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WILMINGTON, DELAWARE 19898

ENGINEERING DEPARTMENT

July 1, 1975

N. D. KRIVANEK
CENTRAL RESEARCH
AND DEVELOPMENT
HASKELL LABORATORY

BLOOD Pb RESPONSE TO AIR Pb

As you requested, I have analyzed data in the literature on the relationship between air Pb and blood Pb. I have concluded that:

- (i) The Azar et al. data indicate that the slope of the log blood Pb - log air Pb relationship is small. For example, a 100% increase in air Pb results in a 11.2% increase in blood Pb.
- (ii) The data by Azar et al. show that blood Pb will increase approximately 1 $\mu\text{gm}/100\text{ ml}$ when air Pb increases 1 $\mu\text{gm}/\text{m}^3$. It should be noted that the blood Pb - air Pb response curve is a log-log relationship and is not linear.
- (iii) Points (i) and (ii) are based on the data reported by Azar et al. It is shown that other data in the literature are consistent with those reported by Azar et al.

A discussion of the analysis is attached. Please contact me if you have any questions.

ENGINEERING SERVICE DIVISION

R. D. Snee
Applied Statistics

RDS:bvt

Atch

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ANALYSIS ON THE RELATIONSHIP
BETWEEN AIR Pb AND BLOOD Pb

1. Introduction

The following analysis of the relationship between air Pb and blood Pb concentrations is based on data reported in the papers listed in the References at the end of this report.

2. The Slope of the Blood Pb - Air Pb Curve is Small

The Azar(1) paper develops a log-log relation between air Pb and Blood Pb of the form

$$\log \text{ blood Pb} = 1.2257 + 0.153 \log \text{ air Pb}$$

$$\text{or blood Pb} = 16.82 (\text{air Pb})^{0.153} \quad (1)$$

Assuming the log-log model is correct, the percent increase in blood Pb which results when air Pb is increased from x_1 to x_2 is given by

$$\text{Blood Pb Increase (\%)} = 100((x_2/x_1)^{0.153} - 1) \quad (2)$$

This relationship shows the small dependence of blood Pb on air Pb. For example, a 100% increase in air Pb (ie the air Pb doubles) results in only an 11.2% increase in blood Pb.

$$\text{Blood Pb Increase (\%)} = 100((2/1)^{0.153} - 1) = 11.2\%$$

Table 1 summarizes increase in blood Pb as a function of air Pb. From equation (2) we see that the % increase in blood Pb is dependent on the slope of the log-log equation (ie, $b = 0.153$). Results using the slope of the Goldsmith and Hexter line (3) (ie, $B = 0.243$) are also included in Table 1 for comparative purposes. It should be noted that the Goldsmith-Hexter slope lies outside the 95% confidence limits of the Azar et al. slope (0.153 ± 0.079). Even with the larger Goldsmith and Hexter slope the % increase in blood Pb is considerably smaller than the corresponding increase in air Pb.

3. Change in Blood Pb ($\mu\text{gm}/100 \text{ ml}$) Resulting from a $1 \mu\text{gm}/\text{m}^3$ Increase in air Pb is approximately $1 \mu\text{gm}/100 \text{ ml}$

The log-log model reported by Azar et al (reference 1, equation 1) can be used to compute the expected change in blood Pb resulting from a $1 \mu\text{gm}/\text{m}^3$ increase in air Pb. The change in blood Pb is given by

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$$A \text{ Blood Pb } (\mu\text{gm}/100 \text{ ml}) = \text{Current Blood Pb } \left(\frac{\text{current air} + 1}{\text{current air Pb}} \right)^{b-1}$$

where b is the slope of the log-log equation. The blood Pb - air Pb response is shown in Table 2 for both the Azar et al slope (b=0.153) and the Goldsmith and Hexter slope (b=0.243). An examination of Table 2 indicates that, in general, a 1 $\mu\text{gm}/\text{m}^3$ increase in air Pb results in a blood Pb increase of 1 $\mu\text{gm}/100 \text{ ml}$.

4. The Azar et al Blood Pb - Air Pb Relationship Is the Best Available Model and Is Consistent with Results from Other Studies

The preceeding information is based on the Azar et al blood Pb - air Pb model. This model was developed from the most comprehensive study conducted to date. The air Pb exposure for 150 subjects was determined 24 hours/day for periods of 2-4 weeks. Blood samples were obtained twice/week from each subject.

The slope of the log-log relationship reported by Azar et al, Goldsmith and Hexter, and calculated from the Albany Study are

<u>Study</u>	<u>Slope (b)</u>	<u>95% Conf. Limits</u>
Azar, et al	0.153	$\pm .079$
Goldsmith and Hexter	0.243	
Albany Study	0.243	

Given two blood Pb values and the corresponding air Pb values, the slope of the log-log model is

$$\text{Slope (b)} = \frac{\log (\text{Blood Pb}_2 / \text{Blood Pb}_1)}{\log (\text{Air Pb}_2 / \text{Air Pb}_1)}$$

where $\text{air Pb}_1 < \text{air Pb}_2$. For example, the slope developed from the Albany Study is

$$\text{Slope} = \frac{\log (35/26)}{\log (10.9/3.2)} = 0.243$$

Log Blood Pb - log air Pb slopes developed from data reported in references 2, 4, 5, and 6 are summarized in Table 3 and shown graphically in Figure 1. The results in Table 3 were developed using assumed ambient air Pb concentrations of 1.2 $\mu\text{gm}/\text{m}^3$ for Kehoe's subjects and 2.0 $\mu\text{gm}/\text{m}^3$ for the Williams, King and Walford and Cole to Lynam data sets. With the exception of the latter two studies the estimated slopes are in general agreement with the Azar et al slope.

Table 1

Percent Increase in Blood Pb as a Function
of Percent Increase in Air Pb

Air Pb Ratio (x_2/x_1)	Air Pb % Increase	% Increase in Blood Pb	
		Slope = 0.153*	Slope = 0.243**
1.10	10	1.5	2.3
1.25	25	3.5	5.6
1.50	50	6.4	10.4
2	100	11.2	18.3
3	200	18.3	30.6
4	300	23.6	40.1
5	400	27.9	47.9
10	900	42.2	75.0

* Azar et al (1)

** Goldsmith and Hexter (3)

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Table 2

Blood Pb Response ($\Delta\mu\text{gm}/100\text{ ml}$) to a
 $1\ \mu\text{gm}/\text{m}^3$ Increase in Air Pb

Data Source	Current Level		$\Delta\text{Blood Pb } (\mu\text{gm}/100\text{ ml})$ $1\ \mu\text{gm}/\text{m}^3\text{ Air Pb Increase}$	
	Air Pb	Blood Pb	Slope=0.153*	Slope=0.243**
Starke, FL*	0.64	16.4	2.5	4.2
Barksdale, WI*	1.01	13.8	1.5	2.5
Philadelphia Cab*	2.62	22.4	1.1	1.8
Los Angeles Office Wkrs.*	3.06	19.9	0.9	1.4
Los Angeles Cab*	6.10	24.6	0.6	0.9
Albany Study	3.2	26	1.1	1.8
Albany Study	10.9	35	0.5	0.8

* Azar et al (1)

** Goldsmith and Hexter (3)

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Table 3

Log Blood Pb - Log Air Pb Slope Estimated From Data in the Literature

Data	$\mu\text{gm}/\text{m}^3$	Exposure Time(HR/WK)	Pb-in-Air		Blood Pb	
			Ambient**	Combined	Unexposed	Exposed
Kehoe (1960)						
Subject FC	75	37.5	1.2	17.67	25	42
						.193
Kehoe (1966) Average						
of Subjects JS&LD	150	10.5	1.2	10.5	23.4	27.2
		21	1.2	19.8	23.4	38.7
		31.5	1.2	29.1	23.4	40.3
		42	1.2	38.4	23.4	44.1
						.193
Kehoe (1966) Average						
of Subjects NK&SS	10	10.5	1.2	1.75	19.5	20.3
		21	1.2	2.30	19.5	21.0
		31.5	1.2	2.85	19.5	21.5
		42	1.2	3.40	19.5	22.4
		52.5	1.2	3.95	19.5	24.7
		63	1.2	4.50	19.5	27.0
		73.5	1.2	5.05	19.5	26.2
						.107
						.114
						.113
						.133
						.198
						.246
						.206
<u>Williams, King and Walford</u>						
Machine Pastors	218	40	2.0	53.4	28+	74
Hand Pastors	150	40	2.0	37.2	28+	63
Forming Dept.	134	40	2.0	33.4	28+	63
						.296
						.277
						.288
<u>Cole and Lynam</u>						
Convicts	10.9	161	2.0	10.5	22	35
						.280

*Slope - $\log (\text{Blood Pb Ratio}) / \log (\text{air Pb Ratio})$

**Estimated ambient exposure

+ Average of Plastic Depts A & B

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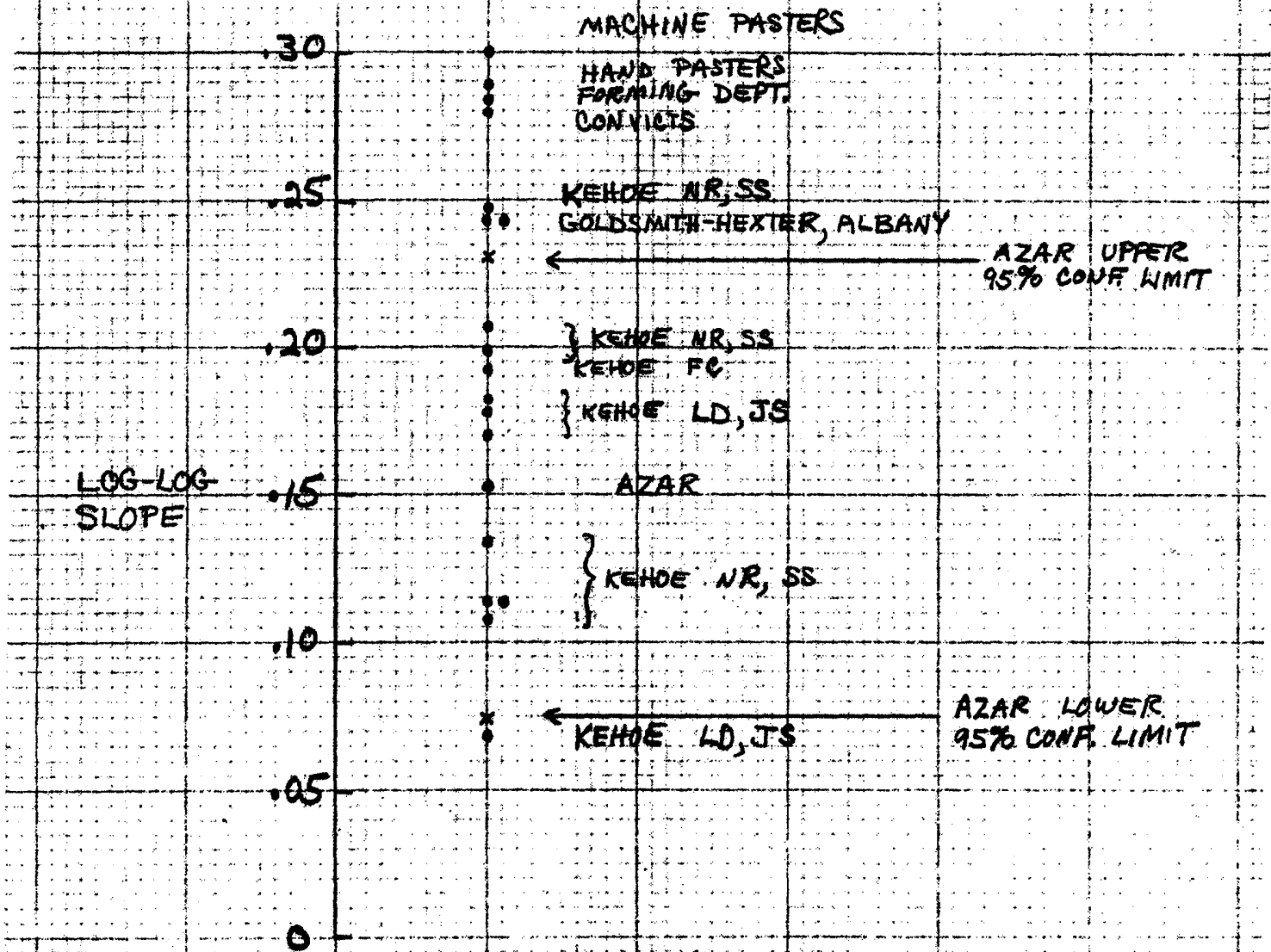
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6. Williams, N. K., King, E., and Walford, J. (1969). An Investigation of Lead Absorption in an Electric Accumulator Factory with the Use of Personal Samplers. Brit J Ind Med 26, 202-216.

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FIGURE 1



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<u>Data Set</u>	<u>X₀</u> <u>Air Pb</u>	<u>y₀</u> <u>Blood Pb</u>	<u>Δ Blood Pb.</u> <u>Δ 1 μg/m³ AIR Pb</u>	
			<u>AZAR</u>	<u>G4H</u>
Starks, Fla.	.64	16.4	2.5	4.2
Berkedale, Wisc.	1.01	13.8	1.5	2.5
Phila. Cal.	2.62	22.4	1.1	1.8
LA. Office	3.06	19.9	0.9	1.4
LA. Cal.	6.10	24.6	0.6	0.9
Albany	3.2	26	1.1	1.8
.Albany	10.9	35	0.5	0.8

Model $y = ax^b$ line thru x_0, y_0 with

slope b is $\log y = \log y_0 + b(\log x - \log x_0)$

$$\text{or } y = y_0 \left(\frac{x}{x_0} \right)^b$$

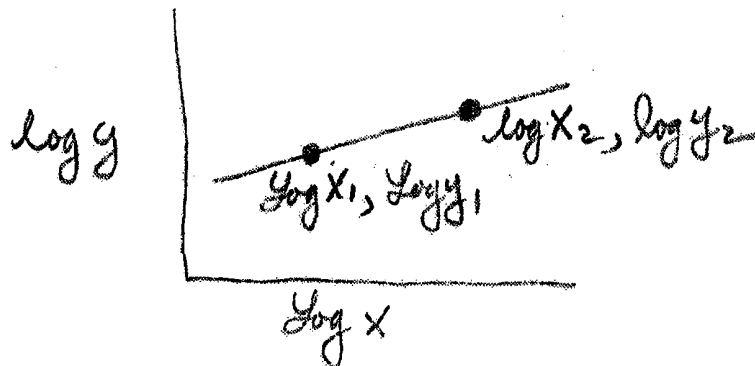
$$\frac{\Delta y}{\Delta x=1} = y_0 \left(\frac{x_0+1}{x_0} \right)^b - y_0 \left(\frac{x_0}{x_0} \right)^b$$

$$= y_0 \left[\left(\frac{x_0+1}{x_0} \right)^b - 1 \right]$$

Poor, Ince, Hahabi Relationship

$$\log \text{Blood Pb} = 1.2257 + .153 \log \text{Air Pb}$$

If $y = ax^b$ then % Increase = $1 - \left(\frac{x_2}{x_1}\right)^b$



$$\frac{y_2}{y_1} = \left(\frac{x_2}{x_1}\right)^b$$

<u>X Ratio ($\frac{x_2}{x_1}$)</u>	<u>DUPONT</u> <u>% Increase</u>	<u>b = .243</u> <u>% Goldsmith-Walker</u>
1	0	0
2 = DOUBLE	11.2	18.3
3 = TRIPLE	18.3	30.6
4	23.6	40.1
5	27.9	47.9
10	42.2	75.0

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$$\underline{\log \text{Blood Pb} = a + b \log \text{air Pb}}$$

<u>Study</u>	<u>b</u>
Azhar et al	$.153 \pm .079$ $\begin{cases} .232 \\ .074 \end{cases}$
Goldsmith & Hexter	.243
Albany Study	.243
$\rightarrow b = \frac{\log (35/26)}{\log (10.9/3.2)} = .243$	

KEHOE FC

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$$\text{Blood Pb} = A (\text{air Pb})^B$$

$$\log \text{blood Pb} = 1.2257 + .153 \log \text{air Pb}$$

<u>Air Pb</u>	<u>Blood Pb</u>	<u>$\frac{\Delta \text{Blood Pb}}{\Delta \text{Air Pb}} = 1 \mu\text{gm}/\text{m}^3$</u>
1	16.8	1.9
5	21.5	0.6
10	23.9	0.4

$$\% \text{ Increase} = \left[\left(\frac{X_0 + 1}{X_0} \right)^b - 1 \right] * 100$$

<u>X₀</u>	<u>$\frac{b}{.153}$</u>	<u>$\frac{b}{.243}$</u>
1	11.2	18.3
5	2.8	4.5
10	1.5	2.3