

FILE NAME: Owens Corning (OC)

DATE: 1935

DOC#: OC294

DOCUMENT DESCRIPTION: Journal Article - Ruck Wool in Relation to Health

AFTER THIS REPORT HAS SERVED YOUR PURPOSE AND IF YOU HAVE NO FURTHER NEED FOR IT. PLEASE RETURN IT TO THE BUREAU OF MINES. THE USE OF THIS MAILING LABEL TO DO SO WILL BE OFFICIAL BUSINESS AND NO POSTAGE STAMPS WILL BE REQUIRED.

UNITED STATES
DEPARTMENT OF THE INTERIOR
BUREAU OF MINES

OFFICIAL BUSINESS

RETURN PENALTY LABEL

THIS LABEL MAY BE USED ONLY FOR
RETURNING OFFICIAL PUBLICATIONS.
THE ADDRESS MUST NOT BE CHANGED

PENALTY FOR PRIVATE USE TO AVOID
PAYMENT OF POSTAGE. \$300

BUREAU OF MINES,

WASHINGTON, D. C.

01 501 0758

ROCK WOOL IN RELATION TO HEALTH*

LAWRENCE T. FAIRHALL, PH.D., STEWART H. WEBSTER, PH.D., AND
GRANVILLE A. BENNETT, M.D.

*From the Laboratory of Physiology, Harvard School of Public Health, and the Department
of Pathology, Harvard Medical School, Boston, Mass.*

THE utilization of furnace slags in the form of wool for insulation purposes has become so extensive that the former metallurgical by-product is now manufactured direct and marketed under various names such as "rock wool," "mineral wool," etc. In insulating houses this material is blown into all spaces in the outer walls and a covering laid down in the attic so that the house is completely insulated. Since individuals living within the houses are surrounded by such material it has been questioned whether the use of rock wool could be a problem of hygienic importance. Inasmuch as this material enters into construction in such a manner that the occupants are separated from it by plastered walls which act as very efficient filters, the only conceivable way in which it could cause injury (assuming harmful ingredients) would be by means of vapor, either direct or as a result of slow decomposition. The following experiments were devised to test this effect.

In this investigation all the raw materials entering into the manufacture were examined as well as the fin-

ished product.† Although the latter is a relatively low silica glass, stable, insoluble and chemically inert, a searching investigation was made of its chemical composition and of the possibility of any vaporization of injurious substances. The presence or absence of any poisonous element would to a large extent establish the material as either dangerous or innocuous—not completely, it is true, for, assuming a very toxic combination of certain elements, this combination might escape detection in the search for the individual elements. The practical test therefore was to determine the effect of the material upon animal life. To this end the following experimental procedure was adopted.

EXPERIMENTAL PROCEDURE

Cats were selected as experimental animals since they are sufficiently sensitive to poisons, can be housed conveniently, and have tissues sufficiently large for extended chemical investiga-

† Our thanks are due the Johns-Manville Company for the materials provided for this investigation. In addition to the raw materials entering into the manufacture of the rock wool, the finished products of three plants geographically distributed as follows were investigated: Manville, N. J., Alexandria, Ind., and Waukegan, Ill.

* Received for publication October 15, 1935.

non. Three types of procedure were followed with the animals: One group (1) was exposed to any fume or vapor that could be evolved by passing air over the rock wool at room temperature. Another group (2) was exposed to air passed through the rock wool so heated that the air current acquired a temperature varying between 67°C. and 82°C. The third

materials and of the rock wool from the various sources showed the following elements to be present and readily detectable—chemically: silicon, calcium, magnesium, sodium, iron, aluminum, sulfur, titanium, manganese and zinc.

When large enough samples were used (10 gm. or more) an extremely minute amount of lead could be de-

TABLE I
CHEMICAL COMPOSITION OF VARIOUS ROCK WOOLS

	A	B ₁	B ₂	C
SiO ₂	43.11	45.20	42.11	42.69
CaO	28.37	21.96	23.93	21.26
MgO	10.33	11.18	13.67	13.88
R ₂ O ₃	9.98	11.83	11.11	14.56
Na ₂ O	7.26	9.19	9.52	7.65
MnO	0.28	0.37	0.19	0.30
S	0.13	0.023	0.028	0.16
Cl	0.039	0.006	0.051	0.025
H ₂ O	0.11	0.155	0.17	0.125
Total	99.969	99.921	100.812	100.650

R₂O₃ includes Fe₂O₃, Al₂O₃ and TiO₂.

B₂ manufactured oil-free.

Oil content

A	B ₁	B ₂	C
per cent	per cent	per cent	per cent
1.18	1.08	Oil-free	0.06

group (3) in addition to the same exposure given to (1) were further exposed for 1 hour daily to a dusty atmosphere of very finely ground rock wool. All these groups received the exposure indicated for 2 months continuously.

At the end of this time the rats were chloroformed and their tissues removed for pathological investigation and chemical and spectrographic analysis.

Preliminary analysis of the raw

material and of the rock wool from the various sources showed the following elements to be present and readily detectable—chemically: silicon, calcium, magnesium, sodium, iron, aluminum, sulfur, titanium, manganese and zinc.

When large enough samples were used (10 gm. or more) an extremely minute amount of lead could be de-

for weight changes or other effects of their environment. At intervals of 3 weeks the rock wool was changed so that the animals were exposed to all three types. The eight cats receiving this exposure were chloroformed at the end of the 2 months and their tissues

rock wool when filled. It was externally heated electrically by means of a winding of nichrome resistance wire (Fig. 2). Air was blown through at the rate of 9 cu. ft. per minute. The air leaving the apparatus at a temperature of 67-82°C. was obviously too

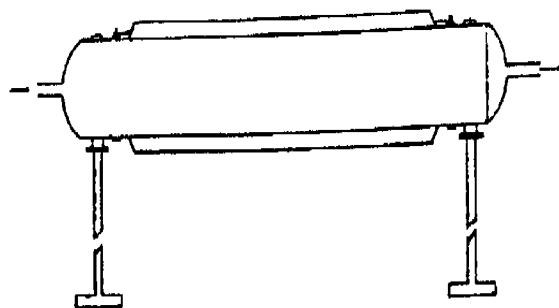
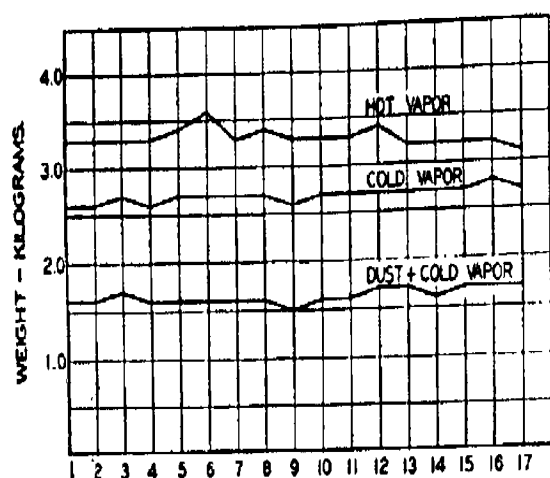


FIG. 2.—Hot vapor cylinder



EFFECT OF ENVIRONMENT UPON THE AVERAGE WEIGHT OF CATS DURING TWO MONTHS EXPOSURE.

FIG. 3.—Weight changes during the experimental period

removed for chemical and pathological examination.

II. THE HOT VAPOR EXPERIMENTS

The apparatus used for heating the rock wool was a metal cylinder 7½ inches in diameter and 4½ ft. in length; the air tubes at either end were 1 inch in diameter. It held 10-12 pounds of

hot to breathe and was therefore passed through a coil of 1-inch-copper tubing immersed in running water and was thereby brought down to a temperature slightly below outside room temperature. The animals were confined to this atmosphere for 2 months, successive lots of rock wool (A), (B) and (C) being used. The space in

which the animals were kept was small and wholly dark, except at such times as when cleaning, feeding and weighing were carried on. Furthermore in spite of a liberal run of cooling water the room

during the remainder of the time to the conditions of group I. The cats were exposed in groups of four at a time in the apparatus shown in Fig. 4. Compressed air (freed from the slight moisture usually present by passing over calcium chloride) was forced through the rock wool which had been repeatedly extracted with carbon tetrachloride and then finely ground in a ball mill. It was found to be rather difficult to produce dust with the com-

course. The average particle size of the floating dust after allowing the coarser material to settle for a few minutes was 2.2μ . Impinger samples of a fixed volume (5 cu. ft. of air) were taken during the actual dusting. The distilled water suspensions thus obtained were evaporated to dryness in platinum dishes and weighed. The dust content in terms of milligrams per cubic meter is shown in the chart (Fig. 5). No attempt was made to

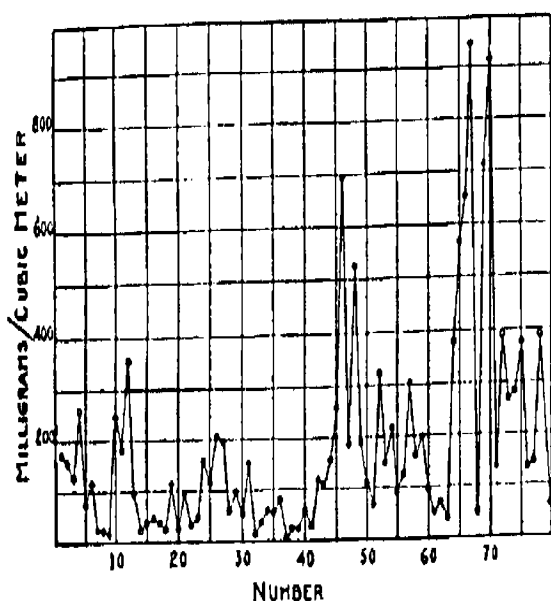


Fig. 5.—Dust concentrations during the experimental period

mercial type of oiled rock wool no matter how finely ground. Furthermore what dust was carried into the air remained in suspension a very short time only, and quickly settled out. The oil very effectively prevents dust formation. When repeatedly extracted with carbon tetrachloride and thoroughly dried, however, a very satisfactory dust was produced which remained in suspension reasonably well. Although finely ground, the particle size varied considerably of

keep the dust at a constant level; on the contrary every effort was made to make the conditions extreme by running the dust content to high levels. The animals were exposed to the various types of rock wool in turn at intervals of 3 weeks. The sole purpose of the dusting experiment was to determine whether the rock wool contained any poisonous substance that had escaped detection by other means. Since, as noted later, the only evidence obtained was of a negative character,

and because of the difficulty we experienced in raising dust from it, one may fairly assume that dustiness in the commercial utilization of rock wool is distinctly a factor of no hygienic importance. Throughout the experiment the animals remained in excellent condition. At the conclusion of the experiment their tissues were removed as in the other cases for further examination.

Discussion

Chemical investigation failed to disclose any single factor of toxic importance. From the raw materials including slag, limestone, sandstone and oil to the finished products including both un-oiled and oil-treated rock wool, nothing but spectrographic traces of any elements of toxic significance could be found and these traces obviously are fixed in the form of silicates and are therefore most inert. The small amount of oil left in the rock wool is also inert, except that it has the valuable property of increasing the filtering action of the rock wool with respect to dust. Not only is it difficult to make a cloud of dust by blowing air through the ground up commercial rock wool, but the rock wool of itself constitutes a filtering medium.

In order to test this, runs were made both with the normal rock wool and with rock wool that had been heated for a month and largely dried out.* The velocity of the air flow through the latter was three times that through the rock wool at room temperature. The heated cylinder (No. 1) was operated 18 hours before testing, the room temperature cylinder (No. 2) was operated

for 1 hour before testing. The air flow through cylinder No. 1 was 9.9 cu. ft. per minute; that through cylinder No. 2 was 3 cu. ft. per minute. The temperatures were as follows:

	°C.
Compressed air	33
Corridor air	32.5
Air leaving Cyl. 1	71
Air leaving Cyl. 2	

TABLE II

CAT NUM- BER	CAT WEIGHT	WEIGHT OF TISSUES			ANALYSES OF TISSUES			
		Lung	Liver	Kidney	Weight used	Arsenic	Mercury	Lead
Cold vapor cats								
23	2.9	23.15	100.80	42.00	Lung 22.15	Nil	Nil	Nil
					Liver 98.40	Questionable	Nil	Nil
					Kidney 41.15	Nil	Nil	Nil
11	4.9	11.75	71.40	47.65	Lung 40.60	Nil	Nil	Nil
					Liver 69.15	Questionable	Nil	Nil
					Kidney 47.00	Nil	Nil	Nil
2	2.9	31.00	91.00	27.20	Lung 20.00	Nil	Nil	Nil
					Liver 89.10	Questionable	Nil	Nil
					Kidney 26.40	Nil	Nil	Nil
6	3.1	19.40	109.35	52.30	Lung 48.40	Nil	Nil	Nil
					Liver 106.90	Questionable	Nil	Nil
					Kidney 31.25	Nil	Nil	Nil
9	2.2	16.25	73.85	20.70	Lung 15.65	Nil	Nil	Nil
					Liver 72.00	Questionable	Nil	Nil
					Kidney 19.60	Nil	Nil	Nil
15	3.0	25.15	93.05	24.40	Lung 23.20	Nil	Nil	Nil
					Liver 90.00	Questionable	Nil	Nil
					Kidney 23.00	Nil	Nil	Nil

TABLE II—*Continued*

Cat	No.	REPORT OF TISSUES			ANALYSES OF TISSUES			
		Lung	Liver	Kidney	Weight used	Arsenic	Mercury	Lead
Dusting experiment— <i>Continued</i>								
7	15	19.35	67.25	11.30	Lung	17.20	Nil	Nil
					Liver	63.75	Nil	Nil
					Kidney	13.05	Nil	Nil
3	27	20.45	83.65	27.10	Lung	19.05	Nil	Nil
					Liver	79.85	Nil	Nil
					Kidney	25.20	Nil	Nil

wool on animal life nothing whatever was evident that could be construed as injuries. The animals either retained their weight, or showed an increase in weight with the exception of those subjected to air passed through heated rock wool and, as has been noted above, the slight average decrease in weight particularly during the last 3 weeks is probably largely owing to the trying living conditions of temperature and humidity to which they were subjected. At the time they were chloroformed the animals looked healthy and macroscopic inspection of their tissues on removal showed nothing unusual.

After chloroforming the cats, their tissues were weighed at once before and after removal of pieces for microscopic examination and the remainder analyzed chemically for arsenic, mercury and lead. For this purpose the tissues were wet ashed by means of nitric acid in order to prevent any possible loss by volatilization of arsenic and mercury. The data thus secured are given in Table 2. It will be noted that the results are wholly negative. No foreign metals of chemical significance were present in the tissues. The very

slight traces of arsenic found were balanced by the trace of arsenic in the nitric acid used in ashing.

MICROSCOPIC EXAMINATION OF TISSUES

Microscopic sections were prepared from the lungs, kidneys and liver of each cat used in these experiments and from four normal control cats. In the experimental animals the tissues were all fixed in Zenker's fluid, embedded in paraffin and the sections were stained with eosin and methylene blue. Tissues from the control cats were fixed in Zenker's fluid and also in 10 per cent formaldehyde and paraffin sections were stained with eosin-methylene blue and hematoxylin and eosin.

Smaller amounts of this particulate matter were observed in the alveolar spaces and interstitial tissues. Regardless of its location, much of this foreign material was contained in phagocytic cells. Many of the small bronchial passages were filled with clumps of macrophages, each containing large numbers of varying sized particles of silicate in addition to the fine, dark pigment observed in cats from the other experimental groups.

With the exception of the presence of numerous particles of silicate, the lungs of five of the eight cats showed no abnormalities not present in the previous groups of animals. In the three remaining cats there were varying numbers of small circumscribed lesions that consisted for the most part of dense accumulations of large, oval and elongated mononuclear cells, lymphocytes, eosinophiles and scattered polymorphonuclear leucocytes. The alveolar walls at the periphery of some of these lesions were moderately thickened and infiltrated with mononuclear and polymuclear inflammatory cells. Silicate particles could be found readily in the majority of these lesions. Four of the cats showed a recent and probably terminal extravasation of blood into the bronchi and pulmonary alveoli. Excessive mucous secretion and an increased desquamation of bronchial epithelium were observed in two cats. These abnormalities, however, did not differ from those observed in the other experimental groups.

Histological examination of the kidneys and livers showed no noteworthy abnormality.

SUMMARY OF MICROSCOPICAL EXAMINATION

No pathological change that could be attributed to the experimental environment was observed in the liver and kidneys of any animal. The renal epithelium of the proximal convoluted tubules of 11 cats (five from Group I, four from Group II and two from Group III) showed slight to moderate degrees of vacuolization. Because 2 of the 4 control cats showed equal or greater amounts of vacuolization and because there were no accompanying signs of retrograde changes in the tubules and no glomerular lesions, this finding was not considered important.

Examination of the lungs from cats in Groups I and II revealed no constant changes that were considered significant. The slight to moderate excess secretion of mucus in the bronchi and bronchioles and the increased epithelial desquamation which were observed in a few cats were not accompanied by any evidence of inflammation of the bronchial walls or peribronchial tissue.

Hemorrhage into the lungs was observed in a few cats from all experimental groups and in one of the four normal control cats. The microscopic appearances clearly indicated that this extravasation of blood had occurred as a terminal event—probably during chloroformization.

Examination of the lungs of cats which had been subjected to the excessive dusting (described earlier in the paper) revealed that large amounts of silicate particles had been inhaled

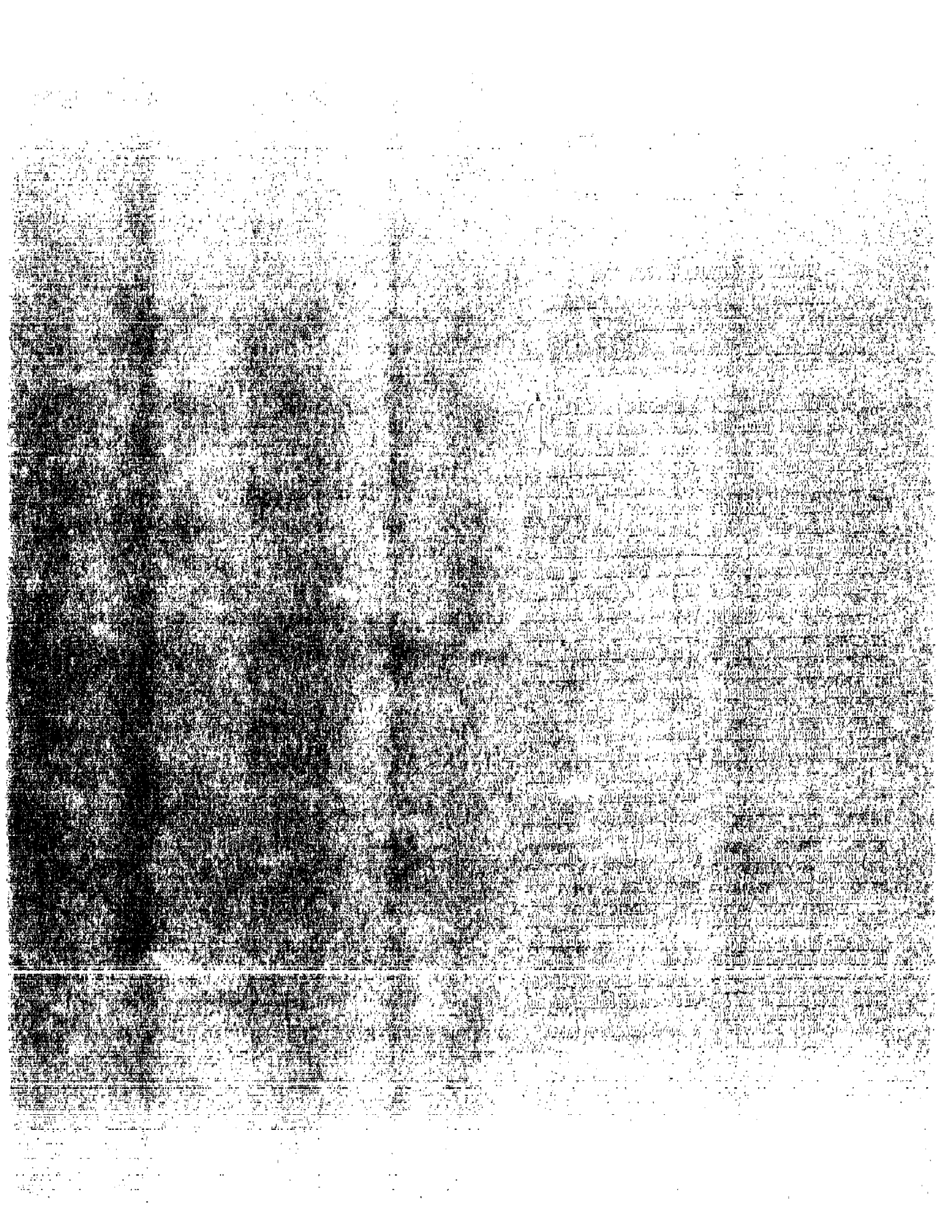
and retained in the smaller air passages and lung tissues. In 3 of the 8 cats from this group there were varying numbers of small cellular lesions in which silicate particles could be detected. Except for the presence of silicate particles, the remaining 5 cats revealed no abnormalities that were not present in animals from the other experimental groups. The absence of wide-spread inflammatory changes in the lungs of all of these animals where silicate particles were in intimate contact with the tissues suggests that no highly irritating substance was present.

SPECTROGRAPHIC EXAMINATION

In addition to the chemical analyses, an aliquot portion of the ash of each tissue was examined spectrographically by means of a Hilger quartz spectrograph with the following results:

Arsenic.—The most sensitive line for arsenic is 2319.8. It can be seen only very faintly if at all in any of the tissues. *Mercury* apparently is not present in any tissue. *Lead*. The typical lead line is extremely faint in all cases with the exception of cat kidney No. 5. In this case other lead lines can be found readily to confirm the first line. The other tissues of this cat show a very faint or no lead line. The lead in the kidney of this cat is doubtless due to some

previous exposure, otherwise some other of the 23 cats subjected to similar conditions would have exhibited similar effects. *Thallium* is absent. *Cadmium* is very faintly present in most cases. No certain indication of *Antimony* is given owing to the presence of iron and traces of silicon.



HAZARDS OF COMPETITIVE PRODUCTS

01 561 0773