING A NEW FOOD

Geneva, 24 June 1972

ACETONE

by

Dr P. S. Elias
(Consultant)

Biological Data

Biochemical aspects

Acetone (CH_3COCH_3) is readily absorbed by solution into the blood after inhalation (Hagdt, 1924) and progressive accumulation in tissues has been noted after 6 hours inhalation (Haggard et al., 1944). Skin absorption occurs only slowly in animals (Lorenz et al., 1931) and probably none through intact human skin (Cesaro and Pinero, 1947). Large oral doses are mostly excreted unchanged through the lungs (40-70%) and the kidneys (15-30%) but small amounts are metabolised slowly to formate and acetate within 24 hours (Williams, 1959). Dogs excrete in their urine some 1-4% of acetone when administered orally at rates of 0.2-1.6 g/kg. Some 60% is excreted through the lungs (Schwarz, 1898). Small doses (1.7 mg/kg body weight) are oxidised mainly to CO_2 (53%) while some 7-14% is exhaled unchanged (Price and Rittenberg, 1950; Sakami and Lafaye, 1950). Some 10% of inhaled acetone is excreted through human skin (Parmeggiani and Sassi, 1954).

Acetone labelled with C^{14} in the methyl group was shown to be converted by rats to liver glycogen, serine, choline and methionine via formate and acetate (Sakami and Lafaye, 1950), while another metabolic pathway involved direct conversion to pyruvate in the liver (Sakami and Lafaye, 1951). Acetone injected into dogs produced high hepatic levels of β -hydroxybutyric acid (Valdiquie, 1935). Of 1 g. administered to man some 24% are excreted through the lungs (Rothkopf, 1936). I.v. injection into rabbits markedly increased acetoacetic acid and

ASI 00004237

The issue of this document does not constitute formal publication. It should not be reviewed, abstracted or quoted without the agreement of the World Health Organization. Authors alone are responsible for views expressed in signed articles.

Ce document ne constitue pas une publication. Il ne doit faire l'objet d'aucun compte rendu ou résumé ni d'aucune citation sans l'autorisation de l'Organisation Mondiale de la Santé. Les opinions exprimées dans les articles signés n'engagent que leurs auteurs.

p-hydroxyphenylpyruvic acid and glucose levels in their blood (Doeber et al., 1937). After i.v. injection at constant rate of 10 g acetone into man over 2 hours dose 110 mg were lost through the lungs (1 normal, 5 diabetic subjects). Total urinary acetone excretion did not rise significantly in 7 fasting normal subjects but some 360 mg acetone were excreted in 24 hours by this route. 12 normal subjects maintained high blood levels of acetone for 4 hours after injection but almost all had disappeared from the blood after 24 hours. Acetone did not lower blood levels in rats but in mice elevated for 4 hours after injection. No p-hydroxyphenylpyruvic acid was detected. Simultaneous i.v. glucose or i.v. insulin did not affect blood acetone levels in 7 normal fasting subjects. i.v. injection of acetone into 12 diabetic showed similar levels as in 19 normal subjects. Hence no impairment in ability to destroy acetone occurs in diabetes. The rate of breakdown is so slow, not very, as to preclude that more than a small part of normal fat metabolism passes through an acetone stage (Doeber et al., 1937).

Acute toxicity

Animal	Route	LD ₅₀ mg/kg bodyweight	LD ₁₀₀ mg/kg bodyweight	References
Mouse	Inhalation	-	46000 ppm	Schultze, 1932
Rat	oral	10700	-	(Smyth et al., 1962)
		9700	-	(Shell Chemical Co., 1969)
	i.v.	-	-	Spector, 1956
	inhalation	-	6-8 ml/kg	Walton et al., 1923
		-	32000 ppm/4 hrs	(Smyth et al., 1962)
		-	-	(Shell Chemical Co., 1969)
Rabbit	intragastric	5340	-	Spector, 1956
		-	5-10 ml/kg	Walton et al., 1923
	i.v.	-	6-8 ml/kg	Walton et al., 1923
	percutaneous	> 20 ml/kg	-	(Smyth et al., 1962)
		-	-	(Shell Chemical Co., 1969)
Dog	intragastric	-	8 g/kg	Albertoni, 1884

The acute effects of acetone are essentially that of a fairly strong anaesthetic (Elkins,). It also causes moderate inhibition of cholinesterase *in vitro* and in cats *in vivo* as shown by increased chemical action potentials in the carotid body nerve after intra-arterial administration (Landgren et al., 1953). Repeated intragastric administration of 8 ml to rabbits for 5-22 days or 8-10 ml to dogs for 8-25 days produced albuminuria and epithelial necrosis of convoluted renal tubules (Albertoni and Pisanti, 1887). Daily administration of 1-2.5 g/kg acetone intragastrically to 5 dogs for 9-12 days produced nephritis with tubular destruction (Poliak, 1925). Intragastric administration to cats caused local irritation of the digestive tract (Taylor and Frodsham, 1943).

Dermatitis from frequent sustained skin contact and irritation of eyes and the nasal mucosa have been observed in man (Browning, 1966). Variable renal effects in man have been reported (Browning, 1966). The LD₅₀ is 1800 ppm (Sax,). 15-20 g given orally daily to man over several days caused only slight drowsiness but no other ill effects (Albertoni, 1934). Acute massive intoxication in man has caused collapse and possible liver and kidney injury (Cox, 1940; S. 1th and Meyers, 1944; Harris and Jackson, 1951). Little absorption occurs through the skin after contact with liquid acetone (Browning, 1966).

Special tests

Mouse

Acetone was painted on the clipped dorsal skin of 60 female mice three times a week for 447 days (6.8 weeks) as part of another experiment. No local papillomata or carcinomas were seen (van Duuren et al., 1966).

Rat

The single injection of 0.1 ml, as part of another experiment, into one or both salivary glands of 25 male and 7 female rats produced local necrosis and complete regenerative repair after 3 weeks. Animals observed for 8 months showed no local or distant tumor formation either macroscopically or histologically (Cherry and Glöckmann, 1966).

Chicken

Injection of acetone into the yolk sac of fresh fertile chicken eggs had little adverse effect on hatchability (80% of normal) and produced no teratogenic lesions (McLauchlin et al., 1964). In another experiment only 37% of embryos survived, the remainder being killed by coagulation due to acetone. Surviving embryos developed normally (Walker, 1967).

Observations in man

Widespread industrial use over many years (e.g. 2000 ppm inhaled for 15 years) only yielded a few cases of mild intoxication but no reports of permanent haematological or organ damage (Browning, 1965; Rowe and Wolf, 1963; Falszett, 1963).

References

- Albertoni, P. (1934) Arch. Exptl. Pathol. Pharmacol., 13, 219
- Albertoni, P. and Filantti, G. (1937) Arch. Exptl. Pathol. Pharmacol., 22, 393
- Browning, E. (1965) Toxicity and Metabolism of Industrial Solvents, Elsevier, Amsterdam
- Caccari, S. (1957) Rivista di Farmac. exp., 21, 163
- Cesaro, A. and Vinciguerra, A. (1947) Riv. di Lavoro, 32, 334
- Cheney, C.P. and Glicksman, A. (1965) Brit. J. Cancer, 19, 287
- Elkins, H.B. () The Chemistry of Industrial Toxicology, New York and London
- Fassett, D.W. (1963) in F.A. Patty, Industrial Hygiene and Toxicology (1963) Vol.11
- Haggard, A.W., Greenberg, L.A. and Turner, J.M. (1944) J. Ind. Hyg. Toxicol., 25, 133
- Harris, L.C. and Jackson, R.W. (1952) B.M.J., 2, 1024
- Kagan, E. (1924) Arch. Hyg. Berlin, 24, 41
- Kochler, A.E., Wilson, E. and Hill, E. (1941) J. Biol. Chem., 140, 811
- Andrgren, S., Liljestrand, G. and Zotterman, Y. (1953) Arch. Exptl. Pathol. Pharmacol., 219, 185
- McLaughlin, J., Marliac, J.-P., Verrett, M.J., Mutchler, M.K. and Fitzhugh, O.G. (1964) Amer. Ind. Hyg. Ass. J., 25, 232
- Parmeggiani, L. and Sassi, C. (1954) Riv. di Lavoro, 45, 431
- Poliak, B. (1935) Arch. Exptl. Path. Pharmacol., 205, 220
- Price, T.O. and Fittenberg, D. (1950) J. Biol. Chem., 185, 449
- Rothkopf, H. (1933) Z. ges. exp. Med., 22, 464
- Rowe, V.K. and Wolf, M.A. (1963) in F.A. Patty, Industrial Hygiene and Toxicology (1963), Vol.11.
- Sack, G. (1940) Arch. Gewerbepathol. Gewerbehyg., 10, 80
- Sakami, W. and Lafaye, J.M. (1950), J. Biol. Chem., 187, 369
- Sakami, W. and Lafaye, J.M. (1951) J. Biol. Chem., 193, 199
- Sax, N.I. () Dangerous Properties of Industrial Materials, New York and London
- Schultze, D. (1932) Dissertation, Univ. Wurzberg
- Schwarz, L. (1898) Arch. Exptl. Pathol. Pharmacol., 40, 168
- Shell Chemicals Co. (1969) Industr. Hyg. Bull., I.C. 69-10
- Smith, A.R. and Mayers, M.R. (1944) Ind. Bull. New York, State Dept. Labor., 23, 174
- Smyth, H.F. Jr., Carpenter, C.P., Weil, C.S., Pozzani, U.C. and Striegel, J.A. (1962) Ind. Hyg. J., 23, 95
- Spector, W.S. (1956) Handbook of Toxicology, Vol.1.

Taylor, H. and Friedman, J. (1948) Report to ICI dated 22.8.48
Valdiquie, P. (1936) Compt. rend. Soc. Biol., 358
van Duuren, B.P., Orrill, L. and Nelson, H. (1966) J. Nat. Cancer Inst.,
35, 707
Walker, R.E. (1967) Toxicol. Pharm., 10, 293
Walton, D.C., Litch, R.E. and Lounsbury, J. G. (1938) J. Pharm. Exptl.
Toxic., 32, 175
Williams, R.A. (1949) Conduction Mechanisms, Chapman and Hall, London.
Lazarus, H.D., Gerschlager, A.J. and Langer, J.H. (1933), Arch.
Gewerbepharm. Toxicol., 2, 41