ETHYL GASOLINE CORPORATION

MANUFACTURING AND TRAFFIC DEPARTMENT

INTER-OFFICE CORRESPONDENCE

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October 4, 1938

Memorandum to:

Mr. O. B. Lewis

Last week Mr. Baldwin brought to the plant can No. H-1216 which he picked up at Ambler Brothers, Philadelphia. The contents were reputedly a sample of straight uncolored tetraethyl lead which one of their acquaintances claims he obtains for his racing cars or boats, as the case may be. Ambler Brothers did not disclose the name of the person from whom the material was obtained, nor the original source. However, from conversation with Mr. Baldwin, it is my understanding that he has an indirect inference that the material originated from the Eastman Kodak Company. This may or may not be true.

Analysis showed the material to have a specific gravity of 1.625 at $\frac{20^{\circ}}{4}$C., and its tetraethyl lead content was 97.65% by the iodine titration method. Unquestionably the material is commercial tetraethyl lead. The size of the sample was approximately 250 cc.

There is one puzzling detail with regard to this sample. Can No. H-1216 contained ETHYL Red Fluid when shipped originally to Ambler Brothers. They report that the can was washed out carefully with gasoline before transferring the special sample to it. Strange to say, the sample was rather blue in color, approximately equal to the color strength of gasoline containing 0.3 to 0.4 grams of our ETHYL Blue dye per 100 gallons. The lead analysis shows that there could have been little contamination from the can itself, and this is further substantiated by the difference in color. This would indicate that the tetraethyl was distributed already dyed.

J. H. Schaefer

BY: Cal

C:LS

Dr. Ralph Wiley - In charge of operation
Dr. Grebe 1st in charge of Physical Lab
Dr. Bousley 2nd in chg. " " "

Labratory reaction c̄ T.E.L. as catalyst
1/2 % to 1 1/2 solution of T.E.L. in Solvent
0.1 % at times
Question of fate of T.E.L. in [illegible]
Determine presence of T.E.L.

KE 0009363

[illegible]

— Operates continuously — capacity ? 700 cu ft
wooden box containing T.E.L. to be put up
at speed. use of solution containing bromine

Regulation – Wash off inside of boot each time
Leak of boot

Handling of gloves

Recommendations

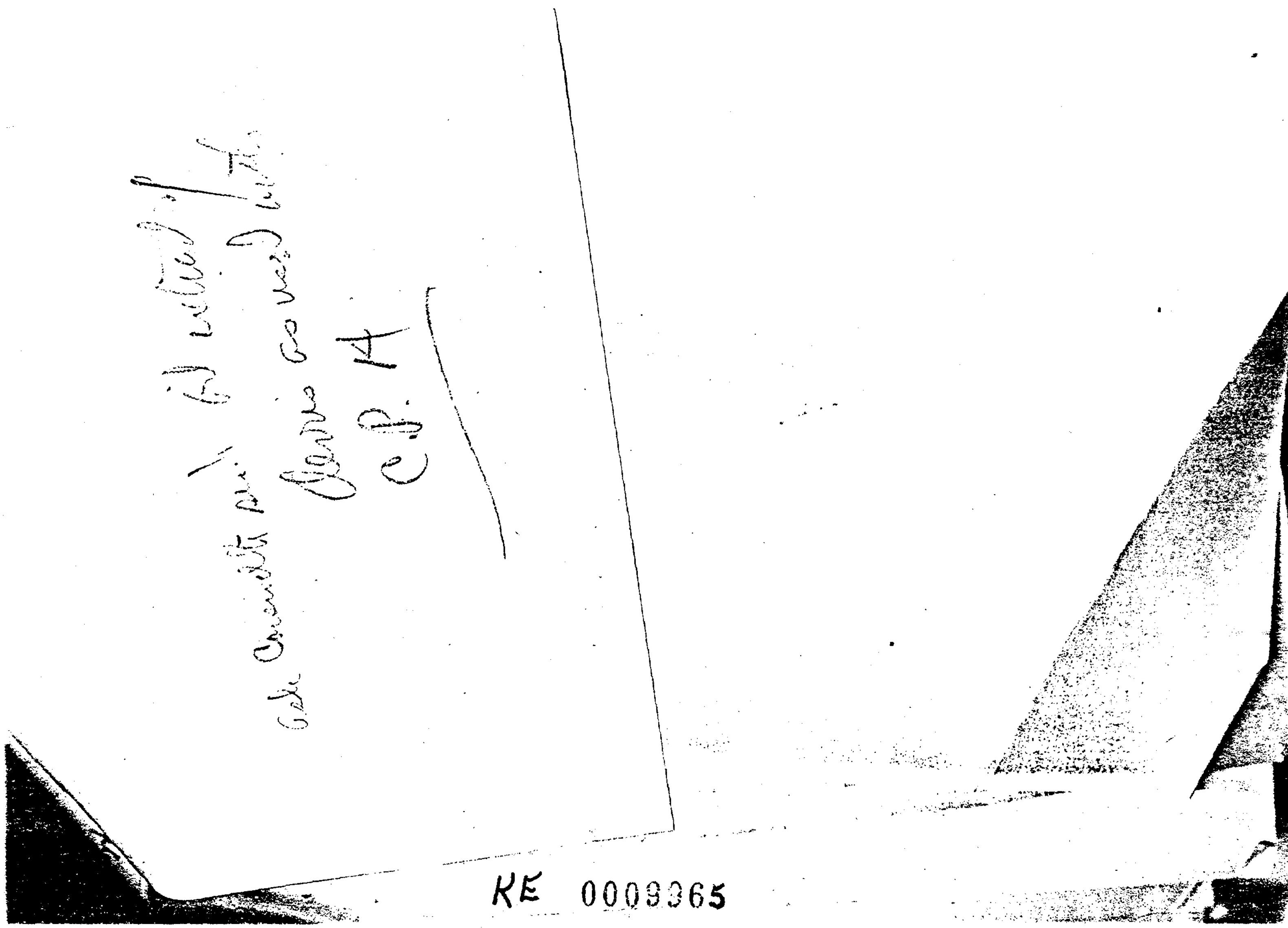
new storage space – one man responsible

Handling gloves 1 man responsible at end of each shift

Send samples containers to Dow

Car Dow Dist. for five men

See analytical results on men in alphabetical lead analyses file binder (miscellaneous)



KE 0009365

ples of material

① Samples of reaction mixture

② Samples of materials from autoclave

(a) Sample of slurry from autoclave

(b) Sample of powder from slurry

(c) Sample of water layer from slurry

(d) Acetone wash - first

(e) Acetone wash - last

(f) Sample of finished polymer

(g)

Reaction mixture will come in dry ice
work rapidly and extract to rid of
T.E. Present