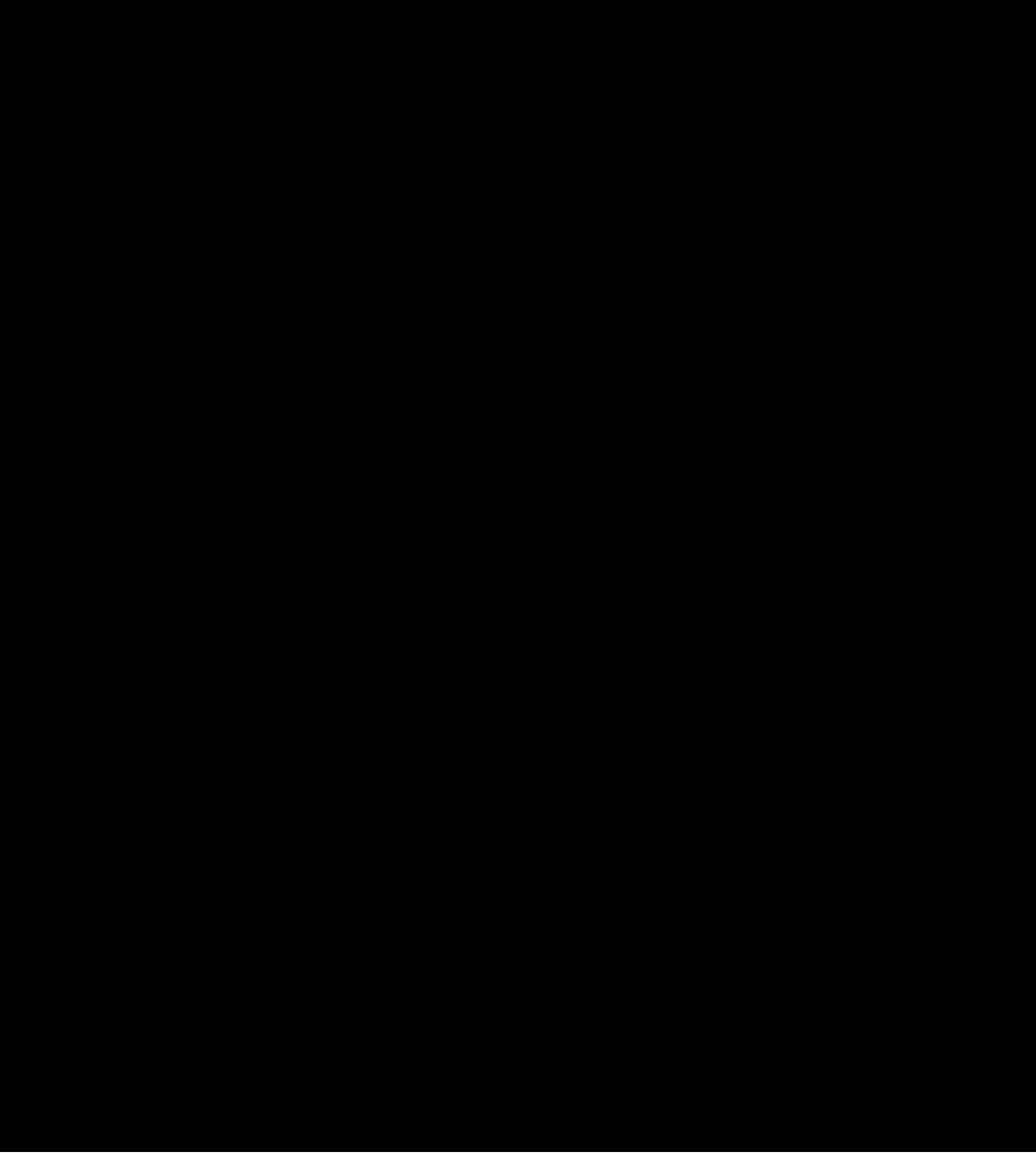




Lead Paints Used On New York Highway Bridges

THE excellent rust-inhibiting properties of red lead paint pigment has made it the standard metal protective paint primer wherever steel is left exposed to the weather. This is particularly true of State Highway Departments, by far the majority of whom specify red lead for the important shop coat of all exposed steel highway structures.

As a result of this widespread use of red lead primer paints, its bright orange color is and will continue to be the motorist's signpost of future highways that will speed him along non-stop, from coast to coast.

New York's Department of Public Works, which has jurisdiction over that state's highway construction, specifies shop coats of red lead over all shop-fabricated steel work. In addition, a field touch-up with the same red lead paint is given after construction on all unpainted surfaces or where the original primer coat has been worn off or has otherwise become defective.

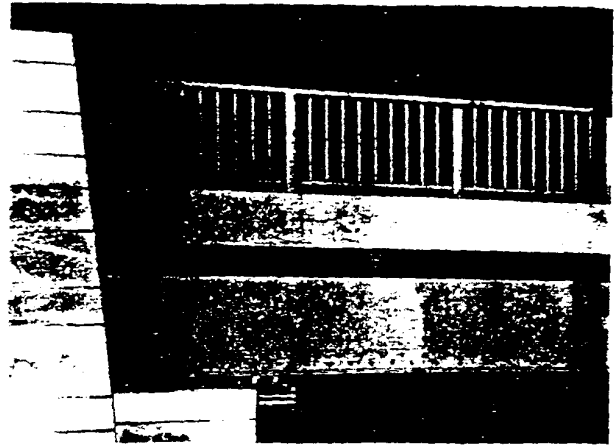
The primer paint specified by New York State is composed of red lead (95 percent grade), 78 to 81 percent

and the vehicle. 19 to 22 percent, the vehicle to consist of a mixture of raw and boiled linseed oil in the proportion of one-third to one-half boiled oil, the balance being raw oil.

One of the many bridges now nearing completion in New York State's highway program is shown in the accompanying photograph. This 110 ft. single-span bridge over the sec-

tion of the New England Thruway in Bronx County, N. Y. at Baychester Ave. is one of several such bridges being constructed by the Central Iron Works, Inc. New York City.

The first and second field coats of paint on this bridge and similar structures is a mixed pigment paint containing a minimum of 50 and a maximum of 70 percent white lead.



A priming coat of red lead protects the steel on the Baychester Ave. bridge over the New England Thruway in Bronx County, N. Y. Top coats are white lead base paints.

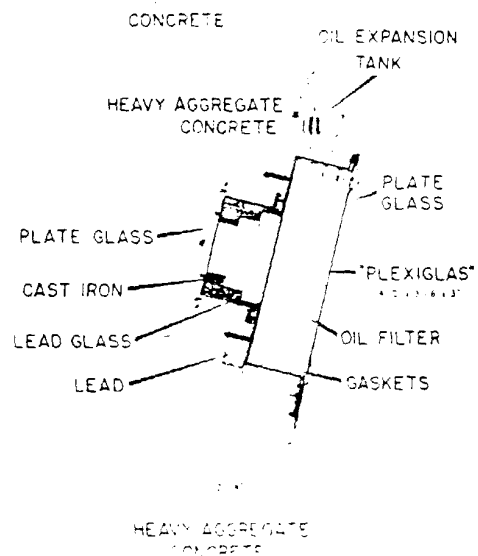
Atomic Reactor Shielding Windows

THE problem of viewing radioactive areas that require operational and maintenance work and at the same time maintaining the necessary shielding against gamma rays and neutron bombardment has been uniquely solved by the engineers at General Electric Co., Richland, Washington, operators of the Hanford Atomic Products Operation.

Because shielding and viewing requirements are usually not compatible and even the requirements for gamma and neutron shielding differ greatly, a combination of materials, which had to be light transparent, was necessary. One material had to be of sufficient den-

Details of construction of the leaded glass window recently installed at the Hanford Atomic Products Operation, Richland, Wash.

REACTOR SHIELDING WINDOW



sity to protect against gamma rays and the other containing sufficient hydrogen to absorb neutrons. How this was accomplished in the minimal space is shown in the accompanying drawing of a window recently installed to view the radioactive rear face of a Hanford reactor. The window is two feet thick and gives the same protection as five feet of concrete.

Construction of the window involved three layers of 4 in. thick lead silicate glass which has good light transmission for all colors except blue and with a density of 6.2 is approximately half as effective as the metal lead in the absorption of gamma rays. The light loss at the surfaces of the

glass was reduced by applying a low reflective coating. Neutron protection was afforded by four sheets of 3 in. thick, the largest commercially available, "Plexiglas" which also has good light transmission as well as 8 per cent hydrogen. To eliminate reflection at the surfaces of the "Plexiglas" a clear mineral oil with the same index of refraction was placed between the layers and confined with a gasket. The two outer surfaces were extensively polished.

In the installation of the windows in the concrete shield, extra sheets of lead were placed around the perimeter of the leaded glass to provide shielding in those areas where the concrete was of insufficient thickness.

By the use of two of these windows, engineers and physicists at the Hanford reactor can view an area approximately 80 ft. square and from a central station operate by remote control many slave devices accurately and with comparative ease and safety.

Alternate to this type of window is one filled with distilled water which has good light transmission and good resolution with monochromatic light. However, this type of window is not feasible where space limitations are a factor. Because of the limited shielding afforded by water, the size of the window becomes objectionable at high radiation levels, ten feet of water being equal to approximately five feet of concrete or eight inches of lead.

Leaded Porcelain Enameled Steel Panels

THE availability of porcelain enameled metal panels for architectural applications has generated almost an entirely new concept in modern construction.

Architects now find it economically feasible to design buildings having interior and exterior enameled surfaces of varying color hues and shades that blend harmoniously with the surrounding terrain; to design with colors that are pleasing to the eye or to serve as company or product identification. The mechanical advantages, such as easy and fast erection and the easily maintained and color-fast surfaces, are also a decisive factor in their selection and use.

Porcelain enameled steel and aluminum sheets are both used for this purpose. The enamels applied to aluminum of necessity have a high lead content to provide a firing temperature below the melting point of aluminum and at the same time result, in a smooth, glossy, tenacious and weather-resistant coating.

Now enamelers are finding the properties of somewhat similar leaded enamels on steel advantageous. By the use of lead in these enamels, the firing temperatures are being reduced from the 1500-1600°F. range to 1000-1100°F. This not only reduces the costs of firing but at the same time results in enamels of superi-



THE O. HOMMEL CO.

Note the remarkable flatness and lack of distortion of these steel panels enameled with low temperature leaded porcelain enamels fired as low as 1,000 to 1,100° F.

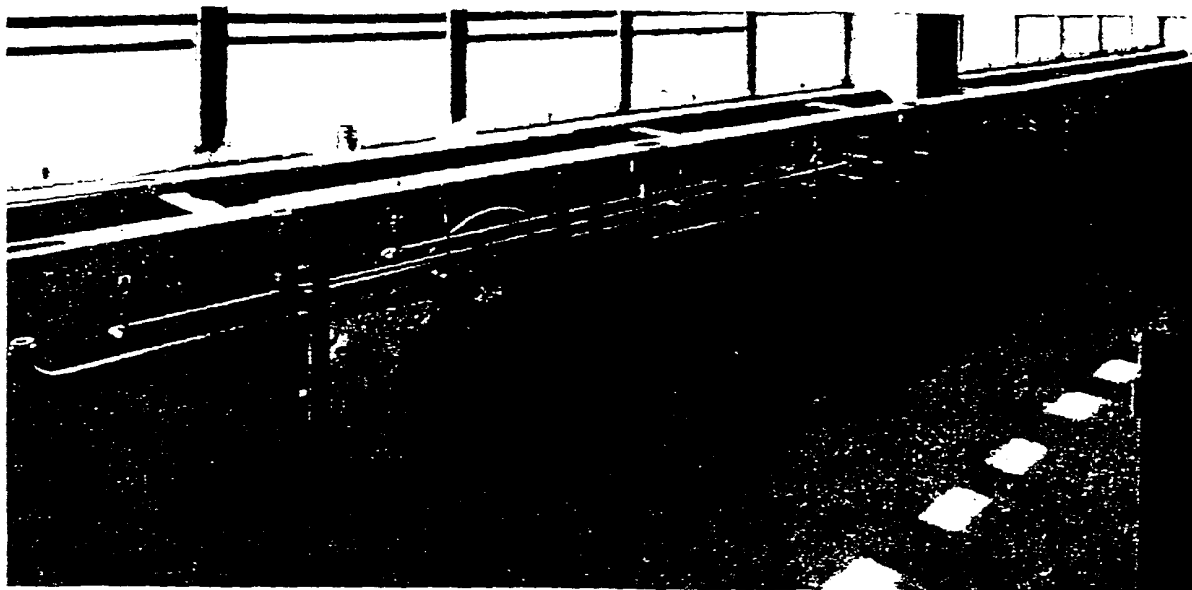
or resistance to chipping and weathering and good elasticity.

Enamels fired in this lower temperature range mature below the transformation temperature (strain point) of the steel, making possible the use of lower cost cold rolled steel as the base metal. Such problems as warping, sagging, boiling and re-boiling, common with higher firing temperatures, are eliminated or materially reduced, and thinner sheets can often be enameled successfully.

The use of lead compounds in

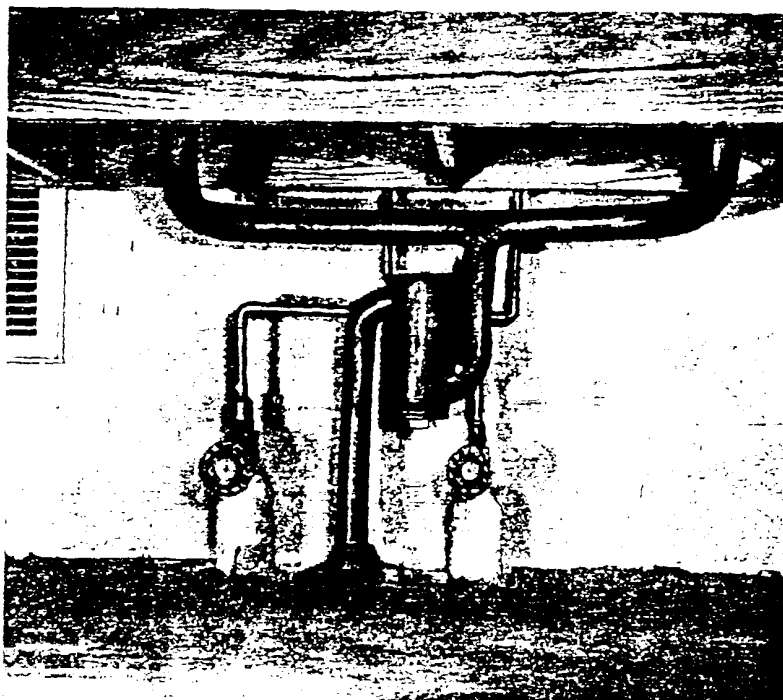
porcelain enamel compositions is primarily as a flux. They have an advantage over most fluxing compounds in that they can be added in relatively large amounts without causing a dull finish. Lead-bearing porcelain enamels are opening new vistas for the porcelain enameeler, the architect and the product designer.

The leaded porcelain enamels used on the steel panels shown in the accompanying photograph were produced by The O. Hommel Co., Pittsburgh, Pa.



Lead waste lines and lead cup sinks for two of the chemical laboratory tables in the new Education Wing of the University of Arkansas' Medical School in Little Rock, Ark. Lead's corrosion resistance and ease of fabrication make its use ideal for such installations.

Lead's Corrosion Resistance Properties Used To Advantage In Chemical Laboratories



LEAD has greater corrosion resistance to more acids or combinations of acids than probably any other common metal. As a result, it is ideally suited for use in chemical laboratories where a variety of corrosive media is constantly being used. In such surroundings, lead's use has been widespread for such things as hood linings, table tops, trough linings and for the chemical plumbing drainage system.

Many of the common corrosive chemicals used in laboratory experiments, such as ammonium and sodium hydroxide, chromic, hydrochloric, hydrofluoric, phosphoric and of course, sulphuric acid, are all safely handled by lead or lead-lined equipment.

In the new Education Wing of the University of Arkansas' Medical School in Little Rock, Ark., a number of chemical laboratories are being installed and lead is used extensively.

In the chemical plumbing system, 1½ and 2-in. lead waste pipe, class "C" weight, (WW-P-325 — Federal

These lead waste lines from two chemical laboratory sinks entering a single lead drum trap require no special fittings for the sharp bends.



When chemical corrosives are to be conveyed in lead, all joints, shown here, should be lead welded.

Lining the troughs in the laboratory tables with 4-lb. sheet lead.

Specification for Pipe, Bends and Traps; Lead (For) Plumbing and Water-Distribution) will carry the chemical wastes from 300 laboratory fixtures to acid resisting cast-iron drainage lines. Lead drum traps as well as lead cup sinks are also being installed.

All joints in the lead pipe and between the lead pipe and fittings are being made by lead welding, the method of joining lead that should always be used when chemical corrosives are apt to come in contact with the jointing material.

Troughs to carry the chemical

wastes from each of the work positions at the laboratory tables are being lined with 4-lb. sheet lead and the joints are also lead welded.

In jobs of this kind, the advantages of lead, other than its corrosion resistance, are its ease of fabrication, need for very few special fittings, and the absence of joints over long lengths and in the sharp bends and turns, advantages that are indicated in the accompanying photographs.

Joints are always a potential source of trouble particularly in corrosive atmospheres and the fact that lead requires a minimum number of

joints can add immeasurably to the life of the installation.

The Kewaunee Co. supplied the lead cup sinks and other laboratory equipment and the chemical plumbing system was installed by the Arkansas Mechanical Contractors, Inc., North Little Rock, Ark.

Copies of the Federal Specification WW-P-325 previously mentioned and a suggested specification for the use of lead in chemical laboratories entitled "Suggested Specifications for Chemical Laboratory Drainage System" may be had from the Association free, upon request.

Another of the many lead waste lines from the lead cup sinks in chemical laboratory tables installed in the new wing of the University of Arkansas' Medical School.



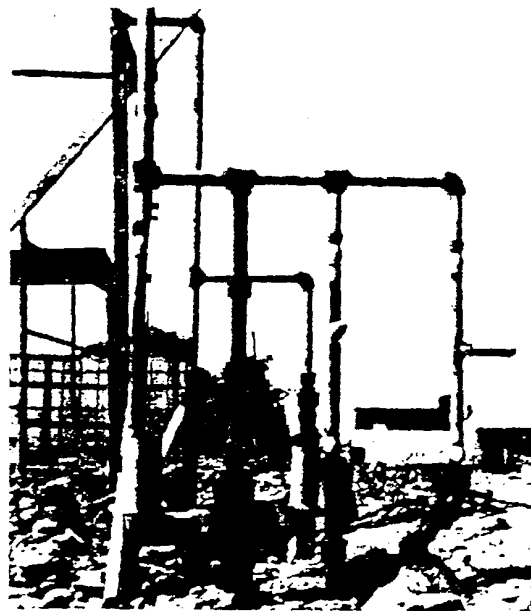
Plumbing In Place Before Building Is Erected

THE concept of small home and building construction wherein the orderly procession of building tradesmen on the construction site would be foundation worker, bricklayer or carpenter, roofer and then the plumber, is undergoing a radical change in many sections of the country. The plumber now finds himself on the job simultaneously with the foundation worker, thus speeding construction and lowering cost.

As the accompanying photographs of recent construction jobs in Tulsa, Okla. and Shreveport, La., would indicate, it has become a common sight to see the plumbing roughing-in completed before any other part of the prospective building is visible.

One of the problems connected with the installation of the plumbing before the house is built is the question of accuracy in placing the plumbing roughing-in. When construction is well enough along to set the fixtures, any slight errors made in the placement of the roughing-in can create a serious and costly problem with rigid pipe connections. However, with lead connections, such inaccuracies are not a serious problem. The flexibility of lead permits bending the connection by hand, if nec-

The roughing-in for a first floor toilet room in the building to be erected around it, in Tulsa, Okla., has lead wastes and vents for the lavatories and lead water closet connections.



essary, to compensate for whatever small errors in measurement that may exist.

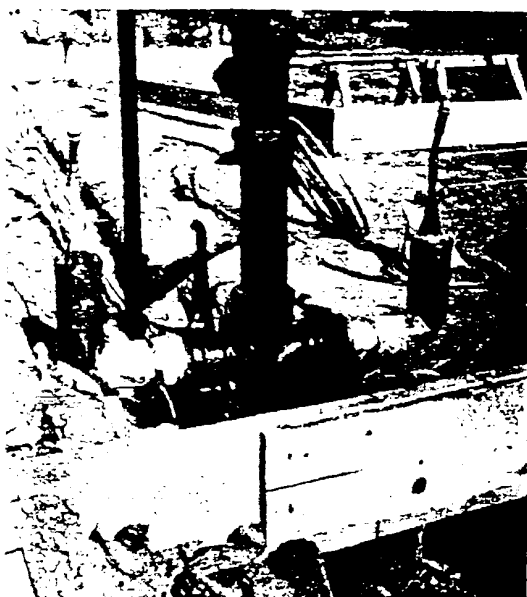
Small ranch-type homes now under construction in Cherokee Park, just outside Shreveport, La., are having the plumbing roughing-in set in place before the concrete slab is poured

and lead is being used for fixture connections.

The Elton J. Wolfe Plumbing Co., of Shreveport, La., is installing 4-in. lead bends for the water closets along with lead stubs for the bathtub wastes in each of the homes' 11½ baths. The flashings for the plumbing vent stack passing through the roof are to be 3-lb. sheet lead.

The same construction procedure is also being used in the new Home Building and Loan Association building in Tulsa, Okla. The plumbing contractor, W. F. Glenn Co., Tulsa, Okla., has completed the installation of the plumbing roughing-in for the first-floor toilet room prior to erection of the building. Lead waste and vent connections for the lavatories and lead bends for the water closets will provide the necessary flexibility to the plumbing system.

The contractor has wisely placed paper packing tubes around the lead bends, see accompanying photograph, to protect them during the pouring of the concrete floor and upon their removal will leave sufficient working space for setting the water closets.



Prior to the pouring of the concrete slab for this small home near Shreveport, La., the plumbing roughing-in, including lead bends, is set in place.

Shipping Radioactive Isotopes By Air

RADIOACTIVE isotopes for physical therapy and medical research are being used in increasing numbers in hospitals throughout the world. Because the half-life of many of these radioactive isotopes is limited, rapid delivery schedules must be maintained and this is accomplished by the use of regular commercial airline flights.

One of the airline companies engaged in this operation is B.O.A.C., British Overseas Airline Corp., which handles several thousand such shipments originating in Great Britain every year. Special regulations laid down by the International Air Transport Association for the handling of



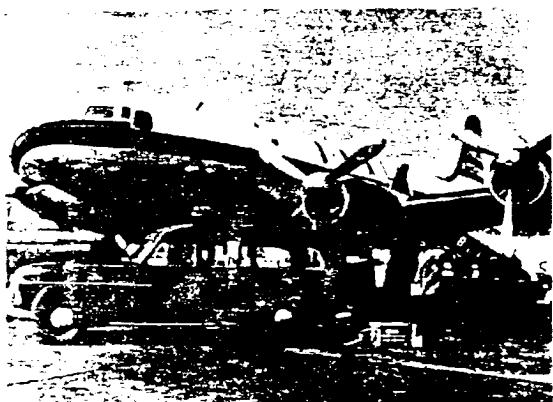
An overseas shipment of radioactive isotopes is removed from its lead box and placed in the tip of airliner's wing to begin its long journey.

radioactive materials assure that there is no risk of contamination to either the crew or passengers.

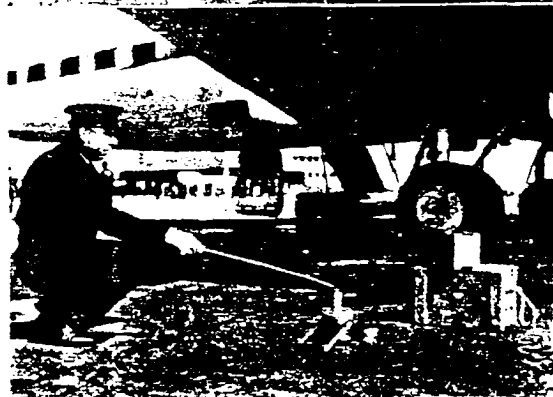
The consignments for overseas destinations arrive at London Airport in lead containers, carried in a

wooden box. The containers are immediately transferred to a safe made of 2 in. thick lead where they await shipment. Shortly before the aircraft is due to leave, the head loader takes the consignment out to the tarmac and transfers the radioactive material in its casing from the lead container to a small compartment in the tips of the airliner's wings. Here it is completely safe being well away from the passengers and crew. Both B.O.A.C. Argonaut and Constellation airliners are equipped to carry consignments in this fashion.

The shipments handled by B.O.A.C. are distributed chiefly by the Radio Chemical Center at Amersham, Bucks. and the Atomic Energy Research Establishment, Harwell, Berkshire, both in England.



The consignments of radioactive isotopes arrive at London airport in lead containers carried in a wooden box.



The head loader on the tarmac at London airport demonstrates the use of a three foot long pole in handling the radioactive materials.

RADIOISOTOPE CONTAINERS USED BY OAK RIDGE NATIONAL LABORATORY

Reprint of a report with the above title, containing descriptions and detailed drawings of shipping containers, prepared by the Oak Ridge National Laboratory, is available, free of charge, upon request to the Lead Industries Association, 60 East 42nd St., New York 17, N. Y.

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HECLA MINING CO.	Wallace, Idaho		
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The following articles and reprints contain information, drawings and specifications on the use of lead products helpful to architects, engineers and others interested. They are available, free of charge, except where otherwise indicated, upon request to the LEAD INDUSTRIES ASSOCIATION, 60 EAST 42nd STREET, NEW YORK 17, N. Y.

MATERIALS OF CHEMICAL CONSTRUCTION ANNUAL REVIEWS — LEAD AND LEAD ALLOYS

Up-to-date reports on new developments in the uses of lead and lead alloys in chemical construction including research progress, engineering applications, and technological advancements. With bibliography. 1956, 1955 and 1954 reviews (editions are not cumulative) available. Reprinted from "Industrial and Engineering Chemistry," September, 1956, October, 1955, and October, 1954.

LEAD REFERENCE SHEET (Parts 1 & 2)

A short outline of the types, grades, and chemical and physical properties of lead in chemical construction, with a summary of the corrosion characteristics of lead in use with a large number of common chemicals and solutions used in various industries. Reprinted from "Chemical Engineering Progress," January and February, 1955.

INSPECTING LEAD LININGS

Latest methods of testing and inspecting sheet lead and bonded lead linings used in the chemical process industries are described. Reprinted from "Chemical Engineering," November, 1956.

BEST DESIGNS FOR LEAD INSTALLATIONS

A special report on the applications of lead and lead alloys in the chemical and metallurgical industries. Recommended practice for joining lead sheets, and the construction of wood stove tanks, laundries, towers and flues where lead is involved are included. Reprinted from "Chemical Engineering," March, April and May, 1956.

PORTABLE RADIATION SHIELDING MATERIALS

A special report on commercially available shielding forms that can be used to build readily assembled, hand stacked shielding walls. Prepared by the editors of "Nucleonics," May, 1955.

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BULLETIN NO. 2

Specification for Lead Shower and Safe Pans

SUGGESTED SPECIFICATIONS FOR CHEMICAL LABORATORY DRAINAGE SYSTEM

A complete guide to the selection and installation of drainage systems in chemical laboratories. Includes laboratory waste and vent piping, traps, sinks, acid trays, troughs, safe pans, neutralizing basins and other topics.

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A 24 page booklet containing descriptive information as well as four pages of detailed drawings for plumbing installations in various type houses.

LEAD WELDING

An informative as well as practical guide to lead welding. Reprinted from the "Welding Handbook, Third Edition" published by the American Welding Society.

LEAD FLASHINGS IN MODERN CONSTRUCTION

Lead flashings as used in an ultramodern house in Randolph, Vt. Reprinted from "National Roofing," October, 1954.

THE STORY OF LEAD—MATERIALS THE RAILROADS BUY (Parts 1 and 2)

Lead from mine to finished product with special reference to its uses by the railroads. Reprinted from "Railway Purchases and Stores," December, 1954 and February, 1955.

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A series of technical letters containing red lead paint formulations. (Available separately.)

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Technical Letter No. 8—Primers for Railroad Structures and Rolling Stock

Technical Letter No. 9—Red Lead Paints for Galvanized Steel, Rusted, Weathered or Treated New

Technical Letter No. 10—E.I.A. Formula 3-7, A Red Lead-Alkyl Vehicle Primer for Industrial and Marine Environments.

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