[7] Wimmer, R., and Reichmann, M. E., *Fed. Proc.*, **27**, 504 (1968).
[8] Rushizky, G. W., Reeves, R. M., and Sober, H. A., *Biochemistry*, **3**, 826 (1964).
[9] Tsuchimoto, R. V., and Tener, G. M., *Biochemistry*, **2**, 807 (1963).
[10] Midgley, J. E. M., *Biochim. Biophys. Acta*, **155**, 210 (1968).
[11] Roi Bhamidary, V. L., *J. Biol. Chem.*, **243**, 350 (1968).
[12] Maxwell, I. H., Wimper, E., and Tener, G. M., *Biochemistry*, **7**, 2620 (1968).
[13] Gillam, I., Blew, D., Warrington, R. C., von Tigerstrom, M., and Tener, G. M., *Biochemistry*, **7**, 3430 (1968).
[14] Bishop, D. W. T., Pace, N. R., and Spiegelman, S., *Proc. US Nat. Acad. Sci.*, **57**, 1780 (1967).
[15] Weith, H. L., Astarmidis, G. T., and Gilham, P. T., *Science*, **160**, 1459 (1968).
[16] Shapiro, L., Franze de Fernandez, M. T., and August, J. T., *Nature*, **220**, 478 (1968).
[17] De Wachter, R., and Fiers, W. J., *Nature*, **221**, 333 (1969).
[18] Weissmann, C., Feix, G., Slor, H., and Fuller, R., *Proc. US Nat. Acad. Sci.*, **57**, 1670 (1967).
[19] De Wachter, R., Verhamme, J.-P., and Fiers, W., *FEBS Lett.*, **1**, 93 (1968).

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Chlorinated Biphenyls in Fish, Mussels and Birds from the River Rhine and the Netherlands Coastal Area

by

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Polychlorinated biphenyls are widely used in industry, and they are now making an appearance in the world's wildlife. Besides being a hazard in their own right, they interfere with the detection of organochlorine insecticides such as DDT.

At the end of 1966 a note in the *New Scientist*[1] mentioned a group of compounds which appear as unknown peaks in gas chromatograms of the tissues of fish and wildlife. In Sweden it was found that these peaks correspond to polychlorinated biphenyls (PCBs).

PCBs are used in industry as lubricants, heat transfer media and insulators. They are compatible with paints, synthetic resins, varnishes and waxes, and improve such properties of these products as their chemical resistance, water resistance, stiffness and flexibility[2,3].

In 1967, PCBs were reported present in fish and wildlife in Great Britain and the Netherlands[4–6]. In these studies the identification was primarily based on the comparison of the gas–liquid chromatography retention times in the sample extracts with those in solutions of technical PCB mixtures supplied by the industry. No information was given about the identity of the individual peaks.

We describe here the occurrence and distribution of PCBs in the River Rhine and the coastal area in the Netherlands. The identification was carried out by gas chromatography and mass spectrometry. For this purpose a CH 4 mass spectrometer (Varian Mat GmbH) was coupled to a Varian Aerograph Oven type *A 550/H*. Particular attention was paid to the design of the interface between the two instruments in order to obtain the best operating conditions[7]. The relative retention times (*r.r.t.*) were estimated with a 204-*1P* Varian Aerograph gas chromatograph with electron capture detection. 'Pyrex' glass columns (5 feet x 1/8 inch), filled with 10 per cent *DC* 200 on Gaschrom *Q* (80–100 mesh), were used in both gas chromatographs, and the columns were operated at 200° C with helium or nitrogen as carrier gas.

The tissues of the indicator organisms used throughout this study were extracted with petroleum ether in a Soxhlet extraction apparatus after drying with anhydrous sodium sulphate. The samples were cleaned up by the

Table 1. MASS NUMBERS AND NUMBERS OF CHLORINE ATOMS/MOLECULE (IN PARENTHESES) OF PEAKS WITH RELATIVE RETENTION TIMES (RELATIVE TO HEPTACHLOR = 1) LONGER THAN 0·20 AND SHORTER THAN 1·00

Sampling area	Type of sample	0·25	0·32	0·33	0·34	0·40	*r.r.t.* 0·48	0·57	0·61	0·65	0·71	0·72	0·83	0·90
1	Roach (*Leuciscus rutilus*) total body extracts 1967, 1968	HCH 292 (6)	256(3)	256 (3)		256 (3) 290 (4)	290 (4)	290 (4)	290 (4)		290 (4)		324 (5)	324 (5)
1	Gravelling (*Gobio gobio*) total body extracts 1968	+	+	+		+	+		+		+		+	+
2	Mussel (*Mytilus edulis*) 1968	+			+	+				+			+	
3	Sand eel (*Ammodytes lancealatus*) total body extracts 1968	+		+		+	+	+	+		+			+
5	Sandwich tern (*Sterna sandvicensis*) tissue and egg extracts 1965, 1966	HCH 292 (6)				256 (3)	290 (4)	290 (4)	Telodrin 406 (8)		290 (4) 376 (6)		324 (5)	
8	Elder (*Somateria mollissima*) tissue and egg extracts 1966, 1967	HCH 292 (6)				256 (3)			Telodrin 406 (8)		290 (4) 376 (6)		324 (5)	
	Phenochlor DP 6											324 (5)	324 (5)	
	Aroclor 1260											324 (5)	324 (5)	
	Clophen A 60											324 (5)	324 (5)	

With the exception of HCB, telodrin and component 376 (6) all compounds identified are confirmable in PCBs. +, At the given retention time a peak is present in the chromatogram. A 10 per cent *DC* 200 on Gaschrom *Q* (80–100 mesh) column was used. T = 205° C. Carrier gas N_2 = 50 ml/min. Molecular calculations are based on Cl^{35}. The sampling places are indicated in Fig. 2. HCH = Hexachlorocyclohexane; telodrin = 1,2,3,4,7,7-hexachloro-1,3a,4,7,7a-hexahydro-4,7-methano-indenoxirane.

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methylformamide partition[8] and column chromatography over activated fluorisil. Column chromatography was performed in two steps. First the apolar compounds including the PCBs were eluted with hexane and then a 10 per cent diethyl ether in hexane solution was used to elute dieldrin and endrin. *Only the results of the first* fraction analyses are reported here.

In Tables 1 and 2 the relative retention times (relative to dieldrin equals 1), the mass numbers and the numbers of chlorine atoms/molecule are given. Mass spectra were taken from all peaks present in the total ion current chromatogram obtained with the double ion source of the CH 4 mass spectrometer[7]. In Fig. 1*A* and 1*B* gas chromatograms are given of an eider extract and a technical PCB mixture (phenochlor DP 6). There is a strong resemblance between the electron capture chromatograms and the ion current chromatograms.

Both from the mass numbers and the numbers of chlorine atoms/molecule we conclude that most of the compounds present in the extracts can be identified as PCBs. The retention times increase in proportion to the increase in the number of chlorine atoms/molecule. A remarkable resemblance was found in the chromatograms of phenochlor DP 6, clophen A 60 and arochlor 1260. Other chlorinated compounds found are the fungicide hexachlorobenzene (HCB), *p,p'*-DDE and the insecticide teledrin. HCB occurs in all places where indicator organisms were collected, and has also been detected in seed-eating and predatory birds in the Netherlands[9]. It does not seem likely that the wide-spread distribution in the aquatic environment is caused merely by the small scale use of the compound in seed dressing practice. On the other hand, it must be emphasized that HCB is a very inert compound which shows a strong tendency to accumulation in animal tissues. Of the DDT derived type of compounds only DDE was detected, although one PCB interfered with its determination. In a former study DDT and DDD were detected in seals from the Wadden Sea[10]. Teledrin was only found in sea birds, and from other studies we can conclude that this insecticide must have originated from a pesticide producing plant near the mouth of the river[9,10]. Both in the roach and the birds a compound with $r_d = 0{\cdot}71$, mass nr 376 and 8 Cl was detected. Further studies concerning the identity of this compound are in progress. Tetrachlorobenzene and pentachlorobenzene were detected in the roach by the mass spectrometer at relative retention times shorter than 0·25.

There is a marked resemblance between the three sampling areas in the identity and pattern of the PCBs with relative retention times longer than 0·71. It seems likely therefore that most of the PCB material present in the fish and sea birds living in the Wadden Sea originated in the *River Rhine*. The coastal water chiefly flows northwards. It has been observed that dumpings of chemicals are diluted to a low degree in the littoral current.

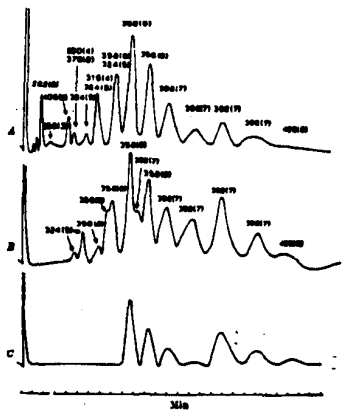

Fig. 1. Gas chromatograms of *A*, an eider extract, *B*, a technical PCB mixture (phenochlor DP 6) and *C*, a quail liver and brain extract (for all preparations, see the text).

In 1965 a clandestine dumping of copper sulphate was displaced over a range of 100 km in a northern direction within a fortnight. It was diluted by a factor of five[11].

As can be seen in the tables, the lower chlorinated PCBs occur more frequently in the tissues of the roach than in the sea birds. This may indicate that the lower chlorinated PCBs are less persistent, and this has been confirmed by an experiment with Japanese quail which were fed a diet containing 2,000 p.p.m. of phenochlor DP 6 of which the chromatogram is given in Fig. 1*B*. All birds (n = 20) died between 6 and 56 days after dosing started. A chromatogram representing the residue in the tissues (liver and brain) of these birds is given in Fig. 1*C*. It can be seen that a number of the compounds present in the original mixture were metabolized, particularly the lower chlorinated ones.

Residues in the brains and livers of the quail were measured on a semiquantitative scale by peak height comparison of peak $r_d = 1{\cdot}45$ using the phenochlor DP 6 mixture as a standard. Residue levels calculated in this way were about twenty times higher in the quail tissues than in the livers and brains of the eiders. The quail

Table 2. MASS NUMBERS AND NUMBERS OF CHLORINE ATOMS/MOLECULE (IN PARENTHESES) OF PEAKS WITH RELATIVE RETENTION TIMES (RELATIVE TO DIELDRIN = 1) OF PEAKS LONGER THAN 1·00

Sampling area	Type of sample	1·01	1·17	1·23	1·45	1·57	1·60	1·66	2·27	2·67	2·89	3·58
1	Roach (*Leuciscus rutilus*) total body extracts 1967, 1968	DDE 318 (4) + 324 (3)	356 (6)	356 (6)	356 (6)	392 (7)	356 (6) 392 (7)	392 (7)	392 (7)	392 (7)	392 (7)	426 (8)
1	Grasspilling (*Gobio gobio*) total body extracts 1968	+	+	+	+	+	+	+	+	+	+	+
2	Mussel (*Mytilus edulis*) 1968	+	+	+	+	+	+	+	+	+	+	+
3	Sand eel (*Ammodytes tobianus*larvae) 1968	+	+	+	+	+	+	+	+	+	+	+
3	Sandwich tern (*Sterna sandvicensis*) tissue and egg extracts 1965, 1966	DDE 318 (4) + 324 (5)		356 (6)	356 (6)		356 (6)	392 (7)	392 (7)	392 (7)	392 (7)	426 (8)
3	Eider (*Somateria mollissima*) tissues and egg extracts 1966, 1967	DDE 318 (4) + 324 (5)		356 (6)	356 (6)		356 (6)	392 (7)	392 (7)	392 (7)	392 (7)	426 (8)
	Phenochlor DP 6	356 (6)	356 (6)	356 (6)	356 (6)	392 (7)	356 (6)	392 (7)	392 (7)	392 (7)	392 (7)	426 (8)
	Aroclor 1260	356 (6)	356 (6)	356 (6)	356 (6)	392 (7)	356 (6)	392 (7)	392 (7)	392 (7)	392 (7)	426 (8)
	Clophen A 60	356 (6)	356 (6)	356 (6)	356 (6)	392 (7)	356 (6)	392 (7)	392 (7)	392 (7)	392 (7)	426 (8)

With the exception of DDE all compounds identified are conformable to PCBs. + : At the given retention time a peak is present in the chromatogram.

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Fig. 2. Location of sampling areas: River Rhine (1), coastal area along the west coast (2) and the Wadden Sea (3).

developed hydropericards with a dose of 2,000 p.p.m. Flick et al.[12] also discovered hydropericards in White Leghorn cockerels dosed with 400 p.p.m. in their diet. No hydropericards were observed in birds which died in the Wadden Sea area. Details of this study will be published separately.

The peculiar physical properties and chemical stability of PCBs make it likely that the application will increase and that their release into the environment will continue. Although fatal concentrations are probably not developed yet along the Dutch coast, we suggest that monitoring of the environment should be carried out regularly in order to keep an eye on the further trends in the level of contamination. Apart from being a potential hazard in the natural environment, these PCBs also represent an important source of interference in the detection of insecticides such as DDE, *o,p'*-DDT, *p,p'*-DDT and DDD.

We thank Mr J. H. Pennings, Miss J. Boeyink and Mr J. M. Eyndhoven for their technical assistance. Autopsy of the quail was carried out by Mr J. G. Vos. The technical PCB mixtures were supplied by Prodelec in France (phenochlor), Monsanto in the US (arochlor) and Bayer in Germany (clophen). Fish from the Rhine were supplied by the Department of Sportfishing and Professional Inland Fishing in Utrecht.

Received December 24, 1968.

1 *New Sci.*, 52, 612 (1966).
2 *Monsanto Company Tech. Bull. No. PL-306*, 50 (1960).
3 *Encyclopedia of Chemical Technology* (edit. by Eller Kirk, R., and Othern, D. F.), second rev. ed., 5, 289 (1964).
4 Holmes, D. C., Simmons, J. H., and Tatton, J. O'G., *Nature*, 216, 227 (1967).
5 Holden, A. V., and Marsden, K., *Nature*, 216, 1274 (1967).
6 Koeman, J. H., Oskamp, A. A. G., Veen, J., Brouwer, E., Roth, J., Zwart, P., v.d. Broek, E., and van Genderen, H., *Meded. Rijksfaculteit Landbouwwetenschappen Gent*, 32, 841 (1967).
7 ten Noever de Brauw, M. C., and Brunnée, C., *Z. Anal. Chem.*, 229, 321 (1967).
8 de Faubert Maunder, M. J., Egan, H., Godly, E. W., Hammond, E. W., Roburn, J., and Thomson, J., *Analyst*, 89, 168 (1964).
9 Vos, J. G., Breeman, Hetty A., and Benschop, H., *Meded. Rijksfaculteit Landbouwwetenschappen Gent*, 33 (in the press).
10 Koeman, J. H., and van Genderen, H., *J. Appl. Ecol.*, 3 (Suppl.), 99 (1966).
11 Koeman, J. H., Veen, J., Brouwer, E., Huisman, de Brouwer, L., and Koolen, J. L., *Helgol. Wissensch. Meeresuntersch.*, 17, 375 (1968).
12 Flick, D. F., O'Dell, R. G., and Childs, V. A., *Poultry Science*, 44, 1480 (1965).

Spontaneous Mutation

The following two articles are concerned with the genetic control of mutation rate. The first describes the anti-mutagenic effect of some DNA polymerase mutations on phage T4 mutation rates, and discusses the implications they hold for the mechanism of action of DNA polymerase. The second compares spontaneous mutation rates in different species, and considers the biological factors which may determine these rates.

Genetic Control of Mutation Rates in Bacteriophage T4

HERITABLE increases in the rates of spontaneous mutation are usually attributed to "mutator genes", the mechanisms of action of which remain obscure. Mutations in the DNA polymerase gene in bacteriophage T4 have, however, been observed to exhibit powerful mutator effects[1-3], suggesting that spontaneous mutation rates depend on the properties of DNA polymerases, and also that these polymerases play an important part in base selection during DNA synthesis.

We have therefore embarked on a detailed examination of the effects of DNA polymerase mutations on T4 mutation rates, to throw light on the genetic control of mutation rates and possibly on the mechanisms of spontaneous mutation and of information transfer in DNA synthesis as well. This article summarizes studies on temperature sensitive (*ts*) mutants of gene 43 (the DNA polymerase gene) obtained from the Caltech collection as a gift of Professor R. S. Edgar. A very large fraction of these mutants turns out to exhibit mutator activity. A further surprising result has been the discovery of several *anti-mutator* alleles of gene 43 which sharply reduce mutation rates at many sites in the T4 genome.

The *ts* mutants have been localized within gene 43 by complementation and recombination. A preliminary map has been prepared from two-factor crosses[2], but its accuracy is limited by the very strong marker effects frequently observed with alleles of gene 43 (ref. 3 and unpublished results of J. W. D. and E. F. A.). The forty-one mutants occupy approximately twenty-one sites, and most of them show little or no dominance, as is to be expected for a catalytic function. Our tests for intracistronic (interallelic) complementation have been uniformly negative or of doubtful significance. These tests

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