

I. Human Ventilation Requirements

Ventilation of residential, commercial, and office spaces is now understood to be necessary to control body odors, tobacco smoke, cooking odors, and other odorous impurities, rather than to maintain the required amount of oxygen or remove the carbon dioxide produced by respiration. This is true because standard construction of buildings for human occupancy cannot prevent the infiltration and exfiltration of substantial amounts of air, even when all window, door, roof, or mechanical-ventilation openings are closed. Published data on the amounts of outdoor air normally available by natural ventilation or infiltration indicate that suffocation by deficiency of oxygen or excess of carbon dioxide as a result of human respiration, is virtually impossible in buildings constructed above ground.

In view of the fact that industrial hygienists are called upon to justify their position to those who are not aware of the background of research and experience upon which present ventilation standards are based, a brief review of the demands of respiration on atmospheric air is in order at this point.

A. AIR COMPOSITION

The approximate composition of air under three conditions is given in Table 1. Simple calculations will demonstrate that human beings do not require more than a few hundred cubic feet of air per hour to satisfy their demand for oxygen.

TABLE 1
Composition of Air
(Volume per cent)

Component	Outdoor air, dry	Indoor air, 70° F. (RH 50)	Expired air, 97° F. (RH 100)
Inert gases (nitrogen, argon, etc.)	79.00	78.00	77.00
Oxygen	20.97	20.69	19.10
Water vapor	0.00	1.25	3.90
Carbon dioxide	0.03	0.06	1.90
Total	100.00	100.00	100.00

and to dilute to a harmless concentration the carbon dioxide produced. A man, even at heavy work, breathes only about 40 liters (1.4 cu. ft.) of air per minute, consumes about two liters (0.07 cu. ft.) of oxygen, and produces about 0.17 liter (0.06 cu. ft.) of carbon dioxide.

¹ *Heating, Ventilating, Air Conditioning Guide, 1967*. American Society of Heating, Ventilating, and Air-Conditioning Engineers, New York, 1967, ch. 11.

Mackey² offered an interesting hypothesis on the physical and chemical changes occurring in indoor air as a result of human occupancy. His analysis was somewhat as follows (modified after communication with Mackey):

An adult at rest uses in one minute about 20 cubic inches of oxygen and produces about 20 cubic inches of carbon dioxide. At 70° F., he loses in one hour about 300 Btu of sensible heat and about 0.1 pound of water vapor. Assuming for simplicity that these rates remain constant (which they would not do in fact), if such an adult were confined in an airtight, insulated space of 1000 cubic feet, initially at 70° F. (ignoring humidity), the dry-bulb temperature would increase to 100° F. in less than two hours, while about 75 hours would be required to reduce the oxygen to 16 per cent and increase the carbon dioxide to 5 per cent. In this extreme case, the physical change is seen to be much more dangerous than the chemical.

CARBON DIOXIDE AS AN INDEX OF VENTILATION REQUIREMENTS

Measurement of carbon dioxide concentrations in spaces occupied by human beings had been standard procedure among ventilating engineers for many years until Yaglou, Riley, and Coggins³ reported in 1936 their results from a series of experiments on ventilation requirements. They concluded that the carbon dioxide concentration "in the air of occupied rooms proved to be an unreliable index of ventilation, from the standpoint of both outdoor air supply and odor intensity," and suggested that "a great deal of unjustified effort would be saved by discontinuing the usual measurements of CO₂ in ordinary ventilation work, except perhaps in instances in which the air flow is well under 10 c.f.m. per person."

Many research workers and writers during the present century have recognized the existence of 0.10 to 0.50 per cent CO₂ in spaces occupied by human beings is not of itself dangerous. Some still express the belief, however, that such an increase in carbon dioxide, from a normal outdoor level of 0.03 to 0.04 per cent, makes the presence of unhealthful constituents as a result of *vitiation of air* by human respiration. This question has not been settled to the satisfaction of the layman. The current use of odors as indexes of ventilation requirements is still regarded by some as neglect of a fundamental hazard to life and health. Nevertheless, public health workers and ventilation engineers are now generally concerned with oxygen and carbon dioxide concentrations in air at normal barometric pressures are not of concern to health in the absence of more substantial disturbances of air composition, as might be caused by industrial processes that consume oxygen or generate carbon dioxide.

The extended use of odor-controlling processes for air conditioning will bring about further revisions in the standards of quantities of outdoor air per capita required for health or comfort.⁴ In some situations it may be possible to approve a mechanical system of 100 per cent recirculation of indoor air, on the strength

² O. Mackey, *Air Conditioning Principles*. International Textbook, Scranton, 1941.

³ C. Yaglou, E. C. Riley, and D. I. Coggins, *Trans. ASHVE*, 42, 133 (1936).

⁴ G. P. McCord and W. N. Witheridge, *Odors: Physiology and Control*. McGraw-Hill, New York, 1949.