

NEW YORK 12, N.Y.

STEEL CASE PRODUCTS DIVISION

January 1, 1951

(See Bulletin No. 3)

In saturators and separators in ammonia plants, many units constructed by the Koppers Company in the Pittsburgh area have been given satisfactory service for two reasons. Nearly one-half the total has exposed lead linings to greater erosion from the hard ammonium sulfate crystals. Greatly increased (48 times) gas flow through the units has brought about much more severe operating conditions. Fewer units handling 48,000,000 cu. ft. per day have used stainless steel with negligible corrosion or erosion for over four years.

Very heavy lead linings have given 17 years good service in the Pittsburgh Coke and Chemical Company saturator and separator units handling 100,000,000 cu. ft. of gas per day. Even these users of lead who should be satisfied are planning to go to stainless steel as lining material for saturators and separators in their by-product ammonium sulfate units.

Stainless steel provides a lining of greater hardness than the crystals formed, thus giving greater abrasion resistance and better service than lead linings which are in a hardness range lower than the crystals. Recent use of tar extractors have greatly increased the deterioration rate of lead linings as the layer of tar used to adhere to the lead forming a protective coating which prevented abrasion by the ammonium sulfate crystals.

Chemical Bulletin No. 3 was based on observations of the by-product coke plant of one of the larger steel companies. Since that time it has been learned that that company has instigated a broad program to determine where their own product, stainless steel, could be used in place of other materials. It may be expected that they will have full access to the stainless information obtained by the Koppers Company.

a. INTRODUCTORY HISTORY (See Chemical Bulletin No. 3, Part 1a)

This bulletin is supplementary to Bulletin No. 3 and presents newly obtained information on the same subject, i.e., the replacement of lead linings by stainless steel in saturators and separators. For years engineers have studied the problem of removing ammonia from coke oven gas and obtaining a better quality ammonia product. It was found necessary to produce a moderately dry, about 90% wet, gas with definite limits for size of ammonia crystal size grading had to be done here as in Europe. To meet these conditions the damage

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acid bath was agitated by the use of air or gas, by mechanical mixers and the use of liquor circulation. Crystal size could thus be increased by bringing about better contact and longer time of contact of the supersaturated solution with the previously formed crystals.

During the late thirties when better quality competitive synthetic sulfate was produced in southwestern Canada, Koppers developed a mixing type saturator which gives better and longer time of contact of the supersaturated solution with the previously formed crystals. The salt produced meets high grade requirements for size and reduces the differential pressure loss. The first saturator of this type to use stainless steel linings and cracker pipe was built in 1941.

In addition to ammonium sulfate from coke plants large amounts are produced from synthetic ammonia which caused a decrease in nitrogen cost from \$0.151 per pound in 1935 to \$0.068 per pound in 1940 (Chemical Engineering, December, 1946). Since then the producing capacity for synthetic nitrogen compounds has been greatly expanded in both the United States and Canada. When European production is resumed further downward trends are probable.

In this highly competitive situation the coke oven operator is faced with the necessity of reducing costs to stay in the ammonium sulfate business or to find new products such as ammonium phosphate, ammonium nitrate or ammonium carbonate through carbon dioxide. These also may be produced from synthetic ammonia. There is apparently no good answer yet and his only recourse is to study recovery methods and make every possible effort to reduce costs. By using stainless 316 to replace lead linings in saturator-separator units their maintenance costs have been reduced from the level of excessive to negligible.

b. PROCESS AND EQUIPMENT DESCRIPTION

Coke oven gas, with or without reheat, enters the outer section of the saturator as shown on the drawing, passes through slots in the lower portion of the cracker pipe and bubbles through (theoretical seal depth, 4-3/4 in.) liquor containing 4 to 6% free acid. The gas leaving the bath level carries with it some sulphuric acid spray which removes additional ammonia. The gas passes through the acid separator, whirling around the inner section through which it finally leaves. The acid particles and the few droplets of tar that may still remain in the gas are thrown outward, impinge upon the walls and drain off through the bottom to the tar skimmer tank as shown.

The cracker pipe forms a ring around the liquid level of the saturator bath and the liquor is circulated as shown, being withdrawn through the top center of the bath from the area of ammonia absorption and supersaturation fast enough to limit its concentration of ammonium sulfate above the normal saturation and thus control the rate of new crystal formation. The circulating liquor passes up through the crystals which collect on the bottom of the saturator. Contact with the crystals desupersaturates the liquor in preparation for additional absorption and supersaturation when the liquor reaches the zone of contact with the gas at the top.

Crystals are pumped from the bottom of the saturator through lead pipe to a lead lined slurry feed tank from which the liquor returns by gravity as shown. A constant level overflow in the saturator maintains the liquor level at 4-3/4 in. above the distributor slots in the cracker pipe and with this seal depth over 99.5% ammonia is removed. The overflow runs off into the tar skimmer

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The Koppers Company engineers have found it is for lining of saturators and separators without success. Brick lining was used and was found to be unsatisfactory for molten liquor tank and for separator. A 1/2-in. depth was formerly used with the old lead cracker pipes and the 1-in. depth for gas piping is obtained with the stainless cracker pipes. Reduction in acid depth (now 3-5 in.)

4. SMALLER PLANT OPERATIONS

The saturators of the Pittsburgh Coke and Chemical Company consist of three units, one of these always being kept in reserve. The saturators are of the same general design as shown in the accompanying drawing and were installed in 1925. After 17 years service they have proved for the time being recently and the 1-in. lead cracker pipes and the 1-in. lead bottom linings were in good condition and have been put back in service. The only replacement was a new 30-lb. lead lining for the vertical section. The saturator is a 10-ft. unit through which is generally handled about 1,000,000 cu. ft. of gas per day.

The General Superintendent of this plant anticipates a lowering of cost of stainless steel and plans to change to this material when the price suits him.

CONCLUSIONS

In general lead has not given satisfactory service in the saturators and separators which are the major portions of the unit for recovery of ammonia. Failure may be traced to tar removal from the coke oven gas and the greatly increased flow of gas in an effort to lower costs and meet competition from synthetic ammonia producers.

1. Lead linings (sides and bottoms) of saturators have failed from abrasion by hard ammonium sulfate crystals since the protective tar is now removed from the gas.
2. Teeth between gas exit slots in the sides of one-inch thick lead cracker pipes break off in a few months under accelerated processing now used in large plants.
3. Lead linings in acid separators have failed by erosion near the entry port where acid particles are thrown from the gas by centrifugal force.
4. Tile was tried for saturator linings and was not well liked.

The competition of synthetic ammonia has caused coke plant engineers to bend every effort to increase production of ammonium sulfate, cut the cost of production and to obtain maximum use from production equipment.

1. Stainless steel, 316, 1/4-in. thick has served since 1942 as cracker pipes with no measurable change in thickness.
2. Saturator bottom linings of 3/16-in. stainless have served since 1942 with no repairs required, though said to be wearing thin in spots.

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3. Even in small plants where operating conditions are relatively light and lead has given good service, plans are to change to stainless.

Lead will continue to be used with more or less satisfactory service in the following units. The amount of lead used in these auxiliary units is small in comparison to the heavy lead covering pipes and linings found in main separators and absorbers.

1. As plates and linings for valves and pumps for handling caustic liquor and sulfuric acids.
2. Lead linings for the slurry feed tanks suffer very little abrasion and will probably continue in use.
3. In small plant operations, where gas flow is not high, lead has given good length of service for cracker pipes and linings for absorbers and separators but will probably be replaced if the plant capacity is increased.
4. Filtrate tanks and tar skimmer tanks will probably continue to use lead linings.
5. Brick linings without lead have been used with success in molten liquor tanks and price will probably be the deciding factor.

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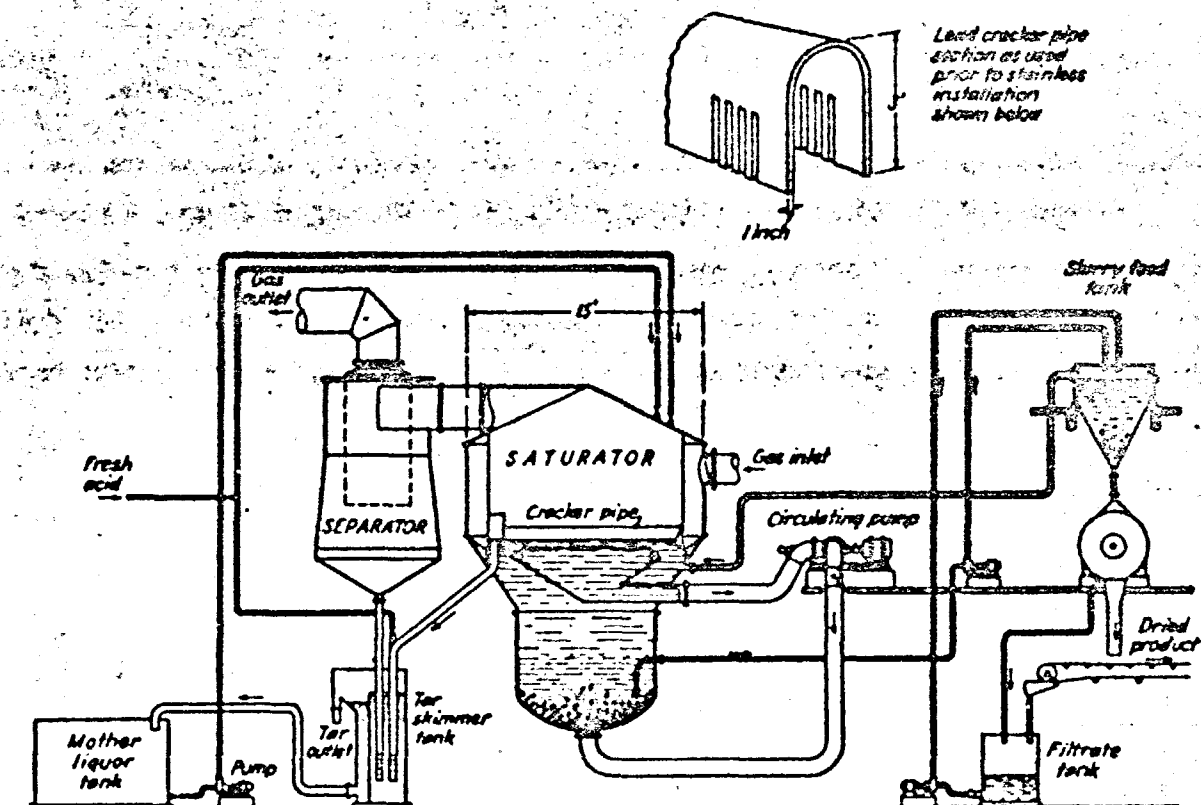
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LEAD INDUSTRIES
ASSOCIATION

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By-Product Coke Plants



Saturator and separator shown above with stainless steel linings-Type 316. For pipe, pumps and other units lead serves satisfactorily. Brick and steel linings have been used for Mother liquor tanks.