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PORTAL HYPERTENSION AND BLEEDING ESOPHAGEAL VARICES*

Their Occurrence in the Absence of Both Intrahepatic and Extrahepatic Obstruction of the Portal Vein

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NEW HAVEN, CONNECTICUT

THERE are a few well documented cases of bleeding from ruptured esophageal varices in the absence of associated portal hypertension.¹⁻³ However, most patients with life-threatening hemorrhage from varices have portal hypertension, proved or assumed, as an underlying cause. Following the example of Whipple,* most authors emphasize the role of venous obstruction in the production of portal hypertension and its consequences, classifying their cases into "intrahepatic" and "extrahepatic" groups depending on the presumed site of obstruction. Although such a classification simplifies the clinical evaluation and management of variceal bleeding, it ignores other important, nonobstructive etiologic factors. Since the pressure within the portal system depends on the volume of blood flow as well as on the resistance to blood flow, it is reasonable to suppose that certain cases of otherwise unexplained portal hypertension may result from abnormalities of flow through the splenoportal axis.

Within the past eight years we have encountered at the Grace-New Haven Community Hospital 5 patients with bleeding esophageal varices and portal hypertension in the absence of demonstrable intrahepatic or extrahepatic venous obstruction. One of these did not have a satisfactory splenoportogram, but the other 4 underwent complete clinical, radiologic and surgical investigation and constitute the basis for this report. A total of 32 patients had portalsystemic-shunt surgery for bleeding varices and portal hypertension in this hospital during the period 1950-1958. These cases are summarized according to etiology in Table 1. Many other patients with gastrointestinal hemorrhage and portal hypertension, usually caused by hepatic cirrhosis, were admitted to the hospital in this period, but were considered un-

suitable for operation. For this reason, the true frequency of nonobstructive portal hypertension with bleeding varices cannot be estimated from our data.

CASE REPORTS

CASE 1. S.S., a 23-year-old single man, was admitted to the Grace-New Haven Community Hospital on January 3, 1955, because of massive hematemesis of 2 hours' duration.

Aside from congenital mental retardation, the patient was well and active until June, 1950, when at the age of 18 years he experienced a sudden massive hematemesis followed by the passage of tarry stools. In another hospital physical examination was within normal limits except for moderate splenomegaly. Laboratory examinations at that time revealed a hemoglobin of 4 gm. per 100 ml., a white-cell count ranging between 3000 and 4900, a slight reduction in platelets and normal results on liver-function tests. A complete gastrointestinal x-ray series was considered to be within normal limits. The patient was treated with transfusions and discharged without a definite diagnosis. Two months later, in August, 1950, he had a similar episode of hematemesis and melena, physical examination and laboratory evaluation being basically as before and treatment being supportive. He returned to his normal activities feeling well, without further gastrointestinal bleeding, until August, 1954, when he again experienced massive hematemesis and melena. On hospitalization at this time splenomegaly was noted once more, and a gastrointestinal x-ray series was again interpreted as normal. He was seen in consultation at this hospital in November, 1954, when physical examination revealed moderate splenomegaly and a gastrointestinal x-ray series demonstrated large varices extending throughout the lower ¾ of the esophagus and upper stomach. Liver-function tests were negative except for an elevation of the indirect-reacting serum bilirubin and a bromsulphalein retention of 9.4 per cent at 45 minutes. After a symptom-free interval of 2 months the patient suddenly experienced painless vomiting of about 2 liters of blood and was immediately admitted to this hospital for care and study.

His weight had been steady, and his diet had been adequate. There was no history of alcoholic intake or exposure to drugs or toxins. He had experienced transient jaundice immediately after a transfusion reaction during a previous admission to another hospital. There was no history of abdominal trauma, ascites or peripheral edema.

The family history was unremarkable except that 3 cousins were reported to have sickle-cell anemia.

Physical examination showed a pale, anxious, well nourished man, who was vomiting small amounts of blood. Significant physical findings included faint scleral icterus, slight cardiomegaly, with a Grade 2 apical systolic murmur, a normal-sized liver by percussion, the lower edge being palpable at the right costal margin, and a firm, nontender, rounded spleen palpable 6 fingerbreadths below the left

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costal margin. No collateral veins were demonstrable over the abdomen, and there was no edema, clubbing, or spider angiomata.

The pulse was 140, and the respirations 20. The blood pressure was 110/70 in the recumbent position.

Initial laboratory studies revealed a hematocrit of 35 per cent, a white-cell count of 12,000, with a normal differential, and adequate platelets on smear. Urinalysis and a serologic test for syphilis were negative. The stools were guaiac positive and tarry in appearance. Liver-function tests at this

TABLE 1. *Etiologic Factors in 32 Patients Undergoing Surgery for Bleeding Esophageal Varices and Portal Hypertension at the Grace-New Haven Community Hospital, 1950-1958.*

FACTOR	No. OF CASES
Intrahepatic cause	24 (75.0%)
Laënnec's cirrhosis	18
Postnecrotic cirrhosis	2
Biliary cirrhosis	2
Hemochromatosis	1
Atypical cirrhosis	1
Extrahepatic portal obstruction	3 (9.4%)
"Nonobstructive" portal hypertension	5 (15.6%)

time were within normal limits except for a total serum bilirubin of 10.08 mg. per 100 ml. (after many transfusions), with a 1-minute, direct-reacting fraction of 0.59 mg. per 100 ml. (Table 2).

The patient continued to vomit blood and to pass tarry stools, receiving 4500 ml. of whole blood, with maintenance of a normal blood pressure. A tamponade of the esophagus with the Sengstaken-Blakemore tube failed to control the bleeding satisfactorily, so that on January 4, ligation of the esophageal varices was carried out, 3 columns of large gas-

operation. Microscopical examination of the tissue demonstrated preservation of general lobular architecture (Fig. 2). The parenchyma was intact throughout except for a few small periportal linear zones in which there was loss of parenchymal cells, sinusoidal congestion, reticulum collapse and small collections of mononuclear leukocytes and pigment-laden macrophages (Fig. 3). There was no increase in connective tissue, no distortion of the vascular pattern and no evidence of disease of the biliary tree. The removed spleen was firm and greatly enlarged, weighing 1400 gm. On microscopical examination the splenic capsule and trabeculae were of normal thickness and cellularity. There was evidence of generalized, but minimal, passive congestion of the sinusoids, with prominence of the sinusoidal endothelium and moderate thickening of the fibrillar reticulum. Scattered, small, perivascular hemorrhages with early organization were noted around some of the thin-walled vessels involved. No excess of iron-containing pigment was demonstrated, and there was no evidence of arteriovenous aneurysms in any portion of the specimen.

The patient recovered rapidly after operation, returning home to his normal activities. He experienced no further gastrointestinal bleeding, although serial esophagrams continued to demonstrate small residual varices. In August he was hospitalized briefly for weakness and malaise. Gastrointestinal x-ray series now demonstrated a medium-sized hiatus hernia and persistent, small esophageal varices. At this time the hemoglobin was 5.2 gm. per 100 ml., and the stools showed a +++ guaiac test for blood. Symptoms abated after transfusion of 3 units of whole blood. Since that time the patient has felt generally well, has been active at home, but has been admitted to the hospital on several occasions with evidence of moderate recurrent gastrointestinal bleeding. Esophagrams have continued to demonstrate varices involving the lower 1/3 of the esophagus and an esophageal hiatus hernia. Treatment has included whole-blood transfusions as necessary, and transesophagosopic injections of the varices with sclerosing solutions.

A transthoracic needle biopsy of the liver was performed on January 31, 1957. As before, the general hepatic architecture was intact. Scattered throughout the specimen, usually confined to the mid-zone portions of individual lobules, were

TABLE 2. *Admission Liver-Function Tests in Cases 1 to 4.*

CASE No.	SERUM BILIRUBIN		BROM- SULFA- LIN RE- TENTION AT 45 MIN.	CEPHALIN FLOCCULATION		THYMOL TURBIDITY units	SERUM ALKALINE PHOSPHA- TASE	SERUM PROTEIN		TEST FOR BILE IN URINE	URINARY UROBI- LINOGEN	PRO- THROM- BIN	
	1-MIN.	TOTAL		in 24 hr.	in 48 hr.			gm./100 ml.	gm./100 ml.				
	mg./100 ml.	mg./100 ml.	%				S.J.R. units	gm./100 ml.	gm./100 ml.		Ehrlich units	%	
1	0.59	10.08	9.4	0	0	1.1	4.5	7.05	3.68	3.37	0	0.45	—*
2	0.08	0.74	3.8	0	+	2.3	6.0	5.47	2.75	2.72	0	—*	100
3	0.13	0.72	20.3	—*	—*	3.6	4.4	6.07	3.02	3.05	0	0.18	90
4	0.09	0.44	1.4	0	0	2.0	7.4	7.10	4.40	2.70	0	—*	100

*Not determined.

tric and esophageal varices being plicated. After this he did fairly well, but continued to bleed slowly from the gastrointestinal tract. On January 27 a percutaneous splenoportogram demonstrated dilated, but patent, splenic and portal veins. The intrahepatic radicles were normal in appearance and emptied rapidly (Fig. 1). Immediately thereafter, a laparotomy was performed, demonstrating a greatly distended splenic vein and artery, a large spleen, and a portal-vein pressure of 315 mm. of saline solution. The liver was normal in size, felt slightly granular and somewhat firmer than normal, but was otherwise unremarkable. At this time a splenectomy and splenorenal shunt were performed, after which the portal-vein pressure dropped to 165 mm. of saline solution.

An adequate surgical biopsy of the liver was obtained at

several small, irregularly outlined areas of sinusoidal dilatation and congestion, without associated hepatocellular destruction. A few clusters of dark-staining mononuclear cells were present in some of the sinusoids, suggesting foci of extramedullary hematopoiesis. Masson and reticulum stains demonstrated normal connective-tissue structure and normal vascular elements.

When last seen in September, 1957, the patient seemed well and had no complaints.

This young man with proved portal hypertension had experienced several bouts of massive gastrointestinal hemorrhage, presumably from large esophageal gastric varices, in the five years before shunt surgery. The associated findings of mental retardation and

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congenital heart disease suggested that the hyper-
tension was due to a developmental anomaly of the
portal vein. However, both a splenoportogram and
careful inspection at laparotomy demonstrated a
patent, somewhat dilated, portal system with definite
elevation of the portal pressure. Two generous liver-



FIGURE 1. Percutaneous Splenoportogram (A) and Tracing (B) in Case 1. Both splenic and portal veins are dilated but patent, and there is flow of dye into the coronary vein and large esophageal varices.

biopsy specimens obtained two years apart demon-
strated only small and inconsistent zones of periportal
congestion, with no apparent progression or altera-
tion in the two-year interval. Serial liver-function
tests have failed to indicate significant hepatic dys-
function. The transient icterus observed on two
occasions was interpreted as being due to post-trans-
fusion hemolysis. The changes in the spleen were
simply those of mild chronic passive congestion.

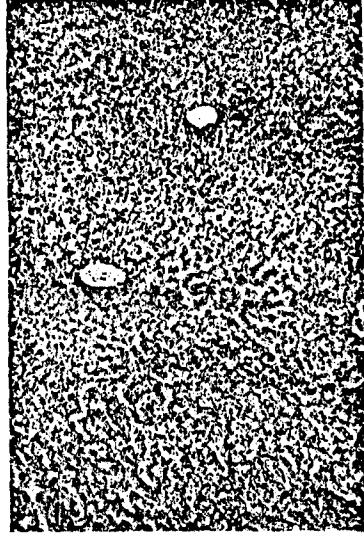


FIGURE 2. Photomicrograph of the Operative Liver-Biopsy Specimen in Case 1, Showing Preservation of the Lobular Pattern, with a Normal Relation of the Portal Triad and Central Vein (Masson Stain — Original Magnification X20).

Thus, definite portal-vein hypertension leading to
rupture of esophageal varices occurred in the absence
of any demonstrable obstruction to portal-blood flow.

CASE 2. E.D., a 58-year-old married male metal polisher,
was admitted to the Grace-New Haven Community Hos-
pital on April 22, 1955, because of massive hematemesis.
He had apparently been well and active until 4 months
previously, when he had an isolated episode of hematemesis
of about 2 cupfuls of red blood, after which he recovered
completely without treatment. Four days before entry, after
drinking some beer, he experienced nausea, dizziness and

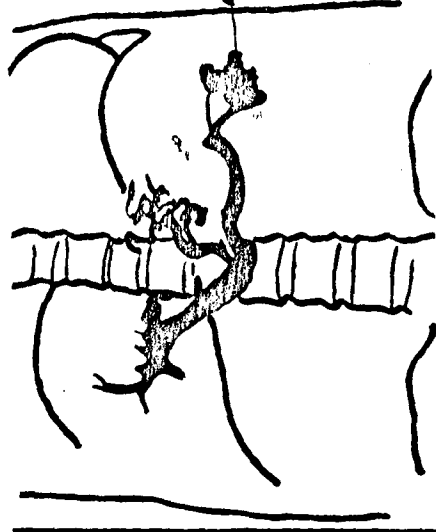


FIGURE 3. Higher Magnification of the Specimen in Figure 2, Showing a Normal Parenchyma, with Dilatation of the Mid-Zonal Sinusoids and Moderate Intrasinusoidal Collections of Mononuclear Cells (Hematoxylin and Eosin Stain — Original Magnification X40).

faintness, followed by the vomiting of large amounts of red
blood. He was admitted to another hospital, where x-ray
studies revealed massive esophageal varices. He received
4 liters of whole blood and was transferred to this hospital
for further evaluation and therapy.

He had drunk 5 or 6 glasses of beer daily for several years,
but he reported a good dietary intake. His weight had been
stable, and he recalled no exposure to drugs or toxins. He

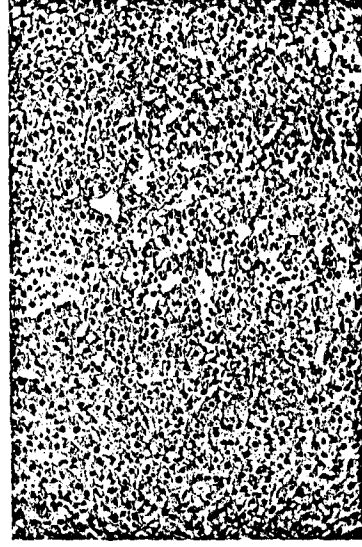


FIGURE 3. Higher Magnification of the Specimen in Figure 2, Showing a Normal Parenchyma, with Dilatation of the Mid-Zonal Sinusoids and Moderate Intrasinusoidal Collections of Mononuclear Cells (Hematoxylin and Eosin Stain — Original Magnification X40).

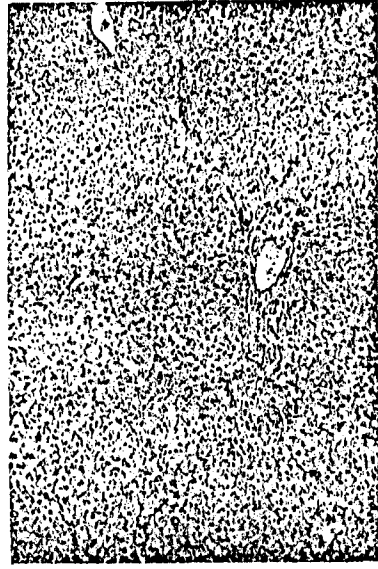
had experienced a brief bout of jaundice in 1914, while
serving with the Army in the Philippines, but had had no
recurrence. There was no history of abdominal pain or
trauma.

The family history was unremarkable.
Physical examination showed a well nourished, somewhat
pale and quite apprehensive man. There was no icterus,

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The pulse was 64, and the respirations 20. The blood pressure was 110/58 in the recumbent position.

Laboratory studies on admission revealed a hematocrit of 42 per cent and a white-cell count of 6300, with a normal differential. Urinalysis and a serologic test for syphilis were negative. The stools were reported as tarry in appearance, and a guaiac test was positive. Complete liver-function studies (Table 2) and serum electrolytes were within normal limits. A transthoracic needle biopsy of the liver



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An end-to-side portacaval shunt was performed at this time, the pressure in an ommental vein falling from 230 to 160 mm. of saline solution. At operation the liver was thought to be "slightly hobnailed," but microscopically it showed a normal lobular architecture with intact parenchyma throughout. Scattered throughout the specimen were small, ill defined, intralobular areas of sinusoidal dilatation. Many of the portal triads, especially those located immediately beneath Glisson's capsule, appeared enlarged by collagen fibers, which often extended in a stellate fashion into adjacent lobules. Some of these strands of connective tissue fused with the overlying capsule, and others reached adjoining triads to produce a "bridging" effect. This periportal fibrosis was minimal in degree and inconstant in distribution. The bile ducts and blood vessels were normal (Fig. 5).

The patient seemed to do well immediately after operation, but on the 3d postoperative day he became slightly drowsy and confused, and 2 days later signs of congestive heart failure developed. Despite vigorous treatment with digitalis, neomycin given by mouth and glutamic acid intravenously, he continued to fail and became comatose. The hematocrit fell, and gastric aspiration revealed blood in the stomach. In spite of several blood transfusions and placement of a Sengstaken-Blakemore tube, he failed and died on the 14th postoperative day.

Post-mortem examination demonstrated marked dilatation of the entire portal venous bed, with no evidence of thrombosis at any point. The portacaval anastomosis was patent. Perfusion of the portal vein with a plastic medium revealed a free flow of fluid throughout the liver. The liver weighed 1900 gm., was light brown and had a finely granular surface. Although the post-mortem specimen showed moderate au-

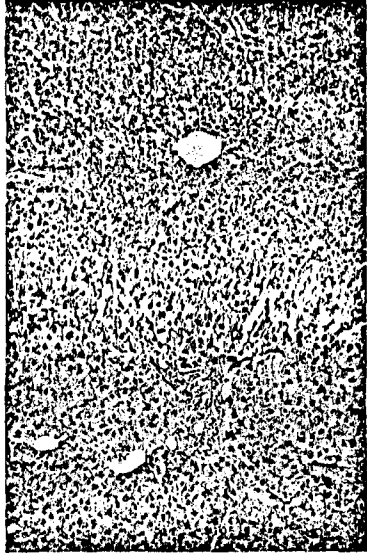


FIGURE 5. Photomicrograph of the Operative Liver-Biopsy Specimen in Case 3, Demonstrating a Normal Lobular Pattern, Moderate Stellate Scarring of a Triad and Minimal Round-Cell Infiltration (Hematoxylin and Eosin Stain—Original Magnification X20).

tolysis, a normal underlying lobular pattern was evident. Glisson's capsule was thickened, edematous and lightly infiltrated with mononuclear inflammatory cells. Many of the portal triads appeared expanded by edema and fibrous tissue, small strands of connective tissue ramifying from the triads into the surrounding parenchyma. In a few periportal areas there were irregular patches of intense sinusoidal congestion, with atrophy and actual necrosis of the intervening parenchymal cells. Inflammatory reaction was minimal in the triads and lobules. Radicles of the portal vein were distended with perfusion solution, and the central veins appeared intact throughout.

The spleen weighed 400 gm., and demonstrated slight capsular thickening, with scattered areas of subcapsular hemorrhage and necrosis. Microscopically, it showed slight sinusoidal congestion, with rare periarterial hemorrhages, definite thickening of the supporting reticulum fibers and a patchy, minimal increase in stainable iron pigment. Deep within the spleen were large, irregular areas of hemorrhagic necrosis, thought to represent recent infarction after injection of contrast medium.

Without previous known or symptomatic liver disease, this elderly man bled suddenly and massively from large esophageal varices. An abnormal retention of bromsulphalein dye, shortly after the most severe bleeding episode, suggested intrinsic liver disease. However, two ante-mortem and the autopsy specimens of the liver showed only inconstant, focal sinusoidal dilatation and a mild degree of periportal and subcapsular fibrosis. At operation the portal pressure was moderately elevated, and on splenopography there was reflux of dye into the coronary and periesophageal plexus of veins, confirming the diagnosis of portal hypertension with the development of portosystemic collateral vessels. Yet neither the splenoportogram nor the examination of the liver gave any evidence of obstruction to the outflow of portal venous blood.

In summary, neither clinical studies nor post-mortem examination indicated a cause of increased resistance to flow within the portal bed. The minimal periportal hepatic fibrosis did not seem extensive enough to distort intrahepatic portal radicles and may have been secondary to the increased portal tension itself.

CASE 4. M.R., a 10-year-old Negro girl, was admitted to the Grace-New Haven Community Hospital on May 10, 1954, because of hematemesis.

She was said to have been well at birth, but at 10 days of age fever and chills had developed and she was noted to have an enlarged spleen. No diagnosis was established at this time, and she recovered completely without treatment, though palpable splenomegaly was recorded on a routine physical examination 6 years later. She had been well and active without gastrointestinal complaints until 4 days before admission, when she experienced a mild headache and transient abdominal cramps. On the day of admission she awoke feeling dizzy and weak, and had a transient fainting episode, followed by the gushing of about 0.5 liter of liquid blood from the nose and mouth.

The past history was unremarkable; she had experienced no jaundice, previous abdominal pain or trauma. She maintained a weight of 34.5 kg. (76 pounds).

Her mother, father and 6 siblings were all living and well.

Physical examination showed a thin, muscular, rather restless girl in no acute distress. There was no jaundice, and, aside from pallor, the skin was within normal limits. The abdomen was flat and relaxed, without evidence of ascites or venous distention. The liver margin was soft and sharp, 1 fingerbreadth below the right costal margin. The spleen was rounded and firm, descending 2 fingerbreadths below the left costal margin. There was no edema or clubbing.

The pulse was 80, and the respirations 18. The blood pressure was 80/30 in the recumbent position.

On entry the hematocrit was 20 per cent, and the white-cell count 13,500, with a normal differential, and platelets were adequate on smear. A serologic test for syphilis was negative, and the nonprotein nitrogen was 43 mg. per 100 ml. The stools were tar colored and gave a + + + + guaiac test. Complete liver-function tests were within normal limits (Table 2).

A gastrointestinal x-ray series on May 15 demonstrated large esophageal and probable gastric varices, without evidence of gastric or duodenal ulceration. Esophagoscopy 5 days later revealed many large varices in the lower 1/3 of the esophagus, without evidence of active bleeding. After receiving 5.5 liters of whole blood the patient was asymptomatic, the hematocrit remaining between 22 and 28 per cent.

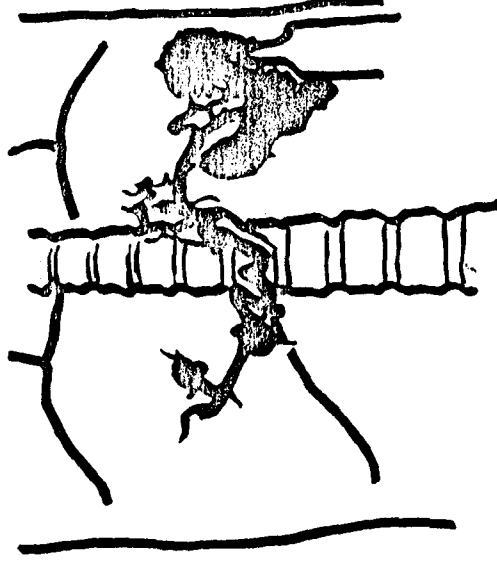
On June 8 a laparotomy was performed. The liver looked normal except for mild subcapsular stellate scarring, but the spleen was described as twice the normal size. An operative splenoportogram demonstrated patent, tortuous splenic and portal veins of normal caliber and rapid flow of dye into the coronary vein and into large esophageal varices, with partial filling of the inferior mesenteric and of several splenic collateral vessels (Fig. 6). Fine and regular serial branch-



A

FIGURE 6. Operative Splenoportogram (A) and Tracing (B) in Case 4.

The splenic and portal veins are tortuous but patent, and there is filling of several collateral vessels, including esophageal varices.



B

ing of the intrahepatic veins was clearly demonstrated. The pressure in the gastropiploic vein was 375 mm. of saline solution. After the performance of a splenorenal shunt and splenectomy, the pressure in this vein fell to 210 mm. of saline solution.

A generous biopsy specimen of the liver was obtained at operation. Microscopically, the hepatic architecture was well preserved throughout, and the parenchymal cells were every-

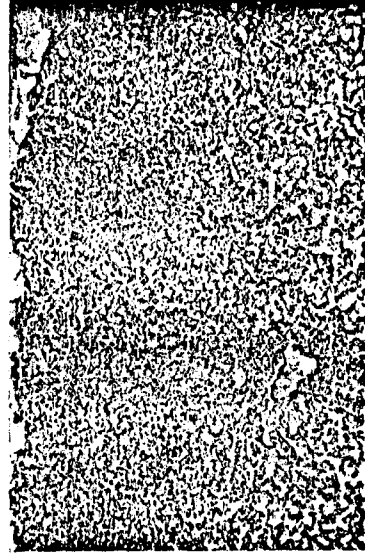


FIGURE 7. Needle-Biopsy Specimen of the Liver in Case 4 Obtained Four Years after Splenorenal Shunt, Demonstrating a Normal Lobular Architecture, with an Intact Central Vein and Moderate Dilatation of the Mid-Zonal Sinusoids (Masson Stain — Original Magnification X20).

where normal. The portal triads were small and contained normal vessels and bile ducts. Radiating from a few of the triads and central veins were thin strands of fibrous tissue that showed no tendency to join one another or to distort the normal lobular architecture. The sinusoids, central veins and portal-vein radicles appeared normal.

The spleen, which measured 5 by 7 by 13 cm., appeared grossly nodular, with a few deep, contracted scars over its anterior border. On microscopic examination its architecture was obscured by fresh hemorrhage, thought to be related to the injection of dye at the time of splenoportography. Otherwise, the splenic architecture was considered to be within normal limits.

The patient made a rapid and complete recovery post-

operatively and had no recurrent bleeding or jaundice. A repeat esophagram in July demonstrated persistence of the varices, but on a subsequent gastrointestinal x-ray study in September, these were no longer evident. She was readmitted to the hospital for evaluation of crampy left-upper-quadrant pain in February, 1958, when physical examination was within normal limits, the hemogram was completely normal, and complete liver-function tests were unremarkable. A barium swallow at this time was reported within normal limits, and a liver biopsy was performed percutaneously with a Vim-Silverman needle. The specimen

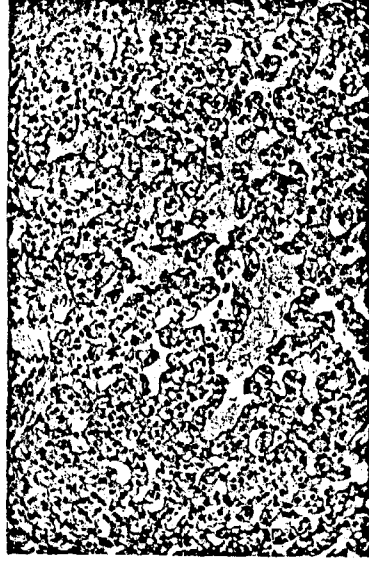


FIGURE 8. Higher Magnification of the Biopsy Specimen Presented in Figure 7, Showing a Normal Parenchyma and Focal Dilatation of the Sinusoids (Masson Stain — Original Magnification X40).

obtained demonstrated a normal lobular pattern in all areas (Fig. 7). In the mid-zonal areas of about half the lobules visualized, there was considerable sinusoidal congestion and distention, without, however, appreciable atrophy or dis-

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ruption of the adjacent liver-cell plates (Fig. 8). Adjoining a portal triad was a single, ill defined collection of small, uniform, mononuclear cells that seemed to compress and partially destroy several parenchymal cells; serial sections revealed no similar lesions. The connective-tissue framework and blood vessels were normal throughout.

On her last follow-up visit, the patient was sturdy, active and without complaints.

When this young Negro girl presented with massive hematemesis and palpable splenomegaly, the past history of a febrile neonatal illness, associated with splenomegaly, suggested that septic inflammation of the portal vein, perhaps from omphalitis, had led to subsequent portal thrombosis, portal hypertension, congestive splenomegaly and the development of varices. Although portal hypertension was confirmed by direct measurement at operation, a beautifully detailed splenoportogram and careful surgical exploration failed to demonstrate any obstruction to the portal outflow, either outside or within the liver. Biopsy on two occasions in a four-year period confirmed the impression that there was no intrinsic liver disease.

OBSERVATIONS

Of the 4 patients described in detail above, 3 were white adult males, and 1 was a young Negro female. Their ages at the onset of symptoms ranged from ten to seventy-two years.

Massive upper gastrointestinal hemorrhage from esophageal varices was the initial manifestation of illness in each case. In none was there any history indicative of antecedent hepatic, portal-system or gastrointestinal disease. The liver was normal in size and consistence, and there were no other manifestations of liver disease. Jaundice appeared after transfusions in Case 1, but this proved to be of hemolytic origin. Except for significant splenomegaly in 2 of the 4 patients, there were no clinical signs of portal hypertension, such as ascites, prominent abdominal-wall collateral veins and periumbilical venous hums.

Except in Case 1, there was no hematologic evidence of "hypersplenism," the blood studies being consistent with recent acute blood loss. Two months before the onset of esophageal bleeding, Case 1 was found to have an elevation of the indirect-reacting fraction of the serum bilirubin, which was interpreted as evidence of a mild hemolytic process.

Results of complete liver-function studies were within the limits of normal, with two exceptions (Table 2): the marked increase in total serum bilirubin, predominantly of the indirect-reacting type, in Case 1, which was compatible with acute hemolysis after recent massive blood replacement and mild hemolytic anemia; and the bromsulfalein retention of 20.3 per cent in Case 3, which in retrospect was probably secondary to massive hemorrhage.

The presence of varices was confirmed radiographically in all 4 cases. In 3, they were demonstrated in both the esophagus and stomach and in

one, were limited to the esophagus. No other potential bleeding site, such as peptic ulcer or hiatus hernia, was demonstrated on the initial films.

Satisfactory radiographic visualization of the splenoportal axis was achieved in all patients, by percutaneous route in Cases 1, 2 and 3, and by cannulation of a mesenteric vein at laparotomy in Case 4. In none was there evidence of narrowing, occlusion or cavernomatous transformation of the splenic or portal vein. However, all patients had one or more abnormalities of the venous bed suggestive of portal hypertension: marked dilatation of the portal vein was seen in Case 1; abnormal reflux of dye into coronary veins, esophageal varices, splenic collateral vessels, or inferior mesenteric branches was seen in Cases 2, 3 and 4; and delayed emptying of major intrahepatic portal radicles was noted in Case 2. At operation no gross arteriovenous fistulas or mechanical distortions involving the portal vein or its tributaries were found.

Portal hypertension was demonstrated by direct measurement at the time of surgery in all patients, the pressures ranging between 230 and 375 mm. of saline solution preoperatively. After establishment of the portosystemic anastomoses, all pressures fell to normal levels.

Although the livers of Cases 1, 3 and 4 were described as slightly scarred or firm at the time of operation, careful microscopical examination of both needle and surgical liver-biopsy specimens failed to confirm the presence of significant scarring or cirrhosis. The most constant lesion demonstrated in all patients was a poorly circumscribed distention of the sinusoids, with occasional compression atrophy of the intervening parenchymal plates, usually localized to the periportal or mid-zonal areas. Sparing of the central zones made it highly unlikely that these changes were due to passive congestion of cardiac disease or of hepatic venous obstructive origin. A second commonly observed pathological finding consisted of occasional thin strands of collagen that projected from slightly thickened portal triads into the adjacent parenchyma. This periportal fibrosis was absent in Cases 1 and 4, minimal in Case 2 and most striking in all the specimens obtained from Case 3. In the last, a few of the fibrous strands connected adjacent portal triads in the subcapsular area, but there was no consistent pattern of scarring, evidence of regenerative activity or pseudolobule formation. Aside from the sinusoidal dilatation described above, there was no distortion of the intrahepatic blood vessels by either connective-tissue bands or parenchymal nodules. None of the liver cells contained fat vacuoles, or gave signs of degeneration, such as "alcoholic" hyaline or acidophilic necrosis. The small zones of extramedullary hematopoiesis in the follow-up biopsy specimen in Case 1, and the single noncaseating granuloma seen in the first of the two specimens ob-

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tained in Case 2, are of unknown significance. Liver-biopsy specimens two and four years after splenectomy shunting and splenectomy in Cases 1 and 4, respectively, demonstrated no progression of the previously found hepatic changes.

Microscopically, the spleen in Cases 1 and 3 showed the diffuse sinusoidal congestion, reticulum thickening and early perivascular fibrosis and hemorrhage considered characteristic of the chronic passive congestion of portal hypertension.⁶ The fibrosis and hemorrhage surrounding the malpighian arteries, termed "fibroadénie" by Banti, and later shown to be nonspecific manifestations of passive congestion of portal hypertension by other workers,⁷ were minimal in Case 1. No abnormal intrasplenic arteriovenous anastomoses were found in any of the spleens.

Discussion

The 4 patients described had clinical and pathological evidence of portal hypertension and bleeding esophageal varices in the absence of either hepatic disease or portal venous obstruction. That this experience is not unique is indicated by several reports in the medical literature. Sir William Osler,⁸ in discussing the syndrome of "splenic anemia," described several patients with chronic splenomegaly and recurrent gastrointestinal hemorrhage in the absence of cirrhosis or gross disease of the portal vein, with apparent cure following splenectomy. Although Rousselot⁹ attributed the portal hypertension of his 15 noncirrhotic patients with "Banti's syndrome" to splenoportal obstruction, he was unable to demonstrate the site of venous block in seven instances. In Whipple's⁴ impressive survey of splenomegaly, 35 cases, or 20 per cent of his patients with "Banti's syndrome," remained with the assumed "obstructive factor" undetermined. Experience at the Mayo Clinic¹⁰ indicated that some patients with the picture of "splenic anemia" and portal hypertension had neither intrahepatic nor extrahepatic venous block. More recently, several similar cases have been reported.¹¹⁻¹⁴ Of interest is the fact that 3 of the patients in the series of Hallenbeck and Shockett¹⁴ had periportal fibrosis and venous distention similar to those described in the present study.

Cases of this type lead necessarily to an examination of the elements that affect portal pressure. Of these, vascular resistance and volume of blood flow appear to be of paramount importance. Partial or total obstruction of the portal vein, in either its extrahepatic or its intrahepatic course, may produce an elevation of portal pressure provided other factors remain constant. Similarly, an increase in the volume of blood traversing the portal vein may raise portal pressure if there is no accompanying change in vascular resistance. Obviously, the effect of either of these factors on portal pressure may be abolished by

reciprocal changes in the other. Thus, the effect of increased blood flow on portal venous pressure may be minimized by dilatation of the portal trunk and the opening of collateral vessels. Similarly, diversion of blood into newly expanded channels may prevent the development of sustained hypertension in portal-vein obstruction.

Most clinical and pathological studies concerned with the pathogenesis of portal hypertension have emphasized the importance of factors that increase resistance to blood flow. Partial or complete obstruction of the portal system in its extrahepatic course by thromboses,¹⁵ scars,⁴ tumor masses¹⁶ or cysts⁴ is known to produce portal hypertension. Within the liver, the constrictive scars¹⁷ and expanding regenerative parenchymal nodules¹⁸ of cirrhosis and invasive tumors¹⁹ commonly raise portal pressure. Finally, the increased intra-abdominal tension accompanying large abdominal cysts²⁰ and exaggerated respiratory movements²¹ may elevate portal pressure.

In contrast to the many reports stressing venous obstruction as a cause of portal hypertension, few studies have dealt with the effect of increased portal blood flow on portal pressure. Theoretically, as stated previously, an increase in the volume of blood entering the portal system may lead to sustained portal hypertension. Practically, this concept is proved by the demonstration of portal hypertension and its consequences in some patients with arteriovenous aneurysms involving the splenoportal axis.²²

Since the blood flow and *vis a tergo* of the portal system are determined largely by the capillary beds and arteriovenous channels of the pancreas, gastrointestinal tract and spleen, functional or structural derangements of their vasculature might permit increased portal blood flow and result in portal hypertension. Little is known about the contribution of the pancreatic circulation to portal blood flow. However, so far as the gastrointestinal tract is concerned, Wornack and Peters²³ have demonstrated in the wall of the bowel humorally controlled arteriovenous anastomoses that help determine the degree of oxygenation of portal blood. Conceivably, these might permit the influx of sufficiently large volumes of blood to raise portal pressure, but further studies are required to establish the importance of this factor in the pathogenesis of portal hypertension.

Both experimental and clinical studies indicate that portal blood flow may be affected dramatically by alterations in splenic circulation. There is evidence that "capillary shunts" exist within the spleen that may permit rapid inflow of blood into the portal bed.²⁴ Ravenna,²⁵ in investigating the pathogenesis of "Banti's syndrome," concluded that primary lesions of the small splenic arteries are responsible for the splenomegaly and portal hypertension characteristic of this syndrome. Clinically, cases of "tropical splenomegaly" of undetermined cause may be associated with

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dilatation of the portal and splenic hypertension.²⁶ Hunt¹⁸ has reported portal pressures of up to 390 mm. of saline with the splenomegaly of myeloid splenic anemia. Although simple splenectomy is rarely considered the operation of splenomegaly and portal hypertension, dramatic lowering of portal pressure and cessation of variceal bleeding may follow this procedure,¹⁶ suggesting that the venous return from the spleen affects the portal pressure directly.

Finally, in considering possible causes of increased portal blood flow that may result in portal hypertension, disturbances of the intrahepatic arterial circulation must be evaluated. That distorted vascular channels within the livers of patients with cirrhosis may produce increased portal blood flow was demonstrated by Herrick.²⁷ Doppler perfusion studies of normal and cirrhotic livers, concluded that increased blood flow through the hepatic artery accounts in large part for the portal hypertension frequently observed in patients with alcoholic cirrhosis. Ligation of the hepatic artery has been performed to decrease intrahepatic blood flow and thus to lower portal pressure.²⁸ It is possible that similar abnormal arterial circulation, perhaps accompanied by increased blood flow through the portal vein or present in the livers of the 4 patients under discussion. If such vascular lesions were small and erratically distributed, they could have escaped detection by both splenoportographic and microscopic examination of liver-biopsy specimens.

We were unable in this study to demonstrate the factors responsible for portal hypertension in our 4 patients. Gross and microscopic examination was capable of producing increased resistance to flow within the portal vein and its intrahepatic radicles appear to have been excluded. However, the possibility of increased vascular resistance due to functional changes, such as increased tone of the portal trunk and its intrahepatic branches, has not been ruled out. Such vasoconstrictor reactions seem to maintain portal venous pressure at normal levels after acute experimental alterations in portal blood flow,³⁰ but their role in clinical portal hypertension has yet to be elucidated.

Increased portal blood flow seems to be the most likely cause of the venous hypertensive cases under discussion. Although anastomoses involving the splenic and portal veins were not found, functional or structural alterations of the vessels of the gastrointestinal tract, spleen and the liver may have permitted increased flow of blood into the portal vein, with a resultant rise in pressure. Despite the compensatory dilatation of the main portal vessels and the opening of numerous venous collaterals, portal pressures remained elevated, with

the eventual production and rupture of esophageal varices.

What is needed is careful study of the nonobstructive factors that may produce portal hypertension. Clinically, the combination of a normal or near normal liver-biopsy specimen and an unobstructed portal vein in a patient with bleeding esophageal varices secondary to portal hypertension should lead to investigation of the factors affecting portal blood flow. Of considerable help would be the direct measurement of blood flow through the portal-vein tributaries at the time of laparotomy. Such critical determinations have been carried out in laboratory animals by means of a square-wave electromagnetic flowmeter.³⁰ Another approach might be the differential temporary occlusion of arteries supplying organs drained by the portal vein, thereby localizing any abnormal arteriovenous unions. Finally, careful post-mortem perfusion studies of the liver, spleen and gastrointestinal tracts of such patients might well reveal important alterations in the structure and function of these blood vessels.

SUMMARY AND CONCLUSIONS

In 4 cases with portal hypertension and massive bleeding from esophageal varices morphologic examinations of the liver and splenoportograms revealed neither intrahepatic nor extrahepatic portal-vein obstruction. The possible etiologic role of functional or undetected structural obstructive alterations of the vessels of the liver remains uncertain.

It is suggested that portal hypertension in the absence of significant venous obstruction may be caused by increased blood flow through the portal vein.

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PATHOGENESIS OF HYPOFIBRINOGENEMIA IN PLACENTAL ABRUPTION*

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THE pathogenesis of hypofibrinogenemia associated with placental abruption continues to be a subject of much conjecture. The most commonly proposed explanation has been intravascular coagulation due to the escape of thromboplastin from the products of gestation into the maternal circulation.¹⁻⁶ Fibrinogen destruction resulting from the activation of circulating plasminogen has also been proposed to account for the fall in circulating fibrinogen.⁷⁻⁹ Recently, Stefanini and Turpini¹⁰ reported hypofibrinogenemia to follow massive hemorrhage in laboratory animals.

Dieckmann,¹¹ who in 1936 first clearly demonstrated that a coagulation defect with severe hypofibrinogenemia occurred with some cases of premature separation of the placenta, suggested that the fibrinogen was lost through hemorrhage and therefore utilized at the site of placental separation. In 1956 Ashworth and Stouffer¹² described in the blood clots from patients with placental abruption a great abundance of material that had the histologic characteristics of fibrin. They too believed that the hypofibrinogenemia resulted from the local deposition of fibrin at the site of placental separation. Very recently, while the present study was in progress, Nilssen,¹³ in Norway, reported his observations on the fibrin content of blood clots from the uterus in cases of placental abruption. He recovered from the blood clots as much as 11.4 gm. of hemoglobin-free, water-insoluble material, which he considered to be fibrin.

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He concluded that factors other than intravascular coagulation may be concerned in defibrination.

Impressed by the large volumes of blood clots within the uterus of patients with this syndrome and by the observations of Ashworth and Stouffer that these appear histologically to be quite rich in fibrin, we undertook studies to quantitate the amount of fibrin that might be lost from the circulation owing to coagulation within the cavity of the uterus. Presented are the results of these studies.

METHODS AND MATERIALS

At the time of delivery as much as possible of the clotted and liquid blood contained within the uterine cavity was collected from 7 patients with placental abruption and varying degrees of hypofibrinogenemia. In each there was total placental abruption, and the fetus had very recently died. The interval from the onset of any symptoms suggesting placental abruption until delivery ranged from four to ten hours. Six patients were delivered vaginally, and 1 by cesarean section. They had received whole blood in a volume not in excess of 250 ml. and no fibrinogen before delivery. Serial measurements were made of the plasma fibrinogen level with the use of the method of Rarnoff and Menzie.¹⁴ Plasma fibrinolytic activity was estimated by determination of the time for complete lysis of oxalated plasma that had been clotted by the addition of an equal volume of 0.025-molar calcium chloride solution and incubated under toluene at 37°C. When the plasma fibrinogen concentration was below 100 mg. per 100 ml., an equal volume of normal plasma was first added to supply fibrinogen. The time for complete lysis normally is longer than two days.

The material was frozen to lyse the erythrocytes, thawed at 37°C. and ground in a Waring blender. It was then centrifuged at room temperature in an

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The Toxicity of Vinyl Chloride as Determined by Repeated Exposure of Laboratory Animals

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Groups of laboratory animals were exposed repeatedly for up to six months to either 500, 200, 100 or 50 ppm vinyl chloride in air. Detectable changes occurred at all but the lowest concentration. The results are discussed and handling precautions suggested.

Introduction

VINYL chloride ($\text{CH}_2=\text{CHCl}$) is a chemical of great industrial importance. It is used in the preparation of polyvinyl chloride resin, as a copolymer in saran and other plastics, as a chemical intermediate and as a solvent. Because of the flammability of vinyl chloride, it has been generally assumed that the greatest hazard associated with vinyl chloride is that due to its flammability rather than its toxicity.¹

The toxicity of vinyl chloride has been reviewed by von Oettingen² and more recently by Mastromatteo *et al.*³ von Oettingen concluded that the gas was anesthetic in high concentrations and that was considerable interest had been shown in the use of vinyl chloride as a surgical anesthetic. However, its effect on the circulatory system has discouraged exploitation of this property. No doubt the hazard from flammability has also hindered this use. Only limited repeated exposures which were reported by Schaumann⁴ were discussed by von Oettingen. These repeated exposures indicated little or no chronic effects even from anesthetic concentrations.

Mastromatteo *et al.*³ also discussed the published data and reported the results of single exposures of mice, rats and guinea pigs which confirmed the low acute toxicity of vinyl chloride. Mastromatteo reported that only two human fatalities due to vinyl chloride had been reported.

It can be concluded from the published toxicological data that anesthesia is the only significant effect of acute exposure. Sufficient repeated exposures have not been reported to draw conclusions about chronic toxicity. The Threshold Limit Value of 500 ppm suggested by the American Conference of Governmental Industrial Hygienists (ACGIH)⁵ is reported by Smyth⁶ to be based on single animal exposures and human experience.

The following report summarizes the results of repeated exposures of laboratory animals to either 500, 200, 100 or 50 ppm of vinyl chloride. The significance of the results is discussed and recommendations for a threshold limit value are made.

Experimental Procedures

Materials Tested

Vinyl chloride, $\text{CH}_2=\text{CHCl}$, is a colorless gas. It has a boiling point of -13.35°C ($+7.93^\circ\text{F}$) and a freezing point of -154°C (-244.82°F). The material polymerizes readily and hence is sometimes inhibited with phenol or tertiary butylcatechol. Vinyl chloride is very flammable, its flashpoint being -78°C (-108°F). The explosive limits are from 4% to 22% by volume. It has little odor although high concentrations may smell faintly sweet.

A single, uninhibited sample was used in these studies. It was shown by mass spectrographic analysis to be essentially pure $\text{CH}_2=\text{CHCl}$, air being the only impurity detected in gas phase samples.

Source and Feeding of Animals

The rats and rabbits used in this study were obtained from the stock colony of this laboratory, the guinea pigs were obtained from a commercial grower⁷ and the dogs were purebred beagles obtained from a local kennel. The rats and dogs were fed Purina Laboratory Chow⁸ or Famo Laboratory Ration.⁹ The rabbits and guinea pigs received Famo Rabbit Breeder Ration. The guinea pig diet was supplemented with carrots.

Experimental Protocol

The experiments were conducted in three phases. In the first phase, groups of 10 male

and 10 female rats were exposed seven 1 day five days per week to 500 ppm months. Exposures were given in a chamber previously described.¹⁰ Mortality growth records were kept. Final organ weights were obtained and tissues were saved for microscopic examination. Five male and five rats served as unexposed controls.

The second phase consisted of repeated daily exposures of 20-24 male and female rats, ten male and eight female guinea pigs, three male and three female rabbits, one male and one female dog to either 500 or 100 ppm of vinyl chloride. These animals and groups of controls were carefully matched on the basis of age, condition and weight. The first group of controls had no exposure and served as unexposed controls. The second group received repeated hourly exposures to room air in a chamber to the one used for exposures to the vinyl chloride. This group is referred to as the air-exposed control group. In addition to the animals having 7-hour daily exposures, eight separate five male rats each were exposed to 500 or 100 ppm for 4, 2, 1 or 0.5 hours per day.

The procedures and equipment used were identical the same as reported previously. And mortality records were kept on a daily basis. The livers of the dogs were biopsied prior to exposure and after 3½ months of exposure the liver of each dog served as its own control. Pre-exposure and terminal hematological determinations were made on all dogs and determinations on representative groups of rats. Urine samples were collected from the representative groups of rats. At the termination of the experiment, part of the rats were kept overnight and then these and all the guinea pigs and rabbits were killed by decapitation the day after their last exposure. The dogs were killed by exsanguination after anesthesia. Thiopental, sodium (Pentothal, Abbott) was used for determination of alkaline phosphatase, serum-urea-nitrogen (SUN), serum-tamie-pyruvic-transaminase (SGPT), serum-glutamic-oxalacetic-transaminase (SGOT). The organs were weighed and tissues were saved for histopathological examination. The rats were not killed the day after exposure stopped were fastured for eight weeks and sacrificed in the previously described manner.

The third phase of this experiment consisted of repeated daily 7-hour exposures of 20-24 female rats, 12 male and 12 female guinea pigs, 3 male and 3 female rabbits and

Results Determined in Laboratory

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For six months to either side of the exposure, but no deaths occurred at all but no precautions suggested.

The report summarizes the results of laboratory animals to which 50 ppm of vinyl chloride, or 50 ppm of vinyl chloride, the results are discussed and a threshold limit value are

Conclusions

$\text{H}_2=\text{CHCl}$, is a colorless gas, boiling at -13.35°C ($+7.93^\circ\text{F}$) and melting at -154°C (-244.82°F). It is readily and hence is used with phenol or tertiary amines. It is very flammable, with a flash point of -108°F . The explosion limits are 4% to 22% by volume, although high concentrations are sweet.

The sample was used in these experiments by mass spectrographic analysis. Initially pure $\text{CH}_2=\text{CHCl}$, air purity detected in gas phase

Use of Animals

The animals used in this study were from a stock colony of this laboratory. The rats were obtained from a local kennel and the dogs were purebred from a local kennel. The rats were from Purina Laboratory Chow[®] and Ration[®]. The rabbits and guinea pigs were from Farnam Rabbit Breeder and pig diet was supplemented

Conclusions

The experiments were conducted in three phases, groups of 10 male

and 10 female rats were exposed seven hours per day five days per week to 500 ppm for 4.5 months. Exposures were given in a 160-liter chamber previously described.¹⁰ Mortality and growth records were kept. Final organ weights were obtained and tissues were saved for microscopic examination. Five male and five female rats served as unexposed controls.

The second phase consisted of repeated 7-hour daily exposures of 20-24 male and 24 female rats, ten male and eight female guinea pigs, three male and three female rabbits and one male and one female dog to either 200 ppm or 100 ppm of vinyl chloride. These animals and two groups of controls were carefully selected and matched on the basis of age, condition and weight. The first group of controls received no exposure and served as unexposed controls. The second group received repeated daily 7-hour exposures in room air in a chamber similar to the one used for exposures to the chemical. This group is referred to as the air-exposed control group. In addition to the animals receiving 7-hour daily exposures, eight separate groups of five male rats which were exposed to either 200 or 100 ppm for 2, 1 or 0.5 hours per day.

The procedure and equipment used were basically the same as reported previously.¹⁰ Growth and mortality records were kept on all groups. The livers of the dogs were biopsied prior to exposure and after 3½ months of exposure, hence the liver of each dog served as its own control. Pre-exposure and terminal hematological determinations were made on all dogs and terminal determinations on representative groups of rats. Urine samples were collected from the dogs and representative groups of rats. At the termination of the experiment, part of the rats were starved overnight and then these and all the guinea pigs and rabbits were killed by decapitation on the day after their last exposure. The dogs were killed by examination after anesthesia with thiopental sodium (Pentothal, Abbott). Samples of blood were taken from representative groups of animals for determination of alkaline phosphatase, serum urea nitrogen (SUN), serum glutamic-pyruvic transaminase (SGPT), and serum glutamic-oxaloacetic transaminase (SGOT). The organs were weighed and tissues fixed for histopathological examination. The rats which were not killed the day after exposures were stopped and were put to rest for eight weeks and then sacrificed in the previously described manner.

The third phase of this experiment consisted of repeated daily 7-hour exposures of 24 male and 24 female rats, 12 male and 12 female guinea pigs, 3 male and 3 female rabbits and one male

and one female dog to 50 ppm vinyl chloride. Equal matched groups served as unexposed and air-exposed controls. Three additional groups of 10 male rats each were exposed for 4, 2 or 1 hour per day to 50 ppm.

The procedures followed were essentially the same as used for exposures to 200 and 100 ppm except that hematological examinations were made on the dogs after three months and liver biopsies were made only prior to exposure. Urine was collected from the rabbits as well as the rats and dogs. The rats which were not sacrificed at the end of the exposures were pastured for six weeks and then sacrificed.

Equipment Used

The exposure equipment used has been described previously.¹⁰ However, in order to gain additional data about the results of repeated short daily exposures, it was necessary to devise a method of introducing additional groups of rats. This was done without opening the doors of the chamber by dropping the rats through chutes made of 3.25-inch stainless steel tubing. These chutes were closed with a rubber stopper except during the brief time it took to insert the rats. The chutes, which were sloped at a 45° angle, ended in covered, screened cages inside the chamber. The additional groups were started 3, 5, 6 and 6.5 hours after the 7-hour exposures started. Since all animals were removed after the 7-hour exposure was completed, this routine resulted in exposures lasting 4, 2, 1 or 0.5 hours.

The vinyl chloride was metered from a saron plastic bag which served as a reservoir for the gas. Metering of the gas was done by Dual Syringe Pumps.¹¹ Total air flow through the chambers was measured by means of calibrated flow meters.

Analysis of the Chamber Atmospheres

Analyses of the air were made by direct combustion of a known volume of air in a heated quartz tube. The resulting chloride was trapped in 1% sodium formate—1% sodium carbonate solution and subsequently titrated by a micro-Volhard technique. The results of the individual analyses were within 15% of the theoretical concentrations during the exposures to 200, 100 and 50 ppm. The average concentrations recovered were 197, 100.5 and 48.7 ppm respectively.

Results of Repeated Exposures

Exposure to 500 ppm Vinyl Chloride

The rats exposed repeatedly at 500 ppm for 4.5 months, grew normally and no changes were

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TABLE I
Summary of Terminal Average Body and Organ Weights of Male Rats Receiving Repeated Exposures to Vinyl Chloride 5 Days per Week

Concentration in ppm	Months on Exposure	Duration of Daily Exposure, hours	Ratio of Survival	Organ Weights, g/100 g Body Weight					Testes
				Final Average Weight, g	Lung	Heart	Liver	Kidney	Spleen
Unexposed control.....	0	0	1.5	304	0.54	0.33	2.52	0.71	0.15
500	4.5	7	7/10	312	0.56	0.33	3.09 ^a	0.73	0.17
Unexposed control.....	0	0	11/12	313	0.53	0.32	2.45	0.69	0.14
Air exposed control.....	6	7	10/12	356	0.52	0.32	2.52	0.70	0.15
200	6	7	9/12	311	0.52	0.32	2.85 ^b	0.68	0.17
200	6	4	4/5	319	0.47	0.29	2.66 ^c	0.61	0.16
200	6	2	4/5	311	0.56	0.30	2.62 ^d	0.61	0.16
200	6	1	3/5	350	0.48	0.28	2.46	0.61	0.13
200	6	0.5	4/5	339	0.51	0.29	2.38	0.67	0.14
100	6	7	7/12	352	0.51	0.29	2.61 ^e	0.63	0.15
100	6	4	3/5	360	0.51	0.28	2.62	0.65	0.15
100	6	2	4/5	331	0.53	0.31	2.61	0.67	0.16
100	6	1	3/5	387	0.49	0.28	2.52	0.64	0.15
100	6	0.5	5/5	363	0.45	0.30	2.42	0.61	0.16
Unexposed* control.....	0	0	10/12	351	0.49	0.31	2.43	0.64	0.14
Air exposed* control.....	6	7	8/12	393	0.59	0.30	2.41	0.63	0.15
200*.....	6	7	8/8	395	0.47	0.29	2.50 ^f	0.61	0.15
100*.....	6	7	6/7	376	0.47	0.31	2.53 ^f	0.65	0.14

* Fastured for 8 weeks after exposures ceased.

(a) $P = 0.001$

(b) $P = <0.001$

(c) $P = 0.07$

(d) $P = 0.1$

(e) $P = 0.02$

(f) $P = 0.13$

apparent grossly at autopsy. Microscopically their livers showed increased central lobular granular degeneration and the kidneys showed interstitial and tubular changes. The average weight increase of the livers of the male rats was statistically significant ($P = 0.001$), Table I. The average weight of the livers of the female rats was above that of the controls but not statistically significant, Table II. SUN, SGPT, SGOT, and alkaline phosphatase determinations were within normal limits (Table III).

Exposure to 200 ppm Vinyl Chloride

All groups exposed seven hours per day 138-144 times in 204 days were normal in appearance, mortality and growth. Hematological, SUN, SGOT, SGPT and alkaline phosphatase values (Table III) and the results of the urinalysis were within normal limits. Gross pathology was normal in all species. Micropathology was normal in the rats, guinea pigs and dogs; however, adverse effects were noted in the livers of

Rabbits of both sexes. In the male rabbits this was characterized by central lobular granular degeneration and necrosis with some foamy vacuolation. In the female rabbits the changes noted were central lobular granular degeneration and necrosis with periportal cellular infiltration. All organ weights were normal except for the livers of the male and female rats which were increased significantly after repeated daily 7-hour exposures (Tables I, II, IV, V, VI, and VII). In the males this increase was still apparent eight weeks after exposure ceased, although the liver weights appeared to be decreasing and returning to normal.

The male rats exposed to 200 ppm of vinyl chloride for either four or two hours per day for six months had elevated average liver weights (Table I). However, the small size of the experimental groups made the finding statistically insignificant ($P = 0.07$ and 0.1). The groups exposed for either 1 or 0.5 hours per day were entirely normal.

TABLE II

Summary of Average Body and Organ Weights of Female Rats Receiving Repeated Exposures to Vinyl Chloride 5 Days per Week

Concentration in ppm	Months on Exposure	Ratio of Survival	Final Average Body Weight, g	Organ Weights, g/100 g Body Weight	
				Lung	Heart
Unexposed control.....	0	5/5	191	0.52	0.4
500	4.5	9/10	195	0.65	0.3
Unexposed control.....	0	10/12	202	0.63	0.3
Air exposed control.....	6	9/12	223	0.66	0.3
200	6	10/12	204	0.63	0.3
100	6	11/12	221	0.60	0.3
Unexposed* control.....	0	8/12	223	0.61	0.3
Air exposed* control.....	6	9/12	210	0.58	0.3
200*.....	6	12/12	233	0.61	0.3
100*.....	6	12/12	235	0.63	0.4
Unexposed control.....	0	11/12	211	0.58	0.3
Air exposed control.....	6	12/12	202	0.59	0.3
50	6	9/12	204	0.61	0.3
Unexposed* control.....	0	11/12	193	0.60	0.4
Air exposed* control.....	6	9/12	221	0.63	0.4
50*.....	6	11/12	233	0.63	0.3

* Fastured for 8 weeks after exposures ceased.

(a) $P = 0.17$

(b) $P = 0.03$

(c) $P = 0.01$

(d) $P = 0.05$

Exposure to 100 ppm Vinyl Chloride

All species exposed to 100 ppm vinyl chloride, seven hours per day, 138-144 days, were judged normal on the following criteria: appearance, hematology, hematological examination, SGPT, alkaline phosphatase determination, gross and microscopic tissue. However, slight increases

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Repeating Repeated Exposures

Weights, g/100 g Body Weight				
	Liver	Kidney	Spleen	Testes
2.52	0.71	0.15	0.89	
3.00 ^a	0.73	0.17	0.95	
2.45	0.69	0.14	0.80	
2.52	0.70	0.15	0.76	
2.85 ^b	0.68	0.17	0.94	
2.66 ^c	0.64	0.16	0.85	
2.62 ^d	0.64	0.16	0.89	
2.46	0.61	0.13	0.78	
2.38	0.67	0.14	0.95	
2.61 ^e	0.63	0.15	0.85	
2.62	0.65	0.15	0.74	
2.64	0.67	0.16	0.93	
2.52	0.64	0.15	0.78	
2.42	0.64	0.16	0.78	
2.43	0.64	0.14	0.84	
2.41	0.63	0.15	0.78	
2.59 ^f	0.64	0.15	0.71	
2.53 ^f	0.65	0.14	0.76	

xy central lobular granular degeneration with some foamy vacuolation in the male rabbits this was not observed. All other organs except the livers of female rats which were injured after repeated daily 7-day exposures to 500 ppm of vinyl chloride were still apparently normal except for the livers of female rats which were injured after repeated daily 7-day exposures to 500 ppm of vinyl chloride. The liver weights of both male and female rats which had shown definite increases after exposures to 200 and 100 ppm were normal or approaching normal after 6-8 weeks of recovery (Tables I, II and VIII).

It is interesting to speculate on the apparent discrepancy between the effects observed at 100 ppm in this experiment and the lack of reported injury in humans using a threshold limit of 500 ppm. Several factors may be contributory. First, the effects noted in animals at 500 ppm were only slight to moderate; growth, mortality, and general appearance were unaffected. Hence, it is possible that if humans were exposed repeatedly at 500 ppm slight injury might occur but due to the lack of subjective symptoms the injury would escape detection. Secondly, the injury was apparently reversible. The liver weights of both male and female rats which had shown definite increases after exposures to 200 and 100 ppm were normal or approaching normal after 6-8 weeks of recovery (Tables I, II and VIII).

TABLE II
Summary of Average Body and Organ Weights of Female Rats Receiving Repeated 7-Hour Exposures to Vinyl Chloride 5 Days per Week

Concentration in Ppm	Months on Exposure	Ratio of Survival	Final Average Body Weight, g	Organ Weights, g/100 g Body Weight			
				Lung	Liver	Kidney	Spleen
Unexposed control	0	5/5	191	0.52/0.41	3.06	0.80	0.19
500	4.5	9/10	195	0.65/0.38	3.22 ^a	0.81	0.20
Unexposed control	0	10/12	202	0.65/0.38	2.92	0.82	0.17
Air exposed control	6	9/12	223	0.66/0.39	2.99	0.81	0.18
200	6	10/12	208	0.63/0.38	3.30 ^b	0.81	0.18
100	6	11/12	221	0.59/0.39	3.35 ^c	0.79	0.20
Unexposed control	0	8/12	223	0.61/0.37	2.67	0.82	0.18
Air exposed control	6	9/12	240	0.58/0.37	2.97	0.83	0.18
200	6	12/12	233	0.61/0.38	3.06	0.81	0.20
100	6	12/12	235	0.63/0.40	3.17	0.83	0.21
Unexposed control	0	11/12	211	0.55/0.39	3.02	0.83	0.20
Air exposed control	6	12/12	202	0.59/0.38	2.93	0.84	0.20
50	6	9/12	206	0.61/0.37	3.02	0.81	0.20
Unexposed control	0	11/12	195	0.66/0.40	2.88	0.91	0.22
Air exposed control	6	9/12	221	0.63/0.40	3.01	0.87	0.19
50	6	11/12	233	0.63/0.38	2.83	0.82/0.20	

^a Pictured for 8 weeks after exposures ceased.

^b Pictured for 6 weeks after exposures ceased.

(a) P = 0.17

(b) P = 0.03

(c) P = 0.01

(d) P = 0.05

Exposure to 100 ppm Vinyl Chloride

All species exposed to 100 ppm of vinyl chloride, seven hours per day, 135-144 times in 204 days, were judged normal on the basis of the following criteria: appearance, mortality, growth, hematological examination, SUN, SGOT, SGPT, alkaline phosphatase determination, urinalysis and gross and microscopic examination of tissue. However, slight increases were found in

the average weights of the livers of male and female rats (Tables I and II). Although not significant statistically, an increase was also seen in the average weight of the livers of the male rats exposed to 100 ppm for either four or two hours per day. The groups exposed for either 1.0 or 0.5 hour per day were entirely normal.

Exposures to 50 ppm Vinyl Chloride

The increase in weight of the rat livers, the only significant finding in animals exposed to 100 ppm, did not occur in rats exposed repeatedly to 50 ppm of vinyl chloride, 130 times in 189 days. All groups of animals were judged normal in all other respects. See Tables II, V, VI, VII, VIII and IX. A statistically significant decrease in kidney weight which was seen in the female rats (Table II) was considered to be an artifact since it was not seen at higher concentrations.

Discussion

Repeated exposure to vinyl chloride at concentrations considerably below the level that has been considered safe for human exposure has been shown to have an effect on the liver and kidney of laboratory animals. Histopathological changes and increased liver weights in male and female rats were noted after repeated exposure at 500 ppm. Repeated exposure at 200 ppm for six months resulted in an increase in the average weight of the livers of male and female rats and neuropathological changes in the livers of the male and female rabbits. The only effect noted after repeated daily 7-hour exposures to 100 ppm for six months was a slight increase in the average weight of the rat livers. Repeated 7-hour exposures to 50 ppm for six months had no effect on any species studied. Hence, the highest concentration without detectable effect on any species was 50 ppm.

It is interesting to speculate on the apparent discrepancy between the effects observed at 100 ppm in this experiment and the lack of reported injury in humans using a threshold limit of 500 ppm. Several factors may be contributory. First, the effects noted in animals at 500 ppm were only slight to moderate; growth, mortality, and general appearance were unaffected. Hence, it is possible that if humans were exposed repeatedly at 500 ppm slight injury might occur but due to the lack of subjective symptoms the injury would escape detection. Secondly, the injury was apparently reversible. The liver weights of both male and female rats which had shown definite increases after exposures to 200 and 100 ppm were normal or approaching normal after 6-8 weeks of recovery (Tables I, II and VIII).

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TABLE III

Summary of Average Biochemical Values Determined for Animals Receiving Repeated Exposures to Vinyl Chloride 5 Days per Week

Species and Dog No.	Sex	Concentration in ppm	Months on Exposure	Duration of Daily Exposure, hours	Number of Animals	Alkaline Phosphate King-Armstrong Units	Blood Urea Nitrogen, mg/100 ml	SGPT	SGOT
								Sigma	Frankel Unit
Dog #101.....	M	Unexposed control	0	0	1	12.2	21	18	43
#113.....		Air exposed control	6	7	1	4.1	19	19	17
#41.....		200	6	7	1	8.6	11	18	48
#98.....		100	6	7	1	4.5	16	21	19
Dog #106.....	F	Unexposed control	0	0	1	5.1	11	10	26
#107.....		Air exposed control	6	7	1	7.1	20	11	17
#103.....		200	6	7	1	7.9	13	16	26
#110.....		100	6	7	1	5.6	10	10	14
Rat.....	M	Unexposed control	0	0	5	32.9*	20.6*	30	
		500	4.5	7	4	18.9	13.8	35	
		200	6	7	5	17.7	16.2	36	
		200	6	4	4	18.4	16.2	32	
		200	6	1	3	20.2	17.7	42	
		100	6	7	4	19.4	15.0	27	
		100	6	4	3	11.7	16.3	36	
		100	6	1	3	19.0	15.3	30	
Rat.....	F	Unexposed control	0	0	4	12.1	20.5	38	
		500	6	7	3	16.1	22.3	26	
		200	6	7	5	13.6	22.2	21	
		100	6	7	6	9.5	22.8	23	
Rabbit.....	M	Control	6	0	5	1.0	25	36	
		200	6	7	3	5.3	21	23	
		100	6	7	2	5.9	26	26	
Rabbit.....	F	Control	6	0	6	4.9	33	38	
		200	6	7	2	5.6	27	35	
		100	6	7	3	4.0	26	35	

* Because of two unusually high individual values in this group these average values are elevated and do not correspond with the values obtained on similar control groups in our laboratory. Normal control values are in the same ranges as those of the experimental groups listed in this table.

TABLE IV
Summary of Average Hematological Values for Animals Receiving Repeated 7-Hour Exposures to Vinyl Chloride

Species or Dog #	Sex	Concentration in ppm	Months on Experiment	No. of Animals	Hemo- globin g/100 g	Hema- tocrit	WBC $\times 10^3$	Neu- trophils	Lym- phocytes	Monocytes	Eosino- phils
Rat.....	M	Air exposed control	0	4	11.5	51	15.8	26.5	71.2	1.5	0.8
Rat.....	M	200	6	5	14.4	50	16.7	19	77.8	0.2	3
Rat.....	F	Air exposed control	6	4	14.0	49.5	13.5	22.5	74.5	0.2	2.8
Rat.....	F	200	6	5	13.5	47.4	12.8	24.8	70.2	1.0	4
Dog #101.....	M	Unexposed control	0	1	—	—	—	—	—	—	—
Dog #101.....	M	Unexposed control	6	1	15	53	14.2	46	51	0	3
Dog #113.....	M	Air exposed control	0	1	14.5	50	13.5	61	23	5	11
Dog #113.....	M	200	6	1	14.5	52	13.2	58	34	4	4
Dog #111.....	M	Air exposed control	0	1	13.5	47	16.2	56	28	3	13
Dog #111.....	M	200	6	1	16.5	58	15.8	56	41	2	1
Dog #111.....	M	100	6	1	15.5	52	19.8	46	46	4	4
Dog #98.....	M	100	6	1	16	54	18.6	61	39	0	0
Dog #106.....	F	Unexposed control	0	1	14.5	51	21.8	42	37	3	18
Dog #106.....	F	Unexposed control	6	1	15.5	51	16.5	55	41	1	4
Dog #107.....	F	Air exposed control	0	1	15	53	11.1	66	23	4	8
Dog #107.....	F	200	6	1	16	57	11.7	66	23	4	7
Dog #103.....	F	Air exposed control	0	1	14	50	17.5	44	48	3	5
Dog #103.....	F	200	6	1	16	51	17.3	46	51	0	3
Dog #110.....	F	100	0	1	12.5	41	16.3	61	37	2	0
Dog #110.....	F	100	6	1	15.5	51	14.4	49	50	1	0

TABLE V

Summary of Terminal Average Body Weights of Guinea Pigs Receiving Daily 7-Hour Exposures to Vinyl Chloride 6 Months

Sex	Concentration in ppm	Ratio of Survival	Final Average Body Weight, g	Organ Weights, g	
				Lung	Heart
M	Unexposed control	8/10	927	0.61	0.28
M	Air exposed control	5/10	1030	0.52	0.26
M	200	7/10	1011	0.53	0.27
M	100	7/10	1089	0.52	0.25
M	Unexposed control	8/12	918	0.60	0.20
M	Air exposed control	11/12	936	0.61	0.27
M	50	10/12	1001	0.63	0.26
F	Unexposed control	6/8	899	0.62	0.27
F	Air exposed control	7/8	867	0.60	0.25
F	200	7/8	897	0.71 ^b	0.27
F	100	7/8	919	0.58	0.26
F	Unexposed control	9/12	728	0.74	0.31
F	Air exposed control	8/12	881	0.65	0.26
F	50	11/12	906	0.69	0.26

(a) $P = 0.05$ (b) $P = 0.21$ (c) $P = 0.23$

TABLE VI

Summary of Terminal Average Body Weights of Rabbits Receiving 7-Hour Exposures to Vinyl Chloride 6 Months

Sex	Concentration in ppm	Ratio of Survival	Final Average Body Weight, g	Organ	
				Lung	Heart
M	Unexposed control	3/3	3.61	0.39	0.1
M	Air exposed control	2/3	3.94	0.31	0.
M	200	3/3	3.14	0.32	0.
M	100	3/3	3.65	0.33	0.
M	Unexposed control	2/3	3.89	0.32	0.
M	Air exposed control	1/3	3.57	0.37	0.
M	50	3/3	3.70	0.36	0.
F	Unexposed control	3/3	1.13	0.31	0.
F	Air exposed control	3/3	3.91	0.30	0.
F	200	2/3	4.10	0.32	0.
F	100	3/3	4.12	0.29	0.
F	Unexposed control	3/3	1.31	0.30	0.
F	Air exposed control	2/3	3.95	0.30	0.
F	50	3/3	1.31	0.32	0.

ing Repeated Exposures to

Real Phos- phatase King- Arm- strong Units	Blood Urea Nitrogen, mg/100 ml	SGPT		Sigma-Frankel Unit
		SGPT	SGOT	
12.2	24	18	43	
4.1	19	19	17	
8.6	11	18	48	
4.5	16	21	19	
5.1	11	10	26	
7.1	20	11	17	
7.9	14	16	26	
5.6	10	10	14	
32.9*	20.6*	30		
18.9	13.8	35		
17.7	16.2	36		
18.4	16.2	32		
20.2	17.7	42		
19.4	15.0	27		
14.7	16.3	36		
10.0	15.3	30		
12.1	20.5	38		
16.1	22.3	26		
13.6	22.2	21		
9.5	22.8	23		
4.0	25	36		
5.3	21	23		
5.9	26	26		
4.0	33	38		
5.6	27	35		
4.0	26	35		

used, or not correspond with the changes as those of the experimental

Receiving

X	Neutro- phils	Lym- pho- cytes	Mono- cytes	Eosino- phils
26.5	71.2	1.5	0.8	
19	77.8	0.2	3	
22.5	74.5	0.2	2.8	
24.8	70.2	1.0	4	
46	51	0	3	
61	23	5	11	
58	34	4	1	
56	28	3	13	
56	41	2	1	
46	46	4	4	
61	39	0	4	
42	37	3	18	
54	41	1	4	
55	33	4	8	
66	23	4	7	
44	48	3	5	
46	51	0	3	
61	37	2	0	
49	50	1	0	

TABLE V

Summary of Terminal Average Body and Organ Weights of Guinea Pigs Receiving Repeated Daily 7-Hour Exposures to Vinyl Chloride 5 Days per Week for 6 Months

Sex	Concentration in ppm	Ratio of Survival	Organ Weights, g/100 g Body Weight				
			Final Average Body Weight, g	Lung	Heart	Liver	Testes
M	Unexposed control	8/10	925	0.61	0.28	3.32	0.70
M	Air exposed control	5/10	1020	0.52	0.28	3.36	0.61
M	200	7/10	1014	0.53	0.27	3.01	0.60
M	100	7/10	1083	0.52	0.25	2.79	0.58
M	Unexposed control	8/12	918	0.60	0.20	3.21	0.68
M	Air exposed control	11/12	934	0.61	0.27	3.31	0.64
M	50	10/12	1001	0.63	0.26	3.11	0.61
F	Unexposed control	6/8	890	0.62	0.27	3.20	0.63
F	Air exposed control	7/8	887	0.66	0.25	3.28	0.67
F	200	7/8	807	0.71	0.27	3.69	0.66
F	100	7/8	919	0.58	0.26	3.50	0.61
F	Unexposed control	9/12	728	0.78	0.31	3.41	0.69
F	Air exposed control	8/12	884	0.65	0.26	3.51	0.65
F	50	11/12	906	0.69	0.28	3.11	0.61

(a) $P = 0.05$ (b) $P = 0.24$ (c) $P = 0.23$

TABLE VI

Summary of Terminal Average Body and Organ Weights of Rabbits Receiving Repeated Daily 7-Hour Exposures to Vinyl Chloride 5 Days per Week for 6 Months

Sex	Concentration in ppm	Ratio of Survival	Organ Weights, g/100 g Body Weight				
			Final Average Body Weight, g	Lung	Heart	Liver	Testes
M	Unexposed control	3/3	3.61	0.39	0.21	2.27	0.43
M	Air exposed control	2/3	3.90	0.31	0.18	2.41	0.46
M	200	3/3	3.18	0.32	0.19	2.31	0.51
M	100	3/3	3.65	0.33	0.18	2.61	0.49
M	Unexposed control	2/3	3.80	0.32	0.17	2.33	0.45
M	Air exposed control	1/3	3.57	0.37	0.18	3.18	0.53
M	50	3/3	3.70	0.36	0.20	2.46	0.47
F	Unexposed control	3/3	4.13	0.31	0.17	2.22	0.49
F	Air exposed control	3/3	3.91	0.30	0.18	2.18	0.41
F	200	2/3	4.10	0.32	0.16	2.26	0.42
F	100	3/3	4.12	0.29	0.18	1.87	0.38
F	Unexposed control	3/3	4.31	0.30	0.17	1.91	0.46
F	Air exposed control	2/3	3.95	0.30	0.17	1.88	0.39
F	50	3/3	4.31	0.32	0.17	2.10	0.38

TABLE VII

Summary of Final Body and Organ Weights of Dogs Receiving Repeated Daily 7-Hour Exposures to Vinyl Chloride 5 Days per Week for 6 Months

Dog No.	Concentration in ppm	Ratio of Survival	Final Body Weight, kg	Organ Weights, g/100 g Body Weight				
				Lung	Heart	Liver	Kidney	Testes
101	Unexposed control	1/1	13.0	0.66	0.62	2.46	0.38	0.32
113	Air exposed control	1/1	8.5	0.68	0.77	2.63	0.47	0.19
111	200	1/1	10.2	1.03	0.96	2.50	0.52	0.25
98	100	1/1	12.0	0.86	0.80	2.50	0.42	0.26
152	Unexposed control	1/1	13.3	0.80	0.75	2.69	0.53	0.23
156	Air exposed control	1/1	10.6	0.77	0.87	3.05	0.45	0.36
162	50	1/1	9.6	0.89	0.78	2.40	0.45	0.21
106	Unexposed control	1/1	9.6	0.71	0.68	2.67	0.43	0.21
107	Air exposed control	1/1	8.7	0.77	0.80	3.15	0.43	0.33
103	200	1/1	10.5	0.78	0.71	2.88	0.45	0.28
110	100	1/1	11.9	0.96	0.78	2.11	0.43	0.31
151	Unexposed control	1/1	8.6	0.69	0.82	1.10	0.45	0.43
155	Air exposed control	1/1	11.3	0.70	0.81	3.31	0.40	0.25
161	50	1/1	7.4	0.87	0.76	2.48	0.47	0.33

TABLE VIII

Summary of Terminal Average Body and Organ Weights of Male Rats Receiving Repeated Exposures to Vinyl Chloride 5 Days per Week

Concentration in ppm	Months on Exposure	Duration of Daily Exposure, hours	Ratio of Survival	Final Average Body Weight, g	Organ Weights, g/100 g Body Weight				
					Lung	Heart	Liver	Kidney	Testes
Unexposed control	0	0	8/8	319	0.50	0.33	2.67	0.75	0.18
Air exposed control	6	7	12/12	317	0.51	0.32	2.41	0.70	0.16
50	6	7	12/12	339	0.50	0.31	2.49	0.68	0.17
50	6	4	8/10	318	0.51	0.33	2.59	0.69	0.16
50	6	2	6/10	322	0.49	0.32	2.57	0.70	0.16
50	6	1	9/10	318	0.47	0.32	2.51	0.73	0.17
Unexposed control	0	0	8/8	336	0.53	0.32	3.37	0.72	0.16
Air exposed control	6	7	12/12	370	0.53	0.30	2.42	0.64	0.11
50*	6	7	12/12	360	0.49	0.33	2.51	0.68	0.18

* Fastured for 6 weeks after exposures ceased.

Thirdly, it is doubtful whether it would be possible for humans to be exposed continuously at 500 ppm in production plants since this high a general concentration would indicate severe leaks in the equipment and extremely high concentrations

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TABLE IX
Summary of Average Hematological Values for Dogs Receiving Repeated 7-Hour
Exposures to Vinyl Chloride

Dog #	Sex	Concentration in ppm	Months on Experiment	No. of Animals	Hemo- globin g/100 g	Hema- to- crit	WBC X ¹⁰ ⁹ /mm ³	Neu- tro- phils	Lym- pho- cytes	Mon- ocytes	Eosino- phils
Dog #152.....	M	Unexposed control	0	1	11	43	12.9	54	41	1	4
Dog #152.....	M	Unexposed control	3	1	15	52	10.4	42	50	4	4
Dog #152.....	M	Unexposed control	6	1	15.5	51	15.4	52	40	5	3
Dog #156.....	M	Air exposed control	0	1	—	—	—	—	—	—	—
Dog #156.....	M	Air exposed control	3	1	14.5	50	13.4	45	50	2	3
Dog #156.....	M	Air exposed control	6	1	14	50	11.1	55	28	5	12
Dog #156.....	M	Air exposed control	0	1	10.5	39	13.6	48	48	2	2
Dog #162.....	M	50	0	1	14.5	51	12.7	48	48	3	1
Dog #162.....	M	50	3	1	15.5	53	11.2	43	50	4	3
Dog #162.....	M	50	6	1	15.5	53	11.2	43	50	3	9
Dog #151.....	F	Unexposed control	0	1	11	48	13.3	59	30	3	0
Dog #151.....	F	Unexposed control	3	1	14.5	52	23.0	67	25	0	8
Dog #151.....	F	Unexposed control	6	1	15.5	58	15.5	46	42	3	9
Dog #155.....	F	Air exposed control	0	1	13	46	14	61	37	0	2
Dog #155.....	F	Air exposed control	3	1	16	59	18.2	60	33	5	2
Dog #155.....	F	Air exposed control	6	1	15.5	56	17.0	53	31	5	11
Dog #161.....	F	50	0	1	10.5	35	15.3	55	40	5	0
Dog #161.....	F	50	3	1	15.5	58	13.4	56	42	2	0
Dog #161.....	F	50	6	1	15.5	56	12.3	43	43	3	11

adjacent to the leaks. The flammability of vinyl chloride precludes such severe leaks if this hazard of vinyl chloride is to be controlled.

Industrial Hygiene Standard

The data presented indicate that little likelihood of injury would be expected if repeated daily 7- to 8-hour exposures are limited to 100 ppm or less.

Although the level of 100 ppm may not appear to offer any margin of safety since rats exposed repeatedly to this level were very slightly affected, the vast amount of human experience that is available while operating under an MAC of 500 ppm indicates that injury is not likely. Likewise the effects of even rather severe overexposure are not serious, hence this level seems reasonable.

A time-weighted average for all exposure should probably not exceed 50 ppm.

Summary

Vinyl chloride ($\text{CH}_2=\text{CHCl}$) is a monomer used in very large quantities in the production of plastics. Repeated exposures of laboratory animals to several concentrations of vinyl chloride in air were conducted to determine the chronic toxicity of this material towards animals in order to assess the hazard to humans. Vinyl chloride was found to have a slight capacity to cause liver and kidney injury on repeated exposures. Male and female rats showed micropathological changes after repeated daily 7-hour exposures at

500 ppm for 4.5 months. Repeated 7-hour exposures at 200 ppm for six months resulted in micropathological changes in the livers of rabbits and statistically significant increases in the average weight of the livers of male and female rats but no detectable changes in dogs and guinea pigs. Repeated 7-hour exposures at 100 ppm resulted in slight increases in the average weight of rat livers, the other species were not affected. All species studied tolerated repeated daily 7-hour exposures to 50 ppm for six months with no detectable injury.

Repeated daily 1-hour exposures at 200 and 100 ppm of vinyl chloride were without effect, longer exposures caused a slight increase in liver weight.

The standard for evaluating regular daily 7- to 8-hour exposures may be defined as the concentration below which practically all analytical results must fall. The value of 100 ppm is suggested as this standard for vinyl chloride, with a time-weighted average for all exposures not to exceed 50 ppm.

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Repeated 7-Hour

Neutrophils	Lymphocytes	Monocytes	Eosinophils
54	41	1	4
42	50	4	4
52	40	5	3
—	—	—	—
45	50	2	3
55	28	5	12
48	48	2	2
48	48	3	1
43	50	4	3
59	30	3	9
67	25	0	8
46	42	3	9
61	37	0	2
60	33	5	2
53	31	5	11
55	40	5	0
56	42	2	0
43	43	3	11

months. Repeated 7-hour exposures for six months resulted in changes in the livers of rabbits. Significant increases in the livers of male and female dogs in the 7-hour exposures at 100 ppm. Increases in the average number of neutrophils were not observed in the 7-hour exposures. Repeated exposures to 50 ppm for six months resulted in changes in the livers of rabbits.

Exposures at 200 and 400 ppm were without effect, but a slight increase in liver weight was observed at 200 ppm.

Evaluating regular daily 7-hour exposures as the concentration of all analytical results at 100 ppm is suggested for vinyl chloride, with a range for all exposures not to

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INDUSTRIAL HYGIENE CERTIFICATION

THE AMERICAN BOARD OF INDUSTRIAL HYGIENE in those who are interested in applying for voluntary certification of industrial hygiene. Certificates will be issued on the basis of those who have eight years of practice, following a suitable bacterial examination will be required of those who have fifteen years of practice to the field, following acceptable education. Certain contributions to the field, following acceptable education. Certain the field will be certified when one-fourth of their time has consisted to application of their specialty to industrial hygiene. A broad certification program, and giving more details of eligibility will be direct inquiries to Henry F. Smyth, Jr., Secretary-Treasurer, Industrial Hygiene, 4400 Fifth Avenue, Pittsburgh 13, Pennsylvania.

The Board was chartered as a non-profit corporation under the laws of Pennsylvania in September 1960. The incorporators were the American Conference of Governmental Industrial Hygienists and Industrial Hygiene Association from among their respective members: John C. Soot, Chairman; Lester V. Cralley, Ph.D., Vice-Chairman; J. R., Ph.D., Secretary-Treasurer; and Edgar C. Barnes, William F. Coleman, Harvey B. Elkins, Ph.D., Louis F. Garber, Thomas J. Kenneth M. Morse, James H. Sterner, M.D., and Charles D. Ya

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s of the Common- e selected by the he American In- rships. They are Henry F. Smyth, Bradley, Allan L. Mancuso, M.D.,

INDUSTRIAL HYGIENE ENGINEERING AND CHEMISTRY

THE DIVISION OF OCCUPATIONAL HEALTH, Public Health Service, will conduct two simultaneous courses January 8-19, 1962, at the Cincinnati Health Research and Training Facility, Cincinnati, Ohio. "Industrial Hygiene Engineering" is designed for industrial hygienists and engineers in the field of occupational health who have a degree in engineering or in the physical sciences. "Industrial Hygiene Chemistry" is designed for chemists in this field who have a degree in chemical engineering, or in the physical sciences with a minor in chemical engineering. During the first week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the second week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the third week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fourth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the tenth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eleventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twelfth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirteenth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fourteenth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifteenth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixteenth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventeenth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighteenth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the nineteenth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twentieth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twenty-first week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twenty-second week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twenty-third week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twenty-fourth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twenty-fifth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twenty-sixth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twenty-seventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twenty-eighth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the twenty-ninth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirtieth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirty-first week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirty-second week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirty-third week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirty-fourth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirty-fifth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirty-sixth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirty-seventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirty-eighth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the thirty-ninth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fortieth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the forty-first week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the forty-second week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the forty-third week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the forty-fourth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the forty-fifth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the forty-sixth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the forty-seventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the forty-eighth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the forty-ninth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fiftieth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifty-first week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifty-second week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifty-third week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifty-fourth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifty-fifth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifty-sixth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifty-seventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifty-eighth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the fifty-ninth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixtieth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixty-first week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixty-second week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixty-third week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixty-fourth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixty-fifth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixty-sixth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixty-seventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixty-eighth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the sixty-ninth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventieth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventy-first week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventy-second week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventy-third week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventy-fourth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventy-fifth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventy-sixth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventy-seventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventy-eighth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the seventy-ninth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eightieth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighty-first week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighty-second week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighty-third week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighty-fourth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighty-fifth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighty-sixth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighty-seventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighty-eighth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the eighty-ninth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninetieth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninety-first week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninety-second week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninety-third week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninety-fourth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninety-fifth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninety-sixth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninety-seventh week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninety-eighth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the ninety-ninth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health. During the one hundredth week, trainees for both courses meet together for instruction in industrial hygiene and environmental health.

It should be noted that applications for the course or request for information should be addressed to the Chief, Training Program, Robert A. Engineering Center, 4676 Columbia Parkway, Cincinnati 26, Ohio or to a DHEW Regional Office Director.

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Основным сырьем при изготовлении хлорвиниловых пластических масс является хлорвиниловая смола, которая получается из винилхлорида — бесцветного газа, конденсирующегося при 12—14° в жидкость с удельным весом 0,86—0,90. Винилхлорид в свою очередь получается при взаимодействии газообразных продуктов разложения паров нефти (винила) и газообразного хлора. В дальнейшем винилхлорид конденсируется с применением дихлорэтана и подвергается полимеризации в автоклавах. Таким образом, полихлорвиниловая смола является продуктом полимеризации винилхлорида. Благодаря стойкости к действию щелочей, разбавленных кислот, спирта, бензина и смазочного масла полихлорвиниловая смола применяется для производства кислотоупорных прокладок, тканей для спецодежды, заменителей кожи и др.

При изготовлении различных изделий полихлорвиниловая смола пластифицируется сложными эфирами фталиевой кислоты (дибутилфталат), фосфорной кислоты (трикрезилфосфат), смесью хлоридифенилов (совол) или хлорированных нафталинов (головакс). Выбор пластификатора зависит от назначения изготавливаемой пластической массы.

Винилхлорид, согласно литературным данным, относится к слабым наркотикам. Наркотические концентрации его для человека находятся в пределах от 75 до 250 мг/л. Согласно данным Петропавловского (1942), летучие вещества, выделяющиеся при нагреве хлорвиниловой смолы и пластикаторов до 130—165°, вызывают у животных в условиях однократной заправки в зависимости от концентрации следующие явления: раздражение слизистых оболочек, нарушение дыхания, поражение центральной нервной системы, расстройство координации движений и гибель при явлениях асфигмических судорог.

Совол и головакс, вводимые в хлорвиниловые пластикаты в качестве мягчителей, обладают общим и местным токсическим действием. По общему воздействию на организм совол и головакс близки к хлорзамещенным углеводородам жирного ряда. В основном они оказывают действие на паренхиматозные органы. Тяжелое отравление протекает по типу острой или подострой желтой атрофии печени. В литературе описаны случаи смертельного исхода при отравлении на производстве хлорированными нафталинами (головаксом) и дифенилами (соволом). Местное действие совол и головакс выражается в поражениях кожи типа акнеформных и эритематозно-везикулярных дерматитов.

Условия труда при работе с хлорвиниловыми пластическими массами изучались нами в заводских условиях в цехе хлорвиниловых пластикаторов при применении хлорвинилового пластиката в качестве заменителя кожи и при применении его для изоляции электропроводов.

Производство хлорированных пластикаторов. В цехе хлорированных пластикаторов хлорвиниловая смола (белая порошкообразная масса) смешивалась с дибутилфталатом или с соволом в смесительных аппаратах при температуре 80—100°. Затем композицию передавали на вальцы, поверхность которых нагревалась до 130—160°.

В конце вальцевания хлорвиниловое полотно на вальцах разрезалось на узкие ленты, которые в дальнейшем сортировали и упаковывали в ящики.

При гигиеническом обследовании цеха хлорвиниловых пластикаторов производились исследования воздуха рабочих помещений на содержание в нем паров совол, хлорорганических соединений и паров соляной кислоты. Полученные при этом результаты приведены в табл. I.

В воздухе рабочих помещений цеха хлорвиниловых пластикаторов, как это видно из данных, приведенных в табл. I, концентрация хлорорганических соединений колебалась от 0,005 до 0,27 мг/л. В подготовительном отделении имело место выделение паров совол при нагреве, слезе и транспортировке его. Концентрации хлорорганических соединений, найденные в этом помещении (0,005 мг/л), с большой вероятностью могут быть отнесены главным образом за счет совол. У смесителей хлорорганические соединения были найдены в количестве от 0,01 до 0,08 мг/л, а у рабочего места вальцовщика — от 0,2 до 0,97 мг/л.

Таблица 1. Содержание хлорорганических соединений, совола и хлористого водорода в воздухе цеха хлорвиниловых пластикатов

Название отделения	Место взятия пробы	Концентрация в мг/л	
		хлорорганические соединения, в том числе совола ¹	хлористый водород
Подготовительное	В шкафу для подогрева совола	0,005	Не обнаружен
Смешения и вальцовки То же	В середине помещения на уровне дыхания рабочего у смесителя во время смешения компонентов . .	0,003	„
	у смесителя во время выгрузки композиций . . .	0,06—0,08	0,025
	На рабочем месте вальцовщика	0,01—0,08	Не обнаружен
	У стола для остывающего пластика	0,2—0,97	0,002
Сверловочное	У рабочего места сверловщика	0,27	Не обнаружен
		0,005	„

В сортировочном отделении концентрации хлорорганических соединений обычно не превышали 0,05 мг/л. Хлористый водород мы находили лишь в отдельных пробах в количествах, не превышающих тысячных долей миллиграмма.

В связи с применением совола серьезную опасность представляет наличие условий для загрязнения кожи ингридиентами, входящими в состав композиции. На коже рук рабочих в конце рабочего дня мы находили хлорорганические соединения.

Для борьбы с загрязнением воздуха цех оборудован приточной и вытяжной вентиляцией с механическим побуждением. Вытяжная вентиляция устроена по принципу местной вытяжки от зонтов, установленных над основными источниками выделений газа.

Эффективность вентиляции, как показала санитарно-техническое обследование, была недостаточной вследствие того, что подача воздуха производилась со скоростью только 7—8 м/сек на уровне плеч рабочего. Неприятные ощущения, вызываемые подобным способом подачи воздуха, наряду с раздуванием и охлаждением композиции

на вальцах, вынуждали рабочих перекрывать шибера душающих патрубков и даже при этом струи воздуха, вырывающиеся через различные неплотности, сбивали работу вытяжных устройств, отклоняя конвекционные потоки, поднимающиеся от вальцов. Вместе с тем за-

¹ В связи с недостаточной разработанностью метода раздельного определения хлорорганических соединений произволилось их суммарное определение.

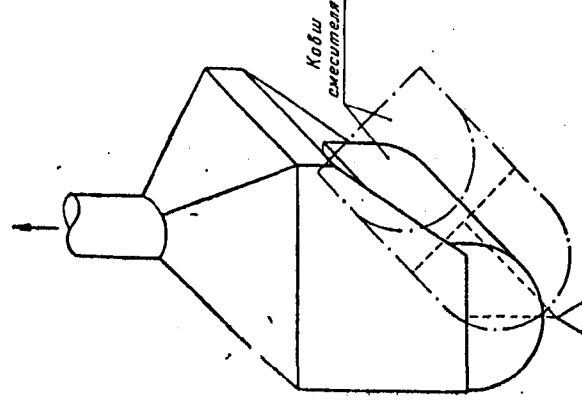


Схема шагового укрытия смеси

крытые приточной тяжкой и влекл воздуха из сосе

Существенно точная эффектн для готовой прк тивным оказал смесителями в е При производнт ностью локализ

Применение в за мен основном сыре чаеся при по: Температура ра катора здесь : стеарат кальция

Композиция для пропитанного днб: композицию уклад ном прессформу. е несколько минут, а

Композиция д: полихлорвиниловой соэреэая на 12 ч при температуре 1 вальцевания хлорж ратуре 160°.

Анализы воз хлорвиниловой

Таблица 2.

Место взятия пробы	
У вальцов . .	
У прессов . .	
В помещении сварочной	

Хлорорганич обработке хлор молекулы хлорнх соединений виниловой обуб 0,14 до 1,2 мг/л в концентрации) о до 0,022 мг/л.

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крытие приточных патрубков нарушало баланс между притоком и вытяжкой и влекло за собой неорганизованное поступление загрязненного воздуха из соседних помещений.

Существенным дефектом вентиляционной системы являлась недостаточная эффективность малометрических зонтов, установленных над столами для готовой продукции и над смесителями. Значительно более эффективными оказались осуществленные по нашему проекту укрытия над смесителями в виде шатров с открытой передней стенкой (см. рисунок). При производительности в 1 000 м³/час эти укрытия практически полностью локализовали выделения вредных веществ.

Применение хлорвиниловых пластикатов в качестве заменителей кожи. В производстве хлорвиниловой обуви основным сырьем является полихлорвиниловая смола, которая получается при полимеризации хлорвинила водноэмульсионным способом. Температура разложения этой смолы 130—150°. В качестве пластификатора здесь применяется дибутилфталат, а стабилизатором служит стеарат кальция.

Композиция для подошв готовится путем смешения текстильного лоскута, пропитанного дибутилфталатом, и хлорвиниловой смолы в смесителе. Полученную композицию укладывают на нагретую до 170—200° и протертую после этого вазелином прессформу. Заполненную таким образом прессформу закрывают, выдерживают несколько минут, а затем ставят на непродолжительное время под горячий пресс.

Композиция для верха обуви получается при смешении определенных количеств полихлорвиниловой смолы и дибутилфталата в мешателе. Массу оставляют здесь для созревания на 12 часов при температуре 70°. После созревания композицию вальцуют при температуре 120—140° с прибавлением стеарата кальция. Полученный после вальцевания хлорвиниловый пластикат перекладывают текстилем и прессуют при температуре 160°.

Анализы воздуха на содержание в нем вредных газов и паров в цехе хлорвиниловой обуви сведены нами в табл. 2.

Таблица 2. Концентрация газов и паров в воздухе рабочих помещений производства хлорвиниловой обуви в мг/л

Место взятия проб	Хлорорганические соединения				HCl				C (суммарно)				CO			
	максимальная	максимальная	средняя	минимальная	максимальная	максимальная	средняя	минимальная	максимальная	максимальная	средняя	минимальная	максимальная	максимальная	средняя	минимальная
У вальцов . .	0,14	0,48	0,35	0,028	0,036	0,028	0,032	0,07	0,35	0,21	—	—	—	—	0,022	—
У прессов . .	0,26	1,13	0,65	0,008	0,036	0,008	0,055	0,21	2,61	1,23	—	—	—	—	0,019	—
В помещении сварочной	0,53	1,2	0,92	0,03	0,17	0,23	0,10	0,68	0,23	—	—	—	—	—	0,01	—

Хлорорганические соединения, которые выделяются при термической обработке хлорвиниловой смолы, представляют собой отщепившиеся молекулы хлорвинила в различных количественных сочетаниях их цепных соединений. В воздухе рабочих помещений на производстве хлорвиниловой обуви эти соединения были найдены нами в количестве от 0,14 до 1,2 мг/л. Здесь были обнаружены также пары соляной кислоты в концентрации от 0,008 до 0,17 мг/л, углеводороды (в суммарном их определении) от 0,21 до 2,61 мг/л и окись углерода в пределах от 0,01 до 0,022 мг/л.

Загрязнение воздушной среды хлорорганическими соединениями парами соляной кислоты возрастало по мере повышения температуры нагрева при различных производственных операциях.

Вентиляция рабочих помещений осуществлялась с помощью нескольких вытяжных установок, первоначально предназначенных для местной вытяжки. Несмотря на сравнительно высокую кратность воздухообмена (10—20), эффективность вентиляции оказалась недостаточной. Основным дефектом ее было отсутствие местного удаления выделений, образующихся при обработке композиции на прессах, вальцах и при сварке обуви.

Установка зонтов над сварочными верстаками, обычно применявшихся в этом производстве, не эффективна и не может рассматриваться как местное вытяжное устройство, так как при сварке хлорвиниловой обуви газы выделяются на уровне колен сидящего сварщика в непосредственной близости от его лица. Наиболее радикальным устройством в данном случае является вертикальный вытяжной канал против каждого из рабочих мест. Удаление воздуха может производиться через отверстие в столе верстака, верхняя часть которого при этом должна быть использована как плоский вытяжной канал.

Удовлетворительный эффект наблюдался и в случае вытяжки через горизонтальные короба, располагаемые на поверхности верстака. Скорость отсоса через вытяжные отверстия должна быть не менее 6 м/сек.

Применение хлорвиниловых пластикаторов для изоляции электропроводов. На заводе, где хлорвиниловый пластик используется в качестве изоляционного материала для электропроводов, покрытие провода хлорвиниловой оболочкой производится на шприцмашинах.

Во время сушки хлорвинилового пластика в сушильных шкафах, а также при плавке его и при выходе провода в горячей оболочке из головки шприцмашины в воздух рабочего помещения выделяются следующие летучие вещества (табл. 3).

Таблица 3

Место взятия проб	Хлорорганические соединения	HCl		C (суммарно)		CO						
		концентрация в мг/л										
		Максимальная	Минимальная	Максимальная	Минимальная		Максимальная	Минимальная				
У шприцмашины в зоне дыхания рабочего . . . Над проводом, покрытым горячей пластмассой . . В середине помещения Над сушильным шкафом	0,003	0,1	0,001	0,06	0,3	0,6	—	—	0,017	0,05	0,02	—
	0,14	0,65	Не обнаружена	—	—	—	—	—	—	—	—	—
	0,004	0,088	То же	—	—	—	0,11	—	—	—	—	—
	0,10	0,14	—	0,016	0,08	0,24	—	—	—	—	—	—

В составе летучих веществ, выделяющихся при нагреве хлорвинилового пластика до 120—140°, содержатся хлорорганические соединения в количестве от 0,003 до 0,6 мг/л, пары свободной соляной кислоты от

0,001 до 0,06 от 0,01 до 0,01

Вентиляция новленных на шины. Газы, в и при остыва шприцмашины, в нем значите

Результ Медицинскому группы рабочих пластикаторов б При рино-лари чих в верхних наружены бел пыли со слизы удалить с пове лена гиперемии ния хроническ со стороны орг гипотония (90- ного, артерио- шечного тракта нического гаси изменения со (гелатитов, соче Со стороны

гемоглобина: у крови и РОЭ б

Обращает н: заболевания ко: Поражения ко: волом, но и у т в том же поме ствием совола

На химичес: имущественно г боты в данной и ных ого-ларинг стых оболочек т только на огран ны у 5 человек, явления плеврит

Со стороны сосудистая гипо женная гиперто рение его грани менее чем в 2 , дали основание дистрофию.

В обследова При этом у нек и желудка. Ни констатировать и желчных пигмен обнаружено неб ствительности п

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0,001 до 0,06 мг/л, углеводороды от 0,08 до 0,6 мг/л и окись углерода от 0,01 до 0,05 мг/л.

Вентиляция цеха осуществлялась с помощью вытяжных зонтов, установленных на телескопическом опуске над каждой головкой шприцмашины. Газы, выделявшиеся при сушке пластика в сушильных шкафах и при остывании хлорвинилового провода после выхода из головки шприцмашины, поступали в воздух рабочего помещения, поддерживая в нем значительные концентрации токсических веществ.

Результаты медицинского обследования рабочих. Медицинскому освидетельствованию были подвергнуты выборочные группы рабочих обследованных предприятий. В цехе хлорвиниловых пластикатов было осмотрено 48 человек, преимущественно вальцовщики. При рино-ларингологическом исследовании у значительной части рабочих в верхних дыхательных путях (в носу, гортани и трахее) были обнаружены белого цвета пленки, образовавшиеся в результате смещения пыли со слизистой. Эти пленки были плотны наощупь, и их трудно было удалить с поверхности слизистой оболочки. У 17 человек была установлена гиперемия слизистой оболочки верхних дыхательных путей. Явления хронического бронхита были найдены у 13 человек. Из изменений со стороны органов кровообращения должна быть отмечена сосудистая гипотония (90—100/45—50) у 13 человек, умеренная гипертония у одного, артерио-кардиосклероз — у 5 человек. Состояние желудочно-кишечного тракта характеризовалось у 10 обследованных явлениями хронического гастрита и у 2 — хронического колита. У 15 человек имелись изменения со стороны печени в форме более или менее выраженных гепатитов, сочетавшихся у 2 обследованных с явлениями холецистита.

Со стороны крови у 29 обследованных имели место низкие цифры гемоглобина: у одного — 46%, у 22—50—55%, у 6—56—60%. Формула крови и РОЭ без особенностей.

Обращает на себя внимание, что у обследованной группы отмечались заболевания кожи (акне, комедоны), в особенности на лице, шее, груди. Поражения кожи наблюдались не только у лиц, соприкасавшихся с соволом, но и у тех, кто прямого контакта с ним не имел, но работал в том же помещении. Видимо, заболевания кожи вызывались воздействием совола при сорбции его паров.

На химическом заводе было обследовано 25 женщин-работниц, преимущественно прессовщицы и сварщицы, с преобладающим стажем работы в данной профессии от 3 до 6 лет. У некоторых из них, осмотренных ото-ларингологом, оказалась различной степени гиперемия слизистых оболочек верхних дыхательных путей на всем их протяжении или только на ограниченных участках. Хронические бронхиты были выявлены у 5 человек, умеренно выраженная эмфизема легких у 3, остаточные явления плеврита у 2 человек.

Со стороны сердечно-сосудистой системы у 14 человек установлена сосудистая гипотония (R/R 90—110/55—65) и у 4 человек умеренно выраженная гипертония (R/R 140—165/80—85). Глухие тоны сердца, расширение его границ и данные электрокардиограммы: низкий зубец T не менее чем в 2 основных отведениях и закругленность интервала S—T, дали основание в нескольких случаях диагностировать миокардиодистрофию.

В обследованной группе у 6 человек был диагностирован гепатит. При этом у некоторых лиц отмечено одновременно заболевание печени и желудка. Ни в одном из случаев заболевания печени мы не могли констатировать в анамнезе или при осмотре признаков желтухи. В моче желчных пигментов также не было. У значительного числа осматривенных обнаружено небольшое увеличение печени без характерных жалоб и чувствительности при пальпации.

Анализируя результаты медицинского обследования, необходимо отметить, что у известной части рабочих были обнаружены изменения со стороны верхних и более глубоких дыхательных путей. Эти наблюдения подтверждаются и литературными данными.

Обращает на себя внимание высокий процент кожных поражений, что этиологически, несомненно, связано с воздействием совола.

Серьезного внимания заслуживает наличие гепатитов у некоторых рабочих, сопрягающихся с хлорвиниловой смолой и хлорвиниловыми пластиками. Необходимо поэтому дальнейшие динамические наблюдения за состоянием здоровья рабочих, сопрягающихся с хлорвиниловой смолой и ее пластикатами, чтобы предупредить развитие серьезных поражений печени.

Выводы

1. В составе летучих веществ, образующихся при термической обработке хлорвиниловой смолы и пластикатов на производстве, входят хлорорганические соединения, углеводороды, пары свободной соляной кислоты и окись углерода.

2. Воздействие этих летучих веществ проявляется в раздражении слизистой оболочки преимущественно верхних дыхательных путей; длительное влияние высокодисперсной пыли и паров хлорвиниловой смолы вызывает, кроме того, сухость и атрофию слизистой оболочки верхних дыхательных путей, а также способствует развитию хронических бронхитов.

3. Значительное число гепатитов среди обследованных рабочих, а также изменения в верхних дыхательных путях ставят вопрос о необходимости дальнейших динамических наблюдений за состоянием здоровья рабочих и проведения ряда лечебно-профилактических мероприятий, в частности, организацию для рабочих тепловлажной (щелочной) ингаляции лекарственных веществ.

4. При применении в качестве пластификатора хлорвинилов совола у большинства рабочих наблюдались кожные поражения типа акне с преимущественной локализацией на лице и шее.

5. В целях профилактики кожных заболеваний при работе с соволом необходим следующий комплекс оздоровительных мероприятий: 1) обеспечение рабочих бельем и спецодеждой; 2) устройство душевой типа санитарного пропускника с отдельными раздевальными для домашней и рабочей одежды. При санитарном пропуске должна быть оборудована паровая камера для пропарки белья и спецодежды при температуре 110—120° в течение часа. В цехе должны быть оборудованы умывальники с подведенной горячей и холодной водой.

6. Для оздоровления воздушной среды необходимо по возможности уменьшить степень разложения хлорвиниловой смолы при термической ее обработке. Этого можно достигнуть путем стандартизации ее термостойкости, а также путем регулирования температуры нагрева вальцов, прессов и другого оборудования, не допуская превышения ее сверх 120—130°.

Чрезвычайно важное значение имеет рационально оборудованная система вентиляции, рассчитанная на максимальное укрытие источников выделения газов, паров и пыли.

В связи с тем, что на некоторых производствах наблюдались случаи воспламенения конденсата, образующегося в вытяжных вентиляционных трубах при соприкосновении с открытым пламенем, вытяжные устройства должны проектироваться с соблюдением условий, принятых строительным законодательством для устройств, предназначенных к перемещению огнеопасных газовых смесей.

О способе от Из каф

Как известно, получившие в видимом пожатии специфическую странность вглубина» рыбы с пия свободных неприя. Одной «ржавления», жжирах. Можно, ния сложных рания белковой-органического и

Обычно степ-органолептическим дополнить нами предлагае: стадии ржавлен

Первое, что вещества «ржав-танных растворящим оказался (кислоты и в то

обладает и тем д-не требует как-Второй проб-рыбы, была пр-рыбы. Альдегид-прогоркании жиления этот прод

Методика по-ривалась, и про-органолептичес-ной, с диффузисных покровов (костей.

От каждой р-разных мест и г-резали ножом н

Мелкую рыбу-скали целиком: части вырезали а для получения участки мяса, п-

Для экстраг-шали в колбы и этилового алког-пробкой. Рыбы-встряхивании. Г-складчатый фил-каучуковой или

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Donc le principal danger résulte de l'introduction de la main dans la zone dangereuse.

Sur les 42 accidents arrivés plus haut :

22 sont arrivés au moment de la fermeture du moule *mais* 10 presses n'avaient pas de sécurité, 5 insuffisamment protégées, 1 protection mal conçue, 4 mauvais fonctionnement de la sécurité, 2 ruptures du vérin de commande, 11 se sont produits à cause d'un autre organe mécanique, 9 furent causés par le montage du moule.

CONCLUSION

L'industrie des plastiques requiert des mesures de précaution qui lui sont propres, dont l'élaboration n'est nullement plus compliquée que celles qui concernent les autres industries, mais qui sont simplement adaptées au caractère particulier découlant du mode d'obtention de ces matériaux.

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2^e PARTIE : TOXICOLOGIE

Aspects toxicologiques de l'industrie des matières plastiques.

Par M. Louis TRUFFERT.

Le problème toxicologique posé par l'industrie des matières plastiques est extrêmement complexe, aussi ne saurait-on le traiter complètement dans cet exposé.

On est amené à rencontrer, dans cette industrie, un nombre considérable de toxiques dont les actions peuvent se contrarier ou s'ajouter et l'on voit apparaître chaque jour de nouveaux produits dont on ignore la nocivité.

Au moment de la fabrication, on pense surtout aux dangers présentés par les matières premières utilisées, mais il faut également tenir compte des produits susceptibles de se former au cours de la réaction. Lors de l'utilisation, le dégagement de nouvelles substances peut être à craindre, du fait par exemple, d'une décomposition par la chaleur, nécessaire au moulage, ou de l'emploi de solvants pour une dissolution.

Le travail mécanique de la matière plastique peut, lui-même, être une source de poussières agressives.

Enfin, des traces d'impuretés peuvent avoir une action physiologique et, alors même qu'il n'y a aucun risque d'intoxication, des accidents allergiques peuvent se manifester.

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Il importe donc d'être circonspect et d'admettre que l'on peut fort bien observer des troubles très différents de ceux auxquels l'étude toxicologique des substances mises en œuvre ou produites par les réactions, permettait de penser.

C'est ainsi par exemple que Devignevielle et Mlle Flucher [1], ont observé que le travail des résines polyvinylques comportait une toxicité faible, mais certaine, ne s'expliquant pas par les données connues. Cette toxicité se manifestait par des troubles gastro-intestinaux, des dermatites de type eczémateux et des troubles sanguins : altérations morphologiques des polynucléaires neutrophiles, éosinophilie et mononucléose.

Cette réserve étant faite, nous passerons rapidement en revue les produits finis et les matières premières utilisées, mais nous ne saurions les étudier tous et nous citerons seulement quelques exemples présentant actuellement de l'intérêt.

RÉSINES

Si les résines peuvent être considérées comme généralement inertes lorsqu'elles sont pures, il arrive cependant qu'on leur impute divers accidents du fait de la présence d'un excès d'un de leurs constituants, ou de la persistance d'un peu de monomère.

La nocivité des constituants des résines polycondensées, ou des monomères des hauts polymères, se manifeste d'ailleurs dès la fabrication et peut réapparaître lors de la décomposition du produit (sous l'influence de la chaleur par exemple).

C'est ainsi que le *polystyrène* est obtenu à partir du styrène, dont la toxicité est notable (voir l'article de P. Dervillée, H. Desoille et L. Truffert dans l'Encyclopédie Médico-Chirurgicale).

On trouvera des renseignements à ce sujet dans un mémoire de R. Lefaux [2] qui traite également des produits suivantes : chloroprène, méthacrylate de méthyle, α -chloracrylate de méthyle, nitrile acrylique, chlorure de vinyle, butadiène, hexaméthylène-diamine, di-isocyanates, polytétrafluoro-éthylène, solvants, plastifiants, etc...

La nocivité des poussières de *polyamides* a été mise en évidence par Massmann et Pilgrim [3] chez le Rat (pneumonie granuleuse par administration intratrachéale, mais rien par voie orale).

Mechels [4] a étudié les effets physiologiques du perlon et son influence sur la flore de la peau humaine.

D'après Schultz et Hermann [5], les eczémas provoqués par les bas de perlon seraient dus à des colorants amino-azoïques ou anthraquinoniques.

Dans la fabrication des *polyuréthanes*, on utilise beaucoup les di-isocyanates, en particulier le di-isocyanate de toluylène ou Desmodur T, dont la toxicité a été signalée en France par Fuchs et Valade [6]. D'autres cas d'intoxication ont été signalés, en particulier par Swenson, Holmquist et Lundgren [7], par Johnstone [8], et par Hama [9]. Zapp [10] effectua une expérimentation animale qui montra le danger d'inhalations répétées pour des teneurs de 1 à 2 p.p.m., qui eurent des suites mortelles.

Pour les *résines époxy*, l'étude effectuée par Hine, Kodama, Anderson, Simonson et Wellington [11] a montré une toxicité relativement basse, mais des dermatites peuvent être causées par des impuretés (notamment des amines).

PRODUITS D'ADDITION. — Ils sont très divers et nous ne pourrions qu'en citer quelques-uns.

Catalyseurs. — Ce sont souvent des acides (sulfurique, chlorhydrique), mais aussi des peroxydes d'aldéhydes ou de cétones. Ceux-ci ont une action très irritante sur la peau et causent des dermatites. Rogers [12] a signalé le danger de combustion spontanée en présence de matières organiques, ce qui nécessite des mesures de prévention d'incendie.

Mallette et Haam [13] ont étudié divers *accélérateurs* dont la toxicité est très variable :

	Toxicité en g/kg
Dithiocarbamate de cyclopentaméthylène pyridinium (Pip-pip)	0,25
N-Isopropylbenzothiazolsulfonamide (Isocyclex)	0,25
Bis 2(benzothiazolylthio éthyl) urée (E.L.66)	6
N-Cyclohexyl-N-diéthylthiocarbonyl sulfonamide (Thiopentex)	1,2

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Il en est de même pour les *activateurs* étudiés également par ces auteurs :

	Toxicité en g/kg
Cyclohexylamine	0,2
Stéarate de cyclohexyl-ammonium	4
Formiate de cyclohexyl-ammonium	58

Tous irritent plus ou moins la peau et ils sont pour la plupart sensibilisants.

Parmi les *stabilisants*, on trouve à côté de sels inoffensifs (stéarates, ricinoléates, laurates, naphthénates de sodium, de lithium ou de calcium), des sels de zinc, de cadmium, d'étain et surtout de plomb dont la toxicité est bien connue.

On utilise également des composés organiques d'étain tétravalent tels que dibutyl-dilaurate d'étain, ou dibutyl-étain-monopropylèneglycol-maléate-2-hydroxypropoxyde) toxiques nerveux pour lesquels il est difficile de transporter de l'animal à l'homme en ce qui concerne les déterminations de la toxicité, ce qui incite à se montrer très prudent.

Les *antioxygènes* sont très nombreux et leur toxicité est également très variable (Mallette et Haam [13] donnent les chiffres suivants :

	Toxicité en g/kg
Diamylphénol	0,62
2-6 dibutyl (tertiaire) para-crésol (Ionol)	8
Phosphite de triphényl (E.F.E.D)	0,25
Sulfure de di-4 butyl (tertiaire) méta-crésol (Santowhite)	5
NN ₁ -di- β naphthyl-p-phénylènediamine (Agerite white)	4,5

Les *plastifiants* sont encore plus nombreux et extrêmement divers. Parmi eux, il faut noter dans les *esters phosphoriques*, le *phosphate de tri-orthocrésyle* qui est respectable d'un nombre considérable d'intoxications. Les *esters phthaliques*, sont beaucoup moins nocifs et l'on peut considérer les homologues supérieurs comme inoffensifs. J'ai administré jusqu'à 50 ml/kg de *phthalate d'heptyle* à des cobayes, sans parvenir à provoquer la mort.

L'emploi comme plastifiants de *caoutchoucs synthétiques* est très intéressant en raison de leur inocuité.

De même on peut utiliser des *esters époxy* qui sont en même temps des stabilisants du chlorure de polyvinyle et dont la toxicité paraît faible. Witnauer, Knight, Palm, Koos, Ault et Swern [14] ont étudié plus d'une vingtaine de ces produits.

Les *colorants* et les *charges* peuvent aussi présenter certains dangers. C'est ainsi que la question du pouvoir cancérogène du *Carbon-Black* s'est posée. Von Haam et Mallette [15] ont extrait du Carbon-Black des produits causant des cancers chez la souris, alors que d'autres auteurs affirmaient que cette matière était sans action sur les ouvriers qui l'employaient. Tout récemment Nau, Neal et Stenbridge [16] ont montré que non seulement le Carbon-Black n'avait pas d'action cancérogène, mais encore que, grâce à son pouvoir absorbant, il empêchait celle d'un corps aussi actif que le méthylcholanthrène.

La présence dans les *pigments* d'éléments toxiques tels que plomb, cadmium, zinc, manganèse, chrome (à l'état de chromate), mercure, baryum, sélénium, peut constituer une source de danger. Mais celle-ci n'est pas toujours réelle, car si le pigment est pur et très insoluble, sa toxicité peut être très faible, ainsi que j'ai pu le constater pour le sulfure et le sulfo-séléniure de cadmium. C'est l'excès de sel de cadmium soluble utilisé pour la préparation, qui paraît être la cause de la toxicité de ces pigments et cet excès étant éliminé, j'ai pu administrer à des animaux, par voie orale, de fortes doses de ces deux pigments (plus de 1 g/kg) sans suite fâcheuse.

En ce qui concerne les solvants utilisés pour dissoudre les matières plastiques, je dois dire que leur nombre augmente tous les jours et que l'usage de nouveaux produits se répand de plus en plus. C'est notamment le cas des *dérivés furanniques*. Le furfural est connu depuis longtemps ; c'est un irritant, dont la concentration maximum tolérable a été fixée à 5 ppm (soit environ 20 mg/m³). Le tétrahydrofurane paraît moins toxique (concentration maximum tolérable = 200 ppm ou 590 mg/m³). L'alcool tétrahydrofurfurylique est encore mal étudié et quant à l'alcool furfurylique, il joint à sa toxicité (concentration maximum tolérable = 50 ppm, soit 200 mg/m³) un danger d'explosion spontanée, en particulier dès qu'il se trouve en présence d'un acide, qui le résinifie (cette réaction permet d'ailleurs de fabriquer certaines matières plastiques).

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PRODUITS DE DÉCOMPOSITION DES MATIÈRES PLASTIQUES

Les matières plastiques peuvent se décomposer sous l'influence de divers facteurs : lumière, chaleur, catalyseurs.

La décomposition peut donner naissance aux constituants d'origine, dont on retrouve la toxicité. C'est le cas de certaines résines formoliques qui résistent mal à la chaleur. Comme exemple de décomposition catalytique on peut citer celui du bioxyde de titane, pigment blanc très utilisé et qui incorporé dans une résine polyamide, considéré, avec raison, comme très résistante, provoquait sa dépolymérisation, avec l'action combinée des rayons solaires.

Des *produits nouveaux* peuvent se produire sous l'action de la chaleur.

Dans le cas des *thiokols*, il suffit d'une élévation de température relativement faible pour provoquer un dégagement de composés organiques soufrés : mercaptans, thiéthers, etc... La toxicité de ceux-ci, que j'ai pu mettre en évidence [17], et qui a été étudiée, notamment par Ljunggren et Norberg [18] n'a généralement pas l'occasion de se manifester, mais leur odeur très désagréable, même à des concentrations extrêmement minimes, est des plus incommodes.

Pour le *chlorure de polyvinyle*, d'après Coleman et Thomas [19], à 300° C, on a environ 30 p. 100 du chlore transformé en acide chlorhydrique, mais il ne se forme que des traces négligeables de phosgène. Au cours de la combustion il y a, en même temps, dégagement de gaz carbonique et d'oxyde de carbone.

Le *celluloïd*, lorsqu'il brûle donne naissance à de l'oxyde de carbone et surtout à des vapeurs nitreuses qui ont fait de nombreuses victimes.

Il suffit de rappeler l'accident d'une clinique de Cleveland, en 1929, où 126 personnes furent mortellement intoxiquées par les gaz résultant de la combustion de films radiographiques.

L'évolution était caractéristique des gaz suffocants du type phosgène avec ses trois périodes : suffocation, rémission, œdème pulmonaire aigu.

Mac Nally [20] a signalé que beaucoup de personnes avaient après l'accident, conduit leur voiture jusqu'à leur domicile pour succomber dans les 36 heures après l'inhalation des gaz toxiques. Il est à noter qu'en dehors de cette affaire, Mac Nally déclare avoir observé 48 cas d'intoxications par vapeurs nitreuses dont 3 mortels.

Easton [21] a relaté un cas d'incendie aux États-Unis ayant fait 492 morts et où des dérivés nitrés provenant de décors en pyroxyline avaient pu donner à la fumée leur effet toxique.

Nous avons eu personnellement connaissance, avec M. Moureu, de deux accidents :

En 1947, l'inflammation de quelques kilogrammes de paillettes de celluloïd, servant à la fabrication de couronnes funéraires, provoqua en 2 à 3 minutes la mort de 22 femmes occupant un atelier d'ortoir.

Les secours furent immédiats, mais sans effet, bien qu'aucune des victimes ne portât des traces de brûlures susceptibles d'entraîner la mort. Nous avons pu établir qu'un gramme de ces paillettes dégageait environ 6 mg d'oxyde de carbone et 37 mg de vapeurs nitreuses. C'était donc à ces dernières, plus toxiques, que l'on devait attribuer la mort.

En 1949, un incendie dans un studio de montage de films cinématographiques faisait trois victimes. Le cadavre de l'une d'elle était retiré intact, sans trace de brûlure, montrant que la mort était due au torrent gazeux produit par la combustion des films nitrocellulosiques.

Avec M. Moureu, nous avons constaté que l'arrosage de ces films, en masses compactes, ne fait qu'aggraver le danger, la combustion faisant place, sous l'eau, à une décomposition thermique dégageant des quantités beaucoup plus considérables de vapeurs nitreuses.

Le *polytétrafluoroéthylène* résiste beaucoup mieux à la chaleur et se décompose à une température élevée, d'une manière qui est encore très discutée.

Il provoque chez l'homme une *fièvre des fumées de polymère*, analogue à la « fièvre des fondeurs », qui a été décrite par divers auteurs (Harris, Sherwood, etc...)

La dégradation thermique du polytétrafluoroéthylène a été étudiée (dans le vide) par Madorsky, Hart et Sedlak [22]. Ils trouvèrent que la pyrolyse entre 423° C et 513° C conduit toujours à 100 p. 100 de tétrafluoroéthylène monomère.

Atkinson et Trenwith [23] ont fait passer un courant de tétrafluoroéthylène gazeux à des températures variant entre 300 et 800° C. Au-dessous de 550° C : dimérisation en octofluorocyclobutane. Entre 600 et 700° on obtient l'hexafluoropropène et deux octofluorobutènes. A 800° C se produit la décomposition des octofluorobutènes avec formation de radicaux libres, puis combinaison pour donner de l'hexafluoroéthane et des molécules non saturées, hautement polymérisées.

L'effet de ces corps est encore mal connu, mais certains peuvent présenter une toxicité assez élevée.

D. R. Hagemeyer et W. Stubblebine [24] ont fait des expériences en chauffant du polytétrafluoroéthylène et ont trouvé que lorsque la concentration de l'atmosphère en halogène ne dépassait pas 10 p.p.m. il n'y avait pratiquement pas d'effet sur la souris. Il fallait atteindre 100 p.p.m. pour provoquer la mort de cet animal et même 1 000 p.p.m. si l'exposition était très courte (30 secondes).

H. H. Schrenk [25] a relaté les expériences suivantes :

Des quantités pesées de Téflon enrobant des fils de cuivre ont été chauffées rapidement dans un four à 800° (le téflon brûle au-dessus de 400° C). Les chats et les chiens, qui se sont montrés les plus résistants, survivaient quand 6,4 g de Téflon étaient consumés dans un courant d'air de 31,8 l/mn, mais mouraient quand 14,8 g étaient brûlés. Les morts, chez les rats et les souris, apparaissaient lorsque seulement 0,92 g de Téflon était consumé et il fallait attendre 3,7 g pour les cobayes.

Il considère que les produits de décomposition du Téflon entre 500° et 800° sont apparemment de même toxicité. Cependant les fumées produites aux températures n'excédant pas 375° ou 390° C sont beaucoup moins toxiques que celles formées aux hautes températures.

Les fumées formées au-dessous de 300° C ne causent la mort d'aucun des animaux.

TABLEAU

EXPÉRIMENTATION SUR L'ANIMAL PAR INHALATION DES PRODUITS
DE DÉCOMPOSITION THERMIQUE DU POLYTÉTRAFLUOROÉTHYLÈNE
UTILISÉ COMME REVÊTEMENT D'USTENSILES DE CUISINE

ANIMAUX		TENEUR EN TÉFLON EN MG/M ³	DURÉE DE CHAUFFAGE EN MN	DURÉE DE L'EXPOSITION EN MN	OBSERVATIONS
COBAYE	POIDS MOYEN EN G				
	452	179	6	30	Aucun symptôme.
	450	330	6	60	—
	587	525	6	60	Légère irritation.
	550	970	8	60	Rémission rapide.
	545	3 340	8	60	Irritation des muqueuses, larmoiement, suffocation.
LAPIN	3 355	515	8	60	Aucun symptôme.
	3 150	974	9	60	Légère irritation, Rémission rapide.
RAT	272	1 023	9	60	Légère irritation. Rémission rapide.
SOURIS	34	978	10	60	Légère irritation. Rémission rapide.
Un animal (sur 4) est mort dans les 36 heures suivantes. On a trouvé à l'autopsie une congestion pulmonaire presque totale et des lésions gastriques.					

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Les produits de décomposition ont provoqué une irritation des voies respiratoires et des dégénérescences du foie, des reins, et du cerveau des animaux. Les tissus des animaux survivants étaient normaux, sauf pour l'irritation résiduelle des poumons et une dégénérescence occasionnelle du foie. La spectrométrie infra rouge des produits de décomposition a montré la présence des bandes du perfluoro-isobutène et du difluorure d'oxygène.

Nous avons effectué des essais sur l'animal en chauffant rapidement à l'aide d'un épiradiateur des plaquettes d'aluminium revêtues de polytétrafluoroéthylène, jusqu'à commencement de fusion du métal (P.F. = 658° C). Les résultats, consignés dans le tableau I, ont montré que la ménagère utilisant des ustensiles culinaires de ce type, ne courrait aucun risque en les oubliant sur le feu, les concentrations atteintes étant très inférieures à la teneur dangereuse.

CONCLUSIONS

Cette étude rapide a montré combien étaient complexes les problèmes toxicologiques qui pouvaient se poser dans l'industrie des matières plastiques. Pour les résoudre, il est nécessaire d'avoir recours aux branches les plus diverses de la toxicologie. C'est pourquoi le médecin d'usine devra toujours garder son attention en éveil, afin d'éviter que des manifestations toxiques nouvelles ne passent inaperçues.

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DISCUSSION

M. G. W. BOWLEY (*Technical Service and Development Department Imperial Chemical Industries Limited Plastics Division*), a donné les réponses suivantes aux questions posées au cours de la discussion :

1° *Avez-vous eu connaissance d'un accident survenu à une personne ayant fumé une cigarette souillée par du Téflon ?*

Je tiens à préciser que mes observations concernent le « Fluon » (P.T.F.E.) fabriqué par I.C.I. en Angleterre et non le « Téflon » qui est, comme vous le savez, fabriqué par Du Pont de Nemours aux U.S.A. Toutefois, ces deux produits sont chimiquement les mêmes et il n'y a pas de raison de croire que leurs toxicités diffèrent en quoi que ce soit. Je propose par conséquent que le produit soit désigné par son nom chimique polytétrafluoroéthylène (P.T.F.E.) et que soit évité l'emploi des désignations commerciales.

Il est à peu près certain, mais non démontré, qu'au moins quelques-uns des cas de fièvre des fumées de polymères signalés dans les articles décrivant les effets toxiques des produits de décomposition du polyfluoroéthylène peuvent être attribués au fait de fumer du tabac souillé de P.T.F.E.

2° *Que pensez-vous de la publicité faite autour de cette question ?*

Beaucoup de publicité indésirable a été faite autour de la mort prétendue survenue aux U.S.A. après avoir fumé une cigarette souillée de P.T.F.E.

Ce renseignement a paru dans un mémoire publié par le *National Advisory Council for Aeronautics* (U.S.A.) en septembre 1956.

Des enquêtes très poussées du Ministère Britannique de la Ressource et de la I.C.I. ont abouti à la conclusion que de tels décès ne se sont pas produits.

Il est très troublant par conséquent de constater que cette rumeur non fondée persiste toujours et beaucoup de tort a été fait au développement du P.T.F.E. par suite de l'exagération de la toxicité que son emploi comporterait.

3° *Avez-vous constaté des troubles à la suite de l'utilisation du Téflon ?*

A ma connaissance, aucun usager au stade de la consommation n'a jamais éprouvé d'effets toxiques.

On ne peut pas en dire autant cependant des gens qui produisent le P.T.F.E. et un certain nombre de cas de fièvre des fumées de polymère ont été signalés, notamment au cours des débuts de la fabrication du produit.

Ces cas ont été provoqués habituellement par l'absence d'une ventilation adéquate pendant le procédé de frittage (chauffage à 380° C). L'expérience a montré que là où le P.T.F.E. est fabriqué dans des ateliers bien ventilés et qu'une ventilation locale aspirante est appliquée aux endroits où le produit est fritté (par exemple four ouvrant sur l'extérieur de l'immeuble); aucun effet pernicieux n'a été observé.

4° *Pensez-vous que l'emploi du Téflon dans l'industrie constitue une source importante de danger ?*

Le P.T.F.E. ne constitue pas un danger important dans l'industrie. Pour des raisons mécaniques ou autres, il n'est pas possible normalement de travailler les consti-

tuants du P.T.F.E. en continu à une température au-dessous de 300° C, bien que de courtes périodes au-dessus de 300° C peuvent être mise en œuvre. Il n'y a aucun rapport que quiconque ait souffert de maux dus au P.T.F.E. en dessous de 350° C.

5° Avez-vous effectivement constaté un état de panique parmi les personnes utilisant et connaissant bien le Téflon ?

Les personnes qui manipulent le P.T.F.E. ne doivent pas avoir de raisons de s'alarmer. Tous les fabricants du polymère recommandent une bonne ventilation quand le produit est fritté, et que l'on prenne soin d'éviter la contamination du tabac. Je suis à même de discerner cependant que bien que beaucoup de fabricants aient installé un système de ventilation convenable, pas tous n'imposent une règle « Interdiction de fumer » dans leurs ateliers.

Pour résumer tout cela il y a une toxicité due au hasard dans la fabrication du P.T.F.E. mais elle peut être évitée en prenant des précautions simples et sûres. Les dangers certainement courus ne justifient pas les rumeurs malveillantes qui continuent à circuler.

Je voudrais profiter de cette occasion pour vous dire combien j'ai été heureux de vous rencontrer à la récente conférence et que j'envie nos futures rencontres dans le même état d'esprit.

La technologie des matières plastiques.

Par M. V. JANS.

Les matières de base de la fabrication des matières plastiques ne présentent qu'exceptionnellement les propriétés requises, à l'état pur. Une préparation de la matière plastique consistant très souvent à opérer un mélange d'une résine synthétique avec des adjuvants qui se présenteront à l'état solide ou liquide, est nécessaire.

L'homogénéisation s'opère dans des appareils de divers types, soit à cuve (mélangeurs-masticateurs, pétrisseurs-malaxeurs, mélangeurs-malaxeurs, dont la plupart sont analogues à ceux utilisés dans l'industrie du caoutchouc), soit à cylindres (mélangeurs à cylindres, employés notamment pour la plupart des thermoplastiques) et ce n'est qu'après cette opération comportant parfois l'incorporation de nombreux ingrédients que sera obtenue la « matière plastique » telle qu'elle sera travaillée.

Dans le cas de résines relativement faciles à amener à l'état liquide, la matière première, par exemple, un phénoplaste ou un aminoplaste, est introduite, avec les ingrédients appropriés, dans des autoclaves verticaux à agitateurs ; le mélange liquide est évacué, il se solidifie, puis il est concassé.

Les poudres à mouler thermodurcissables sont, en général amenées sous forme d'un sable grossier ; celles à base de matières thermoplastiques sont obtenues à partir de feuilles que l'on a déchiquetées et amenées à l'état de rognures, flocons ou petits cubes. Bien souvent également, les matières plastiques sont préformées, c'est-à-dire réduites à l'état de formes simples (pastilles, tablettes, etc.) de poids connu.

Les principaux produits d'addition utilisés à cette phase des opérations sont :

LES STABILISANTS, agents destinés à empêcher, ou, tout au moins retarder, le vieillissement de la matière plastique en permettant à cette dernière de conserver ses propriétés essentielles en résistant aux effets de la lumière, de l'humidité, des variations de température.

Bon nombre de ces stabilisants renferment des produits toxiques et leur emploi pose des problèmes techniques qui n'ont pas toujours reçu une solution entièrement satisfaisante.

LES PLASTIFIANTS tendent à modifier dans le sens de la plasticité le produit auquel on les incorpore, en même temps qu'à apporter certaines améliorations comme celle

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CENTER FOR DISEASE CONTROL



Vol. 23, No. 6

WEEKLY
REPORT

For
Week Ending
February 9, 1974

Morbidity and Mortality

U.S. DEPARTMENT OF HEALTH, EDUCATION, AND WELFARE PUBLIC HEALTH SERVICE

DATE OF RELEASE: FEBRUARY 15, 1974 - ATLANTA, GEORGIA 30333

EPIDEMIOLOGIC NOTES AND REPORTS ANGIOSARCOMA OF THE LIVER AMONG POLYVINYL CHLORIDE WORKERS - Kentucky

Between September 1967 and December 1973, 4 cases of angiosarcoma of the liver were diagnosed among men employed in the polyvinyl chloride polymerization section of a B.F. Goodrich plant near Louisville, Kentucky. This section of the plant began operations in 1938. It employs about 270 persons and produces polyvinyl chloride as well as a variety of copolymers by polymerization of vinyl chloride monomer. All 4 men had worked continuously in the section for at least 14 years prior to onset of illness (Table 1); all 4 had worked directly in various phases of the polymerization process.

Case 1 presented in August 1967 with an epigastric mass and thrombocytopenia. An exploratory laparotomy was per-

formed in September 1967; liver biopsy revealed angiosarcoma. Case 2 presented in January 1970 with gastrointestinal (GI) bleeding. Recurrent bleeding in May 1970 led to an exploratory laparotomy at which time a diagnosis of angiosarcoma was made on liver biopsy. Case 3 presented in January 1964 with GI bleeding which recurred in May 1965 with

CONTENTS

Epidemiologic Notes and Reports	
Angiosarcoma of the Liver Among Polyvinyl Chloride Workers - Kentucky	49
Wild Mushroom Poisoning - Pennsylvania	50
Current Trends	
Influenza - United States	50

TABLE I. CASES OF SPECIFIED NOTIFIABLE DISEASES: UNITED STATES
(Cumulative totals include revised and delayed reports through previous weeks)

DISEASE	6th WEEK ENDING		MEDIAN 1969-1973	CUMULATIVE, FIRST 6 WEEKS		
	February 9, 1974	February 10, 1973		1974	1973	MEDIAN 1969-1973
Septic meningitis	34	29	35	213	238	218
Acicellosis	1	3	1	9	11	11
Chickenpox	3,609	5,305	-	18,811	27,061	-
Diphtheria	1	2	2	7	12	17
Encephalitis:						
Primary: Arthropod-borne and unspecified	19	21	18	86	95	120
Post-Infectious	12	4	5	27	20	26
Hepatitis, Viral:						
Type B	135	120	128	934	805	805
Type A	871	1,027	1,067	4,914	5,775	6,407
Type unspecified	153	6	39	812	17	243
Malaria	6	6	704	18	3,461	3,663
Measles (rubella)	521	704	50	2,414	174	331
Meningococcal infections, total	34	34	49	156	166	307
Civilian	34	34	2	156	8	15
Military	-	-	2,502	-	10,400	12,594
Mumps	1,762	2,145	-	9,148	-	-
Pertussis	72	-	-	182	-	-
Rubella (German measles)	258	441	823	1,098	2,269	2,964
Tetanus	2	1	1	6	6	6
Tuberculosis, new active	529	562	-	3,040	3,027	-
Typhoid fever	2	2	2	10	10	10
Typhus, tick-borne (Rky. Mt. spotted fever)	6	4	4	35	22	26
Veneral Diseases:	-	1	-	12	6	3
Gonorrhea	16,810	14,259	-	97,831	85,951	-
Syphilis, primary and secondary	462	516	-	2,661	2,917	-
Rabies in animals	43	69	67	267	327	347

TABLE II. NOTIFIABLE DISEASES OF LOW FREQUENCY

DISEASE	Cum.		Cum.
	1	2	
Anthrax:			
Botulism: Wash. 2	1	1	-
Congenital rubella syndrome: Ida. 1, Tex. 1	2	2	2
Erysipelas: Hawaii 1	7	7	-
Leptospirosis: Hawaii 1, Ida. 1	4	4	13
Polioencephalitis: total	6	6	2
Paralytic: Tex.	-	-	-
Poliovirus: N.Y. 1, Okla. 1	-	-	-
Typhus, murine: Calif. 1, Va. 1, P.R. 1	-	-	-

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ANGIOSARCOMA — Continued

signs of portal hypertension. A portacaval shunt was performed, and liver biopsy yielded a diagnosis of cirrhosis. Repeat biopsies in October 1970 and September 1972 confirmed this diagnosis. Autopsy in March 1973 revealed angiosarcoma. Case 4 presented in July 1973 with hepatosplenomegaly, weight loss, and jaundice. Two liver biopsies were interpreted as showing severe cirrhosis. Autopsy in December 1973 revealed angiosarcoma.

In each case, pathologic material revealed the presence of extensive cirrhosis of a non-alcoholic type in addition to angiosarcoma. In 2 cases, the diagnosis of angiosarcoma was made only at autopsy, cirrhosis having been diagnosed 7 years before in Case 3 and 5 months before in Case 4. None of the patients gave histories of prolonged alcohol use or exposure to hepatotoxins outside their work place. In particular, none had ever had exposure to thorium dioxide or to arsenic, 2 materials known specifically to induce hepatic angiosarcoma in man (1,2).

(Reported by John Creech, M.D., Plant Physician, B.F. Goodrich Chemical Company, Louisville, Kentucky; Maurice N. Johnson, M.D., Director of Environmental Health, B.F. Goodrich Chemical Company, Akron, Ohio; Bradford Block, M.D., Medical Consultant, Kentucky Occupational Safety and Health Administration, Kentucky State Department of Labor; National Institute for Occupational Safety and Health, and the Cancer and Birth Defects Division, Bureau of Epidemiology, CDC.)

Editorial Note

Angiosarcoma of the liver is an exceedingly rare tumor. It is estimated that only about 25 such cases occur each year in the United States. Four cases, therefore, among a small number of workers at a single plant is a most unusual event, and one which raises the possibility of some work-related carcinogen, conceivably vinyl chloride itself. Although no data are yet available concerning the occurrence of angiosarcoma among workers at other vinyl chloride plants in the United States, it seems distinctly possible that the problem may be industry-wide. Epidemiologic studies have started to

Table 1

Cases of Angiosarcoma of the Liver
among Polyvinyl Chloride Workers
B. F. Goodrich Plant
Louisville, Kentucky

Case	Age at illness onset	Date of		Years worked with PVC before illness
		Illness onset	Death	
1	43	Aug. 1967	Sept. 1967	Jan. 7, 1968
2	36	Jan. 1970	May 1970	Sept. 27, 1971
3	41	Jan. 1964	Mar. 1973	Mar. 3, 1973
4	58	July 1973	Dec. 1973	Dec. 19, 1973

determine the extent of the problem in the United States with respect both to angiosarcoma of the liver and to its possible relationship to post-toxic cirrhosis.

Published data concerning the potential hepatotoxicity and oncogenicity of vinyl chloride are limited. Studies in Germany have suggested a link between hepatic damage and occupational exposure to vinyl chloride (3), while Italian workers have suggested that vinyl chloride may cause a wide variety of tumors in animals (4). The chemical concentration used in these latter experiments, however, far exceed levels likely to be encountered in industrial environments. Effort to confirm such observations and to measure effects at low dose levels are now in progress.

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WILD MUSHROOM POISONING — Pennsylvania

On October 9, 1973, 37 nuns ate their evening meal at a convent in Berks County, Pennsylvania. Between ¾ and 4¼ hours later (median—approximately 2 hours), 17 nuns developed an illness characterized by profuse sweating (76%), watery diarrhea (76%), chills (58%), and abdominal pain (42%) (Table 2). Six of the nuns were hospitalized; none died.

Food-specific attack rates for foods ingested in the 24 hours before illness strongly implicated mushrooms as the intoxicating food (Table 3). Stool specimens from 4 of the ill nuns and bacteriologic cultures of the remaining mushrooms yielded no enteric pathogens. Remnants of the mushrooms were identified as *Clitocybe* sp. and *Lepiota* sp. Some species of the former genus are known to contain muscarine, which produces parasympathomimetic effects.

The mushrooms had been picked on the grounds of the convent on the afternoon of the outbreak. For several years, the nuns had eaten these mushrooms at meals, and no apparent illness had occurred. On October 9, however, the nun who

Table 2
Symptoms in 17 Ill Nuns
Berks County Convent — October 9, 1973

Symptom	Percent with Symptom
Diarrhea	76
Sweating	76
Chills	58
Abdominal pain	42
Nausea	35
Vomiting	23
Excessive salivation	23
Feverish feeling	18
Dry mouth	12
Blurred vision	12
Headache	6
Weakness	6
Faint feeling	6

(Continued on page 51)

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