ORGANIC RESEARCH

COPIES
WHITE - File: PINK - Requester GREEN - Group Mgmt

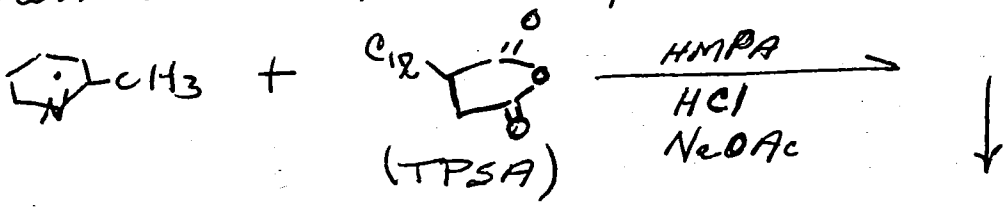
GENERAL PHYSICAL CHEMISTRY
GROUP WORK REQUEST

DATE	CHARGE - Job No.
1/18/66	6674
REQUESTER	LOCATION
Frank Clark	206

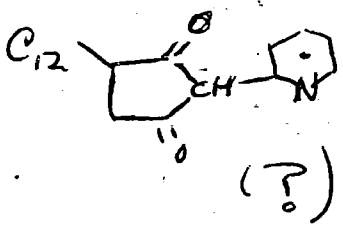
WORK REQUESTED OR ANALYSIS REQUIRED

GLC

REASON: Determine the extent of reaction



COMMENTS (e.g. Accuracy requirements, timing situation)



SAMPLE INDENTIFICATION (notebook page numbers, lot numbers, dates, or other information that will be unequivocal indefintely.)

~~70139~~ 70139
final solid

SPECIAL INFORMATION (Data that will facilitate our work such as known solvents for molecular weight samples, melting point if properties in solid and liquid state are needed, maximum process temperature for thermal stability samples, boiling points for gas chromatography, chemical reaction involved, etc.)

Unstable ____, *Explosive* ____, *Toxic* ____, *Stench* ____, *sublimes* ____, *Hygroscopic* ____, *Use dry box* ____

M. P. ____, *B.P.* ____ *mm, Solvents* ____

Other ____

DSW 298156

O-627

STLCOPCB4068067

This side to filled in by general physical chemistry group

REPORT OF RESULTS

DATE 1-23-67

TO:

CC:

No positive results by G.L.C.

Most of starting material gone but no product eluted

DSW 298157

G. Gasser

SIGNATURE

STLCOPCB4068068

ORGANIC RESEARCH

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WHITE - File: PINK - Requester GREEN - Group Mee

GENERAL PHYSICAL CHEMISTRY
GROUP WORK REQUEST

DATE: 12 | 22 | 66

CHARGE - Job No.: JFQ Dept. 809.3?

REQUESTER: Dick Wagner

LOCATION: W - Bldg.

WORK REQUESTED OR ANALYSIS REQUIRED

Naphthalene Assay on Choc. Bayou Naphthalene Concentrate Samples

REASON: For Blending of MCS-327 EDM Fluid (Mixture of Naph. Conc. and Santowax OM)

COMMENTS (e.g. Accuracy requirements, timing situation)

SAMPLE INDENTIFICATION (notebook page numbers, lot numbers, dates, or other information that will be unequivocal indefintely.)

Lot QH 20009 Drum (1) & (2)

" " Completely melted drum

SPECIAL INFORMATION (Data that will facilitate our work such as known solvents for molecular weight samples, melting point if properties in solid and liquid state are needed, maximum process temperature for thermal stability samples, boiling points for gas chromatography, chemical reaction involved, etc.)

Unstable _____, *Explosive* _____, *Toxic* _____, *Stench* _____, *sublimes* _____, *Hygroscopic* _____, *Use dry box* _____

M. P. _____, *B.P.* _____ *mm, Solvents* _____

Other _____

DSW 298158

Q-627

STLCOPCB4068069

This side to filled in by general physical chemistry group

REPORT OF RESULTS | DATE

TO: | CC:

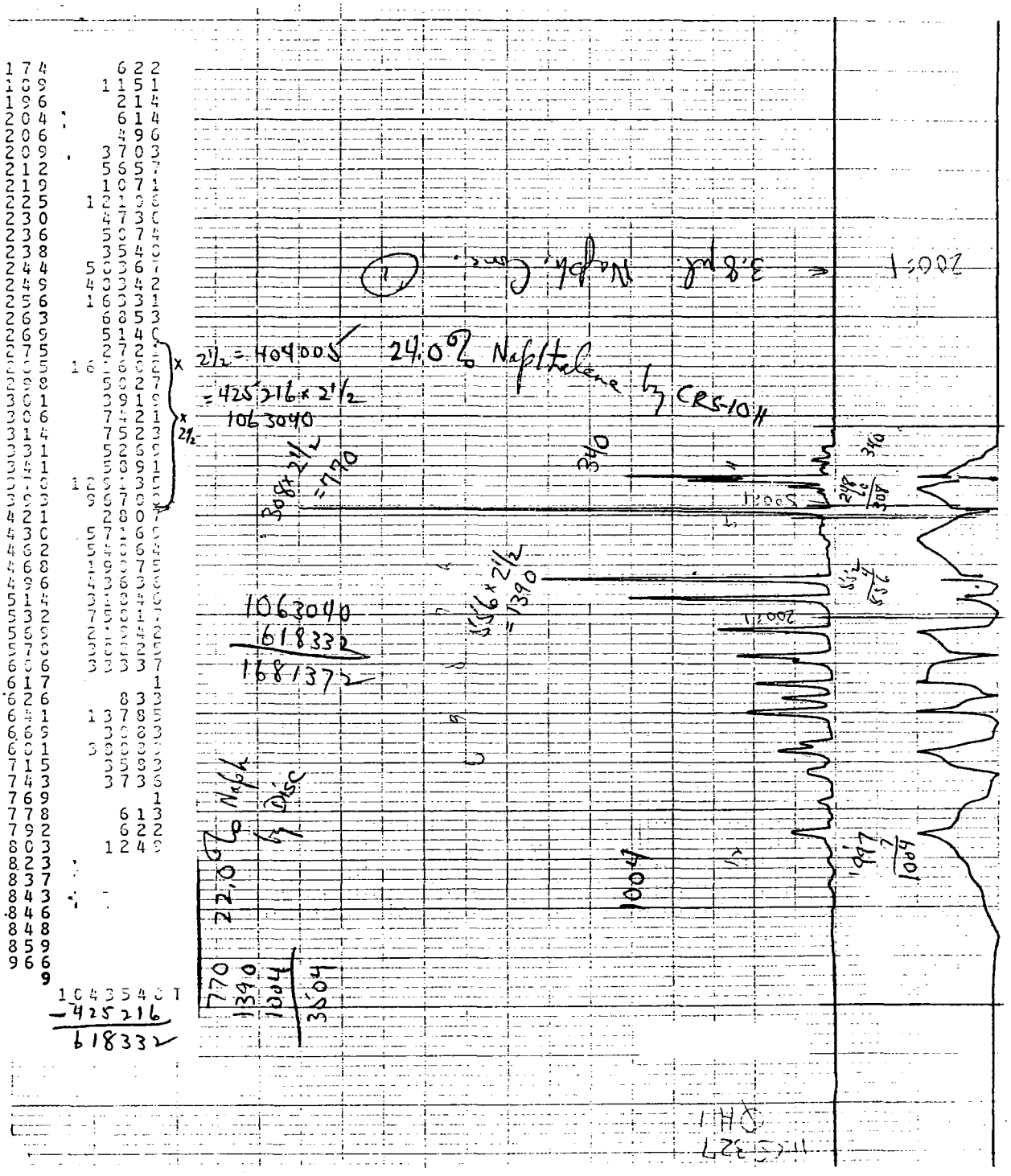
Sample	Naphthalene Content (Wt. %)
QH20009 - (1)	24.0 %
" - (2)	24.8 %
" - Completely melted	32.2 %

DSW 298159

E. M. Emery

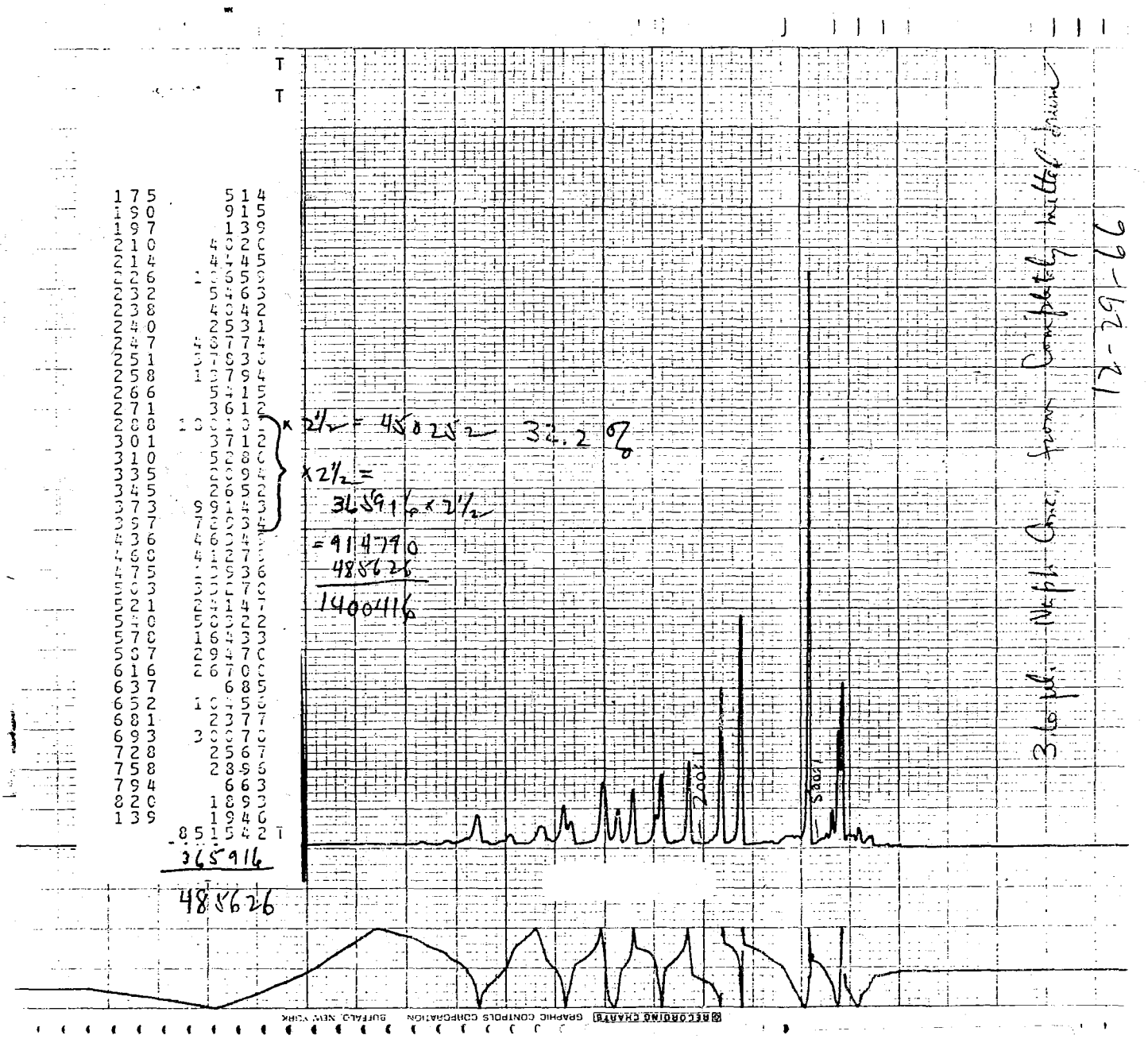
SIGNATURE

STLCOPCB4068070



DSW 298160

STLCOPCB4068071



DSW 298161

STLCOPCB4068072