

Aroclor

Aroclor is Monsanto's brand name for a series of 3 capacitor impregnants composed of chlorinated diphenyl. The 3 grades are:

Aroclor 1242

(Commercially known as trichlorodiphenyl)

This grade now predominates and is used exclusively in many countries. It is the most economically attractive grade and is essential for capacitors installed outside in colder climates.

Aroclor 1248

(Commercially known as tetrachlorodiphenyl)

This grade continues to be used to a small extent, although it is less economic than Aroclor 1242.

Aroclor 1254

(Commercially known as pentachlorodiphenyl)

Although this grade was the first to be used in the capacitor industry, it has now been almost entirely superseded by Aroclor 1242.

These 3 grades of Aroclor are classified as 'capacitor askarels'. The term 'askarel' is defined by the American Society for Testing and Materials (ASTM) as 'a generic term for a group of nonflammable chlorinated hydrocarbons used as electrical insulating media'.

Aroclor was first used in the U.S.A. in 1931 and is now extensively used by the capacitor industry throughout the world. Monsanto has been producing Aroclor at its Newport (U.K.) plant since 1951.

Aroclor has been used in impregnated paper capacitors since its introduction into the industry. In more recent years its use has been extended to capacitors in which polypropylene film forms part of the dielectric.

Although most of the Aroclor impregnated capacitors operate at power frequency, some are used—with water cooling—at high frequencies, and some—with suitable additives—in d.c. circuits.

Advantages of Aroclor

1. Capacitors having the least volume and mass per unit of reactive power can be built.
2. Raw materials costs can be reduced because of this higher efficiency.
3. Labour costs can be reduced because smaller windings are required and plant throughput is increased.
4. Overload security is provided by uniform permittivity of impregnant and paper.
5. Long lasting capacitors result from high stability and resistance to oxidation.
6. Reduced fire hazard.

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Specified Properties of Aroclor

PROPERTY	TEST METHOD	UNIT	AROCLOR 1242	AROCLOR 1248	AROCLOR 1254
Appearance	ASTM D1702	—	Clear liquid free from suspended matter or sediment.		
Colour	ASTM D1702	APHA	50 max.	50 max.	50 max.
Density	ASTM D1810 and ISO/R649	g/cm ³	1.380 to 1.390 at 20°C	1.450 to 1.470 at 20°C	1.470 to 1.490 at 90°C
Refractive Index at 20°C	ASTM D1807	—	1.6250 to 1.6270	1.6305 to 1.6325	1.6380 to 1.6400
Kinematic Viscosity	ASTM D445	cSt	55 to 71 at 20°C	3.4 to 4.5 at 90°C	7 to 10 at 90°C
Flash point	ASTM D92	°C	170 min.	190 min.	None to BP
Fire point	ASTM D92	°C	None to BP	None to BP	None to BP
Pour point	ASTM D97	°C	-18 max.	-6 max.	+12 max.
Neutralization Number	Monsanto	mg KOH/g	0.01 max.	0.01 max.	0.01 max.
Inorganic Chlorides	ASTM D1821	ppm Cl	0.05 max.	0.05 max.	0.05 max.
Hydrolyzable Chlorides	ASTM D2441	ppm Cl	1.0 max.	1.0 max.	1.0 max.
Water Content	ASTM D1533	ppm	30 max.	30 max.	30 max.
Resistivity at 100°C, 500V after 1 minute	Monsanto	Tacm	2 ⁺ min.	2 min.	2 min.
Dissipation Factor at 100°C, 60 Hz	Monsanto	%	2 ⁺ max.	2 max.	2 max.
Relative Permittivity at 100°C, 60 Hz	Monsanto	—	4.7 to 4.9	4.5 to 4.8	4.15 to 4.35

*The value given in the table applies to bulk deliveries. For deliveries over long distances in drums the dissipation factor is increased to 4%, maximum, and the resistivity decreased to 1 Tacm, minimum.

Non-specified Properties of Aroclor

(Typical values are given for information only)

PROPERTY	UNIT	AROCLOR 1242	AROCLOR 1248	AROCLOR 1254
Mean coefficient of expansion over temperature range 20 to 90 °C*	10 ⁻⁴ /°C	6.8	6.8	6.6
Mean specific heat capacity over temperature range 20 to 90 °C	J/g °C	1.269	1.190	1.128
Thermal conductivity at 28 °C at 81 °C	W/m °C	0.1005 0.0879	0.0988 0.0967	0.0959 0.0925
Vapour Pressure at 90 °C	10 ⁻³ torr	10	4	1.4
Surface tension at 20 °C at 25 °C	dyne/cm	45.3 —	44.2 —	— 48.4

*Calculated from the formula $\bar{\alpha} = \frac{1n(P_1/P_2)}{\Theta_2 - \Theta_1}$

where P_1 = density at temperature Θ_1
 P_2 = density at temperature Θ_2

Handling Precautions

Aroclor drums and containers carry the following label which gives general instructions for handling:

IMPORTANT

CHLORODIPHENYLS (42%-54%) (or 80%) (PCB)

This product contains POLYCHLORINATED BIPHENYLS (PCB's) which some studies have shown may be persistent, an environmental contaminant and, possibly, injurious to certain forms of bird, aquatic and animal life.

PREVENT ANY ENTRY INTO THE ENVIRONMENT THROUGH SPILLS, LEAKAGE, DISPOSAL, VAPOURISATION, RE-USE OF CONTAINERS OR OTHERWISE.

SPILLS, LEAKAGES AND WASTE PRODUCT MUST BE COLLECTED.

Use of this product must be restricted to applications which can be controlled so that entry into the environment does not occur and to applications in which it cannot come into contact with food, animal feedstuffs or pharmaceuticals.

HARMFUL SUBSTANCE IF TAKEN INTERNALLY OR IF IN PROLONGED CONTACT WITH THE SKIN. CAUSES IRRITATION OF SKIN AND EYES.

DURING SHIPMENT AVOID SPILLS AND LEAKAGE INTO INLAND WATERWAYS AND THE SEA.

Store away from food, animal feedstuffs and pharmaceuticals.

Keep container tightly closed.

Do not eat or smoke when using.

Avoid breathing vapours, mist or fumes.

Do not wear contaminated clothing.

Wash product from skin with soap and water.

Wash eyes by flushing with water.

In addition to the foregoing we can add that:

1. Animal toxicity tests (small mammals) indicate that acute oral toxicity is low but that acute dermal toxicity may be moderate.
2. When handling the fluid minimise contact with the skin and wash any off with soap and water. People with sensitive skins should wear gloves resistant to chlorinated solvents or apply a silicone hand cream.
3. Should Aroclor accidentally splash or be rubbed into the eyes, they should be rinsed thoroughly with water for a few minutes. It is suggested that all operators be provided with safety glasses as Aroclor will cause pain if it gets into the eyes.
4. Elementary rules of hygiene should be applied—such as not smoking, eating or drinking during handling and contaminated clothing should be discarded.
5. Aroclor should be handled at ambient temperatures in well ventilated areas. If an Aroclor-filled capacitor is to be opened, it should be allowed to cool down to room temperature before opening it. If hot Aroclor has to be handled, a respirator with an active carbon canister should be worn.

Empty drums should not be re-used for anything except scrap Aroclor. Unwanted empty drums should be disposed of by a specialised waste disposal contractor familiar with PCB's who will crush the drums and bury them on an approved toxic dump in accordance with local regulations. Handling of Aroclor in bulk will minimise the problems associated with handling of drums. Waste

Aroclor should be disposed of by special high temperature incineration at 1000° C minimum with a residence time of 1.5 seconds or longer or alternatively by a method conforming to local regulations.

Care must be exercised during such disposal and the ultimate use of Aroclor to avoid pollution, particularly of sewers and waterways.

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The information, recommendations and suggestions made herein are offered in the light of prevailing legislation and on the basis of present scientific information (1972) available to Monsanto.

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Aroclor 1242 E1

It is well known that the working life of a.c. paper capacitors impregnated with askarel can be extended by the addition to the askarel of small amounts of suitable epoxy compounds, known as 'scavengers'. This effect is especially beneficial in the case of capacitors operating under arduous service conditions, such as those subjected to frequent surges.

However, some epoxy compounds have been found to be very difficult to control under the practical conditions obtained in existing capacitor impregnation plants.

Aroclor 1242 E1 consists of electrical quality Aroclor 1242 to which a specially purified epoxy compound has been added at just above the concentration at which it will be used. This product has been developed to provide convenient processing in the capacitor impregnation plant as well as sufficient activity to give an effective scavenging action in the finished capacitor.

The vapour pressure of the additive itself is very much lower than that of Aroclor 1242 so that the concentration of the additive will not be reduced during the degassing stage of the process when this is carried out over the usual ranges of temperature and pressure.

The concentration of the additive will be slightly reduced by a batch treatment process with Fullers' earth or other adsorbent. In order to allow for this reduction, the initial concentration of the additive has been set at a slightly higher level than would otherwise be necessary. In this way, after a batch treatment with an adsorbent of normal activity at a concentration up to about 1% by weight, the final concentration of the additive will be sufficient to give protection to the capacitors.

If the impregnant is purified by percolation through a refining column the additive will be initially heavily depleted by the adsorbent. However, the adsorbent later becomes 'saturated' with the additive whilst continuing to remove ionic impurities from the impregnant. The volume of impregnant required to saturate the adsorbent depends upon its mass and activity.

The properties of Aroclor 1242 E1 (see Table 1) are the same as those of Aroclor 1242 except for the following:

Scavenger Content: equivalent to 1.9 to 2.1 mg HCl/g
Dissipation Factor (tan δ) at 100°C, 50 Hz: <3%
Resistivity at 100°C, 500V, 1 min.: > 1 T a cm.

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Table 1

Aroclor 1242 E1. Tentative Data

PROPERTY	TEST METHOD	UNITS	VALUE
Appearance	ASTM D 1702	—	clear liquid free from suspended matter and sediment.
Colour	ASTM D 1702	APHA	50 max.
Density at 20°C	ASTM D 1810 & ISO/R 649	g/cm ³	1.38 to 1.39
Refractive Index at 20°C	ASTM D 1807	—	1.6250 to 1.6270
Kinematic Viscosity at 20°C	ASTM D 446	cSt	55 to 71
Flash Point	ASTM D 92	°C	153 min.
Fire Point	ASTM D 92	°C	none to boiling point.
Pour Point	ASTM D 97	°C	-18 max.
Neutralization No.	Monsanto	mg KOH/g	0.01 max.
Inorganic Chlorides	ASTM D 1821	ppm	0.05 max.
Water Content	ISO/R 760	ppm	30 max.
Hydrolyzable Chlorides	ASTM D 2441	ppm	1.0 max.
Scavenger Content	Monsanto (Appendix 1)	mg HCL/g	1.9 to 2.1
Volume Resistivity at 100°C, 500V, 1 min.	Monsanto	T $\pm$ cm	> 1 min.
Dispersion Factor (tan δ) at 100°C, 50 Hz	Monsanto	%	< 3 max.
Relative Permittivity at 100°C, 50 Hz	Monsanto	—	4.7 to 4.9

A suitable method for estimating the scavenger content is given in Appendix 1. It has sometimes been found difficult to clean the dielectric test cells that are used to measure the dispersion factor and resistivity of epoxide-containing askarels. We have found that it is helpful first to standardise the cell with pure Aroclor 1242 whose dissipation factor is already known. If this measurement is satisfactory, the cell is then rinsed twice with Aroclor

1242 E1 and the measurement proceeds in the usual way.

This product should be treated from the toxicity and safety aspects in the same way as Aroclor 1242 (see Appendix 2). For further information contact Monsanto at 10-18 Victoria Street, London SW1H 0NQ, Telephone 222 5678 or at Place Madou 1, 1030 Brussels, Telephone 19.11.01.

Appendix 1

Test Method for determining the quantity of hydrogen chloride scavenging additive in Aroclor

1. Principle

The method involves titrating the oxirane oxygen in glacial acetic acid with a standard perchloric acid solution to a crystal violet end-point. The titration is carried out in an excess of quaternary ammonium bromide, which, on contact with perchloric acid, produces hydrogen bromide. The hydrogen bromide, in turn, reacts with the oxirane oxygen.

2. Reagents

- Glacial acetic acid
- Chloroform
- Perchloric acid solution—0.1 N in glacial acetic acid (standardize against potassium hydrogen phthalate).
- Crystal violet—1 % solution in glacial acetic acid.
- Tetraethyl ammonium bromide solution—dissolve 100 g in 400 cm³ of glacial acetic acid.

3. Apparatus

- 10 cm³ burette
- Usual laboratory glassware

4. Procedure

- Weigh 2.5 to 3.0 of the askarel to the nearest 0.05 g into a 125 cm³ Erlenmeyer flask.
- Add 10 cm³ of chloroform, 40 cm³ of glacial acetic acid and 10 cm³ of the tetraethyl ammonium bromide solution and swirl to dissolve.
- Add 1 drop of crystal violet indicator and titrate with 0.1 N perchloric acid to the greatest colour range (violet to green-blue).

5. Calculation

$$\text{Hydrogen chloride equivalence (mg/g)} = \frac{36.5 \times T \times N}{W}$$

where T = titre of perchloric acid (cm³)
N = normality of perchloric acid
W = weight of sample (g)

6. Additional Notes

- The titrant is very stable and does not require frequent restandardization.
- No blank correction should be required if good quality reagents are used.

Appendix 2

Toxicity and Safe Handling

- Avoid skin contact. Protect the hands, preferably by oil-resistant gloves.
 - If skin contact should occur, wash off with soap and water.
 - If hot Aroclor must be handled in a closed or confined area provide exhaust ventilation or wear an approved respirator.
 - If Aroclor gets in the eyes, flush with large amounts of water. In the event that irritation persists refer the patient to a Physician.
 - Exposure to vapours or mists should be avoided.
- If these precautions are neglected acne may develop and excessive exposure may cause liver damage.

Caution

Aroclor is composed of polychlorinated biphenyls, which some studies have shown may be an environmental contaminant. Extreme care should be taken to prevent any entry into the environment through spills, leakage, use, disposal, vaporization or otherwise.

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